Stereoselective Radical Transformations by Co(II)-Based Metalloradical Catalysis

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Chapter 1. Co(II)-Based Metalloradical Catalysis for Stereoselective Radical Cyclopropanation of Alkenes

This Account summarizes our group's recent efforts in developing metalloradical catalysis as a one-electron approach for catalytic radical cyclopropanation of alkenes with diazo compounds.

Chapter 2. Asymmetric Radical Process for General Synthesis of Chiral Heteroaryl Cyclopropanes

We have developed a Co(II)-based metalloradical system that is highly effective for asymmetric radical cyclopropanation of alkenes with in situ-generated α heteroaryldiazomethanes. Through fine-tuning the cavity-like environments of newlydeveloped D_2 -symmetric chiral amidoporphyrins as the supporting ligand, the optimized Co(II)-based metalloradical system is broadly applicable to pyridyl and other heteroaryldiazomethanes for asymmetric cyclopropanation of a wide range of alkenes, providing general access to valuable chiral heteroaryl cyclopropanes in high yields with excellent diastereoselectivities and enantioselectivities.

Chapter 3. Enantioselective Metalloradical 1,6-C–H Alkylation of In Situ-Generated Alkyldiazomethanes for Synthesis of Chiral Piperidines

We have disclosed an effective Co(II)-based metalloradical system as a fundamentally different approach to harness the potential of 1,6-HAA radical process, enabling asymmetric 1,6-C–H alkylation of in situ-generated α -alkyldiazomethanes to construct

chiral piperidines. Supported by an optimal D_2 -symmetric chiral amidoporphyrin ligand, the Co(II)-catalyzed alkylation system is capable of activating a wide array of α alkyldiazomethanes containing C(sp³)–H bonds with varied steric and electronic properties, providing access to chiral α -substituted piperidines in good to high yields with high enantioselectivities from readily accessible 4-aminobutanal derivatives. In addition to practical attributes, such as operational simplicity and mild conditions, the metalloradical system is highlighted by its tolerance to different functional groups as well as compatibility with heteroaryl units.

Chapter 4. Design and Synthesis of A Novel *D*₂-Symmetric Chiral Porphyrin for Co(II)-Based Metalloradical Catalysis

A novel D_2 -symmetric chiral amidoporphyrin derived from chiral cyclopropanecarboxamide containing diphenyl units has been effectively constructed based on Co(II)-catalyzed asymmetric cyclopropanation of alkenes.

DEDICATION

This thesis is dedicated to the loving memory of my grandfather

Zhirui Yang

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CHAPTER 1

COBALT(II)-BASED METALLORADICAL CATALYSIS FOR STEREOSELECTIVE RADICAL CYCLOPROPANATION OF ALKENES 1.1. INTRODUCTION

Over the past decades, there has been a rapid growth in the development of radical reactions as powerful tools for modern organic synthesis owing to their rich reactivities and unique attributes.¹ Despite significant advancements, general strategies for addressing the longstanding issues of controlling reactivity and selectivity in radical reactions, especially enantioselectivity, remain to be developed and pose a formidable task in the field.² Among considerable endeavors,³ metalloradical catalysis (MRC) represents a

¹ For selected books, see: (a) Zard, S. Z. *Radical Reactions in Organic Synthesis*; Oxford University Press, 2003. (b) Chatgilialoglu, C.; Studer, A., *Encyclopedia of Radicals in Chemistry, Biology, and Materials*; John Wiley & Sons, 2012. For selected reviews, see: (c) Zard, S. Z. *Chem. Soc. Rev.* **2008**, *37*, 1603–1618. (d) Narayanam, J. M.; Stephenson, C. R. *Chem. Soc. Rev.* **2011**, *40*, 102–113. (e) Quiclet-Sire, B.; Zard, S. Z. *Pure Appl. Chem.* **2011**, *83*, 519–551. (f) Prier, C. K.; Rankic, D. A.; MacMillan, D. W. *Chem. Rev.* **2013**, *113*, 5322–5363. (g) Studer, A.; Curran, D. P. *Angew. Chem., Int. Ed.* **2016**, *55*, 58–102.

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fundamentally different approach for achieving stereoselective radical reactions by exploiting metalloradical complexes as open-shell catalysts for catalytic generation and regulation of metal-stabilized organic radicals (Figure 1.1A).^{4,5,6} To this end, Co(II) complexes of porphyrins [Co(Por)], as stable 15e-metalloradicals with a well-defined low spin d⁷ electronic configuration, have emerged as a new class of privileged catalysts for homolytic activation of diazo compounds to generate unprecedented α -Co(III)-alkyl radicals upon release of N₂ as the sole byproduct (Figure 1.1B).⁷ Through the employment

⁴ For selected reviews and highlights on Co(II)-based MRC, see: (a) Lu, H. J.; Zhang, X. P. *Chem. Soc. Rev.* **2011**, *40*, 1899–1909. (b) Pellissier, H.; Clavier, H. *Chem. Rev.* **2014**, *114*, 2775–2823. (c) Demarteau, J.; Debuigne, A.; Detrembleur, C. *Chem. Rev.* **2019**, *119*, 6906–6955. (d) Huang, H.-M.; Garduño-Castro, M. H.; Morrill, C.; Procter, D. J. *Chem. Soc. Rev.* **2019**, *48*, 4626–4638. (e) Singh, R.; Mukherjee, A. *ACS Catal.* **2019**, *9*, 3604–3617. (f) Roy, S.; Das, S. K.; Khatua, H.; Das, S.; Chattopadhyay, B. Acc. Chem. Res. **2021**, *54*, 4395–4409.

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of modularly-designed D_2 -symmetric chiral amidoporphyrins with tunable environments as the supporting ligands, the initially-generated C-centered radicals, which remain covalently bonded with the cobalt complex, retain the reactive nature of radicals and can be precisely governed to engage in common radical reactions, such as radical addition and H-atom abstraction as well as radical substitution, leading to the development of new catalytic processes of stereoselective radical transformations.^{7g-i, 8}

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Figure 1.1. Co(II)-Based Metalloradical Catalysis as A Conceptually Different Approach Toward Controlling Radical Reactions

 $d_{x^{2}-v^{2}}$ ^{1}R \mathbf{R}^2 vs. [Co(II)(Por)] "Free" "Confined" Alkyl Radicals α-Metalloalkyl Radicals Metalloradical Catalvst **B. Metalloradical Activation of Diazo Compounds** ^{1}R \mathbb{R}^2 [Co(II)(Por)] α-Co(III)-Alkyl Radicals M = Co(II) $M = Rh_{2}(II), Cu(I)...$ Half n Bonding SOMO sp² sp² dσ d_{xz/xy} С **Fischer Carbenes Co-Carbene Radicals** Co π

A. Co(II) Complexes of Porphyrins as Metalloradical Catalysts

Chiral cyclopropanes are the smallest class of carbocyclic compounds with its threemembered ring structure. In addition to their fundamental significance, chiral cyclopropanes are of practical importance as they exist as recurring units in numerous natural products and bioactive compounds, in addition to serving as versatile chiral synthons for stereoselective organic synthesis (Figure 1.2).⁹ Among synthetic methods, metal-catalyzed asymmetric cyclopropanation of alkenes with diazo compounds represents the most general approach for stereoselective synthesis of chiral cyclopropanes.¹⁰ While a

⁹ Talele, T. T. J. Med. Chem. 2016, 59, 8712–8756. (b) Ebner, C.; Carreira, E. M. Chem. Rev. 2017, 117, 11651–11679.

¹⁰ (a) Doyle, M. P. *Chem. Rev.* **1986**, *86*, 919–939. (b) Doyle, M. P.; Forbes, D. C. *Chem. Rev.* **1998**, *98*, 911–935. (c) Lebel, H.; Marcoux, J.-F.; Molinaro, C.; Charette, A. B. *Chem. Rev.* **2003**, *103*, 977–1050. (d) Intrieri, D.; Carminati, D. M.; Gallo, E. *Dalton Trans.* **2016**, *45*, 15746–16048.

Figure 1.2. Selected Examples of Natural Products and Bioactive Compounds Containing Cyclopropane Motifs



number of transition metal-based catalytic systems have been successfully developed for asymmetric cyclopropanation with acceptor-substituted diazo compounds, such as diazoacetates, ¹¹ other important classes of diazo compounds, including acceptor/acceptor ¹² - and donor-substituted diazo compounds, ¹³ have proven to be challenging as carbene sources for the catalytic transformation presumably due to the

¹¹ (a) Evans, D. A.; Woerpel, K. A.; Hinman, M. M.; Faul, M. M. J. Am. Chem. Soc. **1991**, *113*, 726–728. (b) Doyle, M. P.; Winchester, W. R.; Hoorn, J. A. A.; Lynch, V.; Simonsen, S. H.; Ghosh, R. J. Am. Chem. Soc. **1993**, *115*, 9968–9978. (c) Nishiyama, H.; Itoh, Y.; Matsumoto, H.; Park, S.-B.; Itoh, K. J. Am. Chem. Soc. **1994**, *116*, 2223–2224. (d) Lo, M. M. C.; Fu, G. C. J. Am. Chem. Soc. **1998**, *120*, 10270–10271. (e) Che, C.-M.; Huang, J.-S.; Lee, F.-W.; Li, Y.; Lai, T.-S.; Kwong, H.-L.; Teng, P.-F.; Lee, W.-S.; Lo, W.-C.; Peng, S.-M.; Zhou, Z.-Y. J. Am. Chem. Soc. **2001**, *123*, 4119–4129. (f) Davies, H. M. L.; Antoulinakis, E. G. Org. React. **2001**, *57*, 1. (g) Pellissier, H. Tetrahedron **2008**, *64*, 7041–7095. (h) Caballero, A.; Prieto, A.; Díaz-Requejo, M. M.; Pérez, P. J. Eur. J. Org. Chem. **2009**, 1137–1144. (i) Coelho, P. S.; Brustad, E. M.; Kannan, A.; Arnold, F. H. Science **2013**, *339*, 307–310. (j) Wei, Y.; Tinoco, A.; Steck, V.; Fasan, R.; Zhang, Y. J. Am. Chem. Soc. **2018**, *140*, 1649–1662. (k) Shaw, S.; White, J. D. Chem. Rev. **2019**, *119*, 9381–9426.
¹² Doyle, M. P. Angew. Chem., Int. Ed. **2009**, *48*, 850–852.

¹³ Allouche, E. M. D.; Charette, A. B. Synthesis **2019**, *51*, 3947–3963.

electrophilic nature of the metallocarbene intermediates involved in the existing catalytic systems. For the same reason, asymmetric cyclopropanation of electron-deficient olefins, such as α , β -unsaturated carbonyl compounds and nitriles, remains a major challenge in the field. In addition to the limited substrate scope, existing catalytic systems typically require the use of excess olefins and slow addition of diazo compounds as dimerization of diazo compounds is a common side reaction. To address these and other issues, it calls for the development of new catalytic approach that is fundamentally different from existing catalytic systems. This Account aims to summarize Zhang group's recent efforts in developing Co(II)-based MRC as a one-electron approach for asymmetric radical cyclopropanation of various alkenes with different classes of diazo compounds (Figure 1.3).

Figure 1.3. Classes of diazo compounds (EWG = electron withdrawing group; EDG = electron donating group)



EWG: carbonyl; sulfonyl; cyano; nitro; phospho EDG: alkyl; aryl; vinyl; alkynyl; heteroaryl

1.2. COBALT(II) COMPLEXES OF *D*₂-SYMMETRIC AMIDOPORPHYRINS AS METALLORADICAL CATALYSTS

Inspired by the extraordinary catalytic capability of heme-containing enzymes in nature, metalloporphyrins have been recognized for their potential utility as effective catalysts in organic synthesis. Owing to the unique ligand environment and metal coordination mode, metalloporphyrin-based systems have been shown to exhibit exceptional selectivities and high catalytic turn overs.¹⁴ Since the first demonstration of rhodium porphyrins for catalytic cyclopropanation in 1980,¹⁵ a number of metalloporphyrins, including iron, ruthenium, and osmium complexes of porphyrins, have been utilized to catalyze selective cyclopropanation and related carbene transfer reactions.^{11e,16} Only until 2003, Zhang¹⁷ and Cenini and coworkers¹⁸ independently disclosed for the first time the catalytic capability of Co(II) complexes of porphyrins [Co(Por)] for olefin cyclopropanation with diazo compounds. In addition to the disclosure of the new catalytic activity, Zhang and coworkers further showed that [Co(Por)] are even superior catalysts for olefin cyclopropanation in comparison with other common metalloporphyrins, including Fe(III), Ru(II) and Rh(III) complexes of porphyrins.¹⁷ It was found that the common side products resulting from competitive carbene dimerization of diazo compounds was minimized in [Co(Por)]-based catalytic system for olefin cyclopropanation. Consequently, the Co(II)-based catalytic system can operate in a one-pot fashion with alkenes as the limiting reagents and without the need of slow addition of diazo compounds, a practically desirable protocol that is atypical for most catalytic systems. With the support of chiral porphyrins, asymmetric induction was demonstrated for Co(II)-catalyzed olefin cyclopropanation. More importantly, a significant ligand effect on reactivity as well as diastereoselectivity and

¹⁴ Kadish, K. M.; Smith, K. M.; Guilard, R., The Porphyrin Handbook. Academic Press: San Diego, 2003.

¹⁵ Callot, H. J.; Piechocki, C. *Tetrahedron Lett.* **1980**, *21*, 3489–3492.

¹⁶) (a) Maxwell, J. L.; O'Malley, S.; Brown, K. C.; Kodadek, T. Organometallics 1992, 11, 645–652. (b) Smith, D. A.; Reynolds, D. N.; Woo, L. K. J. Am. Chem. Soc. 1993, 115, 2511–2513. (c) Wolf, J. R.; Hamaker, C. G.; Djukic, J.-P.; Kodadek, T.; Woo, L. K. J. Am. Chem. Soc. 1995, 117, 9194–9199. (d) Che, C.-M.; Huang, J.-S. Coord. Chem. Rev. 2002, 231, 151–164. (e) Morandi, B.; Carreira, E. M. Science 2012, 335, 1471–1474. (f) Allouche, E. M. D.; Al-Saleh, A.; Charette, A. B. Chem. Commun. 2018, 54, 13256–13259. (g) Damiano, C.; Sonzini, P.; Gallo, E. Chem. Soc. Rev. 2020, 49, 4867–4905.

¹⁷ Huang, L.; Chen, Y.; Gao, G.-Y.; Zhang, X. P. J. Org. Chem. 2003, 68, 8179-8184.

¹⁸ Penoni, A.; Wanke, R.; Tollari, S.; Gallo, E.; Musella, D.; Ragaini, F.; Demartin, F.; Cenini, S. *Eur. J. Org. Chem.* **2003**, 1452–1460.

enantioselectivity was revealed in Co(II)-based catalytic system for asymmetric cyclopropanation (Scheme 1.1).





For example, with D_4 -symmetric chiral esterporphyrin **P1** as the supporting ligand, Co(II)-based catalytic system could catalyze asymmetric cyclopropanation of styrene with ethyl diazoacetate (EDA) under the practical protocol, affording the desired cyclopropane in good yield (73%) as *cis*-dominant product (*cis:trans* = 64:36) with moderate enantioselectivities (77% ee and 62% ee for *cis*- and *trans*-cyclopropane, respectively). When the supporting ligand was replaced by D_2 -symmetric chiral etherporphyrin **P2**, the Co(II)-catalyzed reaction produced the cyclopropane in higher yield (84%) as *trans*dominant product (*cis:trans* = 48:52) with significantly lower enantioselectivities (31% ee and 10% ee for *cis*- and *trans*-cyclopropane, respectively).¹⁷ Mechanistically, Zhang and coworkers proposed a catalytic cycle for the Co(II)-catalyzed cyclopropanation involving "a Co(III)-carbene complex with Co–C single bond and carbon-based radical character".¹⁷ Through both experimental and computational studies, the proposed Co(III)-carbene radicals, which is systematically termed as α -Co(III)-alkyl radicals, have since been well established as key intermediates in diverse catalytic radical processes via Co(II)-based metalloradical catalysis. On the basis of these initial results, it was envisioned that the design and synthesis of new chiral porphyrin ligands would improve both reactivity and stereoselectivities of radical cyclopropanation by [Co(Por)] and might lead to the discovery of new catalytic radical transformations.

As stable 15e-metalloradicals, Co(II) complex of porphyrins exhibit several unique features that render them versatile and efficient catalysts, such as excellent thermal and metal coordination stability as well as minimal complication resulting from potential *cis*coordination. Moreover, the introduction of peripheral substituents within the porphyrin ligands essentially permits flexibility in design of [Co(Por)] with varied electronic, steric, and chiral environments. However, the preparation of porphyrin derivatives bearing different substituents often requires tedious purification and typically gives low yields.¹⁹ Within this context, in 2004, Zhang and coworkers developed a general and efficient method for the modular construction of *D*₂-symmetric chiral amidoporphyrins (*D*₂-Por^{*}) via Pd-catalyzed amidation reactions of 5,10-bis(2',6'-dibromophenyl)porphyrin synthons with chiral amides (Scheme 1.2).²⁰ A series of 5,10- dibromophenyl)porphyrins containing different *meso*-R groups at 10 and 20 position can be readily accessed by MacDonald porp-

¹⁹ Kadish, K. M.; Smith, K. M.; Guilard, R., The Porphyrin Handbook. Academic Press: San Diego, 2003; pp. 75–132.

²⁰ Chen, Y.; Fields, K. B.; Zhang, X. P. J. Am. Chem. Soc. 2004, 126, 14718–14719.



Scheme 1.2. Modular Construction of Co(II) Complexes of D₂-Symmetric Chiral Amidoporphyrins

-hyrin synthesis using Lindsey's condition.²¹ The resulting tetrabromoporphyrin synthons can be efficiently coupled with optically pure amides under the optimized Pd-catalyzed quadruple amidation conditions, ²² enabling the construction of a "toolbox" of D_2 symmetric chiral amidoporphyrins with varied *ortho*-chiral R^{*} and *meso*-achiral R groups (Scheme 1.3). Followed by facile cobalt metalation, the corresponding Co(II) complexes of D_2 -symmetric chiral amidoporphyrins can be furnished as stable purple solids in overall high yields.²⁰ The nearly perpendicular arrangement between the *meso*-phenyl rings and the porphyrin plane, in combination with the *trans*-amide conformation, could direct the *ortho* chiral amide units toward the center of the porphyrins, providing the requisite rigidity

²¹ Lindsey, J. S.; Hsu, H. C.; Schreiman, I. C. Tetrahedron Lett. 1986, 27, 4969–4970.

²² Gao, G.-Y.; Chen, Y.; Zhang, X. P. Org. Lett. 2004, 6, 1837–1840.

for asymmetric induction. Though the combined use of chiral R^{*} and *meso*-R groups, it may be possible to control diastereoselectivity as well as enantioselectivity. Furthermore, it was hypothesized that the N–H moieties of the amide units are capable of engaging in potential H–bonding interactions with the substrates, which could play an important role in controlling reactivity and selectivity of Co(II)-based metalloradical systems that will be discussed later in details.

Scheme 1.3. Selected Examples of Co(II) Complexes of D₂-Symmetric Chiral Amidoporphyrins





[Co(**P6**)] (**P6** = 3,5-Di^tBu-QingPhyrin)



[Co(**P4**)] (**P4** = 2,6-DiMeO-ChenPhyrin)



[Co(P7)](P7 = 3,5-Di^tBu-Xu(2'-Naph)Phyrin)



[Co(**P5**)] (**P5** = 2,6-DiMeO-ZhuPhyrin)



[Co(**P8**)] (**P8**= 3,5-Di^{*t*}Bu-Tao(^{*t*}Bu)Phyrin)

1.3. COBALT(II)-CATALYZED INTERMOLECULAR RADICAL

CYCLOPROPANATION

1.3.1. Intermolecular Radical Cyclopropanation with Acceptor-Substituted Diazo Compounds

To demonstrate the catalytic capability of the new class of metalloradical catalyst $[Co(D_2-Por^*)]$, asymmetric cyclopropanation of styrene with diazoacetate was evaluated as the model reaction (Scheme 1.4).²⁰ The first-generation metalloradical catalyst [Co(P3)] (P3 = 3,5-Di/Bu-ChenPhyrin) proved to be effective in catalyzing the asymmetric radical cyclopropanation with *tert*-butyl diazoacetate (*t*-BDA). Moreover, during the course of the studies, a significant *trans*-effect of potential coordinating additives²³ was revealed in influencing the stereoselectivity of the radical cyclopropanation process. In the presence of substoichiometric amount of DMAP (4-(dimethylamino)pyridine) as additives, both the diastereoselectivity and enantioselectivity could be substantially enhanced, affording the desired cyclopropane product up to 85% yield with >99:1 dr and 98% ee. In addition to high yield and stereoselectivity, the [Co(P3)]-catalyzed asymmetric cyclopropanation can operate effectively in a one-pot fashion with styrene as the limiting reagent and without slow addition of the diazo compound.

After developing the highly asymmetric cyclopropanation of styrene with diazoacetate using [Co(P3)], Zhang and coworkers explored the catalytic reactivity of [Co(P3)] toward more challenging substrates, such as electron-deficient olefins (Scheme 1.5). As discussed earlier, asymmetric cyclopropanation of electron-deficient olefins has proven to be a formidable challenge in existing catalytic systems presumably attributed to the electrophil-

²³Chen, Y.; Zhang, X. P. Synthesis **2006**, 10, 1697–1700.

Scheme 1.4. Asymmetric Cyclopropanation of Styrene Catalyzed by *D*₂-Symmetric Chiral Co(II) Porphyrin [Co(P3)]

1.0 equiv		[Co(P3)] (1 mol %) DMAP (0.5 equiv) N ₂ CHCO ₂ R (1.2 equiv) toluene; temp; 20 h		H H_{1} $CO_{2}R$ $+$ $cis-(1S,2R)$ H H_{2} $+$ $trans-(1R,2R)$		H CO_2R cis-(1R,2S) H H T^2 CO_2R CO_2R CO_2R CO_2R CO_2R CO_2R CO_2R
entry	R	temp	yield (%)	trans:cis	ee (%)	config
1	Et	RT	82	97:3	78	1 <i>R</i> , 2 <i>R</i>
2	^t Bu	RT	84	>99:1	95	1 <i>R</i> , 2 <i>R</i>
3	^t Bu	−20 °C	85	>99:1	98	1 <i>R</i> , 2 <i>R</i>

-lic nature of the metallocarbene intermediates.¹⁰ Considering that neutral radicals typically react under mild conditions with a high degree of tolerance toward electronic properties of the substrates, a radical-mediated asymmetric cyclopropanation could potentially address the aforementioned issues with electron-deficient olefins. In 2007, Zhang and coworkers found that under a practical one-pot protocol using 1 mol % of [Co(**P3**)] in the presence of DMAP, a wide range of electron-deficient olefins, including unsaturated esters, amides, ketones, and nitriles could be effectively cyclopropanated with EDA and *t*-BDA, delivering the corresponding electrophilic cyclopropane derivatives in moderate to high yields with high stereoselectivities.²⁴ Evidently, the unique profile of reactivity and selectivity exhibited by the Co(II)-based metalloradical cyclopropanation toward electron-deficient olefins suggests the involvement of catalytic intermediates that are distinctly different from the previously reported systems.

²⁴ Chen, Y.; Ruppel, J. V.; Zhang, X. P. J. Am. Chem. Soc. 2007, 129, 12074–12075.

Scheme 1.5. [Co(P3)]-Catalyzed Asymmetric Radical Cyclopropanation of Electron-

Deficient Olefins



In addition to alkyl diazoacetates, the Co(II)-based MRC could be successfully applied for asymmetric cyclopropanation with succinimidyl diazoacetate, leading to the general synthesis of biologically important cyclopropyl carboxamides (Scheme 1.6).²⁵ After extensive ligand screening, it was found that the steric bulkiness of the succinimidyl diazoacetate essentially governed the reactivity difference of the metalloradical catalysts. Among different $[Co(D_2-Por^*)]$, [Co(P3)] was identified as the optimal catalyst for catalyzing the asymmetric cyclopropanation of styrene with succinimidyl diazoacetate in the presence of DMAP. Under the optimized conditions, it was revealed that the Co(II)based catalytic system exhibits a high degree of functional group tolerance as demonstrated with the reactions of acetoxyl- and nitro-substituted styrenes to form the corresponding cyclopropanes in moderate to high yields with excellent stereoselectivities. Electron-

²⁵ Ruppel, J. V.; Gauthier, T. J.; Snyder, N. L.; Perman, J. A.; Zhang, X. P. Org. Lett. 2009, 11, 2273–2276.

deficient olefins, which are typically challenging substrates, could be effectively cyclopropanated by [Co(P3)] as well, furnishing the functionalized electrophilic cyclopropanes with excellent control of stereoselectivities albeit in lower yields. Subsequently, the authors have demonstrated the potential applications of the enantioenriched succinimidyl cyclopropyl carboxylate derivatives as versatile synthesis of chiral cyclopropyl carboxamides (Scheme 1.6).

Scheme 1.6. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with Succinimidyl Diazoacetates



In 2008, Zhang and coworkers reported a Co(II)-based metalloradical system that is highly effective for asymmetric radical cyclopropanation employing diazosulfones as the

carbene precursor (Scheme 1.7).²⁶ Unexpectedly, [Co(**P3**)], which was previously shown to be the optimal catalyst for the cyclopropanation with diazoacetates, was found to be **Scheme 1.7. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with**

Diazosulfones



ineffective enantioselectivity in terms of in catalyzing the reaction with tosyldiazomethanes. To this end, Zhang and coworkers designed and synthesized a new chiral metalloradical catalyst [Co(P5)] (P5 = 2,6-DiMeO-ZhuPhyrin)] featuring enhanced conformational rigidity as a result of unique intramolecular H-bonding interactions in (S)-2-tethydrafurancarboxamide units. The new metalloradical catalyst [Co(P5)] could effectively activate tosyldiazomethane for asymmetric cyclopropanation with styrene, delivering the desired product in 99% yield with >99:1 dr and 92% ee. In addition to styrene,

²⁶ Zhu, S.; Ruppel, J. V.; Lu, H.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2008, 130, 5042–5043.

the [Co(P5)]-catalyzed cyclopropanation was shown to be generally applicable to a broad range of styrene derivatives as well as electron-deficient olefins, resulting in the highyielding formation of the corresponding cyclopropyl sulfones with high diastereoselectivities and enantioselectivities.

In an effort to gain insights into the working details as well as the basis of unique reactivity and selectivity profiles of Co(II)-catalyzed asymmetric radical cyclopropanation, the Zhang group collaborated with Prof. de Bruin and coworkers and conducted a series of computational and experimental studies. Accordingly, density functional theory (DFT) calculations in combination with electronic paramagnetic resonance (EPR) and high-resolution mass spectrometry (HRMS) disclosed a stepwise radical mechanism involving α - and γ -metalloalkyl radicals as the key intermediates.^{7a}

As illustrated in Scheme 1.8, the catalytic cycle begins with the metalloradical activation of the diazo compound by [Co(Por)] to generate α -Co(III)-alkyl radical intermediate with release of dinitrogen as the byproduct. The resulting Co-supported C-centered radical is capable of undergoing radical addition to the C=C double bond of the olefin substrate to form γ -Co(III)-alkyl radical intermediate (Figure 1.4). Finally, the last step of 3-*exo-tet* cyclization of γ -Co(III)-alkyl radical intermediate is a nearly barrierless process, leading to the formation of cyclopropane product while regenerating the metalloradical catalyst [Co(Por)]. Scheme 1.8. Proposed Catalytic Cycle of Co(II)-Catalyzed Asymmetric Radical Cyclopropanation







To probe the existence of α -Co(III)-alkyl radical intermediate in the proposed mechanism, ethyl styryldiazoacetate was employed as the carbene source for a stoichiometric reaction with [Co(TPP)] (TPP = 5,10,15,20-tetraphenylprophyrin) in the absence of olefin substrates (Scheme 1.9).^{7c} The initially formed α -Co(III)-allylic radical,

which could exist in its resonance form of γ -Co(III)-allylic radical, was found to undergo radical dimerization, forming a dinuclear Co(III) porphyrin complex in 90% yield whose structure was further confirmed by X-ray crystallography. Alternatively, the γ -Co(III)-allylic radical could be effectively captured by an excess amount of TEMPO (2,2,6,6-tetramethyl-1-piperidinyloxy) to afford a mononuclear Co(III) complex via C–O bond formation in 74% yield. Taken together with the previous DFT calculations, these experimental results provided another corroborating evidence for the proposed stepwise radical mechanism of Co(II)-catalyzed asymmetric radical cyclopropanation.







1.3.2. Intermolecular Radical Cyclopropanation with Acceptor/Acceptor-Substituted Diazo Compounds

In 2008, Zhang and coworkers initiated their effort to develop a new Co(II)-based catalytic system for asymmetric cyclopropanation with acceptor/acceptor-substituted diazo

compounds in the hope of addressing the longstanding challenges associated with existing catalytic systems involving electrophilic metallocarbene intermediates.¹⁰

Scheme 1.10. Co(II)-Catalyzed Asymmetric Z-Cyclopropanation with α -Nitrodiazoacetates



Their first target was α -nitrodiazoacetates considering that the resulting cyclopropanes may serve as valuable precursor for α -amino acids and aminocyclopropanes (Scheme 1.10).²⁷ Through systematic ligand evaluations, [Co(**P3**)] was identified to be the optimal catalyst for the asymmetric cyclopropanation with α -nitrodiazoacetates, even at room temperature. ²⁸ It was rationalized that the effectiveness of [Co(**P3**)] toward α nitrodiazoacetate is presumably attributed to the two potential H-bonding interactions between the chiral amido N–H moieties on the **P3** ligand and the respective N=O (-NO₂ group) and C=O (-CO₂R group) units of the diazo compound. Such noncovalent attractive

²⁷ (a) Wurz, R. P.; Charette, A. B. J. Org. Chem. **2004**, 69, 1262–1269. (b) Wurz, R. P.; Charette, A. B. Org. Lett. **2005**, 7, 2313–2316.

²⁸ Zhu, S.; Perman, J. A.; Zhang, X. P. Angew. Chem., Int. Ed. **2008**, 47, 8460–8463.

interactions are postulated to be essential in promoting the formation of the corresponding α -Co(III)-alkyl radical intermediate while enhancing its rigidity for subsequent addition reaction with the olefin substrate, which would lead to a more effective and selective catalytic process.²⁸ As one salient feature of the Co(II)-based metalloradical system, electron-deficient olefins, such as α , β -unsaturated esters and amides could be utilized for asymmetric cyclopropanation by [Co(**P3**)] albeit with diminished diastereoselectivities. Gratifyingly, [Co(**P3**)] proved to be similarly effective for catalyzing the asymmetric cyclopropanation of more challenging aliphatic olefins, forming the alkyl-substituted cyclopropane nitroesters in moderate yields with good stereoselectivities. Moreover, it is noted that the Co(II)-catalyzed radical cyclopropanation process exhibits an atypical (*Z*)-diastereoselectivity in contrast to the previously reported systems.

To further exploit the unique potential of Co(II)-based MRC for asymmetric cyclopropanation with acceptor/acceptor-substituted diazo compounds, Zhang and coworkers sought to explore the feasibility of employing α -cyanodiazoacetate as the carbene source, which represents a class of acceptor/acceptor-substituted diazo compounds that has not been successfully utilized for asymmetric olefin cyclopropanation. In 2010, Zhang and coworkers reported the first catalytic system for asymmetric cyclopropanation with α -cyanodiazoacetate (Scheme 1.11).²⁹ Given that a cyano group is generally considered as a stronger H-bonding acceptor than a nitro group, we envisioned the existence of similar double H-bonding interactions in the resulting α -Co(III)-alkyl radical intermediate.

²⁹ Zhu, S.; Xu, X.; Perman, J. A.; Zhang, X. P. J. Am. Chem. Soc. **2010**, 132, 12796–12799.

Scheme 1.11. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with α -

Cyanodiazoacetates



Using styrene as a model substrate, [Co(P3)] was found to be equally effective for the cyclopropanation reaction with α -cyanodiazoacetate. Further optimization of the reaction conditions revealed that the use of *tert*-butyl α -cyanodiazoacetate in *n*-hexane at -20 °C would afford the desired 1,1-cyclopropanenitrile ester in high yield with complete (*E*)-diastereoselectivity and 98% ee. In addition to styrene derivatives, the [Co(P3)]-based cyclopropanation is highlighted by a broad scope of olefins, including both electron-rich and electron-deficient substrates containing various functionalities. In this work, the resulting enantioenriched cyclopropanenitrile esters were also utilized in several stereospecific transformations.

While attempting to extend the application of Co(II)-based cyclopropanation to other acceptor/acceptor-substituted diazo compounds, Zhang and coworkers uncovered that dicarbonyl diazo compound, such as α -ketodiazoacetates bearing both ketone and ester functionalities could also be utilized as viable carbene sources (Scheme 1.12).³⁰ At the beginning of the investigations, it was unclear whether the metalloradical catalyst $[Co(D_2 - D_2 - D_2)]$ Por^{*})] could still render effective control of both enantioselectivity and diastereoselectivity for the cyclopropanation reaction with α -ketodiazoacetate. In view of the similar sizes of the two geminal carbonyl groups, could the incoming olefin substrate effectively discriminate the two prochiral faces of α -Co(III)-alkyl radical intermediate during the first C-C bond forming step? For the same consideration, additional uncertainty might arise from the subsequent 3-exo-tet cyclization of γ -Co(III)-alkyl radical intermediate. Through the identification of the optimal metalloradical catalyst [Co(P3)], it was demonstrated that tert-butyl acetodiazoacetate could be effectively activated, even at room temperature for the cyclopropanation of a wide range of olefins. In addition to high enantioselectivity, the Co(II)-based cyclopropanation system exhibits a distinct sense of diastereoselectivity from the previously reported systems involving electrophilic metallocarbenes, enabling for the first time the direct synthesis of chiral (E)-1,1-cyclopropane ketoesters. In this work, an iodide-promoted stereospecific epimerization process was also disclosed, which allows practical access to the corresponding (Z)-diastereomers with retention of the original optical purity.

³⁰ Xu, X.; Zhu, S.; Cui, X.; Wojtas, L.; Zhang, X. P. Angew. Chem., Int. Ed. **2013**, 52, 11857–11861.

Scheme 1.12. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with α -

Ketodiazoacetates



Recently, Zhang developed a new Co(II)-based catalytic system that is highly effective in activating α -formyldiazoacetates for asymmetric radical cyclopropanation, providing a general access to 1,1-cyclopropaneformylesters in high yields with excellent control of both diastereoselectivity and enantioselectivity (Scheme 1.13).³¹ Given that many existing systems are incompatible with the aldehyde functionality,³² this work represents the first application of α -formyl diazo compounds for metal-catalyzed asymmetric cyclopropanation. During the initial evaluation, a dramatic difference in catalytic performance was observed between the optimal metalloradical [Co(**P3**)] and [Co(TPP)],

³¹ Xu, X.; Wang, Y.; Cui, X.; Wojtas, L.; Zhang, X. P. Chem. Sci. 2017, 8, 4347–4351.

³² Wenkert, E.; Ananthanarayan, T. P.; Ferreira, V. F.; Hoffmann, M. G.; Kim, H. S. *J. Org. Chem.* **1990**, *55*, 4975–4976.

which is consistent with the hypothesized role of the double H-bonding interactions in activating the diazo compound while stabilizing the resulting C-centered radical intermediate for exquisite stereocontrol. Under the optimized conditions, it was further demonstrated that the Co(II)-catalyzed cyclopropanation features a broad scope of olefins as well as high functional group tolerance. As potential applications, the resulting enantioenriched 1,1-cyclocylopropaneformyl esters could be readily transformed into various chiral cyclopropane derivatives and 2,3-dihydrofurans while retaining high enantiopurity.

Scheme 1.13. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with α-Formyldiazoacetates

$^{1}R \longrightarrow R^{2}$ + 1.0 equiv	$N_2 \xrightarrow[CO_2R^3]{CO(P3)] (2 \text{ mol } \%)}_{CO_2R^3} \xrightarrow[toluene; 40 °C; 24]{toluene; 40 °C; 24}$	$\frac{1}{h} \xrightarrow{3} RO_2C_{,,} \\ OHC \\ R^1$		
	(TPP = Tetraphenylporphyrin)	Me Me Me Me O N-H H D O N-H H-N		
Ó	[Co(P3)]			
< 10% yield (with styrene; ³ R = Et)	Rigidification through Double H-bonding Interactions	_{Me} ∕ [™] Me ^{Me} ^{Me} 84% yield 95:5 dr; 96% ee (with styrene; ³ R = ⁴ Bu)		
⁷ BuO ₂ C., H Br OHC	^t BuO ₂ C,,,,,,,,,,,H OHC OEt	^t BuO ₂ C., H OHC		
86% yield 92:8 dr; 98% ee	74% yield 60:40 dr; 96% ee	89% yield 95:5 dr; 80% ee		
^t BuO ₂ C,,,,H Ph EtO ₂ C	BuO ₂ C, H Ph	⁰ ,Ph(<i>o</i> -Br) ^t BuO ₂ C		
95:5 dr; 96% ee	95:5 dr; 94% ee	74% yield; 82% ee		

1.3.3. Intermolecular Radical Cyclopropanation with Donor-Substituted Diazo Compounds

Having recognized the remarkable catalytic capability of $[Co(D_2-Por^*)]$ in activating different acceptor- and acceptor/acceptor-substituted diazo compounds for asymmetric radical cyclopropanation, Zhang and coworkers aimed to explore whether the Co(II)-based catalytic system could be applied to other types of diazo compounds with substituents beyond carbonyl groups. Specifically, it would be highly appealing if the Co(II)-based metalloradical system could be applied for the asymmetric cyclopropanation with donorsubstituted diazo compounds, which remains as an ongoing challenge in the field. This underdevelopment is largely attributed to the inherent instability of donor-substituted diazo compounds as well as their propensity towards unwanted dimerization.³³ To circumvent this issue, Zhang and coworkers envisioned the use of *N*-sulfonyl hydrazones as stable diazo precursors for the in situ-generation of less stabilized donor-substituted diazo compounds in the presence of base (Scheme 1.14).³⁴

³³ (a) Doyle, M. P.; High, K. G.; Oon, S.-M.; Osborn, A. K. *Tetrahedron Lett.* **1989**, *30*, 3049–3052. (b) Regitz, M.; Maas, G. *Diazo Compounds: Properties and Synthesis*; Academic Press: London, 1996. (c) Aggarwal, V. K.; de Vicente, J.; Bonnert, R. V. *Org. Lett.* **2001**, *3*, 2785–2788.

³⁴ (a) Fulton, J. R.; Aggarwal, V. K.; de Vicente, J. *Eur. J. Org. Chem.* **2005**, 1479–1492. (b) Xia, Y.; Wang, J. *Chem. Soc. Rev.* **2017**, *46*, 2306–2362.
Scheme 1.14. Proposed Mechanism of Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with In Situ-Generated α-Aryldiazomethanes



However, it was unclear whether the Co(II)-based catalytic system would be compatible with the basic condition required for the in situ-generation protocol. Additionally, it would be crucially important to match the rate of generation of diazo compounds with the ensuing metalloradical activation in order to inhibit dimerization and azine formation.^{33c} On the basis of our previous studies, the potential H-bonding interactions between the chiral amide units of the porphyrin ligands and the carbonyl groups of the diazo compounds have been postulated to facilitate effective asymmetric induction. In the absence of carbonyl functionalities, what element could be exploited for the stereoselective control of the radical cyclopropanation processes would be another elusive question.

To address these and related issues, in 2017, Zhang reported the first Co(II)-based metalloradical system that is highly effective for the asymmetric cyclopropanation of alkenes with α -aryldiazomethanes, generated in situ from the corresponding *N*-sulfonyl hydrazones in the presence of base (Scheme 1.15).^{7d} Initial experiments were performed to

Scheme 1.15. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with In Situ-

Generated *α*-Aryldiazomethanes



examine the feasibility of Co(II)-based MRC for asymmetric cyclopropanation of styrene with different substituted benzaldehyde tosylhydrazones under basic conditions. Using second-generation metalloradical catalyst [Co(P7)] (P7 = 3,5-Di'Bu-Xu(2'-Naph)Phyrin) containing naphthyl groups in the chiral amide units, *o*-methoxybenzaldehyde tosylhydrazone was found to be an effective radical precursor to provide the desired 1,2bisaryl cyclopropane in good yield (78%) with high diastereoselectivity (95:5 dr) and excellent enantioselectivity (99% ee). Conversely, the asymmetric induction of the cyclopropanation process was significantly diminished when the *o*-OMe substituent was replaced by the sterically comparable *o*-Et group. Moreover, non-substituted phenyldiazomethane as well as *m*- and *p*-methoxyphenyldiazomethanes proved to be less effective radical precursors for the cyclopropanation reaction (Scheme 1.15). Based on these observations, we hypothesized that the installation of H-bonding acceptors such as methoxy group at the *ortho*-position is necessary for engaging in potential H-bonding interactions with the amido group of the chiral ligand, which could rigidify the conformation of radical intermediates for effective control of asymmetric induction. In addition to styrene, we revealed that different classes of alkanes, including the more challenging electron-rich and electron-deficient nonaromatic olefins, could be effectively cyclopropanated by [Co(P7)] under the optimized conditions.

Guided by the hypothesis of H-bonding interactions, the scope of sulfonyl hydrazones was also evaluated as radical precursors for [Co(**P7**)]-catalyzed asymmetric cyclopropanation (Scheme 1.16). Notably, fluoroarene-based tosylhydrazones were shown to be more reactive radial precursors, allowing for the productive formation of fluorinated cyclopropanes even at room temperature.^{7d} Interestingly, the use of the corresponding trishydrazone (2,4,6-triisopropylphenylsulfonyl hydrazone) enabled the cyclopropanation process to proceed at 0 °C, affording the cyclopropanes with improved stereoselectivities.^{7d}

Scheme 1.16. Asymmetric Cyclopropanation of Styrene with Different Sulfonyl Hydrazones Catalyzed by [Co(P7)]



More recently, the reported procedure of Co(II)-catalyzed cyclopropanation of alkenes with in situ generated α -aryldiazomethanes can be extended to the use of dehydroaminocarboxylates for the direct synthesis of chiral cyclopropyl α -amino acid derivatives, which may serve as important non-proteinogenic α -amino acid building blocks for the design and preparation of peptides with restricted conformations.³⁵ In addition to the inherent steric hindrance of 1,1-disubstituted olefins, the synthetic challenge associated with the preparation of such cyclopropanes is largely due to the electronic incompatibility of dehydroaminocarboxylates as electron-deficient olefins with the existing catalytic systems involving electrophilic metallocarbene intermediates. Considering that the metalloradical system would be less sensitive to the electronic requirements of the substrates, Zhang and coworkers explored the suitability of dehydroaminocarboxylates for Co(II)-catalyzed asymmetric radical cyclopropanation (Scheme 1.17).^{7e} In the preliminary studies, a similar positive effect of installing -OMe group at the ortho-position of the in situ-generated α -aryldiazomethane was observed, which could be effectively activated by [Co(P7)] (P7 = 3,5-Di'Bu-Xu(2'-Naph)Phyrin) for the cyclopropanation reaction of dehydroaminocarboxylate, affording the desired chiral cyclopropyl α -amino ester in excellent yield (98%) with high (Z)-diastereoselectivity (82:18 dr) and excellent enantioselectivity (93% ee). Having established the optimal reaction conditions, a broad combination of α -aryldiazomethanes and dehydroaminocarboxylates with varied steric and electronic properties was evaluated for the [Co(P7)]-based catalytic system, providing

³⁵ (a) Jiménez, A. I.; Marraud, M.; Cativiela, C. *Tetrahedron Lett.* **2003**, *44*, 3147–3150. (b) Brackmann, F.; de Meijere, A. *Chem. Rev.* **2007**, *107*, 4493–4537.

practical access to different α -amino- β -arylcyclopropanecarboxylates in high yields with excellent stereoselectivities.

Scheme 1.17. Asymmetric Radical Cyclopropanation of Dehydroaminocarboxylates with α-Aryldiazomethanes



Furthermore, Zhang and coworkers performed detailed DFT calculations to gain insights into the basis of reactivity and stereoinduction, including the revelation of a network of multiple noncovalent attractive interactions, such as multiple H-bonding and π -stacking interactions in the optimized structures of intermediates and transition states.^{7e} In addition to computational elucidation, the proposed stepwise radical mechanism is further supported by EPR spectroscopy as well as trapping experiments with thiol and TEMPO (Scheme 1.18). To demonstrate the synthetic applications of this new catalytic system, Zhang and coworkers also showcased the resulting enantioenriched α -amino- β arylcyclopropanecarboxylates can serve as useful building blocks for the synthesis of dipeptides that may possess interesting biological properties (Scheme 1.18).

Scheme 1.18. Mechanistic Studies on Co(II)-Catalyzed Radical Cyclopropanation of

Dehydroaminocarboxylates with α-Aryldiazomethanes



A. Trapping of α-Co(III)-Benzyl Radical Intermediate by Thiol

In 2021, Zhang and coworkers leveraged the analogous α -heteroaryldiazomethanes for the stereoselective preparation of valuable chiral heteroaryl cyclopropanes,³⁶ which was anticipated to be challenging in view of the potential competitive coordination of the heteroaryl moieties to the metal center.^{33b,37} To date, there have been only a few reports on metal-based catalytic systems for direct synthesis of heteroaryl cyclopropanes from α heteroaryldiazomethanes.^{16f,38} Systematic reaction and catalyst development led to the

³⁶ (a) Liu, H.; Kerdesky, F. A.; Black, L. A.; Fitzgerald, M.; Henry, R.; Esbenshade, T. A.; Hancock, A. A.; Bennani, Y. L. *J. Org. Chem.* **2004**, *69*, 192–194. (b) Butcher, K. J.; Denton, S. M.; Field, S. E.; Gillmore, A. T.; Harbottle, G. W.; Howard, R. M.; Laity, D. A.; Ngono, C. J.; Pibworth, B. A. *Org. Process Res. Dev.* **2011**, *15*, 1192–1200.

³⁷ Bajaj, P.; Sreenilayam, G.; Tyagi, V.; Fasan, R. Angew. Chem., Int. Ed. 2016, 55, 16110–16114.

³⁸ (a) Liu, Z.; Zhang, X.; Zanoni, G.; Bi, X. *Org. Lett.* **2017**, *19*, 6646–6649. (b) Roy, S.; Das, S. K.; Chattopadhyay, B. *Angew. Chem., Int. Ed.* **2018**, *57*, 2238–2243. (c) Sharland, J.; Wei, B.; Hardee, D.; Hodges, T.; Gong, W.; Voight, E.; Davies, H. *Chem. Sci.* **2021**, *12*, 11181–11190.

identification of a newly-synthesized C₆-bridged metalloradical catalyst [Co(**P9**)] (**P9** = 2,6-DiPhO-Hu(C₆)Phyrin) as the optimal catalyst for the cyclopropanation reaction, resulting in almost quantitative formation of 2-pyridylcyclopropane (99%) with high diastereoselectivity (92% de) as essentially one enantiomer of the major (*E*)-isomer (99% ee).^{7f} Furthermore, the [Co(**P9**)]-based catalytic system was demonstrated to exhibit a broad substrate scope, including several classes of challenging olefin substrates, enabling a general method for the stereoselective synthesis of highly enantioenriched heteroaryl cyclopropanes (Scheme 1.19).

Scheme 1.19. Co(II)-Catalyzed Asymmetric Radical Cyclopropanation with α-Heteroaryldiazomethanes



In addition, Zhang and coworkers revealed that the newly-developed bridged catalyst [Co(P9)] could be applied to α -aryldiazomethanes beyond those containing H-bonding acceptors at the *ortho*-position, which is perceived as a major limitation in the previous systems.^{7d,7e} Accordingly, a wide array of α -aryldiazomethanes without H-bonding acceptors could also be employed as effective radical precursors for asymmetric cyclopropanation by [Co(P9)], forming the corresponding arylcyclopropanes in similarly high yields with the same high level of stereoselectivities (Scheme 1.20).^{7f} Together with the wide scope of α -heteroaryldiazomethanes, [Co(P9)]-catalyzed asymmetric radical cyclopropanation constitutes a general protocol for the stereoselective preparation of both aryl- and heteroarylcyclopropanes.





1.4. COBALT(II)-CATALYZED INTRAMOLECULAR RADICAL CYCLOPROPANATION

In parallel with the development of asymmetric intermolecular cyclopropanation, its intramolecular variant has also attracted growing research interest as a powerful strategy for the stereoselective construction of complex [n.1.0] bicyclic ring structures, which may serve as versatile structural motifs in many biologically active compounds and natural products.³⁹ Despite significant progress, the realization of asymmetric intramolecular cyclopropanation remains challenging, with success being reported mostly with acceptor-and acceptor/donor-substituted diazo compounds.⁴⁰ Inspired by the recent demonstration of Co(II)-based MRC for asymmetric intermolecular cyclopropanations, Zhang and coworkers further explored the metalloradical approach for asymmetric intramolecular cyclopropanation with acceptor/acceptor-substituted diazo compounds.⁴¹ Considering the more demanding steric requirements associated with the double cyclization reactions, it was unclear whether the confined chiral cavity environment of [Co(D_2 -Por*)] could accommodate the intramolecular catalytic process.

³⁹ (a) Monn, J. A.; Valli, M. J.; Massey, S. M.; Wright, R. A.; Salhoff, C. R.; Johnson, B. G.; Howe, T.; Alt, C. A.; Rhodes, G. A.; Robey, R. L.; Griffey, K. R.; Tizzano, J. P.; Kallman, M. J.; Helton, D. R.; Schoepp, D. D. *J. Med. Chem.* **1997**, *40*, 528–537. (b) Li, J.; Lowary, T. L. Org. Lett. **2008**, *10*, 881–884.

⁴⁰ (a) Doyle, M. P.; Pieters, R. J.; Martin, S. F.; Austin, R. E.; Oalmann, C. J.; Mueller, P. J. Am. Chem. Soc. **1991**, *113*, 1423–1424. (b) Doyle, M. P.; Davies, S. B.; Hu, W. Org. Lett. **2000**, *2*, 1145–1147. (c) Uchida, T.; Saha, B.; Katsuki, T. Tetrahedron Lett. **2001**, *42*, 2521–2524. (d) Xu, Z.-J.; Fang, R.; Zhao, C.; Huang, J.-S.; Li, G.-Y.; Zhu, N.; Che, C.-M. J. Am. Chem. Soc. **2009**, *131*, 4405–4417. (e) Shen, J.-J.; Zhu, S.-F.; Cai, Y.; Xu, H.; Xie, X.-L.; Zhou, Q.-L. Angew. Chem., Int. Ed. **2014**, *53*, 13188–13191. (f) Nakagawa, Y.; Chanthamath, S.; Shibatomi, K.; Iwasa, S. Org. Lett. **2015**, *17*, 2792–2795. (g) Marichev, K. O.; Ramey, J. T.; Arman, H.; Doyle, M. P. Org. Lett. **2017**, *19*, 1306–1309. (h) Ren, X.; Chandgude, A. L.; Fasan, R. ACS Catal. **2020**, *10*, 2308–2313.

⁴¹ Xu, X.; Lu, H. J.; Ruppel, J. V.; Cui, X.; de Mesa, S. L.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. **2011**, *133*, 15292–15295.

The initial efforts were focused on the identification of a suitable D_2 -symmetric chiral amidoporphyrin ligand for the asymmetric intramolecular cyclopropanation with cinnamyl α -cyanodiazoacetate as a model substrate (Scheme 1.21).

Scheme 1.21. Asymmetric Intramolecular Cyclopropanation of Acceptor-Substituted Diazoacetates by Co(II)-MRC



[Co(P3)], which was previously shown to be the optimal catalyst for asymmetric intermolecular cyclopropanation with α -cyanodiazoacetate²⁹ was found to be capable of activating cinnamyl α -cyanodiazoacetate for the intramolecular cyclopropanation process, affording the desired bicyclic product in nearly quantitative yield (99%) as a single diastereomer (99% de) while exhibiting significant enantioselectivity (55% ee). To further improve the enantioselectivity of this catalytic system, a new generation of metalloradical catalyst [Co(P6)] (P6 = 3,5-Di/Bu-QingPhyrin) containing cyclopropanecarboxyamides

with two contiguous stereogenic centers was developed based on [Co(P3)]-catalyzed asymmetric intermolecular cyclopropanation (Scheme 1.22).⁴¹

Switching [Co(P3)] to second-generation metalloradical catalyst [Co(P6)] led to significant enhancement in enantioselectivity (99%) without affecting the yield and diastereoselectivity. It is noted that the sense of asymmetric induction also switched in favor of the opposite enantiomer with the use of [Co(P6)]. Similarly, the new metalloradical catalyst [Co(P6)] could be successfully applied for the asymmetric intramolecular cyclopropanation with various α -substituted allyl diazoacetates, enabling a streamlined synthesis of highly enantioenriched 3-oxabicyclo[3.1.0]hexan-2-one derivatives that may serve as valuable intermediates for stereoselective organic synthesis.

Scheme 1.22. Synthetic Approach to Construct [Co(P6)] (P6 = 3,5-Di'Bu-QingPhyrin)



1.5. NEXT-GENERATION *D***₂-SYMMETRIC CHIRAL PORPHYRINS:**

CATALYST ENGINEERING BY DISTAL BRIDGING

Since the first introduction in 2004,²⁰ Co(II)-complexes of D_2 -symmetric chiral amidoporphyrins [Co(D_2 -Por^{*})] have emerged as a privileged class of metalloradical catalysts capable of controlling reactivity and stereoselectivity in various radical catalytic systems based on Co(II)-MRC. The key to the success of these radical processes lies in the

judicious modulation of D_2 -symmetric chiral amidoporphyrin ligands to adopt proper steric, electronic, and chiral environments that maximize noncovalent attractive interactions with the reacting substrates and active radical intermediates. On the basis of the remarkable ligand effect observed with existing [Co(D_2 -Por^{*})], Zhang and coworkers reported the synthesis of D_2 -symmetric chiral amidoporphyrins containing alkyl bridges across two chiral amide units on both sides of the porphyrin plane.⁴² The new-generation metalloradical catalyst [Co(HuPhyrin)] whose cavity can be fine-tuned by varying the length of alkyl bridges features a more rigid chiral environment as a result of the bridging effect.

Using asymmetric cyclopropanation of styrene with EDA as a model reaction, the catalytic performance of C₄-bridged metalloradical catalyst [Co(P11)] (P11 = 3,5-Di'Bu-Hu(C₄)Phyrin) and C₆-bridged metalloradical catalyst [Co(P12)] (P12 = 3,5-Di'Bu-Hu(C₆)Phyrin) was evaluated (Scheme 1.23). Both bridged catalysts proved to be superior in terms of reactivity and stereoselectivity as compared to the open-analogue [Co(P10)] (P10 = 3,5-Di'Bu-Tao(Et)Phyrin). Furthermore, the difference in performance between [Co(P11)] and [Co(P12)] demonstrates that even a subtle modification to the distal alkyl bridges can give rise to significant improvements in reactivity as well as stereoselectivity, manifesting the immense power of catalyst development in controlling radical processes.

⁴² Hu, Y.; Lang, K.; Tao, J.; Marshall, M. K.; Cheng, Q.; Cui, X.; Wojtas, L.; Zhang, X. P. *Angew. Chem., Int. Ed.* **2019**, *58*, 2670–2674.

Scheme 1.23. Bridging Effect on Co(II)-Catalyzed Asymmetric Cyclopropanation of

Styrene with EDA



1.6. CONCLUSIONS AND OUTLOOK

This Account outlined Zhang and coworkers' recent endeavors toward developing metalloradical catalysis as a conceptually different approach for catalytic radical cyclopropanation with diazo compounds in the past decades. Attributed to the underlying stepwise radical mechanism, the Co(II)-catalyzed cyclopropanation exhibits a unique profile of reactivity and selectivity that is fundamentally distinct from the existing ionic systems involving metallocarbenes. Through the identification of suitable D_2 -symmetric chiral amidoporphyrins as the supporting ligands, the Co(II)-based metalloradical system has been shown to effectively activate different classes of diazo compounds as carbene sources for asymmetric radical cyclopropane derivatives in high yields with excellent stereoselectivities. In particular, $[Co(D_2-Por^*)]$ exhibit capability in activating various acceptor/acceptor-substituted diazo compounds, which are attractive but typically

challenging candidates for asymmetric olefin cyclopropanation. More broadly, the Co(II)based cyclopropanation process is highlighted by its exceptional catalytic reactivity toward electron-deficient olefins owing to the salient features of radical- mediated reaction pathway. In parallel, the application of Co(II)-based MRC has been extend to the intramolecular variant of asymmetric radical cyclopropanation, allowing for the stereoselective construction of densely functionalized 3-oxabicyclo[3.1.0]hexan-2-one derivatives, which may serve as valuable chiral building blocks in a myriad of potential synthetic and biological applications. Using N-sulfonyl hydrazones as stable diazo surrogates, the feasibility of employing in situ-generated α -aryldiazomethanes and α heteroaryldiazomethanes has been demonstrated as effective radical precursors for Co(II)catalyzed olefin cyclopropanation. Furthermore, detailed computational and experimental results have shed light on the underlying stepwise radical mechanism as well as the basis of asymmetric induction. Theses mechanistic insights may have far-reaching implications for future catalyst design and optimization of novel radical reactions. While significant advances have been achieved to date, the development of Co(II)-based asymmetric radical cyclopropanation still confronts foreseen challenges. For instance, the utilization of other types of diazo compounds, especially with substituents beyond C(sp²)-based carbonyl and aryl groups has remained elusive. The formidable challenges associated with controlling reactivity and selectivity of radical reactions can be essentially translated into a solvable problem of rational catalyst design and development. It is our hope that this Account will provide a useful perspective and stimulate further exploration of untapped potential of radical reactions in stereoselective organic synthesis.

CHAPTER 2

ASYMMETRIC RADICAL PROCESS FOR GENERAL SYNHTESIS OF CHIRAL HETEROARYL CYCLOPROPANES

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2.1. INTRODUCTION

Radical chemistry has been increasingly explored for the development of new synthetic tools in modern organic synthesis. ¹ Despite tremendous endeavors, long-standing challenges associated with control of reactivity and enantioselectivity remain largely unresolved for many radical reactions.² Among recent advances,³ metalloradical catalysis (MRC), which involves the generation and utilization of metal-stabilized organic radicals as catalytic intermediates to harness the potential of radical chemistry, has emerged as a

¹ For selected books, see: *Radical Reactions in Organic Synthesis*; Oxford University Press, 2003. (b) Chatgilialoglu, C.; Studer, A., *Encyclopedia of Radicals in Chemistry, Biology, and Materials*; John Wiley & Sons, 2012. For selected reviews, see: (c) Zard, S. Z. *Chem. Soc. Rev.* **2008**, *37*, 1603–1618. (d) Narayanam, J. M.; Stephenson, C. R. *Chem. Soc. Rev.* **2011**, *40*, 102–113. (e) Quiclet-Sire, B.; Zard, S. Z. *Pure Appl. Chem.* **2011**, *83*, 519–551. (f) Prier, C. K.; Rankic, D. A.; MacMillan, D. W. *Chem. Rev.* **2013**, *113*, 5322–5363. (g) Studer, A.; Curran, D. P. *Angew. Chem., Int. Ed.* **2016**, *55*, 58–102.

² For selected reviews, see: (a) Bar, G.; Parsons, A. F. *Chem. Soc. Rev.* 2003, *32*, 251–263. (b) Sibi, M. P.;
Manyem, S.; Zimmerman, J. *Chem. Rev.* 2003, *103*, 3263–3295. (c) Brimioulle, R.; Lenhart, D.; Maturi, M. M.; Bach, T. *Angew. Chem., Int. Ed.* 2015, *54*, 3872–3890.

³ For selected examples on approaches to controlling radical reactivity and stereoselectivity, see: (a) Du, J. N.; Skubi, K. L.; Schultz, D. M.; Yoon, T. P. *Science* **2014**, *344*, 392–396. (b) Huo, H.; Shen, X.; Wang, C.; Zhang, L.; Röse, P.; Chen, L.-A.; Harms, K.; Marsch, M.; Hilt, G.; Meggers, E. *Nature* **2014**, *515*, 100–103. (c) Kainz, Q. M.; Matier, C. D.; Bartoszewicz, A.; Zultanski, S. L.; Peters, J. C.; Fu, G. C. *Science* **2016**, *351*, 681–684. (d) Zhang, W.; Wang, F.; McCann, S. D.; Wang, D. H.; Chen, P. H.; Stahl, S. S.; Liu, G. S. *Science* **2016**, *353*, 1014–1018. (e) Kern, N.; Plesniak, M. P.; McDouall, J. J. W.; Procter, D. J. *Nat. Chem.* **2017**, *9*, 1198–1204. (f) Morrill, C.; Jensen, C.; Just-Baringo, X.; Grogan, G.; Turner, N. J.; Procter, D. J. *Angew. Chem., Int. Ed.* **2018**, *57*, 3692–3696. (g) Proctor, R. S. J.; Davis, H. J.; Phipps, R. J. *Science* **2018**, *360*, 419–422. (h) Biegasiewicz, K. F.; Cooper, S. J.; Gao, X.; Oblinsky, D. G.; Kim, J. H.; Garfinkle, S. E.; Joyce, L. A.; Sandoval, B. A.; Scholes, G. D.; Hyster, T. K. *Science* **2019**, *364*, 1166–1169. (i) Huang, H.-M.; McDouall, J. J. W.; Procter, D. J. *Nat. Catal.* **2019**, *2*, 211–218. (j) Nakafuku, K. M.; Zhang, Z.; Wappes, E. A.; Stateman, L. M.; Chen, A. D.; Nagib, D. A. *Nat. Chem.* **2020**, *12*, 697–704.

conceptually new approach to guide the discovery of catalytic solutions toward controlling the reactivity and stereoselectivity of radical processes.^{4,5,6} As stable 15e-metalloradicals, cobalt(II) complexes of porphyrins ([Co(Por)]) exhibit unique capability of homolytically activating diazo compounds to generate α -Co(III)-alkyl radicals as key intermediates for various radical transformations.⁷ Specifically, with D_2 -symmetric chiral amidoporphyrins (D_2 -Por^{*}) as the supporting ligands, these Co-stabilized carbon-centered radicals can engage in asymmetric radical cyclopropanation of alkenes for the preparation of optically active three-membered carbocycles.⁸ While donor-substituted diazo compounds such as in

⁴ For selected reviews and highlights on Co(II)-based MRC, see: (a) Lu, H. J.; Zhang, X. P. *Chem. Soc. Rev.* **2011**, *40*, 1899–1909. (b) Pellissier, H.; Clavier, H. *Chem. Rev.* **2014**, *114*, 2775–2823. (c) Demarteau, J.; Debuigne, A.; Detrembleur, C. *Chem. Rev.* **2019**, *119*, 6906–6955. (d) Huang, H.-M.; Garduño-Castro, M. H.; Morrill, C.; Procter, D. J. *Chem. Soc. Rev.* **2019**, *48*, 4626–4638. (e) Singh, R.; Mukherjee, A. *ACS Catal.* **2019**, *9*, 3604–3617.

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situ-generated α -aryldiazomethanes have recently been demonstrated as suitable radical precursors for Co(II)-based asymmetric radical cyclopropanation,^{8k,8n} the analogous α -heteroaryldiazomethanes have remained underexploited for stereoselective synthesis of valuable chiral heteroaryl cyclopropanes.⁹ With this in mind, we sought to explore the feasibility of developing a catalytic process that would employ α -heteroaryldiazomethanes for asymmetric cyclopropanation of alkenes via Co(II)-MRC (Scheme 2.1).

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⁹ For selected examples of stereoselective synthesis of heteroaryl cyclopropanes, see: (a) Charette, A. B.; Molinaro, C.; Brochu, C. J. Am. Chem. Soc. 2001, 123, 12168–12175. (b) Davies, H. M. L.; Townsend, R. J. J. Org. Chem. 2001, 66, 6595–6603. (c) Marcin, L. R.; Denhart, D. J.; Mattson, R. J. Org. Lett. 2005, 7, 2651–2654. (d) Chuprakov, S.; Kwok, S. W.; Zhang, L.; Lercher, L.; Fokin, V. V. J. Am. Chem. Soc. 2009, 131, 18034–18035. (e) Grimster, N.; Zhang, L.; Fokin, V. V. J. Am. Chem. Soc. 2010, 132, 2510–2511. (f) Jin, C.; Decker, A. M.; Huang, X.-P.; Gilmour, B. P.; Blough, B. E.; Roth, B. L.; Hu, Y.; Gill, J. B.; Zhang, X. P. ACS Chem. Neurosci. 2014, 5, 576–587. (g) Bajaj, P.; Sreenilayam, G.; Tyagi, V.; Fasan, R. Angew. Chem., Int. Ed. 2016, 55, 16110–16114. (h) Fu, L.; Mighion, J. D.; Voight, E. A.; Davies, H. M. L. Chem. Eur. J. 2017, 23, 3272–3275. (i) Moore, E. J.; Steck, V.; Bajaj, P.; Fasan, R. J. Org. Chem. 2018, 83, 7480–7490. (j) Vargas, D. A.; Khade, R. L.; Zhang, Y.; Fasan, R. Angew. Chem., Int. Ed. 2019, 58, 10148–10152. (k) Wei, B.; Sharland, J. C.; Lin, P.; Wilkerson-Hill, S. M.; Fullilove, F. A.; McKinnon, S.; Blackmond, D. G.; Davies, H. M. L. ACS Catal. 2020, 10, 1161–1170. (l) Nam, D.; Steck, V.; Potenzino, R. J.; Fasan, R. J. Am. Chem. Soc. 2021, 143, 2221–2231. (m) Sharland, J. C.; Wei, B.; Hardee, D. J.; Hodges, T. R.; Gong, W.; Voight, E. A.; Davies, H. M. L. Chem. Sci. 2021, 12, 11181–11190.

Scheme 2.1. Working Proposal for Synthesis of Heteroaryl Cyclopropanes from Alkenes via Co(II)-Based MRC



In view of the intrinsic properties of heteroaryl moieties, the proposed Co(II)-based catalytic process presented several fundamental challenges. Besides the concerns with efficiency of metalloradical activation of α -heteroaryldiazomethanes **1'** generated in situ from the corresponding hydrazones **1** in the presence of base, whether the subsequent radical addition of the initially-formed α -Co(III)-heterobenzyl radicals **I** to the alkene substrates **2** could be rendered enantioselective is an unanswered question, primarily owing to the potential competitive coordination of the heteroaryl moieties to the metal center. On this basis, additional uncertainty of controlling reactivity and diastereoselectivity might also arise from the following 3-*exo-tet* cyclization of the resulting γ -Co(III)-alkyl radicals **II** while forging the second C–C bond (Scheme 2.1). Furthermore, the presence of heteroatoms was anticipated to engage in potential H-bonding interactions with the amide

units of the amidoporphyrin ligands that could pose potential complication in controlling both reactivity and selectivity in these Co(II)-based radical processes. To address these and related issues, we envisioned the prospect of designing a suitable D_2 -symmetric chiral amidoporphyrin ligand with proper steric, electronic and chiral environments that could direct the Co(II)-based catalysis for productive cyclopropanation with effective stereocontrol. If realized, it would enable the development of a new catalytic system for asymmetric olefin cyclopropanation with in situ generated α -heteroaryldiazomethanes to furnish valuable chiral heteroaryl cyclopropanes.

Figure 2.1. Selected Examples of Important Bioactive Compounds Containing Heteroaryl Cyclopropane Motifs



Heteroaryl cyclopropanes are ubiquitous structural motifs in many pharmaceuticals and biologically important molecules (Figure 2.1).¹⁰ Nevertheless, the methods for their stereoselective synthesis remain rather limited.⁹ Among advances in the realm, Charette and coworkers disclosed an enantioselective Simmons-Smith cyclopropanation of indolyl-substituted allylic alcohol using bis(iodomethyl)zinc and titanium-TADDOLate complex (Scheme 2.2).^{9a} Mattson and coworkers later explored an alternative strategy involving asymmetric cyclopropanation of vinylindole with diazoacetate catalyzed by pybox-Ru(II) complex (Scheme 2.3).^{9c}

Scheme 2.2. Enantioselective Simmons-Smith Cyclopropanation





While most of these previous systems employed heteroaryl olefins for cyclopropanation with carbene precursors,^{9,10b} transition-metal catalyzed asymmetric cyclopropanation of

¹⁰ (a) Liu, H.; Kerdesky, F. A.; Black, L. A.; Fitzgerald, M.; Henry, R.; Esbenshade, T. A.; Hancock, A. A.; Bennani, Y. L. *J. Org. Chem.* **2004**, *69*, 192–194. (b) Butcher, K. J.; Denton, S. M.; Field, S. E.; Gillmore, A. T.; Harbottle, G. W.; Howard, R. M.; Laity, D. A.; Ngono, C. J.; Pibworth, B. A. *Org. Process Res. Dev.* **2011**, *15*, 1192–1200. (c) MacKinnon, C. H.; Lau, K.; Burch, J. D.; Chen, Y.; Dines, J.; Ding, X.; Eigenbrot, C.; Heifetz, A.; Jaochico, A.; Johnson, A.; Kraemer, J.; Kruger, S.; Krülle, T. M.; Liimatta, M.; Ly, J.; Maghames, R.; Montalbetti, C. A.; Ortwine, D. F.; Pérez-Fuertes, Y.; Shia, S.; Stein, D. B.; Trani, G.; Vaidya, D. G.; Wang, X.; Bromidge, S. M.; Wu, L. C.; Pei, Z. *Bioorg Med. Chem. Lett.* **2013**, *23*, 6331–6335. (d) Kohn, T. J.; Du, X.; Lai, S.; Xiong, Y.; Komorowski, R.; Veniant, M.; Fu, Z.; Jiao, X.; Pattaropong, V.; Chow, D.; Cardozo, M.; Jin, L.; Conn, M.; DeWolf, W. E.; Kraser, C. F.; Hinklin, R. J.; Boys, M. L.; Medina, J. C.; Houze, J.; Dransfield, P.; Coward, P. *ACS Med. Chem. Lett.* **2016**, *7*, 666–670. (e) Talele, T. T. *J. Med. Chem.* **2016**, *59*, 8712–8756.

alkenes with heteroaryldiazomethanes represents an appealing approach for the general synthesis of heteroaryl cyclopropanes with the potential to control both diastereoselectivity

Scheme 2.3. Asymmetric Cyclopropanation of Vinylindole with Diazoacetate

Mattson, R. J., 2005



and enantioselectivity.¹⁰ In contrast to the well-precedented asymmetric cyclopropanation with other types of diazo compounds,¹¹ only a few catalytic systems involving the use of heteroaryldiazomethanes have been reported.^{12,13} This underdevelopment is largely attributed to the inherent instability of α -heteroaryldiazomethanes as well as their high propensity for unwanted formal dimerization.^{12,14} Moreover, it is known that rhodium- and other existing metal-based catalytic systems of cyclopropanation could suffer from the

¹¹ For general reviews on metal-catalyzed olefin cyclopropanation with diazo compounds, see: (a) Doyle, M.

P. Chem. Rev. 1986, 86, 919–939. (b) Doyle, M. P.; Forbes, D. C. Chem. Rev. 1998, 98, 911–935. (c) Intrieri,

D.; Carminati, D. M.; Gallo, E. Dalton Trans. 2016, 45, 15746–16048.

¹² Allouche, E. M. D.; Charette, A. B. Synthesis **2019**, *51*, 3947–3963.

¹³ For selected examples of olefin cyclopropanation with in situ-generated α-heteroaryldiazomethanes, see: (a) Zimmerman, H. E.; Ignatchenko, A. J. Org. Chem. **1999**, 64, 6635–6645. (b) Barluenga, J.; Quiñones, N.; Tomás-Gamasa, M.; Cabal, M.-P. *Eur. J. Org. Chem.* **2012**, 2312–2317. (c) Roda, N. M.; Tran, D. N.; Battilocchio, C.; Labes, R.; Ingham, R. J.; Hawkins, J. M.; Ley, S. V. Org. Biomol. Chem. **2015**, *13*, 2550– 2554. (d) Liu, Z.; Zhang, X.; Zanoni, G.; Bi, X. Org. Lett. **2017**, *19*, 6646–6649. (e) Allouche, E. M. D.; Al-Saleh, A.; Charette, A. B. Chem. Commun. **2018**, *54*, 13256–13259. (f) Roy, S.; Das, S. K.; Chattopadhyay, B. Angew. Chem., Int. Ed. **2018**, *57*, 2238–2243. (g) Zhang, Z.; Yadagiri, D.; Gevorgyan, V. Chem. Sci. **2019**, *10*, 8399–8404. (h) Shang, Z.-H.; Zhang, Z.-X.; Weng, W.-Z.; Wang, Y.-F.; Cheng, T.-W.; Zhang, Q.-Y.; Song, L.-Q.; Shao, T.-Q.; Liu, K.-X.; Zhu, Y.-P. Adv. Synth. Catal. **2021**, *363*, 490–496.

¹⁴ (a) Doyle, M. P.; High, K. G.; Oon, S.-M.; Osborn, A. K. *Tetrahedron Lett.* **1989**, *30*, 3049–3052. (b) Regitz, M.; Maas, G. Academic Press: London, 1996. (c) Aggarwal, V. K.; de Vicente, J.; Bonnert, R. V. Org. Lett. **2001**, *3*, 2785–2788. (d) Fulton, J. R.; Aggarwal, V. K.; de Vicente, J. Eur. J. Org. Chem. **2005**, 1479–1492.

notorious catalyst poisoning effect in the presence of nitrogen- and sulfur-containing heterocycles.^{9g,10b}

Scheme 2.4. [ClFe(TPP)]-Catalyzed Olefin Cyclopropanation with α-Heteroaryldiazomethanes

Charette, A. B., 2018



One notable example of cyclopropanation of alkenes with α -heteroaryldiazomethanes could be found in a recent report by Charette and coworkers on Fe-based catalytic system involving in situ-generation of diazo compounds from *N*-nosylhydrazones (Scheme 2.4).^{13e} Chattopadhyay and coworkers also developed a [Co(TPP)]-catalyzed (TPP = 5,10,15,20tetraphenylporphyrin) metalloradical cyclopropanation with 2-pyridyldiazomethanes, which could be generated in situ from readily accessible *N*-tosylhydrazone precursors in the presence of base (Scheme 2.5).^{13f} While these in situ-generation protocols offer a novel alternative for catalytic synthesis of heteroaryl cyclopropanes in their racemic forms, the

Scheme 2.5 [Co(TPP)]-Catalyzed Olefin Cyclopropanation with α-Heteroaryldiazomethanes



enantioselective variant of this transformation is an attractive process that remains elusive. Moreover, it would be desirable to develop new catalytic systems that are generally applicable for stereoselective synthesis of diverse types of chiral heteroaryl cyclopropanes. We herein report the development of a new Co(II)-based catalytic system that is highly efficient for asymmetric cyclopropanation of alkenes with in situ-generated α heteroaryldiazomethanes. Through the support of a new bridged D_2 -symmetric chiral amidoporphyrin ligand, the Co(II)-catalyzed system allows for efficient activation of 2pyridyldiazomethanes and other common α -heteroaryldiazomethanes for asymmetric cyclopropanation of a broad range of alkenes, affording the valuable chiral heteroaryl cyclopropanes in high yields with both excellent diastereoselectivities and enantioselectivities. Furthermore, we present detailed computational and experimental studies that shed light on the underlying stepwise radical mechanism.

2.2. RESULTS AND DISCUSSION

2.2.1. Condition Optimization for Asymmetric Radical Cyclopropanation of Alkenes with α-Heteroaryldiazomethanes

At the outset of this project, 2-pyridyldiazomethane (1a'), which was in situ generated from the corresponding tosylhydrazone 1a in the presence of Cs₂CO₃, was investigated as the representative α -heteroaryldiazomethane for asymmetric radical cyclopropanation of styrene (2a) by a series of Co(II) complexes of D_2 -symmetric chiral amidoporphyrin ligands [Co(D_2 -Por^{*})] (Figure 2.2). It was found that first-generation chiral metalloradical catalyst [Co(P3)] (P3 = 3,5-Di'Bu-ChenPhyrin)^{8a} could promote the conversion to 2pyridylcyclopropane 3a in 20% yield with moderate diastereoselectivity (89:11 dr) and insignificant asymmetric induction (8% ee) (Figure 2.2; entry 1). Other than 3a, [1,2,3]triazolo[1,5-*a*]pyridine (4) was also formed in 53% yield (entry1), presumably through ring-chain tautomerism of in situ-generated 2-pyridyldiazomethane (1a').^{13f,13g} To facilitate formation of open-form 1a', elevated reaction temperature (80 °C) was employed to afford cyclopropane product 3a in 95% yield with 86:14 dr and 5% ee (entry 2).





In addition, we performed a control experiment with pyridyltriazole **4** as diazo precursor in the absence of Cs₂CO₃ (Scheme 2.6). Interestingly, the desired cyclopropane **3a** could be obtained in a similarly high yield and stereoselectivities, suggesting that pyridyltriazole **4** could undergo ring-chain tautomerism at elevated temperature to deliver its diazo tautomer **1a'**, which upon metalloradical activation would generate the reactive α -Co(III)pyridyl radical intermediate by [Co(Por)].

Switching to second-generation metalloradical catalyst [Co(P8)] (P8 = 3,5-Di'Bu-Tao('Bu)Phyrin)¹⁵ bearing chiral amide units with ester moieties resulted in the formation of **3a** in 86% yield with further improved diastereoselectivity (94:6 dr) and a significant level of enantioselectivity (40% ee) (entry 3). To further enhance the asymmetric induction

¹⁵ Hu, Y.; Lang, K.; Tao, J.; Marshall, M. K.; Cheng, Q.; Cui, X.; Wojtas, L.; Zhang, X. P. Angew. Chem., Int. Ed. **2019**, *58*, 2670–2674.

of this catalytic system, we then turned our attention to new-generation metalloradical catalysts [Co(HuPhyrin)], the Co(II) complexes of bridged D₂-symmetric chiral amidoporphyrins featuring more rigid cavity-like environments. When the C₄-bridged [Co(P11)] (P11 = 3,5-Di^tBu-Hu(C₄)Phyrin)¹⁵ was employed as the catalyst under the same conditions, it indeed enhanced both reactivity and stereoselectivities of the cyclopropanation reaction substantially, generating 3a in high yield (92%) with both excellent diastereoselectivity (99:1 dr) and enantioselectivity (90% ee) (entry 4). When the alkyl bridge was further extended to C₆-linker in [Co(P12)] (P12 = 3,5-Di'Bu- $Hu(C_6)$ Phyrin).¹⁵ cyclopropane **3a** was produced with a higher enantioselectivity (92% ee) while maintaining the excellent yield (94%) and diastereoselectivity (99:1 dr) (entry 5). Subsequent use of analogous catalyst [Co(P13)] (P13 = 2,6-DiMeO-Hu(C₆)Phyrin),¹⁶ which bears 2,6-dimethoxyphenyl instead of 3,5-di-*tert*-butylphenyl groups as the 5,15diaryl substituents, led to the production of **3a** in a comparable yield (91% yield) with the same high level of stereoselectivities (99:1 dr and 92% ee) (entry 6). Aiming at further improving the catalytic system, we synthesized a new C₆-bridged catalyst [Co(P9)] (P9 = 2,6-DiPhO-Hu(C_6)Phyrin) by replacing the methoxy groups in **P13** with phenoxy groups. Gratifyingly, [Co(P9)] could catalyze the cyclopropanation reaction to afford 2pyridylcyclopropane **3a** in almost quantitative yield (99%) with high diastereoselectivity (96:4 dr) and outstanding enantioselectivity (99% ee) (entry 7).

¹⁶ Lang, K.; Torker, S.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2019, 141, 12388–12396.

NNHTs						
	, ⊢н		[Co(Por)] (2 mol %)	_ N [,] ,		N
N	+		Cs ₂ CO ₃ ; toluene; temp	5 H		N-N
	1a	2a			3a 🧹	4
entry	[Co(Por)]	temp (°C)	3a yield (%)	3a (<i>E</i>):(<i>Z</i>)	3a ee (%)	4 yield (%)
1	[Co(P2)]	40	20	89:11	8	53
2	[Co(P2)]	80	95	86:14	5	-
3	[Co(P8)]	80	86	94:6	40	-
4	[Co(P11)]	80	92	99:1	90	-
5	[Co(P12)]	80	94	99:1	92	-
6	[Co(P13)]	80	91	99:1	92	-
7	[Co(P9)]	80	99	96:4	99	-
Me Me N-H						

Figure 2.2. Evaluation of Reaction Conditions^a



[Co(P12)] $(P12 = 3,5-Di^{t}Bu-Hu(C_{6})Phyrin)$

$$\label{eq:comparameters} \begin{split} & [Co(\textbf{P9})] \\ (\textbf{P9} = 2,6\text{-DiPhO-Hu}(C_6)\text{Phyrin}) \end{split}$$
$$\label{eq:compared} \begin{split} & [Co(\textbf{P13})] \\ (\textbf{P13} = 2,6\text{-}DiMeO\text{-}Hu(C_6)\text{Phyrin}) \end{split}$$
^aCarried out with 1a (0.10 mmol), 2a (0.15 mmol) and Cs₂CO₃ (0.20 mmol) using [Co(Por)] (2 mol %) in

toluene (1.0 mL) for 16 h; Isolated yields; Diastereomeric excess (de) determined by ¹H NMR of crude reaction mixture; Enantiomeric excess (ee) of the major (E)-isomer determined by chiral HPLC; Ts = 4toluenesulfonyl.

2.2.2. Asymmetric Radical Cyclopropanation of Different Alkenes with 2-Pyridyldiazomethanes

Under the optimized conditions, the scope and versatility of [Co(P9)]-catalyzed asymmetric cyclopropanation with in situ-generated 2-pyridyldiazomethane (1a') were explored by employing different types of alkenes as the substrates (Table 2.1). Like formation of **3a** from styrene (Table 2.1; entry 1), its derivatives bearing electron-donating and electron-withdrawing aryl substituents could also be effectively cyclopropanated by [Co(P6)] with 1a', producing the desired cyclopropanes 3b and 3c in similarly high yields and stereoselectivities (entries 2 and 3). The absolute configuration of the major enantiomer of **3b** was established as (R,R). Additionally, this [Co(P6)]-based metalloradical system was shown to tolerate various functional groups as exemplified by the stereoselective formation of 3d-3f containing aryl substituents of halogen, pinacolborane, and formyl functionalities at different positions (entries 4-6). Besides mono-substituted olefins, 1,1disubstituted olefins like α -substituted styrenes could serve as suitable substrates as well, affording the trisubstituted cyclopropanes 3g and 3h with excellent control of the newlygenerated quaternary stereogenic centers (entries 7 and 8). In addition to the extended aromatic olefins such as formation of **3i** from 2-vinylnaphthalene (entry 9), both conjugated dienes and envnes could be regio- and chemoselectively cyclopropanated to form cyclopropanes **3j**-**3l** in high yields with excellent stereoselectivities (entries 10–12).

Table 2.1. Asymmetric Radical Cyclopropanation of Styrene Derivatives with In Situ-

Generated 2-Pyridyldiazomethanes by [Co(P9)]^a



^{*a*}Carried out with **1a** (0.10 mmol), **2** (0.15 mmol) and Cs_2CO_3 (0.20 mmol) at 80 °C for 16 h using [Co(**P9**)] (2 mol %) in toluene (1.0 mL); Ts = 4-toluenesulfonyl; Isolated yields; Diastereomeric ratio (dr) determined by ¹H NMR of crude reaction mixture; Enantiomeric excess (ee) of the major isomer determined by chiral HPLC. ^{*b*}Absolute configuration determined by X-ray crystallography.

The Co(II)-based cyclopropanation was further highlighted by its unique reactivity towards various heteroaromatic olefins as shown with the highly stereoselective synthesis of 1,2-bisheteroaryl cyclopropanes 3m-3t containing pyridine, thiophene, benzofuran, benzothiophene, indole, and quinoline (Table 2.2, entries 1–8). Given that both heteroarene and cyclopropane are prevalent structural motifs in bioactive compounds,¹⁰ the access of

bisheteroaryl cyclopropanes in high enantiopurity may find potential applications in drug research and development.

 Table 2.2. Asymmetric Radical Cyclopropanation of Heteroaromatic Olefins with In

 Situ-Generated 2-Pyridyldiazomethanes by [Co(P9)]^a



^aCarried out with **1a** (0.10 mmol), **2** (0.15 mmol) and Cs_2CO_3 (0.20 mmol) at 80 °C for 16 h using [Co(**P9**)] (2 mol %) in toluene (1.0 mL); Ts = 4-toluenesulfonyl; Isolated yields; Diastereomeric ratio (dr) determined by ¹H NMR of crude reaction mixture; Enantiomeric excess (ee) of the major isomer determined by chiral HPLC.

Moreover, electron-deficient olefins such as acrylketones, acrylates, acrylamides and acrylonitriles, which are known to be challenging substrates,¹⁷ could all be utilized for asymmetric cyclopropanation by [Co(**P9**)], furnishing the functionalized electrophilic cyclopropanes 3u-3z in high yields with excellent control of stereoselectivities (Table 2.3;

¹⁷ Doyle, M. P. Angew. Chem., Int. Ed. 2009, 48, 850–852.

entries 1–6). Similar to electron-deficient olefins, the catalytic cyclopropanation could also be applied to electron-rich olefins such as vinyl benzoate and vinyl propyl ether for highly **Table 2.3. Asymmetric Radical Cyclopropanation of Electron-Rich and Electron**-

Deficient Olefins with In Situ-Generated 2-Pyridyldiazomethanes by [Co(P9)]^a



^{*a*}Carried out with **1a** (0.10 mmol), **2** (0.15 mmol) and Cs_2CO_3 (0.20 mmol) at 80 °C for 16 h using [Co(**P9**)] (2 mol %) in toluene (1.0 mL); Ts = 4-toluenesulfonyl; Isolated yields; Diastereomeric ratio (dr) determined by ¹H NMR of crude reaction mixture; Enantiomeric excess (ee) of the major isomer determined by chiral HPLC. ^{*b*}With **2** (0.30 mmol). ^{*c*}With **2** (1.0 mmol).

stereoselective formation of cyclopropyl ester **3aa** and cyclopropyl ether **3ab** albeit in relatively lower yields (entries 7 and 8). Notably, the [Co(P9)]-based system proved to be similarly effective for the asymmetric cyclopropanation of aliphatic olefins, affording the

alkyl-substituted pyridylcyclopropanes **3ac** and **3ad** in moderate yields with moderate to excellent stereoselectivities (entries 9 and 10). Gratifyingly, internal olefins such as indene and benzofuran, which are typically challenging substrates for asymmetric cyclopropanation due to steric factors,^{10j} could also be cyclopropanated to form the fused cyclopropanes **3ae** and **3af** in moderate yields with high enantioselectivities despite varied diastereoselectivities (entries 11 and 12). It is worth mentioning that the Co(II)-based system was amenable to late-stage derivatization of biologically complex molecules as exemplified by the high-yielding formation of cyclopropane derivative of estrone **3ag** with excellent diastereoselectivity (Scheme 2.7).

Scheme 2.7. Late-Stage Derivatization



2.2.3. Asymmetric Radical Cyclopropanation of Alkenes with Different α-Heteroaryldiazomethanes

In addition to the representative 2-pyridyldiazomethane (1a'), it was demonstrated that metalloradical catalyst [Co(P9)] could effectively activate different types of α -heteroaryldiazomethanes for asymmetric cyclopropanation of alkenes (Table 2.4). For instance, 3-pyridyldiazomethane (1b') generated from the corresponding trishydrazone (2,4,6-triisopropylbenzenesulfonyl hydrazone) was found to be a competent radical precursor even at room temperature for the Co(II)-based asymmetric cyclopropanation. As shown with styrene (monosubstituted olefin), α -bromostyrene (1,1-disubstituted olefin), 1-

phenyl-1,3-butadiene (conjugated diene), and methyl acrylate (electron-deficient olefin) as representative substrates, [Co(**P9**)] could effectively activate in situ-generated **1b'** at room temperature for highly asymmetric cyclopropanation reactions, leading to productive formation of the corresponding 3-pyridylcyclopropanes **3ah**, **3ai**, **3aj** and **3ak** with exceptional control of stereoselectivities (Table 2.4; entries 1–4). Likewise, other heteroaryldiazomethanes, including those generated in situ from the trishydrazones derived from 5-bromo-3-pyridyl, 4-pyridyl, 3-thienyl, 3-indolyl and 3-quinolinyl carboxaldehydes, were all shown to be effective radical precursors for [Co(**P6**)]-catalyzed asymmetric olefin cyclopropanation as exemplified with the room temperature reactions of styrene as the model substrate, affording the corresponding heteroaryl cyclopropanes **3al–3ap** in moderate to high yields with excellent stereoselectivities (entries 5–9). The absolute configurations of the newly-generated stereogenic centers in **3al** and **3ao** were both established as (*R*,*R*) by X-ray crystallography.

In addition, 3-quinolinyldiazomethane (1g') was also investigated for Co(II)-based cyclopropanation reactions of selected alkenes ranging from α -chlorostyrene to electrondeficient olefins. Gratifyingly, almost all of these alkene substrates could be effectively cyclopropanated, allowing for the high-yielding formation of 3-quinolinylcyclopropanes **3aq–3at** with excellent stereoselectivities (entries 10–13). The only exception was observed for the reaction of methyl vinyl ketone, which afforded the corresponding cyclopropane **3ar** with excellent enantioselectivity but in lower yield with diminished diastereoselectivity (entry 11). The absolute configuration of the major enantiomer of **3aq** was determined to be (*S*,*S*) by X-ray crystallography.

Table 2.4. Asymmetric Olefin Cyclopropanation with Different In Situ-Generated α-

Heteroaryldiazomethanes by [Co(P9)]^a



^aCarried out with **1b–1g** (0.10 mmol), **2** (0.15 mmol) and Cs_2CO_3 (0.20 mmol) at 22 °C for 16 h using [Co(**P9**)] (2 mol %) in toluene (1.0 mL); Tris = 2,4,6-triisopropylbenzensulfonyl; Isolated yields; Diastereomeric ratio (dr) determined by ¹H NMR of crude reaction mixture; Enantiomeric excess (ee) of the major isomer determined by chiral HPLC. ^bWith **2** (0.30 mmol). ^cAbsolute configuration determined by X-ray crystallography. ^dWith Ts = 4-toluenesulfonyl at 60 °C.

2.2.4. Asymmetric Radical Cyclopropanation of Alkenes with Different α-Aryldiazomethanes

Considering that the [Co(P9)]-based catalytic system could productively utilize various α -heteroaryldiazomethanes containing heteroatom at different positions, we sought to explore the possibility of employing α -aryldiazomethanes for asymmetric cyclopropanation (Table 2.5), which has been largely limited to those with α -aryl groups containing H-bonding acceptors at the *ortho*-position (Scheme 2.8).^{8k,8n}

Scheme 2.8. Limitation to α-Aryldiazomethanes Containing Ortho-Methoxy Group





To our delight, it was found that [Co(P9)] could effectively activate α phenyldiazomethane (1h') derived from benzaldehyde trishydrazone (1h) as the radical precursor for asymmetric cyclopropanation of styrene (2a) under the standard conditions, affording the desired 1,2-diphenylcyclopropane (3au) in high yield with high diastereoselectivity and excellent enantioselectivity (Table 2.5; entry 1). Encouraged by this positive outcome, we then evaluated a wide array of α -aryldiazomethanes without Hbonding acceptors for asymmetric cyclopropanation by [Co(P9)].

Table 2.5. Scope of Co(II)-Catalyzed Asymmetric Radical Cyclopropanation of Styrene with α-Aryldiazomethanes^a



^aCarried out with **1h–1s** (0.10 mmol), **2a** (0.15 mmol) and Cs_2CO_3 (0.20 mmol) using [Co(**P9**)] (2 mol %) in toluene (1.0 mL) at 22 °C for 16 h; Isolated yields; Diastereomeric ratio (dr) determined by ¹H NMR of crude reaction mixture; Enantiomeric excess (ee) of the major (*E*)-isomer determined by chiral HPLC; Tris = 2,4,6-triisopropylbenzene sulfonyl.

In addition to non-substituted α -phenyldiazomethane, α -aryldiazomethanes bearing methyl substituent at different aryl positions, including *p*-Me (1i'), *m*-Me (1j'), and *o*-Me (1k'), could all be efficiently activated by [Co(P9)] for cyclopropanation of 2a, furnishing the corresponding arylcyclopropanes **3av**–**3ax** in similarly high yields with the same high level of stereoselectivities (Table 2.5; entries 2–4). Notably, the sterically-encumbered *o*-

ethylphenyldiazomethane (11') was found to be also suitable for the catalytic reaction, forming cyclopropane **3ay** with high stereoselectivities albeit in lower yield (entry 5). It was further shown that the [Co(P9)]-based system could use α -aryldiazomethanes containing substituents with varied electronic properties at different aryl positions, such as p-OMe (1m'), p-CF₃ (1n'), p-CO₂Me (1o'), and m-NO₂ (1p'), for the reaction, enabling high-yielding formation of the desired cyclopropanes 3az-3bc with excellent stereoselectivities (entries 6-9). Additionally, halogenated aryldiazomethanes were also suitable for the catalytic process as exemplified by the stereoselective synthesis of cyclopropane **3bd** with *o*-bromophenyldiazomethane (**1q**') (entry 10). Furthermore, the Co(II)-based catalytic system could be applicable to α -aryldiazomethanes bearing extended aromatic systems, including *p*- biphenyldiazomethane (1r') and 2-naphthyldiazo methane (1s'), delivering the corresponding cyclopropanes **3be** and **3bf** in high yields with excellent control of stereoselectivities (entries 11-12). Evidently, [Co(P9)] represents a powerful new catalyst that is generally applicable for asymmetric olefin cyclopropanation with both α -heteroaryldiazomethanes and α -aryldiazomethanes.

2.2.5. Mechanistic Studies on Co(II)-Catalyzed Radical Olefin Cyclopropanation with α-Heteroaryldiazomethanes

To gain insight into the proposed stepwise radical mechanism (Scheme 2.1), a combination of computational and experimental studies was conducted with collaboration of fellow labmates Jing Ke and Yiling Zhu. First, density functional theory (DFT) calculations were performed to elucidate the details of the catalytic pathway and associated energetics for the cyclopropanation reaction of styrene (**2a**) with 2-pyridyldiazomethane (**1a'**) by [Co(**P9**)] (Scheme 2.9; see Scheme S1 in Experimental Section for details). The geometry
optimizations were performed with the Gaussian 16¹⁸ at the BP86^{19,20}/lanl2dz^{16,21} level of theory in the gas phase at room temperature. To further improve the accuracy of energies, single point energies were carried out at the B3LYP²²/def2-tzvp^{16,21} level of theory along with Grimme's dispersion correction²³ (D3BJ) and SMD²⁴ solvation model (in toluene).

Scheme 2.9. DFT Calculations on Energetics of [Co(P6)]-Catalyzed Cyclopropanation^a



^aApplied BP86/LANL2DZ for geometry optimization and B3LYP/def2-tzvp for calculations of single point energies (kcal/mol) along with Grimme's dispersion correction and SMD (toluene) solvation model.

The DFT calculations reveal the initial formation of intermediate **B** between the catalyst and 2-pyridyldiazomethane through multiple noncovalent attractive interactions, including

¹⁸ Frisch, M. J. et al., Gaussian 16, Revision A.03, Gaussian, Inc., Wallingford CT, 2016.

¹⁹ Schultz, N. E.; Zhao, Yan; Truhlar, D. G. J. Phys. Chem. A 2005, 109, 11127–11143.

²⁰ Suarez, A. I. O.; Jiang, H.-L.; Zhang, X. P.; de Bruin, B. *Dalton Trans.* **2011**, *40*, 5697–5705.

²¹ Eichkorn, K.; Weigend, F.; Treutler, O.; Ahlrichs, R. *Theor. Chem. Acc.* **1997**, *97*, 119–124.

²² Furche, F.; Perdew, J. P. J. Chem. Phys. 2006, 124, 044103.

²³ Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. J. Chem. Phys. 2010, 132, 154104.

²⁴ Marenich, A. V.; Cramer, C. J.; Truhlar, D. G. J. Phys. Chem. B 2009, 113, 6378–6396.

H-bonding and π -stacking interactions (Figure 2.3; see Scheme S2 in Experimental Section). The noncovalent complexation, which is exergonic by 8.7 kcal/mol, positions the α -carbon atom of diazo **1a'** in close proximity to the Co(II)-metalloradical center of [Co(**P9**)] (C---Co: ~2.91 Å) for further interactions. The ensuing metalloradical activation, which is slightly exergonic by 1.2 kcal/mol, is found to be associated with a relatively high but accessible activation barrier (**TS1**: $\Delta G^{\ddagger} = 19.7$ kcal/mol), affording α -Co(III)-pyridyl radical intermediate **C** with the release of dinitrogen as the byproduct. The subsequent radical addition of the resulting radical intermediate **C** to alkene **2a**, which is highly exergonic by 22.4 kcal/mol, proceeds through an exceedingly low activation barrier (**TS2**: $\Delta G^{\ddagger} = 2.3$ kcal/mol), delivering γ -Co(III)-alkyl radical intermediate **D**.





As illustrated in the DFT-optimized structure of **TS2** (Figure 2.3; see Scheme S2 in Experimental Section), there exists a network of noncovalent attractive interactions, such as multiple H-bonding and π -stacking interactions, between the substrates and the catalyst that synergistically lower the activation barrier of the transition state. According to the DFT calculations, the final step of 3-*exo-tet* cyclization of γ -Co(III)-alkyl radical intermediate **D**, which is exergonic by 12.6 kcal/mol, is a nearly barrierless process, leading to the formation of cyclopropane product **3a** while regenerating the metalloradical catalyst [Co(**P6**)].

In an effort to directly trap α -Co(III)-heterobenzyl radical intermediate **I**, the metalloradical activation of 3-pyridyldiazomethane (**1b**') by [Co(**P14**)] (**P14** = 3,5-Di'Bu-IbuPhyrin)] ²⁵ was carried out in the presence of TEMPO (2,2,6,6-tetramethyl-1-piperidinyloxy) without alkene substrates, resulting in the isolation of bis-TEMPO-trapped product **5** in 19% yield (Scheme 2.10). The observation of compound **5** evidently implies the initial formation of α -Co(III)-pyridyl radical **I**_{1b}, which was presumably captured by TEMPO through radical recombination to generate Co(III)-alkyl intermediate **III**_{1b}. Subsequent radical substitution reaction of intermediate **III**_{1b} with a second molecule of TEMPO was likely responsible for the final formation of **5**. Similarly, bis-TEMPO-trapped product **6** was isolated in 12% yield from the metalloradical activation of 3-quinolinyldiazomethane (**1g**') by [Co(**P14**)] in the presence of TEMPO, indicating the existence of α -Co(III)-heterobenzyl radical intermediate **II**_{1g} and the following Co(III)-alkyl intermediate **III**_{1g} (Scheme 2.10).

²⁵ Ruppel, J. V.; Jones, J. E.; Huff, C. A.; Kamble, R. M.; Chen, Y.; Zhang, X. P. *Org. Lett.* **2008**, *10*, 1995–1998.



Scheme 2.10. Trapping of α-Co(III)-Heterobenzyl Radicals by TEMPO

To probe the involvement of γ -Co(III)-alkyl radical intermediate **II** in the proposed mechanism (Scheme 2.1), both isotopomers of β -deuterostyrene (*E*)-**2a**_D and (*Z*)-**2a**_D were employed as substrates for Co(II)-catalyzed cyclopropanation with 2-pyridyl tosylhydrazone (**1a**) (Schemes 2.11 and 2.12). Unlike a concerted mechanism that results in stereospecific cyclopropane products, a stepwise radical mechanism may give rise to the formation of four possible diastereomers of cyclopropanes due to the potential rotation of the β -C–C bond in γ -Co(III)-alkyl radical intermediate **II** before cyclization. As expected, both reactions of (*E*)-**2a**_D and (*Z*)-**2a**_D with **1a** afforded the cyclopropane products as a mixture of four different diastereomers: (*E*,*E*)-**3a**_D, (*Z*,*Z*)-**3a**_D, and (*E*,*Z*)-**3a** (Schemes 2.11 and 2.12). Among them, (*E*,*Z*)-**3a** and (*Z*,*Z*)-**3a**_D are formed due to β -C–C bond rotation in γ -Co(III)-alkyl radical intermediate **II** for cyclopropanation reaction of

Scheme 2.11 Probing of γ -Co(III)-Alkyl Radicals by Cyclopropanation of (E)- β -





Scheme 2.12. Probing of γ -Co(III)-Alkyl Radicals by Cyclopropanation of (Z)- β -

Deuterostyrene



(*E*)-2**a**_D. The isotopic ratio between the diastereomers of (*E*,*Z*)-3**a** and (*Z*,*Z*)-3**a**_D and that of (*E*,*E*)-3**a**_D and (*Z*,*E*)-3**a**_D could be determined by ¹H NMR analysis. Similarly, diastereomers of (*E*,*E*)-3**a**_D and (*Z*,*E*)-3**a**_D are formed for reaction of (*Z*)-2**a**_D, and the isotopic ratio between diastereomers of (*E*,*E*)-3**a**_D and (*Z*,*E*)-3**a**_D and that of (*E*,*Z*)-3**a** and (*Z*,*Z*)-3**a**_D could also be analyzed.

When the bridged [Co(P9)] (P9 = 2,6-DiPhO-Hu(C₆)Phyrin)] was used as the catalyst, the isotopic ratio between the diastereomers of (E,Z)-**3a** and (Z,Z)-**3a**_D and that of (E,E)-**3a**_D and (Z,E)-**3a**_D was determined to be 6:94 for the cyclopropanation reaction of (E)-**2a**_D. Accordingly, the isotopic ratio between diastereomers of (E,E)-**3a**_D and (Z,E)-**3a**_D and that of (E,Z)-**3a** and (Z,Z)-**3a**_D was determined to be 7:93 for the reaction of (Z)-**2a**_D (Scheme 2.13). The observation of both (E)- and (Z)- isotopomers of (E)-**3a** in both reactions evidently suggested the rotation of β -C–C bond in the corresponding γ -Co(III)-alkyl radical intermediates $\Pi_{1a/(E)-2aD}$ and $\Pi_{1a/(Z)-2aD}$.

Scheme 2.13. ¹H Spectrum for Cyclopropane Isotopomers 3a_D from [Co(P9)]-Catalyzed Cyclopropanation





When the non-bridged [Co(P3)] (P3 = 3,5-Di'Bu-ChenPhyrin)] was employed as the catalyst, the isotopic ratios changed for both reactions of (*E*)-2a_D (from 6:94 to 13:87) and (*Z*)-2a_D (from 7:93 to 13:87), indicating a higher degree of the β -C–C bond rotation in the less-hindered catalyst environment (Scheme 2.14).

Scheme 2.14. ¹H NMR Spectrum for Cyclopropane Isotopomers 3a_D from [Co(P3)]-Catalyzed Cyclopropanation





Accordingly, the use of even less-hindered catalyst [Co(P14)] (P14 = 3,5-Di'Bu-IbuPhyrin)] allowed a further increase in the degree of the β -C–C bond rotation, changing the isotopic ratios to 16:84 and 16:84, respectively, for the two reactions (Scheme 2.15). Collectively, these experimental results, together with the DFT calculations, provided corroborating evidence for the proposed stepwise radical mechanism of the Co(II)-catalyzed asymmetric cyclopropanation with heteroaryldiazomethanes.

Scheme 2.15. ¹H NMR Spectrum for Cyclopropane Isotopomers 3a_D from [Co(P14)]-

Catalyzed Cyclopropanation



2.3. CONCLUSIONS AND OUTLOOK

In summary, we have applied Co(II)-based metalloradical catalysis (MRC) for the successful development of asymmetric radical cyclopropanation of alkenes with heteroaryldiazomethanes. With the newly-synthesized bridged D_2 -symmetric chiral

amidoporphyrin 2,6-DiPhO-Hu(C₆)Phyrin as the optimal supporting ligand, the Co(II)can based metalloradical system effectively activate different types of heteroaryldiazomethanes even at room temperature for olefin cyclopropanation, offering a general approach for stereoselective synthesis of chiral heteroaryl cyclopropanes. In addition to styrene derivatives, the Co(II)-catalyzed cyclopropanation is highlighted by an extraordinarily broad scope of alkenes, including several types of challenging substrates, affording a diverse range of heteroaryl cyclopropanes in high yields with both excellent diastereoselectivities and enantioselectivities. Furthermore, both computational and experimental studies have provided several lines of evidence in elucidating the underlying stepwise radical mechanism of the Co(II)-based olefin cyclopropanation involving α - and γ -metalloalkyl radicals as the key intermediates. In view of the ubiquity of the resulting enantioenriched heteroaryl cyclopropanes in biologically important compounds, we hope this Co(II)-catalyzed asymmetric radical cyclopropanation process will find wide applications in organic synthesis related to drug discovery.

2.4. EXPERIMENTAL SECTION

2.4.1. General Considerations

All cyclopropanation reactions were performed in anhydrous solvents under N₂ atmosphere in an oven-dried glassware following standard Schlenk techniques. Gas tight syringes were used to transfer liquid reagents and solvents in catalytic reactions. Solvent was freshly distilled/degassed prior to use unless otherwise noted. Thin layer chromatography was performed on Merck TLC plates (silica gel 60 F254). Flash column chromatography was performed with ICN silica gel (60 Å, 230-400 mesh, 32-63 µm). ¹H NMR spectra were acquired using Varian INOVA 400 (400 MHz), Bruker 500 (500 MHz), or Varian INOVA 600 (600 MHz) spectrometer. Chemical shifts were internally referenced to residual solvent peak (CHCl₃ δ = 7.26 ppm). Data were reported as follows: chemical shift (ppm), integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, p = pentet, hept = heptet, br = broad, m = multiplet), and coupling constants J (Hz). ¹³C NMR spectra were acquired using Varian INOVA 400 (100 MHz), Bruker 500 (125 MHz), or INOVA 600 (150 MHz) spectrometer with complete proton decoupling. Chemical shifts were reported in ppm with residual solvent peak (CDCl₃ δ = 77.16 ppm or (DMSO-*d*₆ δ = 39.52 ppm) as the internal standard. In ¹³C NMR analysis, peaks for carbon atoms adjacent to a boron center were generally not observed owing to quadrupolar broadening. ¹¹B spectra were externally referenced to a standard of BF₃·Et₂O ($\delta = 0.0$ ppm). ¹⁹F NMR spectrum was acquired using Varian Bruker 500 (470 MHz) or INOVA 600 (564 MHz) spectrometer. Infrared spectra were measured with a Nicolet Avatar 320 spectrometer with a Smart Miracle accessory. Optical rotations were measured on a Rudolph Research Analytical AUTOPOL® IV digital polarimeter. HPLC measurements were carried out on a Shimadzu HPLC system with Chiralcel OD-H, OJ-H, AD-H, AS-H, IA, IB, IC, ID, IE and IF columns. High-resolution mass spectrometry (DART and ESI) was performed at the Mass Spectrometry Facility, Boston College, Chestnut Hill, MA. The X-ray diffraction data were collected using Bruker-AXS SMART-APEXII CCD diffractometer. All reagents were purchased either from Aldrich, Alfa Aesar, Acros, Ak Sci, Oakwood Chemicals, Strem Chemicals or TCI and were used without further purification.

2.4.2. Synthesis and Characterization of Catalysts

2.4.2.1. Experimental Procedure for Preparation of 2,6-DiPhO-Tao('Bu)Phyrin



2,6-DiPhO-bromosynthon

2,6-DiPhO-Tao(^tBu)Phyrin

2,6-DiPhO-Tao('Bu)Phyrin was synthesized according to our previous reported procedure.^{8a} 2,6-DiPhO-bromosynthon^{8m} (300 mg, 0.231 mmol, 1.0 equiv), the chiral amide (1R,2R)-*tert*-butyl-2-carbamoylcyclopropane-1-carboxylate¹⁵ (685 mg, 3.70 mmol, 16.0 equiv), Pd(OAc)₂ (20.74 mg, 0.0924 mmol, 0.4 equiv), Xantphos (107 mg, 0.184 mmol, 0.8 equiv) and Cs₂CO₃ (1.20 g, 3.70 mmol, 16.0 equiv) were placed in an ovendried Schlenk tube. The tube was capped with a Teflon screw cap, evacuated, and backfilled with nitrogen. Under nitrogen atmosphere, the screw cap was replaced with a rubber septum. Dioxane (0.02 M) was added via a gas tight syringe. The tube was then purged with nitrogen for 1 min and sealed with Teflon screw cap. After stirring at 100 °C for 72 h, the resulting mixture was cooled down to room temperature, diluted with EtOAc,

filtered through a short pad of celite and concentrated under vacuum. The crude mixture was further purified by flash column chromatography (eluent: 4:1 to 2:1 Hexanes/EtOAc) to afford the title compound (238 mg, 60% yield). ¹H NMR (500 MHz, CDCl₃) δ 9.14 (d, J = 4.8 Hz, 4H), 8.80 (d, J = 4.8 Hz, 4H), 8.47 (br, 4H), 7.86 (t, J = 8.4 Hz, 2H), 7.65 (t, J = 8.5 Hz, 2H), 7.07 – 6.98 (m, 12H), 6.82 (t, J = 7.4 Hz, 4H), 6.78 – 6.74 (m, 8H), 6.64 (br, 4H), 1.79 (ddd, J = 9.1, 5.8, 3.7 Hz, 4H), 0.95 (s, 36H), 0.85 – 0.79 (m, 4H), 0.47 (br, 4H), 0.24 (br, 4H), -2.53 (s, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 170.35, 169.10, 159.20, 156.28, 139.15, 132.58, 130.73, 130.37, 129.53, 123.61, 123.38, 121.99, 119.45, 118.06, 112.73, 112.16, 107.12, 80.55, 77.36, 27.75, 23.97, 22.94, 14.83. UV-Vis (CHCl₃), λ_{max} nm (log ϵ): 423 (5.54), 515 (4.31), 545 (3.64), 589 (3.81), 643 (3.26). HRMS (ESI) ([M+H]⁺) Calcd. For C₁₀₄H₉₉N₈O₁₆⁺: 1715.71736, Found: 1715.71326.

2.4.2.2. Experimental Procedure for Preparation of 2,6-DiPhO-Tao(But-3-en-1yl)Phyrin



2,6-DiPhO-Tao(^tBu)Phyrin

2,6-DiPhO-Tao(But-3-en-1-yl)Phyrin

2,6-DiPhO-Tao(But-3-en-1-yl)Phyrin was synthesized according to our previous reported procedure.¹⁵ TFA (1.50 mL, 19.5 mmol, 100.0 equiv) was added to a solution of **2,6-DiPhO-Tao('Bu)Phyrin** (335 mg, 0.195 mmol, 1.0 equiv) in DCM (0.05 M), and the reaction was stirred at room temperature overnight prior to evaporation of all the volatiles.

The crude mixture was transferred to an oven-dried Schlenk tube followed by the addition of anhydrous K₂CO₃ (1.35 g, 9.76 mmol, 50.0 equiv). The tube was capped with a Teflon screw cap, evacuated, and backfilled with nitrogen. Under nitrogen atmosphere, but-3-enyl 4-methylbenzenesulfonate (883 mg, 3.90 mmol, 20.0 equiv) was added, followed by the addition of DMF (0.05 M) via a gas tight syringe. The reaction mixture was heated at 100 °C for 12 h. After cooling down to room temperature, the reaction mixture was diluted with EtOAc and water. The organic layer was separated and concentrated under vacuum. Further purification by flash column chromatography (eluent: 3:1 to 2:1 Hexanes/EtOAc) to afford the title compound (206 mg, 62% yield). ¹H NMR (600 MHz, CDCl₃) δ 9.18 (d, J = 4.7 Hz, 4H), 8.82 (d, J = 4.8 Hz, 4H), 8.51 (br, 4H), 7.88 (t, J = 8.4 Hz, 2H), 7.67 (t, J = 8.5 Hz, 2H), 7.08 - 7.01 (m, 12H), 6.84 (t, J = 7.3 Hz, 4H), 6.82 - 6.76 (m, 8H), 6.66(br, 4H), 5.34 (dd, J = 17.2, 9.1 Hz, 4H), 4.72 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 3.60 - 3.44 (m, 8H), 1.94 - 4.67 (m, 8H), 3.60 - 3.44 (m, 8H), 3.60 -1.82 (m, 12H), 0.95 – 0.89 (m, 4H), 0.50 (br, 4H), 0.34 (br, 4H), -2.52 (s, 2H). ¹³C NMR (150 MHz, CDCl₃) δ 171.14, 168.65, 159.22, 156.17, 139.05, 133.65, 130.82, 130.51, 129.58, 123.71, 122.99, 121.76, 119.56, 117.86, 116.99, 112.77, 112.02, 107.01, 63.43, 32.64, 24.35, 21.82, 14.83. UV-Vis (CHCl₃), λ_{max} nm (log ε): 423 (5.67), 515 (4.44), 543 (3.81), 587 (3.96), 642 (3.45). HRMS (ESI) ([M+H]⁺) Calcd. For C₁₀₄H₉₁N₈O₁₆⁺: 1707.65476, Found: 1707.65015.

2.4.2.3. Experimental Procedure for Preparation of [H₂(P9)]



2,6-DiPhO-Tao(But-3-en-1-yl)Phyrin

[H₂(**P9**)]

2,6-DiPhO-Hu(C₆)Phyrin [H₂(**P9**)] was synthesized according to our previous reported procedure.¹⁵ Grubbs 2nd generation catalyst (10.3 mg, 12.2 µmol, 0.1 equiv) was added to a solution of 2,6-DiPhO-Tao(But-3-en-1-yl)Phyrin (206 mg, 0.122 mmol, 1.0 equiv) in DCM (0.001 M), and the reaction mixture was stirred at 40 °C for 12 h. After the reaction finished, the resulting mixture was concentrated under vacuum and purified by flash column chromatography (1:1:1 to 1:2:1 Hexanes/EtOAc/DCM) to afford the mixture of trans and cis isomers, which was in turn dissolved in EtOAc-toluene (V/V=2/1, 0.02 M) in the presence of 10% Pd/C (1 mg per mg of porphyrin). Hydrogen gas was bubbled through the reaction mixture until the reaction was completed based on the crude ¹H NMR (typically for 15 min). The reaction mixture was filtered through a short pad of celite and concentrated under vacuum. The resulting crude was purified by flash column chromatography (eluent: 1:2:1 Hexanes/EtOAc/DCM) to afford the title compound (145 mg, 71% yield). ¹H NMR (600 MHz, CDCl₃) δ 9.22 (d, J = 4.7 Hz, 4H), 8.90 (d, J = 4.7 Hz, 4H), 8.38 (d, J = 8.4 Hz, 4H), 7.90 (t, J = 8.5 Hz, 2H), 7.66 (t, J = 8.6 Hz, 2H), 7.14 – 7.07 (m, 8H), 7.01 (d, J = 8.6 Hz, 4H), 6.91 (t, J = 7.4 Hz, 4H), 6.87 – 6.81 (m, 8H), 6.62

(br, 4H), 3.56 - 3.52 (m, 4H), 3.34 - 3.30 (m, 4H), 1.92 (ddd, J = 9.3, 5.9, 3.8 Hz, 4H), 0.95 - 0.91 (m, 4H), 0.83 - 0.74 (m, 8H), 0.62 - 0.54 (m, 12H), 0.27 - 0.22 (m, 4H), -2.58(s, 2H). ¹³C NMR (150 MHz, CDCl₃) δ 170.70, 168.53, 159.33, 156.10, 138.75, 130.88, 130.45, 129.64, 123.90, 122.85, 122.59, 119.73, 119.48, 112.98, 111.43, 107.59, 64.16, 27.28, 24.49, 23.75, 22.31, 14.68. UV-Vis (CHCl₃), λ_{max} nm (log ε): 423 (5.51), 515 (4.26), 547 (3.64), 588 (3.78), 641 (3.26). HRMS (ESI) ([M+H]⁺) Calcd. For C₁₀₀H₈₇N₈O₁₆⁺: 1655.6235, Found: 1655.61731.



2.4.2.3. Experimental Procedure for Preparation of [Co(P9)]

[Co(P9)] was synthesized according to our previous reported procedure.^{8a} Free base porphyrin $[H_2(P9)]$ (145 mg) and anhydrous CoCl₂ (10.0 equiv) were placed in an ovendried Schlenk tube. The tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen. Under nitrogen atmosphere, 2,6-Lutidine (20.0 equiv) and anhydrous dioxane (0.02 M) were added. The tube was purged with nitrogen for 1 min and sealed with Teflon screw cap. The reaction mixture was stirred at 100 °C for 48 h prior to being cooled down to room temperature. The reaction mixture was diluted with EtOAc, washed with water 2 times and then concentrated under vacuum. The compound [Co(P9)] was isolated as a purple solid (118 mg, 79% yield) after purification by flash column chromatography (eluent: 1:2:1 Hexanes/EtOAc/DCM). UV-Vis (CHCl₃), λ_{max} nm (log ε): 415 (5.19), 528 (4.01). HRMS (ESI) ([M]⁺) Calcd. For C₁₀₀H₈₄CoN₈O₁₆⁺: 1711.53373, Found: 1711.52722.

2.4.3. Synthesis and Characterization of N-Sulfonyl Hydrazones

2.4.3.1. Experimental Procedure for Preparation of N-Tosylhydrazones

To a stirred solution of tosylhydrazide (1.0 mmol) in methanol (10.0 mL) at room temperature, aldehyde (1.0 equiv) was added dropwise (or portionwise if solid).^{8k} After the reaction was stirred overnight, the solvent was removed directly under reduced pressure, and the crude mixture was further purified by trituration.

2.4.3.2. Experimental Procedure for Preparation of *N*-2,4,6-Triisopropylbenzenesulfonyl Hydrazones

To a stirred solution of 2,4,6-triisopropylbenzenesulfonyl hydrazide (1.0 mmol) in THF (10.0 mL) at room temperature, aldehyde (1.0 equiv) was added dropwise (or portionwise if solid).^{8k} After the reaction was stirred overnight, the solvent was removed directly under reduced pressure, and the crude mixture was further purified by trituration.

2-Pyridinecarboxaldehyde tosylhydrazone (1a) Yield: 78%. $R_f = 0.43$ (1:3)

O=S=O N H H 1a Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 8.56 (d, *J* = 5.5 Hz, 1H), 7.91 (d, *J* = 7.9 Hz, 1H), 7.88 – 7.84 (m, 3H), 7.71 (td, *J* = 7.7, 1.7 Hz, 1H), 7.30 (d, *J* = 8.2 Hz, 2H), 7.28 – 7.25 (m, 2H), 2.40 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 152.52, 149.37, 147.17, 144.54, 136.85, 135.47, 129.90, 128.07, 124.57, 120.83, 21.74. IR (neat, cm⁻

¹): 3189.65, 1586.65, 1494.53, 1360.06, 1166.01, 1067.70, 941.03, 671.04. HRMS (DART) ([M+H]⁺) Calcd. for C₁₃H₁₄N₃O₂S⁺: 276.08012, Found: 276.08104.

3-Pyridinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1b) Yield:





CDCl₃) δ 153.85, 151.56, 150.91, 149.01, 142.63, 133.71, 131.25, 129.73, 124.10, 123.77, 34.33, 30.20, 25.01, 23.65. IR (neat, cm⁻¹): 3172.82, 2958.71, 1600.00, 1462.37, 1318.69, 1165.36, 944.50, 673.65. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₃₀N₃O₂S⁺: 388.20532, Found: 388.20490.





(1c) Yield: 68%. $R_f = 0.32$ (2:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.62 (d, J = 2.2 Hz, 1H), 8.56 (d, J = 1.8 Hz, 1H), 8.14 (s, 1H), 8.11 – 8.10 (m, 1H), 7.71 (s, 1H), 7.20 (s, 2H), 4.23 (hept, J = 6.8 Hz, 2H), 2.93 – 2.89 (m, 1H), 1.32 (d, J = 6.7 Hz, 12H), 1.26 (d, J = 6.8 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 154.13,

151.67, 151.46, 146.55, 140.41, 136.26, 131.36, 130.99, 124.21, 121.34, 34.36, 30.30, 25.02, 23.65. IR (neat, cm⁻¹): 3168.68, 2952.85, 1602.02, 1460.03, 1321.22, 1166.46, 1054.56, 964.21. 868.25. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₂₉N₃O₂SBr⁺: 466.11584, Found: 466.11543.



Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 8.63 (d, *J* = 6.1 Hz, 2H), 8.08 (br, 1H), 7.87 (d, *J* = 8.4 Hz, 2H), 7.68 (s, 1H), 7.44 (d, *J* = 6.1 Hz, 2H), 7.34 (d, *J* = 7.9 Hz, 2H), 2.43 (s, 3H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 150.33, 144.43, 143.91, 140.92, 136.00, 129.92,

127.33, 120.80, 21.12. IR (neat, cm⁻¹): 3032.92, 1595.62, 1484.90, 1372.66, 1168.01, 1075.72, 941.71, 816.12. HRMS (DART) ([M+H]⁺) Calcd. for C₁₃H₁₄N₃O₂S⁺: 276.08012, Found: 276.07983.

4-Pyridinecarboxaldehyde tosylhydrazone (1d) Yield: 72%. $R_f = 0.23$ (1:3)

3-Thiophenecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1e) Yield:



83%. R_f = 0.41 (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 7.82 – 7.81 (m, 2H), 7.42 (s, 1H), 7.35 (d, *J* = 5.1 Hz, 1H), 7.26 (d, *J* = 5.3 Hz, 1H), 7.18 (s, 2H), 4.26 (hept, *J* = 6.8 Hz, 2H), 2.90 (hept, *J* = 6.9 Hz, 1H), 1.30 (d, *J* = 7.9 Hz, 12H), 1.25 (d, *J* = 8.1 Hz, 6H). ¹³C NMR (150 MHz, CDCl₃) δ 153.64, 151.52, 141.98, 136.67,

131.35, 127.35, 126.74, 125.30, 124.04, 34.33, 30.19, 25.02, 23.67. IR (neat, cm⁻¹): 3172.18, 2951.18, 1600.04, 1456.70, 1315.85, 1165.34, 1035.93, 788.04. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₉N₂O₂S₂⁺: 393.16650, Found: 393.16684.





Yield: 76%. $R_f = 0.45$ (2:1 Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 8.08 (d, J = 8.4 Hz, 1H), 8.02 – 7.98 (m, 2H), 7.95 (s, 1H), 7.72 (s, 1H), 7.33 (ddd, J = 8.5, 7.2, 1.3 Hz, 1H), 7.22 – 7.17 (m, 3H), 4.32 (hept, J = 6.7 Hz, 2H), 2.89 (hept, J = 6.9 Hz, 1H), 1.66 (s, 9H), 1.33 (d, J = 6.7 Hz, 12H), 1.24 (d, J = 6.9 Hz, 6H). ¹³C NMR

(125 MHz, CDCl₃) δ 153.64, 151.45, 149.25, 141.59, 135.99, 131.30, 128.78, 126.89, 125.61, 124.00, 123.62, 122.95, 116.12, 115.03, 84.70, 34.35, 30.20, 28.27, 25.10, 23.67. IR (neat, cm⁻¹): 3210.74, 2961.08, 1737.56, 1370.03, 1150.43, 1095.34, 909.03, 694.54. HRMS (DART) ([M+H]⁺) Calcd. for C₂₉H₄₀N₃O₄S⁺: 526.27340, Found: 526.27256.

3-Quinolinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1g) Yield:



71%. $R_f = 0.52$ (1:1 Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 9.15 (d, J = 2.1 Hz, 1H), 8.21 (s, 1H), 8.15 (br, 1H), 8.09 (d, J =8.4 Hz, 1H), 7.93 (s, 1H), 7.79 (d, J = 7.8 Hz, 1H), 7.75 – 7.71 (m, 1H), 7.60 – 7.52 (m, 1H), 7.21 (s, 2H), 4.30 (hept, J = 6.8 Hz, 2H), 2.90 (hept, J = 7.0 Hz, 1H), 1.34 (d, J = 6.8 Hz, 12H), 1.25 (d, J =

6.8 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 153.87, 151.62, 148.51, 148.31, 143.25, 135.05, 131.26, 130.70, 129.30, 128.34, 127.57, 127.55, 126.86, 124.13, 34.33, 30.26, 25.07, 23.64. IR (neat, cm⁻¹): 3189.20, 2959.43, 1620.26, 1462.31, 1314.83, 1165.48, 940.50, 882.11. HRMS (DART) ([M+H]⁺) Calcd. for C₂₅H₃₂N₃O₂S⁺: 438.22097, Found: 438.22048.

Benzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1h) Yield: 79%. $R_f = 0.42$



(3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.03 (br, 1H),
7.76 (s, 1H), 7.57 – 7.53 (m, 2H), 7.36 – 7.30 (m, 3H), 7.18 (s, 2H),
4.29 (hept, J = 6.8 Hz, 2H), 2.90 (hept, J = 7.0 Hz, 1H), 1.32 (d, J =
6.8 Hz, 12H), 1.25 (d, J = 7.0 Hz, 6H). ¹³C NMR (150 MHz, CDCl₃)
δ 153.64, 151.54, 146.47, 133.49, 131.39, 130.39, 128.73, 127.43,

124.04, 34.33, 30.22, 25.02, 23.67. IR (neat, cm⁻¹): 3184.35, 2961.81, 1601.24, 1442.75, 1375.14, 1167.70, 1036.03, 953.32.

4-Methylbenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1i) Yield: 63%.



 $R_f = 0.40$ (3:1 Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 8.03 (br, 1H), 7.74 (s, 1H), 7.47 – 7.41 (m, 2H), 7.18 (s, 2H), 7.14 (d, J = 8.0 Hz, 2H), 4.29 (hept, J = 6.7 Hz, 2H), 2.90 (hept, J = 6.9 Hz, 1H), 2.34 (s, 3H), 1.31 (d, J = 6.8 Hz, 12H), 1.25 (d, J = 6.9 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 153.54, 151.51, 146.76, 140.70, 131.47,

130.80, 129.43, 127.39, 123.99, 34.30, 30.19, 25.00, 23.65, 21.56. IR (neat, cm⁻¹): 3180.13, 2958.55, 1601.72, 1451.06, 1376.12, 1322.84, 1166.29, 1036.59, 965.47. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₃₃N₂O₂S⁺: 401.22573, Found: 401.22592.

3-Methylbenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1j) Yield: 76%.



 $R_f = 0.50$ (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 7.82 (s, 1H), 7.71 (s, 1H), 7.43 (s, 1H), 7.31 (d, J = 7.5 Hz, 1H), 7.23 – 7.21 (m, 1H), 7.18 – 7.15 (m, 3H), 4.28 (hept, J = 6.8 Hz, 2H), 2.89 (hept, J = 7.0 Hz, 1H), 2.32 (s, 3H), 1.31 (d, J = 6.8 Hz, 12H), 1.25 (d, J = 6.9 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 153.62, 151.56,

146.68, 138.50, 133.38, 131.46, 131.28, 128.63, 127.74, 124.95, 124.04, 34.34, 30.24, 25.03, 23.67, 21.34. IR (neat, cm⁻¹): 3174.70, 2956.20, 1599.97, 1432.69, 1301.83, 1152.29, 1035.53, 900.84. HRMS (DART) ($[M+H]^+$) Calcd. for C₂₃H₃₃N₂O₂S⁺: 401.22573, Found: 401.22521.



2-Methylbenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1k) Yield: 79%.

NMR (150 MHz, CDCl₃) δ 153.63, 151.50, 145.40, 137.01, 131.53, 131.38, 130.96, 130.08, 127.28, 126.19, 124.00, 34.34, 30.21, 25.01, 23.68, 19.78. IR (neat, cm⁻¹): 3165.07, 2957.59, 1598.55, 1422.18, 1294.00, 1151.53, 1034.42, 927.28. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₃₃N₂O₂S⁺: 401.22573, Found: 401.22674.

2-Ethylbenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (11) Yield: 71%. R_f



= 0.58 (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.02 (s, 1H), 7.98 (s, 1H), 7.69 (d, J = 7.8 Hz, 1H), 7.27 – 7.21 (m, 1H), 7.18 -7.09 (m, 4H), 4.25 (hept, J = 7.0 Hz, 2H), 2.88 (hept, J = 6.9 Hz, 1H), 2.68 (q, J = 7.5 Hz, 2H), 1.28 (d, J = 6.7 Hz, 12H), 1.23 (d, J =8.5 Hz, 6H), 1.09 (t, J = 8.4 Hz, 3H). ¹³C NMR (150 MHz, CDCl₃) δ 153.63, 151.47, 145.06, 143.20, 131.38, 130.76, 130.36, 129.35, 127.26, 126.24, 124.01, 34.34, 30.19, 26.08, 24.99, 23.67, 15.98. IR (neat, cm⁻¹): 3208.30, 2960.31, 1599.28, 1426.46, 1320.70, 1152.03, 1036.88, 943.17. HRMS (DART) ([M+H]⁺) Calcd. for C₂₄H₃₅N₂O₂S⁺:

415.24138, Found: 415.24276.

4-Methoxybenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1m) Yield:



83%. $R_f = 0.35$ (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 7.89 (br, 1H), 7.72 (s, 1H), 7.52 – 7.47 (m, 2H), 7.18 (s, 2H), 6.88 – 6.81 (m, 2H), 4.28 (hept, J = 6.8 Hz, 2H), 3.80 (s, 3H), 2.90 (hept, J = 6.9 Hz, 1H), 1.31 (d, J = 6.7 Hz, 12H), 1.25 (d, J = 6.9 Hz, 6H). ¹³C NMR (150 MHz, CDCl₃) δ 161.46, 153.52, 151.50, 146.76,

131.51, 129.01, 126.25, 124.00, 114.19, 55.49, 34.32, 30.20, 25.03, 23.67. IR (neat, cm⁻¹): 3198.45, 2959.56, 1605.94, 1514.36, 1300.45, 1253.45, 1165.49, 1035.04, 944.05. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₃₃N₂O₃S⁺: 417.22064, Found: 417.22008.





Yield: 89%. $R_f = 0.31$ (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.13 (br, 1H), 7.78 (s, 1H), 7.65 (d, J = 8.1 Hz, 2H), 7.59 (d, J = 8.2 Hz, 2H), 7.19 (s, 2H), 4.26 (hept, J = 6.9 Hz, 2H), 2.90 (hept, J = 6.9 Hz, 1H), 1.31 (d, J = 6.7 Hz, 12H), 1.25 (d, J = 7.0 Hz, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 153.95, 151.60, 144.23,

136.86, 131.89 (q, J = 34.0 Hz), 131.16, 127.50, 126.12 (q, J = 273.4 Hz), 125.73 (q, J = 3.8 Hz), 124.14, 34.35, 30.26, 25.01, 23.66. ¹⁹F NMR (564 MHz, CDCl₃) δ –62.91. IR (neat, cm⁻¹): 3450.22, 3176.27, 2960.65, 1457.91, 1324.32, 1275.41, 1163.74, 1128.97, 1066.13. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₃₀N₂O₂F₃S⁺: 455.19746, Found: 455.19733.

4-(Methoxycarbonyl)benzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (10)



Yield: 73%. $R_f = 0.52$ (2:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.14 (s, 1H), 8.00 (d, J = 8.2 Hz, 2H), 7.77 (s, 1H), 7.61 (d, J = 8.1 Hz, 2H), 7.19 (s, 2H), 4.26 (hept, J = 6.8 Hz, 2H), 3.91 (s, 3H), 2.90 (hept, J = 6.6 Hz, 1H), 1.31 (d, J = 6.7 Hz, 12H), 1.25 (d, J = 6.9 Hz, 6H). ¹³C NMR (150 MHz, CDCl₃) δ 166.62,

153.88, 151.59, 144.72, 137.60, 131.47, 131.23, 130.00, 127.20, 124.12, 52.42, 34.34, 30.26, 25.02, 23.66. IR (neat, cm⁻¹): 3178.94, 2959.44, 1723.13, 1602.35, 1277.52, 1168.72, 1055.95, 946.47. HRMS (DART) ([M+H]⁺) Calcd. for C₂₄H₃₃N₂O₄S⁺: 445.21555, Found: 445.21508.

3-Nitrobenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1p) Yield: 68%. R_f



= 0.44 (2:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.40
(s, 1H), 8.21 – 8.16 (m, 2H), 7.85 (d, J = 7.8 Hz, 1H), 7.82 (s, 1H),
7.52 (t, J = 7.9 Hz, 1H), 7.21 (s, 2H), 4.27 (hept, J = 6.8 Hz, 2H),
2.91 (hept, J = 7.1 Hz, 1H), 1.33 (d, J = 6.5 Hz, 12H), 1.25 (d, J =
6.9 Hz, 6H). ¹³C NMR (150 MHz, DMSO-d₆) δ 152.85, 150.46,

148.13, 142.34, 135.69, 133.13, 132.34, 130.37, 123.94, 123.66, 120.05, 33.34, 29.23, 24.67, 23.33. IR (neat, cm⁻¹): 3170.57, 2960.81, 1600.68, 1533.84, 1345.34, 1294.55, 1150.78, 1036.76. HRMS (DART) ([M+H]⁺) Calcd. for C₂₂H₃₀N₃O₄S⁺: 432.19515, Found: 432.19583.

2-Bromobenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1q) Yield: 79%.



¹): 3188.79, 2958.28, 1598.44, 1426.49, 1323.03, 1165.31, 1037.35, 931.68. HRMS (DART) ([M+H]⁺) Calcd. for C₂₂H₃₀N₂O₂SBr⁺: 465.12059, Found: 465.11968.

4-Phenylbenzaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1r) Yield: 74%.



 $R_f = 0.48$ (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 7.93 (s, 1H), 7.79 (s, 1H), 7.64 – 7.61 (m, 2H), 7.59 – 7.56 (m, 4H), 7.45 – 7.41 (m, 2H), 7.38 – 7.33 (m, 1H), 7.19 (s, 2H), 4.30 (hept, J =6.8 Hz, 2H), 2.90 (hept, J = 7.0 Hz, 1H), 1.33 (d, J = 6.8 Hz, 12H), 1.25 (d, J = 6.9 Hz, 6H). ¹³C NMR (150 MHz, CDCl₃) δ 153.67,

151.54, 146.12, 143.17, 140.32, 132.42, 131.41, 129.01, 127.94, 127.88, 127.42, 127.18, 124.06, 34.34, 30.24, 25.04, 23.68. IR (neat, cm⁻¹): 3185.70, 2960.10, 1602.06, 1376.42, 1321.51, 1168.65, 1052.42, 1036.60, 943.11. HRMS (DART) ($[M+H]^+$) Calcd. for C₂₈H₃₅N₂O₂S⁺: 463.24138, Found: 463.23993.

2-Naphthaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1s) Yield: 73%. $R_f =$



0.42 (3:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ 8.05 (s, 1H), 7.91 (s, 1H), 7.85 (s, 1H), 7.84 – 7.75 (m, 4H), 7.51 – 7.46 (m, 2H), 7.20 (s, 2H), 4.33 (hept, *J* = 6.8 Hz, 2H), 2.90 (hept, *J* = 6.8 Hz, 1H), 1.35 (d, *J* = 6.7 Hz, 12H), 1.25 (d, *J* = 6.9 Hz, 6H). ¹³C

NMR (150 MHz, CDCl₃) δ 153.68, 151.57, 146.61, 134.41, 133.11, 131.45, 131.21, 129.01, 128.61, 128.46, 127.99, 127.30, 126.78, 124.07, 123.12, 34.33, 30.28, 25.07, 23.67. IR (neat, cm⁻¹): 3177.75, 2956.13, 1601.90, 1378.37, 1317.63, 1167.14, 1037.27, 954.15. HRMS (DART) ([M+H]⁺) Calcd. for C₂₆H₃₃N₂O₂S⁺: 437.22573, Found: 437.22626.

2.4.4. Synthesis and Characterization of Aryl- and Heteroaryl Cyclopropanes

2.4.4.1. Experimental Procedure for [Co(Por)]-Catalyzed Asymmetric Cyclopropanation

A 10 mL oven-dried Schlenk tube was charged with *N*-sulfonyl hydrazone (0.10 mmol, 1.0 equiv), [Co(Por)] (2 mol %) and Cs₂CO₃ (0.20 mmol, 2.0 equiv). The Schlenk tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen 3 times. Under nitrogen atmosphere, olefin (1.5–10.0 equiv) and anhydrous toluene (1.0 mL) were added. The Schlenk tube was then purged with nitrogen for 1 min and sealed with the Teflon screw cap. The reaction mixture was stirred at 22–80 °C for 16 h. Following completion of the reaction, the reaction mixture was filtered through a pad of silica gel, concentrated under vacuum and purified by flash column chromatography.





procedure using 1.5 equiv of olefin with 2-pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 99%. dr: 96:4. $R_f = 0.34$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-355.7^\circ$ (c = 2.1, CHCl₃). ¹H NMR

(600 MHz, CDCl₃) δ 8.50 (d, J = 5.0 Hz, 1H), 7.54 (td, J = 7.7, 1.8 Hz, 1H), 7.33 – 7.27 (m, 2H), 7.22 – 7.15 (m, 4H), 7.07 (ddd, J = 7.5, 4.9, 1.2 Hz, 1H), 2.54 (ddd, J = 8.9, 6.1, 4.2 Hz, 1H), 2.29 (ddd, J = 8.6, 5.6, 4.2 Hz, 1H), 1.79 (ddd, J = 8.8, 5.6, 4.5 Hz, 1H), 1.48 (ddd, J = 8.5, 6.1, 4.6 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.24, 149.55, 142.43, 135.98, 128.50, 126.02, 125.94, 122.13, 120.75, 29.52, 28.25, 19.06. IR (neat, cm⁻¹): 1595.08, 1567.49, 1473.29, 1442.90, 1207.26, 745.52, 697.05. HPLC analysis (*E*)-isomer: ee = 99%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 8.15$ min, $t_{minor} = 6.94$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₄N⁺: 196.11208, Found: 196.11143.

2-((1R,2R)-2-(4-Methoxyphenyl)cyclopropyl)pyridine (3b) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 98%. dr: 95:5. $R_f = 0.30$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

392.7° (c = 2.1, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.49 (d, J = 5.1 Hz, 1H), 7.54 (td,



J = 7.6, 1.8 Hz, 1H), 7.18 (d, *J* = 7.8 Hz, 1H), 7.14 – 7.08 (m, 2H), 7.05 (ddd, *J* = 7.5, 4.9, 1.2 Hz, 1H), 6.93 – 6.75 (m, 2H), 3.79 (s, 3H), 2.49 (ddd, *J* = 9.0, 6.1, 4.2 Hz, 1H), 2.20 (ddd, *J* = 8.5, 5.4, 4.2 Hz, 1H), 1.72 (ddd, *J* = 8.9, 5.5, 4.5

Hz, 1H), 1.41 (ddd, J = 8.6, 6.1, 4.5 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.46,

158.04, 149.52, 135.95, 134.42, 127.14, 122.04, 120.65, 113.99, 55.47, 29.12, 27.61, 18.70. IR (neat, cm⁻¹): 1594.80, 1514.56, 1473.47, 1246.04, 1035.94, 905.42, 823.41, 745.89. HPLC analysis (*E*)-isomer: ee = 99%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 13.88$ min, $t_{minor} = 11.39$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₆NO⁺: 226.12264, Found: 226.12349. The absolute configuration of the major (*E*)isomer was assigned as (1*R*,2*R*) by X-ray crystallography.

2-((1R,2R)-2-(4-(Trifluoromethyl)phenyl)cyclopropyl)pyridine (3c) was obtained



through the general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 95%. dr: 94:6. $R_f = 0.43$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

322.8° (c = 2.2, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.50 (d, J = 6.6 Hz, 1H), 7.59 – 7.49 (m, 3H), 7.25 (d, J = 8.0 Hz, 2H), 7.20 (dd, J = 7.9, 1.1 Hz, 1H), 7.08 (ddd, J = 7.4, 4.9, 1.2 Hz, 1H), 2.60 (ddd, J = 9.6, 5.8, 4.2 Hz, 1H), 2.31 (ddd, J = 9.5, 4.9, 3.7 Hz, 1H), 1.86 – 1.83 (m, 1H), 1.53 – 1.46 (m, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 160.49, 149.65, 146.75, 136.11, 128.16 (q, J = 32.4 Hz), 126.22, 125.43 (q, J = 3.9 Hz), 124.49 (q, J = 271.5 Hz), 122.25, 121.05, 30.00, 27.85, 19.42. ¹⁹F NMR (564 MHz, CDCl₃) δ –62.30 (s, 3F). IR (neat, cm⁻¹): 1618.51, 1592.84, 1474.04, 1323.46, 1162.73, 1067.82, 906.69, 825.82, 745.93. HPLC analysis (*E*)-isomer: ee = 99%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 6.32$ min, $t_{minor} = 5.64$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₃NF₃⁺: 264.09946, Found: 264.09941.

2-((1R,2R)-2-(2-Bromophenyl)cyclopropyl)pyridine (3d) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 97%. dr: 90:10. $R_f = 0.52$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -122.7° (c = 1.9, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.52 (d, J = 3.5 Hz, 1H), 7.62 – 7.53 (m, 2H), 7.30 – 7.23 (m, 2H), 7.14 (dd, J = 7.8, 1.6 Hz, 1H), 7.10 – 7.06 (m, 2H), 2.76 (ddd, J = 8.8, 6.2, 4.6 Hz, 1H), 2.20 (ddd, J = 8.5, 5.4, 4.5 Hz, 1H), 1.81 (ddd, J = 8.8, 5.5, 4.6 Hz, 1H), 1.48 (ddd, J = 8.6, 6.3, 4.6 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.09, 149.50, 141.34, 135.97, 132.63, 127.68, 127.51, 127.43, 126.45, 122.18, 120.84, 28.88, 28.27, 17.36. IR (neat, cm⁻¹): 1587.49, 1473.06, 1270.11, 1148.66, 1023.52, 907.20, 794.01, 743.88. HPLC analysis (*E*)-isomer: ee = 97%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 7.57$ min, $t_{minor} = 6.96$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₃NBr⁺: 274.02259, Found: 274.02201.

2-((1R,2R)-2-(3-(4,4,5,5-Tetramethyl-1,3,2-dioxaborolan-2-



yl)phenyl)cyclopropyl)pyridine (3e) was obtained through the general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield:

90%. dr: 93:7. $R_f = 0.41$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-250.0^\circ$ (c = 1.7, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.48 (d, J = 6.9 Hz, 1H), 7.63 (d, J = 6.8 Hz, 1H), 7.61 (s, 1H), 7.53 (td, J = 7.7, 1.8 Hz, 1H), 7.32 – 7.26 (m, 2H), 7.18 (d, J = 7.8 Hz, 1H), 7.05 (ddd, J = 7.5, 4.9, 1.2 Hz, 1H), 2.55 (ddd, J = 9.0, 6.1, 4.2 Hz, 1H), 2.29 (ddd, J = 8.6, 5.5, 4.2 Hz, 1H), 1.79 – 1.73 (m, 1H), 1.53 (ddd, J = 8.6, 6.1, 4.5 Hz, 1H), 1.34 (s, 12H). ¹³C NMR (150 MHz, CDCl₃) δ 161.35, 149.52, 141.62, 135.94, 132.43, 132.28, 129.20, 127.91, 122.10, 120.69, 83.91, 29.52, 28.30, 25.01, 18.73. ¹¹B NMR (160 MHz, CDCl₃) δ 31.24. IR (neat, cm⁻¹): 2977.20, 1587.80, 1473.44, 1355.23, 1143.53, 964.23, 845.56, 707.17, 679.34. HPLC analysis (*E*)-isomer: ee = 97%. IC (95% hexanes: 5% isopropanol, 0.8

mL/min): $t_{major} = 7.22$ min, $t_{minor} = 6.78$ min. HRMS (DART) ([M+H]⁺) Calcd. for $C_{20}H_{25}BNO_2^+$: 322.19729, Found: 322.19686.





general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 87%. dr: 94:6. $R_f = 0.33$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

349.5° (c = 1.6, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 9.99 (s, 1H), 8.50 (d, J = 6.7 Hz, 1H), 7.71 – 7.66 (m, 1H), 7.65 (s, 1H), 7.57 (td, J = 7.7, 1.8 Hz, 1H), 7.45 (d, J = 5.9 Hz, 2H), 7.21 (d, J = 9.0 Hz, 1H), 7.09 (dd, J = 7.5, 4.8 Hz, 1H), 2.67 – 2.59 (m, 1H), 2.34 (ddd, J = 9.2, 5.6, 4.3 Hz, 1H), 1.85 – 1.82 (m, 1H), 1.52 (ddd, J = 8.7, 6.0, 4.7 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 192.56, 160.57, 149.46, 143.69, 136.72, 136.27, 132.65, 129.14, 127.84, 126.50, 122.25, 121.05, 29.59, 27.72, 19.23. IR (neat, cm⁻¹): 3005.00, 1697.11, 1587.92, 1473.88, 1437.79, 1213.25, 1148.71, 785.45, 690.37. HPLC analysis (*E*)-isomer: ee = 96%. ID (80% hexanes: 20% isopropanol, 0.8 mL/min): $t_{major} = 15.18$ min, $t_{minor} = 12.97$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₄NO⁺: 224.10699, Found: 224.10687.

2-((1R,2R)-2-Methyl-2-phenylcyclopropyl)pyridine (3g) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 82%. dr: 98:2. $R_f = 0.45$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-242.0^\circ$

(*c* = 1.4, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.57 (d, *J* = 7.3 Hz, 1H), 7.60 (td, *J* = 7.7, 1.9 Hz, 1H), 7.41 – 7.37 (m, 2H), 7.36 – 7.29 (m, 3H), 7.24 – 7.19 (m, 1H), 7.11 (ddd, *J* = 7.5, 4.9, 1.2 Hz, 1H), 2.51 (dd, *J* = 8.6, 6.3 Hz, 1H), 1.74 (dd, *J* = 6.3, 4.9 Hz, 1H), 1.53

(dd, J = 8.7, 4.9 Hz, 1H), 1.24 (s, 3H). ¹³C NMR (150 MHz, CDCl₃) δ 159.23, 149.06, 147.69, 135.82, 128.48, 127.03, 125.98, 124.40, 120.88, 33.28, 29.17, 19.73, 19.00. IR (neat, cm⁻¹): 3005.53, 1589.02, 1495.62, 1118.95, 1030.52, 876.82, 765.55, 698.79. HPLC analysis (*E*)-isomer: ee = 98%. IC (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 6.87$ min, $t_{minor} = 7.31$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₆N⁺: 210.12773, Found: 210.12697.

2-((1S,2S)-2-Bromo-2-phenylcyclopropyl)pyridine (3h) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 86%. dr: 99:1. $R_f = 0.36$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-115.5^\circ$

 $(c = 0.9, \text{CHCl}_3)$. ¹H NMR (500 MHz, CDCl}_3) δ 8.65 (d, J = 7.3 Hz, 1H), 7.71 (td, J = 7.7, 1.9 Hz, 1H), 7.64 – 7.56 (m, 2H), 7.43 – 7.33 (m, 3H), 7.33 – 7.27 (m, 1H), 7.22 (ddd, J = 7.6, 4.8, 1.1 Hz, 1H), 2.73 (dd, J = 9.7, 7.6 Hz, 1H), 2.29 (dd, J = 7.5, 6.7 Hz, 1H), 1.95 (dd, J = 9.7, 6.6 Hz, 1H). ¹³C NMR (150 MHz, CDCl}_3) δ 157.24, 149.19, 144.22, 136.17, 128.80, 128.74, 128.26, 124.18, 122.08, 41.33, 32.63, 20.85. IR (neat, cm⁻¹): 3058.53, 1591.26, 1494.96, 1149.08, 1038.78, 954.47, 741.78, 696.26. HPLC analysis (*Z*)-isomer: ee = 97%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 16.21$ min, $t_{minor} = 8.98$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₃NBr⁺: 274.02259, Found: 274.02183.



90%. dr: 95:5. $R_f = 0.31$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

466.0° (c = 1.4, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.52 (d, J = 4.2 Hz, 1H), 7.84 –

7.72 (m, 3H), 7.62 (s, 1H), 7.56 (td, J = 7.6, 1.8 Hz, 1H), 7.49 – 7.37 (m, 2H), 7.29 (dd, J = 8.3, 1.6 Hz, 1H), 7.22 (d, J = 7.8 Hz, 1H), 7.08 (ddd, J = 7.5, 4.9, 1.1 Hz, 1H), 2.71 (ddd, J = 8.9, 6.1, 4.2 Hz, 1H), 2.39 (ddd, J = 8.6, 5.5, 4.2 Hz, 1H), 1.85 (ddd, J = 8.8, 5.6, 4.6 Hz, 1H), 1.60 (ddd, J = 8.5, 6.1, 4.6 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.22, 149.61, 139.90, 136.03, 133.66, 132.20, 128.13, 127.76, 127.48, 126.24, 125.25, 124.89, 124.27, 122.19, 120.81, 29.58, 28.52, 19.01. IR (neat, cm⁻¹): 3052.65, 1593.09, 1473.31, 1269.73, 1148.24, 990.13, 814.59, 769.19. HPLC analysis (*E*)-isomer: ee = 98%. IA (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 11.56$ min, $t_{minor} = 8.18$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₁₆N⁺: 246.12773, Found: 246.12781.

2-((1R,2S)-2-((E)-Styryl)cyclopropyl)pyridine (3j) was obtained through the general



procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 93%. dr: 96:4. $R_f = 0.45$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

477.7° (c = 2.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.47 (d, J = 5.8 Hz, 1H), 7.54 (td, J = 7.6, 2.3 Hz, 1H), 7.35 – 7.31 (m, 2H), 7.30 – 7.27 (m, 2H), 7.21 – 7.16 (m, 2H), 7.07 – 7.03 (m, 1H), 6.53 (d, J = 15.7 Hz, 1H), 5.94 (dd, J = 15.7, 8.8 Hz, 1H), 2.23 – 2.14 (m, 2H), 1.65 (ddd, J = 8.5, 5.6, 4.4 Hz, 1H), 1.25 (ddd, J = 8.5, 5.8, 4.4 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.06, 149.51, 137.63, 135.96, 132.59, 128.88, 128.64, 126.94, 125.84, 122.02, 120.69, 27.69, 27.33, 17.77. IR (neat, cm⁻¹): 3023.41, 1591.42, 1473.47, 1242.33, 1148.60, 957.85, 765.10, 692.81. HPLC analysis (*E*)-isomer: ee = 98%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 9.82$ min, $t_{minor} = 8.54$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₆N⁺: 222.12773, Found: 222.12748.

2-((1R,2R)-2-((E)-1-Phenylprop-1-en-2-yl)cyclopropyl)pyridine (3k) was obtained



through the general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 98%. dr: 99:1. $R_f = 0.52$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

284.7° (c = 2.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.49 (d, J = 5.2 Hz, 1H), 7.56 (td, J = 7.7, 1.8 Hz, 1H), 7.34 – 7.31 (m, 2H), 7.28 – 7.23 (m, 2H), 7.22 – 7.15 (m, 2H), 7.06 (ddd, J = 7.5, 4.9, 1.1 Hz, 1H), 6.43 (s, 1H), 2.26 – 2.20 (m, 2H), 1.85 (s, 3H), 1.51 (ddd, J = 8.6, 5.6, 4.5 Hz, 1H), 1.39 (ddd, J = 8.3, 6.5, 4.5 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.76, 149.50, 138.41, 138.01, 135.97, 128.98, 128.18, 126.05, 124.63, 121.83, 120.64, 32.82, 25.32, 15.90, 15.74. IR (neat, cm⁻¹): 2918.57, 1588.00, 1473.50, 1211.59, 1148.50, 918.63, 743.14, 698.42. HPLC analysis (*E*)-isomer: ee = 93%. ID (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 8.06$ min, $t_{minor} = 6.17$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₁₈N⁺: 236.14338, Found: 236.14311.

2-((1R,2R)-2-(Phenylethynyl)cyclopropyl)pyridine (31) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 98%. dr: 99:1. $R_f = 0.53$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

424.5° (c = 2.3, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.45 (d, J = 4.8 Hz, 1H), 7.56 (td, J = 7.7, 1.8 Hz, 1H), 7.45 – 7.37 (m, 2H), 7.32 – 7.21 (m, 4H), 7.07 (ddd, J = 7.5, 4.9, 1.2 Hz, 1H), 2.43 (ddd, J = 8.5, 5.7, 4.2 Hz, 1H), 2.11 (ddd, J = 8.8, 5.9, 4.2 Hz, 1H), 1.67 (ddd, J = 8.8, 5.8, 4.0 Hz, 1H), 1.43 (ddd, J = 8.5, 5.9, 4.0 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 159.51, 149.54, 136.03, 131.78, 128.33, 127.75, 123.83, 122.57, 121.17, 91.98, 77.12, 27.86, 18.87, 12.51. IR (neat, cm⁻¹): 2226.07, 1588.89, 1473.99, 1210.20, 1148.93,

952.80, 854.76, 755.45, 690.95. HPLC analysis (*E*)-isomer: ee = 98%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): t_{major} = 7.58 min, t_{minor} = 7.13 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₄N⁺: 220.11208, Found: 220.11143.

(1R,2R)-1,2-Di(pyridin-2-yl)cyclopropane (3m) was obtained through the general



procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 92%. dr: 95:5. $R_f = 0.32$ (1:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-452.0^\circ$

 $(c = 0.7, \text{CHCl}_3)$. ¹H NMR (600 MHz, CDCl}3) δ 8.49 (d, J = 5.8 Hz, 2H), 7.54 (td, J = 7.6, 1.8 Hz, 2H), 7.24 (d, J = 7.8 Hz, 2H), 7.06 (ddd, J = 7.5, 4.8, 1.2 Hz, 2H), 2.68 – 2.61 (m, 2H), 1.80 – 1.74 (m, 2H). ¹³C NMR (125 MHz, CDCl_3) δ 161.12, 149.52, 135.98, 122.56, 120.83, 29.39, 19.67. IR (neat, cm⁻¹): 1590.15, 1448.89, 1434.78, 1334.78, 1200.03, 1149.00, 909.01, 799.87. HPLC analysis (*E*)-isomer: ee = 99%. AD-H (70% hexanes: 30% isopropanol, 0.8 mL/min): $t_{major} = 7.03$ min, $t_{minor} = 5.64$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₃H₁₃N₂⁺: 197.10732, Found: 197.10720.

2-((1R,2R)-2-(Pyridin-3-yl)cyclopropyl)pyridine (3n) was obtained through the general



procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 86%. dr: 98:2. $R_f = 0.30$ (1:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-368.6^\circ$

 $(c = 0.9, \text{CHCl}_3)$. ¹H NMR (600 MHz, CDCl}3) δ 8.55 – 8.37 (m, 3H), 7.55 (td, J = 7.7, 1.8 Hz, 1H), 7.40 (d, J = 7.8 Hz, 1H), 7.20 – 7.19 (m, 2H), 7.07 (dd, J = 7.5, 4.9 Hz, 1H), 2.54 (ddd, J = 9.7, 6.1, 4.2 Hz, 1H), 2.34 – 2.23 (m, 1H), 1.82 – 1.79 (m, 1H), 1.46 (ddd, J = 8.6, 6.0, 4.7 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 160.48, 149.65, 148.53, 147.38, 137.87, 136.12, 132.97, 123.37, 122.20, 121.06, 29.17, 25.43, 18.72. IR (neat, cm⁻¹):

1588.00, 1473.87, 1431.53, 1212.60, 1046.43, 904.82, 804.72, 712.35. HPLC analysis (*E*)isomer: ee = 99%. OD-H (70% hexanes: 30% isopropanol, 0.8 mL/min): t_{major} = 8.43 min, t_{minor} = 7.56 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₃H₁₃N₂⁺: 197.10732, Found: 197.10836.

2-((1R,2R)-2-(Pyridin-4-yl)cyclopropyl)pyridine (30) was obtained through the general



procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 94%. dr: 95:5. $R_f = 0.34$ (EtOAc only). $[\alpha]_D^{20} = (-)-377.6^\circ$ (c = 0.8,

CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.49 (d, *J* = 3.3 Hz, 1H), 8.46 (d, *J* = 6.2 Hz, 2H), 7.56 (td, *J* = 7.6, 1.8 Hz, 1H), 7.20 (d, *J* = 7.8 Hz, 1H), 7.09 (dd, *J* = 7.5, 4.9 Hz, 1H), 7.04 (d, *J* = 6.2 Hz, 2H), 2.50 (ddd, *J* = 9.3, 5.9, 4.2 Hz, 1H), 2.34 (ddd, *J* = 9.4, 5.7, 4.1 Hz, 1H), 1.89 – 1.85 (m, 1H), 1.55 – 1.46 (m, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 160.02, 151.89, 149.74, 149.67, 136.15, 122.32, 121.20, 121.13, 30.26, 27.19, 19.65. IR (neat, cm⁻¹): 1599.23, 1473.72, 1429.94, 1204.91, 1049.42, 908.49, 813.37, 758.28. HPLC analysis (*E*)-isomer: ee = 99%. AD-H (70% hexanes: 30% isopropanol, 0.8 mL/min): *t_{major}* = 11.33 min, *t_{minor}* = 7.91 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₃H₁₃N₂⁺: 197.10732, Found: 197.10702.

2-((1R,2R)-2-(Thiophen-2-yl)cyclopropyl)pyridine (3p) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 97%. dr: 96:4. $R_f = 0.48$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-351.4^\circ$

(*c* = 1.6, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.48 (d, *J* = 4.5 Hz, 1H), 7.55 (td, *J* = 7.6, 1.9 Hz, 1H), 7.21 (d, *J* = 8.9 Hz, 1H), 7.10 – 7.04 (m, 2H), 6.92 (dd, *J* = 5.1, 3.5 Hz, 1H),

6.85 (d, J = 3.5 Hz, 1H), 2.78 – 2.71 (m, 1H), 2.31 (ddd, J = 8.6, 5.6, 4.2 Hz, 1H), 1.80 (ddd, J = 8.8, 5.6, 4.5 Hz, 1H), 1.47 (ddd, J = 8.6, 6.0, 4.5 Hz, 1H). ¹³C NMR (150 MHz, CDC1₃) δ 160.56, 149.56, 146.91, 136.02, 126.99, 123.06, 122.39, 122.28, 120.90, 30.20, 23.50, 20.01. IR (neat, cm⁻¹): 3006.31, 1590.29, 1473.22, 1440.26, 1214.88, 1148.74, 848.51, 691.90. HPLC analysis (*E*)-isomer: ee = 98%. ID (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 6.75$ min, $t_{minor} = 6.15$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₂H₁₂NS⁺: 202.06850, Found: 202.06905.

2-((1R,2R)-2-(Benzofuran-2-yl)cyclopropyl)pyridine (3q) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 96%. dr: 99:1. $R_f = 0.50$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-493.2^\circ$

 $(c = 2.1, CHCl_3)$. ¹H NMR (600 MHz, CDCl_3) δ 8.50 (d, J = 4.4 Hz, 1H), 7.56 (td, J = 7.6, 1.8 Hz, 1H), 7.50 – 7.43 (m, 1H), 7.40 (d, J = 8.1 Hz, 1H), 7.27 – 7.16 (m, 3H), 7.08 (ddd, J = 7.5, 4.9, 1.2 Hz, 1H), 6.46 (s, 1H), 2.70 (ddd, J = 9.0, 6.1, 4.2 Hz, 1H), 2.58 (ddd, J = 8.7, 5.7, 4.3 Hz, 1H), 1.81 (ddd, J = 9.0, 5.7, 4.2 Hz, 1H), 1.71 (ddd, J = 8.7, 6.0, 4.2 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 160.07, 158.65, 154.39, 149.62, 136.07, 129.21, 123.24, 122.72, 122.52, 121.06, 120.16, 110.78, 101.34, 27.23, 21.62, 17.01. IR (neat, cm⁻¹): 1592.79, 1473.63, 1453.51, 1251.80, 1168.18, 948.66, 747.00, 670.27. HPLC analysis (*E*)-isomer: ee = 99%. ID (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 7.48$ min, $t_{minor} = 6.76$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₄NO⁺: 236.10699, Found: 236.10707.
2-((1R,2R)-2-(Benzo[b]thiophen-3-yl)cyclopropyl)pyridine (3r) was obtained through



the general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 96%. dr: 75:25. $R_f = 0.47$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

165.6° (c = 1.8, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.54 (d, J = 4.5 Hz, 1H), 7.88 – 7.80 (m, 2H), 7.58 (td, J = 7.6, 1.8 Hz, 1H), 7.38 – 7.32 (m, 2H), 7.25 (d, J = 8.3 Hz, 1H), 7.11 (ddd, J = 7.5, 4.8, 1.1 Hz, 1H), 7.08 (s, 1H), 2.78 – 2.73 (m, 1H), 2.29 – 2.23 (m, 1H), 1.79 (ddd, J = 9.0, 5.3, 4.3 Hz, 1H), 1.53 (ddd, J = 8.5, 6.2, 4.3 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.23, 149.64, 140.53, 139.79, 137.64, 136.12, 124.57, 124.13, 122.92, 122.28, 122.22, 120.89, 120.17, 26.96, 21.84, 16.94. IR (neat, cm⁻¹): 1591.23, 1472.90, 1426.70, 1351.16, 1048.29, 928.66, 761.24, 731.26. HPLC analysis (*E*)-isomer: ee = 98%. ID (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 8.40$ min, $t_{minor} = 6.67$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₄NS⁺: 252.08415, Found 252.08426.

tert-Butyl 3-((1R,2R)-2-(pyridin-2-yl)cyclopropyl)-1H-indole-1-carboxylate (3s) was



obtained through the general procedure using 1.5 equiv of olefin with 2-pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 93%. dr: 95:5. $R_f = 0.42$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-$

)-175.3° (c = 2.6, CHCl₃). ¹H NMR (600 MHz, CDC₃) δ 8.52 (d, J = 4.5 Hz, 1H), 8.12 (s, 1H), 7.61 – 7.55 (m, 2H), 7.39 – 7.29 (m, 2H), 7.25 – 7.20 (m, 2H), 7.09 (dd, J = 7.5, 4.8 Hz, 1H), 2.57 (ddd, J = 8.9, 6.7, 4.4 Hz, 1H), 2.30 – 2.25 (m, 1H), 1.74 (ddd, J = 9.1, 5.3, 4.2 Hz, 1H), 1.67 (s, 9H), 1.46 (ddd, J = 8.4, 6.1, 4.2 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 161.37, 149.93, 149.57, 136.09, 131.09, 124.61, 122.60, 122.52, 122.13, 121.52, 120.81, 119.42, 115.38, 83.61, 28.38, 26.82, 18.92, 17.10. IR (neat, cm⁻¹): 1729.33, 1591.53,

1473.98, 1372.36, 1253.74, 1156.82, 1077.36, 744.41. HPLC analysis (*E*)-isomer: ee = 99%. OD-H (98% hexanes: 2% isopropanol, 0.5 mL/min): $t_{major} = 13.56$ min, $t_{minor} = 12.77$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₂₃N₂O₂⁺: 335.17540, Found: 335.17544.

4-((1R,2R)-2-(Pyridin-2-yl)cyclopropyl)quinoline (3t) was obtained through the general



procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 90%. dr: 85:15. $R_f = 0.22$ (1:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

121.2° (c = 1.2, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.83 (d, J = 4.5 Hz, 1H), 8.57 (d, J = 4.0 Hz, 1H), 8.14 (d, J = 8.4 Hz, 1H), 8.11 (d, J = 8.4 Hz, 1H), 7.70 (ddd, J = 8.2, 6.9, 1.2 Hz, 1H), 7.63 – 7.57 (m, 1H), 7.51 (td, J = 7.0, 1.2 Hz, 1H), 7.28 – 7.25 (m, 1H), 7.17 (d, J = 4.5 Hz, 1H), 7.14 (ddd, J = 7.5, 3.5, 2.4 Hz, 1H), 3.15 (ddd, J = 8.8, 6.2, 4.7 Hz, 1H), 2.36 – 2.29 (m, 1H), 1.92 (ddd, J = 10.0, 6.1, 3.9 Hz, 1H), 1.69 – 1.62 (m, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 160.29, 150.24, 149.81, 148.59, 147.99, 136.28, 129.93, 129.45, 128.75, 126.66, 124.31, 122.52, 121.27, 117.39, 28.40, 24.19, 17.42. HPLC analysis (*E*)-isomer: ee = 98%. IC (80% hexanes: 20% isopropanol, 0.8 mL/min): $t_{major} = 20.97$ min, $t_{minor} = 17.35$ min. IR (neat, cm⁻¹): 3005.23, 1587.98, 1567.02, 1473.25, 1428.53, 1211.04, 1149.37, 908.62, 760.17. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₁₅N₂⁺: 247.12297, Found: 247.12285.

1-((1*R*,2*R*)-2-(Pyridin-2-yl)cyclopropyl)propan-1-one (3u) was obtained through the $H \rightarrow H$ general procedure using 3.0 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 94%. dr: 99:1. $R_f = 0.32$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-378.6^\circ$

(*c* = 0.8, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.44 (d, *J* = 7.6 Hz, 1H), 7.55 (td, *J* = 7.7,

1.8 Hz, 1H), 7.21 (d, J = 7.8 Hz, 1H), 7.08 (ddd, J = 7.3, 4.8, 1.1 Hz, 1H), 2.71 – 2.53 (m, 4H), 1.64 – 1.59 (m, 2H), 1.09 (t, J = 7.3 Hz, 3H). ¹³C NMR (150 MHz, CDCl₃) δ 209.95, 159.37, 149.49, 136.24, 122.74, 121.42, 37.31, 31.65, 29.39, 19.63, 8.01. IR (neat, cm⁻¹): 1698.77, 1593.43, 1392.77, 1120.21, 1031.18, 983.81, 889.73, 746.83. HPLC analysis (*E*)-isomer: ee = 98%. IF (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 10.67$ min, $t_{minor} = 9.89$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₁H₁₄NO⁺: 176.10699, Found: 176.10595.

Methyl (1R,2R)-2-(pyridin-2-yl)cyclopropane-1-carboxylate (3v) was obtained through



the general procedure using 3.0 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 91%. dr: 99:1. $R_f = 0.30$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-312.4^\circ$

 $(c = 1.8, CHCl_3)$. ¹H NMR (500 MHz, CDCl_3) δ 8.44 (d, J = 6.6 Hz, 1H), 7.55 (td, J = 7.7, 1.8 Hz, 1H), 7.22 (d, J = 7.8 Hz, 1H), 7.08 (dd, J = 8.7, 4.8 Hz, 1H), 3.71 (s, 3H), 2.58 (ddd, J = 9.6, 6.1, 3.9 Hz, 1H), 2.26 (ddd, J = 9.0, 5.5, 3.9 Hz, 1H), 1.64 (ddd, J = 8.3, 6.1, 3.8 Hz, 1H), 1.59 (ddd, J = 9.2, 5.5, 3.9 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 173.96, 158.89, 149.55, 136.15, 122.65, 121.45, 52.01, 27.40, 24.24, 17.43. IR (neat, cm⁻¹): 2951.64, 1724.32, 1594.50, 1475.14, 1435.87, 1334.28, 1200.40, 1169.81, 775.25. HPLC analysis (*E*)-isomer: ee = 95%. IE (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} =$ 13.58 min, $t_{minor} = 15.45$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₀H₁₂NO₂⁺: 178.08626, Found: 178.08713.

(1R,2R)-2-(Pyridin-2-yl)cyclopropane-1-carboxamide (3w) was obtained through the



general procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 70%. dr: 99:1. $R_f = 0.28$ (EtOAc only). $[\alpha]_D^{20} = (-)-246.0^\circ$ (c = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.41 (d, J = 4.8 Hz, 1H), 7.55 (ddd, J = 9.3, 6.9, 1.6 Hz, 1H), 7.25 (d, J = 8.2 Hz, 1H), 7.09 – 7.06 (m, 1H), 5.77 (d, J = 66.5 Hz, 2H), 2.57 (ddd, J = 9.5, 5.0, 3.4 Hz, 1H), 2.15 – 2.12 (m, 1H), 1.54 – 1.51 (m, 1H), 1.53 (ddd, J = 9.2, 5.7, 2.3 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 174.53, 159.49, 149.43, 136.24, 123.06, 121.38, 26.69, 25.50, 17.37. IR (neat, cm⁻¹): 338.142, 3198.69, 2921.27, 1644.94, 1431.26, 1365.07, 933.96, 772.19. HPLC analysis (*E*)-isomer: ee = 98%. ID (70% hexanes: 30% isopropanol, 0.8 mL/min): $t_{major} = 10.63$ min, $t_{minor} = 8.49$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₉H₁₁N₂O⁺: 163.08659, Found: 163.08633.

(1*R*,2*R*)-*N*,*N*-Dimethyl-2-(pyridin-2-yl)cyclopropane-1-carboxamide (3x) was



obtained through the general procedure using 1.5 equiv of olefin with 2-pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 95%. dr: 99:1. $R_f = 0.27$ (1:3 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

270.6° (c = 1.1, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.43 (d, J = 5.9 Hz, 1H), 7.54 (td, J = 7.7, 1.8 Hz, 1H), 7.26 (d, J = 7.7 Hz, 1H), 7.06 (dd, J = 7.6, 4.8 Hz, 1H), 3.14 (s, 3H), 2.98 (s, 3H), 2.58 (ddd, J = 9.3, 5.8, 4.0 Hz, 1H), 2.45 (ddd, J = 8.2, 5.5, 4.0 Hz, 1H), 1.58 (ddd, J = 8.8, 5.5, 3.4 Hz, 1H), 1.52 (ddd, J = 8.8, 5.7, 3.3 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 172.12, 160.09, 149.48, 136.10, 122.99, 121.17, 37.49, 35.97, 26.58, 23.06, 17.59. IR (neat, cm⁻¹): 2926.78, 1636.03, 1592.86, 1495.69, 1373.06, 1263.96, 1139.35, 795.19. HPLC analysis (*E*)-isomer: ee = 99%. ID (80% hexanes: 20% isopropanol, 0.8 mL/min): $t_{major} = 33.05$ min, $t_{minor} = 15.83$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₁H₁₅N₂O⁺: 191.11789, Found: 191.11757.

(1R,2R)-2-(Pyridin-2-yl)cyclopropane-1-carbonitrile (3y) was obtained through the



(–)-3z

general procedure using 3.0 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 94%. dr: 98:2. $R_f = 0.33$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-277.1^\circ$ (c =

0.9, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.43 (d, *J* = 7.0 Hz, 1H), 7.60 (td, *J* = 7.7, 1.8 Hz, 1H), 7.29 (d, *J* = 7.8 Hz, 1H), 7.15 – 7.11 (m, 1H), 2.63 (ddd, *J* = 8.9, 6.1, 4.3 Hz, 1H), 2.05 (ddd, *J* = 9.7, 5.8, 4.3 Hz, 1H), 1.71 (ddd, *J* = 9.0, 6.1, 4.6 Hz, 1H), 1.58 (ddd, *J* = 8.8, 5.9, 4.6 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 156.37, 149.76, 136.46, 123.09, 122.19, 121.27, 25.56, 16.52, 6.81. IR (neat, cm⁻¹): 2238.09, 1590.82, 1475.05, 1452.24, 1212.54, 1150.57, 941.64, 763.13. HPLC analysis (*E*)-isomer: ee = 97%. OD-H (95% hexanes: 5% isopropanol, 0.8 mL/min): *t_{major}* = 13.27 min, *t_{minor}* = 11.21 min. HRMS (DART) ([M+H]⁺) Calcd. for C₉H₉N₂⁺: 145.07602, Found: 145.07647.

(1R,2R)-1-Methyl-2-(pyridin-2-yl)cyclopropane-1-carbonitrile (3z) was obtained through the general procedure using 3.0 equiv of olefin with 2-



283.3° (c = 1.6, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 8.50 (d, J = 3.9 Hz, 1H), 7.64 (td, J = 7.7, 1.8 Hz, 1H), 7.35 (d, J = 7.8 Hz, 1H), 7.16 (ddd, J = 7.6, 4.9, 1.2 Hz, 1H), 2.83 (dd, J = 9.2, 7.0 Hz, 1H), 1.82 (dd, J = 7.1, 5.2 Hz, 1H), 1.67 (dd, J = 9.2, 5.3 Hz, 1H), 1.20 (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 154.47, 149.13, 136.41, 125.35, 124.35, 122.22, 30.80, 18.40, 14.67, 12.33. IR (neat, cm⁻¹): 2232.26, 1590.85, 1474.90, 1446.52, 1150.77, 1086.40, 802.77, 756.34. HPLC analysis (*E*)-isomer: ee = 98%. OD-H (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 10.83$ min, $t_{minor} = 9.30$ min. HRMS (DART) ([M+H]⁺)

Calcd. for C₁₀H₁₁N₂⁺: 159.09167, Found: 159.09178.

(1R,2S)-2-(Pyridin-2-yl)cyclopropyl benzoate (3aa) was obtained through the general



procedure using 1.5 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 49%. dr: 92:8. $R_f = 0.34$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

132.5° (c = 0.7, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.47 (d, J = 4.7 Hz, 1H), 8.03 (d, J = 8.4 Hz, 2H), 7.59 – 7.54 (m, 2H), 7.47 – 7.41 (m, 2H), 7.25 (d, J = 8.0 Hz, 1H), 7.10 – 7.07 (m, 1H), 4.69 – 4.67 (m, 1H), 2.42 (ddd, J = 10.0, 6.9, 2.7 Hz, 1H), 1.71 – 1.66 (m, 1H), 1.53 (ddd, J = 11.0, 6.5, 4.3 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 167.12, 159.16, 149.51, 136.12, 133.25, 129.97, 129.74, 128.52, 122.32, 121.17, 56.98, 24.81, 15.51. IR (neat, cm⁻¹): 1723.78, 1589.76, 1321.83, 1474.75, 1268.71, 1137.34, 1094.97, 709.61. HPLC analysis (*E*)-isomer: ee = 98%. IF (80% hexanes: 20% isopropanol, 0.8 mL/min): $t_{major} = 8.97$ min, $t_{minor} = 8.32$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₄NO₂⁺: 240.10191, Found: 240.10166.

2-((1S,2R)-2-Propoxycyclopropyl)pyridine (3ab) was obtained through the general



procedure using 10.0 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 55%. dr: 94:6. $R_f = 0.42$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-77.1^\circ$

 $(c = 0.8, \text{CHCl}_3)$. ¹H NMR (600 MHz, CDCl}_3) δ 8.41 (d, J = 6.0 Hz, 1H), 7.52 (td, J = 7.6, 1.8 Hz, 1H), 7.15 (d, J = 8.9 Hz, 1H), 7.02 (ddd, J = 7.3, 4.8, 1.1 Hz, 1H), 3.62 (ddd, J = 6.3, 3.8, 2.3 Hz, 1H), 3.51 (t, J = 6.7 Hz, 2H), 2.18 (ddd, J = 9.7, 6.0, 2.2 Hz, 1H), 1.63 – 1.57 (m, 2H), 1.39 – 1.33 (m, 1H), 1.31 (ddd, J = 9.5, 5.5, 3.8 Hz, 1H), 0.92 (t, J = 7.4 Hz, 3H). ¹³C NMR (150 MHz, CDCl}_3) δ 160.79, 149.28, 135.92, 121.78, 120.55, 72.77, 62.13,

25.36, 22.93, 17.02, 10.72. IR (neat, cm⁻¹): 2962.34, 2924.53, 1589.74, 1474.29, 1450.77, 1376.03, 1169.24, 1084.12. HPLC analysis (*E*)-isomer: ee = 91%. ID (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 6.47$ min, $t_{minor} = 5.73$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₁H₁₆NO⁺: 178.12264, Found: 178.12218.

2-((1R,2R)-2-Phenethylcyclopropyl)pyridine (3ac) was obtained through the general



procedure using 10.0 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 55%. dr: 88:12. $R_f = 0.45$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

45.4° (c = 0.9, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.42 (d, J = 6.9 Hz, 1H), 7.50 (td, J = 7.6, 1.8 Hz, 1H), 7.27 – 7.22 (m, 2H), 7.19 – 7.14 (m, 3H), 7.04 – 6.97 (m, 2H), 2.81 – 2.71 (m, 2H), 1.80 – 1.66 (m, 3H), 1.45 – 1.37 (m, 1H), 1.23 – 1.20 (m, 1H), 0.82 (ddd, J = 8.4, 5.8, 4.2 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 162.73, 149.26, 142.41, 135.82, 128.59, 128.38, 125.79, 121.31, 120.30, 36.19, 35.88, 25.15, 24.12, 16.92. IR (neat, cm⁻¹): 3024.45, 2921.01, 2853.10, 1588.98, 1474.17, 1148.26, 744.79, 698.73. HPLC analysis (*E*)-isomer: ee = 49%. OD-H (98% hexanes: 2% isopropanol, 0.8 mL/min): $t_{major} = 9.94$ min, $t_{minor} = 8.50$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₈N⁺: 224.14338, Found: 224.14298.

2-((1R,2S)-2-Cyclohexylcyclopropyl)pyridine (3ad) was obtained through the general



procedure using 10.0 equiv of olefin with 2pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 43%. dr: 94:6. $R_f = 0.53$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-87.7^\circ$

(*c* = 0.8, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.41 (d, *J* = 6.7 Hz, 1H), 7.50 (td, *J* = 7.7, 1.8 Hz, 1H), 7.07 (d, *J* = 7.9 Hz, 1H), 6.99 (dd, *J* = 7.4, 4.9 Hz, 1H), 1.82 – 1.79 (m, 3H),

1.76 – 1.58 (m, 3H), 1.24 – 1.05 (m, 7H), 0.90 – 0.77 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 163.26, 149.16, 135.93, 121.08, 120.18, 42.77, 33.10, 32.64, 31.12, 26.64, 26.39, 26.37, 23.74, 15.78. IR (neat, cm⁻¹): 2921.66, 2850.07, 1590.12, 1473.99, 1207.77, 1148.01, 885.72, 743.97. HPLC analysis (*E*)-isomer: ee = 94%. IC (99% hexanes: 1% isopropanol, 0.5 mL/min): t_{major} = 12.84 min, t_{minor} = 13.40 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₂₀N⁺: 202.15903, Found: 202.15983.

2-((1*R*,1a*R*,6a*S*)-1,1a,6,6a-Tetrahydrocyclopropa[*a*]inden-1-yl)pyridine (*exo*-3ae)



was obtained through the general procedure using 10.0 equiv of olefin with 2-pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield (*exo* + *endo*): 43%. dr: 58:42. $R_f = 0.48$ (6:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-192.8^{\circ} (c = 0.6, CHCl_3)$. ¹H NMR (500 MHz, CDCl₃) δ 8.48 (d, J = 7.3 Hz, 1H), 7.53 (td, J = 7.6, 1.9 Hz, 1H), 7.37 – 7.31 (m, 1H), 7.24 – 7.19 (m, 1H), 7.18 – 7.08 (m, 3H), 7.07 – 7.03 (m, 1H), 3.37 (dd, J = 17.5, 6.5 Hz, 1H), 3.15 (d, J = 17.4 Hz, 1H), 2.98 (dd, J = 6.4, 2.0 Hz, 1H), 2.52 – 2.49 (m, 1H), 1.63 (t, J = 3.0 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 161.16, 149.36, 145.81, 142.72, 135.99, 126.29, 125.94, 125.43, 123.83, 121.95, 120.56, 36.53, 36.28, 35.82, 28.13. IR (neat, cm⁻¹): 3041.44, 2912.87, 1715.18, 1590.13, 1566.88, 1472.65, 1440.22, 750.89. HPLC analysis (*exo*-isomer): ee = 81 %. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 7.84$ min, $t_{minor} = 7.30$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₄N⁺: 208.11208, Found: 208.11160.

2-((1*R*,1a*S*,6a*R*)-1,1a,6,6a-Tetrahydrocyclopropa[*a*]inden-1-yl)pyridine (*endo*-3ae)



was obtained through the general procedure using 10.0 equiv of olefin with 2-pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield (*exo* + *endo*): 43%. dr: 58:42. $R_f = 0.43$ (6:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-37.3^{\circ} (c = 0.5, CHCl_3)$. ¹H NMR (600 MHz, CDCl_3) δ 8.43 (d, J = 5.0 Hz, 1H), 7.39 (d, J = 7.5 Hz, 1H), 7.28 – 7.24 (m, 1H), 7.08 (t, J = 7.4 Hz, 1H), 6.97 – 6.89 (m, 2H), 6.79 (d, J = 7.5 Hz, 1H), 6.63 (d, J = 8.1 Hz, 1H), 3.16 (dd, J = 17.5, 7.0 Hz, 1H), 3.00 (t, J = 7.8 Hz, 1H), 2.88 (d, J = 17.5 Hz, 1H), 2.61 (t, J = 8.3 Hz, 1H), 2.48 – 2.41 (m, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 156.39, 148.85, 143.05, 142.86, 135.63, 126.16, 125.84, 125.65, 124.75, 124.57, 120.97, 32.18, 30.02, 28.89, 22.86. IR (neat, cm⁻¹): 3042.87, 2916.07, 1588.90, 1477.24, 1433.25, 1021.26, 794.71, 758.72. HPLC analysis (*endo*-isomer): ee = 93%. IA (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 8.55$ min, $t_{minor} = 8.17$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₄N⁺: 208.11208, Found: 208.11183.

2-((1R,1aS,6bR)-1a,6b-Dihydro-1H-cyclopropa[b]benzofuran-1-yl)pyridine (exo-3af)



was obtained through the general procedure using 10.0 equiv of olefin with 2-pyridinecarboxaldehyde tosylhydrazone (1a) at 80 °C. Yield: 56%. dr: 96:4. $R_f = 0.48$ (6:1 Hexanes/EtOAc). $[\alpha]_D^{20}$

= (-)-224.4° (c = 1.1, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.47 (d, J = 5.0 Hz, 1H), 7.57 - 7.52 (m, 1H), 7.39 (d, J = 7.9 Hz, 1H), 7.19 (dd, J = 7.8, 1.2 Hz, 1H), 7.17 - 7.13 (m, 1H), 7.08 (ddd, J = 7.5, 4.9, 1.2 Hz, 1H), 6.96 - 6.88 (m, 2H), 5.13 (dd, J = 5.4, 1.3 Hz, 1H), 3.32 (dd, J = 5.4, 3.3 Hz, 1H), 1.68 (dd, J = 3.2, 1.4 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 160.31, 159.47, 149.56, 135.84, 130.72, 127.42, 124.29, 122.65, 120.96, 120.70, 110.47, 69.14, 31.78, 27.57. IR (neat, cm⁻¹): 1589.29, 1566.75, 1473.70, 1462.31, 1245.27, 1065.90, 1001.41, 747.17. HPLC analysis (*exo*-isomer): ee = 97%. ID (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 7.01 \text{ min}, t_{minor} = 7.46 \text{ min}.$ HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₂NO⁺: 210.09134, Found: 210.09097.

(8R,9S,13S,14S)-13-Methyl-3-vinyl-6,7,8,9,11,12,13,14,15,16-decahydro-17H-



cyclopenta[*a*]phenanthren-17-one (2ag) was prepared according to the literature.²⁶ Known compound. Yield: 59%. $R_f = 0.33$ (4:1 Hexanes/EtOAc). ¹H NMR (600 MHz, CDCl₃) δ

7.28 – 7.23 (m, 1H), 7.20 (d, J = 8.2 Hz, 1H), 7.13 (s, 1H), 6.66 (dd, J = 17.6, 10.9 Hz, 1H), 5.69 (d, J = 17.6 Hz, 1H), 5.18 (d, J = 10.7 Hz, 1H), 2.91 (dd, J = 9.2, 4.2 Hz, 2H), 2.50 (dd, J = 19.0, 8.8 Hz, 1H), 2.45 – 2.38 (m, 1H), 2.32 – 2.27 (m, 1H), 2.19 – 1.92 (m, 4H), 1.63 – 1.44 (m, 6H), 0.90 (s, 3H). ¹³C NMR (150 MHz, CDCl₃) δ 221.01, 139.66, 136.70, 135.35, 127.01, 125.67, 123.74, 113.32, 50.65, 48.13, 44.59, 38.30, 36.00, 31.73, 29.52, 26.64, 25.86, 21.73, 13.98.

(8R,9S,13S,14S)-13-Methyl-3-((1R,2R)-2-(pyridin-2-yl)cyclopropyl)-



6,7,8,9,11,12,13,14,15,16-decahydro-17*H*-

cyclopenijta[a]phenanthren-17-one (3ag) was obtained through the general procedure using 1.5 equiv of olefin

with 2-pyridinecarboxaldehyde tosylhydrazone (**1a**) at 80 °C. Yield: 93%. dr: 99:1. $R_f = 0.41$ (3:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-149.8^\circ$ (c = 2.6, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.49 (d, J = 5.9 Hz, 1H), 7.54 (td, J = 7.6, 1.8 Hz, 1H), 7.23 (d, J = 8.2 Hz, 1H), 7.17 (dd, J = 7.9, 1.1 Hz, 1H), 7.06 (ddd, J = 7.5, 4.9, 1.2 Hz, 1H), 6.96 (dd, J = 8.1, 2.0

²⁶ Crespin, L.; Biancalana, L.; Morack, T.; Blakemore, D. C.; Ley, S. V. Org. Lett. 2017, 19, 1084–1087.

Hz, 1H), 6.92 (s, 1H), 2.90 (dd, J = 9.1, 4.2 Hz, 2H), 2.56 – 2.46 (m, 2H), 2.46 – 2.38 (m, 1H), 2.34 – 2.21 (m, 2H), 2.19 – 1.94 (m, 4H), 1.75 (ddd, J = 8.9, 5.5, 4.5 Hz, 1H), 1.66 – 1.42 (m, 7H), 0.91 (s, 3H). ¹³C NMR (150 MHz, CDCl₃) δ 221.01, 161.31, 149.46, 139.87, 137.51, 136.60, 136.02, 126.77, 125.56, 123.42, 122.11, 120.70, 50.60, 48.12, 44.42, 38.37, 35.99, 31.72, 29.54, 29.38, 27.92, 26.66, 25.90, 21.71, 18.79, 13.96. IR (neat, cm⁻¹): 2928.26, 1737.11, 1590.21, 1473.50, 1212.59, 1007.18, 748.32. HRMS (DART) ([M+H]⁺) Calcd. for C₂₆H₃₀NO⁺: 372.23219, Found: 372.23197.

3-((1R,2R)-2-Phenylcyclopropyl)pyridine (3ah) was obtained through the general



procedure using 1.5 equiv of olefin with 3pyridinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1b**) at 22 °C. Yield: 91%. dr: 95:5. $R_f = 0.40$ (1:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-252.5^{\circ} (c = 1.3, CHCl_3)$. ¹H NMR (500 MHz, CDCl₃) δ 8.49 (s, 1H), 8.44 (d, J = 4.6 Hz, 1H), 7.41 – 7.38 (m, 1H), 7.34 – 7.28 (m, 2H), 7.25 – 7.18 (m, 2H), 7.18 – 7.13 (m, 2H), 2.24 – 2.14 (m, 2H), 1.55 – 1.44 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 148.36, 147.26, 141.81, 132.84, 129.26, 128.64, 126.23, 125.95, 123.48, 27.96, 25.37, 17.98. IR (neat, cm⁻¹): 3027.05, 1603.41, 1572.98, 1481.13, 1425.29, 1193.12, 1025.40, 803.66. HPLC analysis (*E*)-isomer: ee = 99%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 24.42$ min, $t_{minor} = 28.73$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₄N⁺: 196.11208, Found: 196.11254.

3-((1S,2S)-2-Bromo-2-phenylcyclopropyl)pyridine (3ai) was obtained through the



general procedure using 1.5 equiv of olefin with 3pyridinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1b**) at 22 °C. Yield: 58%. dr: 99:1. $R_f = 0.32$ (2:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-90.5^{\circ} (c = 1.3, CHCl_3)$. ¹H NMR (600 MHz, CDCl₃) δ 8.68 (s, 1H), 8.56 (d, J = 4.8 Hz, 1H), 7.66 (d, J = 7.9 Hz, 1H), 7.56 (d, J = 8.2 Hz, 2H), 7.42 – 7.36 (m, 2H), 7.35 – 7.29 (m, 2H), 2.49 (dd, J = 9.8, 7.7 Hz, 1H), 2.01 (dd, J = 10.5, 7.3 Hz, 1H), 1.89 (dd, J = 7.7, 5.6 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 150.96, 148.24, 143.77, 136.69, 133.88, 128.91, 128.62, 128.43, 123.09, 41.24, 28.61, 21.22. IR (neat, cm⁻¹): 3028.16, 1574.56, 1480.61, 1446.80, 1418.02, 1160.47, 1025.83, 752.06. HPLC analysis (*Z*)-isomer: ee = 99%. IE (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 27.74$ min, $t_{minor} = 21.56$ min. HRMS (DART) ([M+H]⁺) Calcd. For C₁₄H₁₃NBr⁺: 274.02259, Found: 274.02239.

3-((1R,2S)-2-((E)-Styryl)cyclopropyl)pyridine (3aj) was obtained through the general



procedure using 1.5 equiv of olefin with 3pyridinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1b**) at 22 °C. Yield: 83%. dr: 91:9. $R_f = 0.30$ (2:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-328.4^{\circ} (c = 2.1, CHCl_3)$. ¹H NMR (500 MHz, CDCl₃) δ 8.45 (d, J = 2.4 Hz, 1H), 8.42 (dd, J = 4.8, 1.7 Hz, 1H), 7.37 – 7.27 (m, 5H), 7.23 – 7.17 (m, 2H), 6.51 (d, J = 15.7 Hz, 1H), 5.91 (dd, J = 15.8, 8.6 Hz, 1H), 2.04 (ddd, J = 8.8, 5.8, 4.3 Hz, 1H), 1.89 – 1.83 (m, 1H), 1.36 – 1.27 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 148.26, 147.16, 137.80, 137.37, 132.77, 132.04, 129.09, 128.70, 127.14, 125.88, 123.40, 27.30, 23.23, 16.94. IR (neat, cm⁻¹): 3023.71, 1647.37, 1596.12, 1480.67, 1424.92, 1174.53, 959.18, 712.17. HPLC analysis (*E*)-isomer: ee = 98%. ID (80% hexanes: 20% isopropanol, 0.8 mL/min): $t_{major} = 13.84$ min, $t_{minor} = 11.95$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₆N⁺: 222.12773, Found: 222.12880.





through the general procedure using 3.0 equiv of olefin with 3pyridinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1b**) at 22 °C. Yield: 91%. dr: 90:10. $R_f = 0.33$ (1:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-233.6^{\circ} (c = 1.0, CHCl_3)$. ¹H NMR (500 MHz, CDCl_3) δ 8.48 - 8.40 (m, 2H), 7.37 - 7.33 (m, 1H), 7.20 (dd, J = 7.9, 4.8 Hz, 1H), 3.72 (s, 3H), 2.52 (ddd, J = 9.3, 6.5, 4.2 Hz, 1H), 1.96 - 1.88 (m, 1H), 1.70 - 1.59 (m, 1H), 1.33 (ddd, J = 8.4, 6.2,4.7 Hz, 1H). ¹³C NMR (125 MHz, CDCl_3) δ 173.44, 148.55, 147.94, 135.74, 133.54, 123.47, 52.19, 23.70, 23.67, 16.70. IR (neat, cm⁻¹): 2952.54, 1726.14, 1437.88, 1274.44, 1174.66, 1199.69, 905.51, 712.81. HPLC analysis (*E*)-isomer: ee = 97%. AS-H (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 14.24$ min, $t_{minor} = 20.24$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₀H₁₂NO₂⁺: 178.08626, Found: 178.08604.

3-Bromo-5-((1*R***,2***R***)-2-phenylcyclopropyl)pyridine (3al)** was obtained through the general procedure using 1.5 equiv of olefin with 5-bromo-3-pyridinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1c) at 22 °C. Yield: 90%. dr: 97:3. $R_f = 0.49$ (3:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-236.0^{\circ} (c = 1.7, CHCl_3)$. ¹H NMR (600 MHz, CDCl₃) δ 8.50

(s, 1H), 8.39 (s, 1H), 7.54 (s, 1H), 7.34 – 7.29 (m, 2H), 7.25 – 7.19 (m, 1H), 7.14 (d, J = 6.9 Hz, 2H), 2.21 (ddd, J = 8.9, 6.0, 4.5 Hz, 1H), 2.14 (ddd, J= 8.8, 5.8, 4.5 Hz, 1H), 1.57– 1.54 (m, 1H), 1.49 – 1.45 (m, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 148 17–146 45–141 21–140 25–135 49–128 71



148.17, 146.45, 141.21, 140.25, 135.49, 128.71, 126.46, 125.96, 120.96, 28.36, 24.88,

18.15. IR (neat, cm⁻¹): 3028.07, 1603.97, 1579.73, 1499.28, 1424.85, 1097.55, 1017.67, 913.04. HPLC analysis (*E*)-isomer: ee = 99%. OD-H (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 16.75$ min, $t_{minor} = 11.65$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₃NBr⁺: 274.02259, Found: 274.02312. The absolute configuration of the major (*E*)-isomer was assigned as (1*R*,2*R*) by X-ray crystallography.

4-((1R,2R)-2-Phenylcyclopropyl)pyridine (3am) was obtained through the general



procedure using 1.5 equiv of olefin with 4pyridinecarboxaldehyde tosylhydrazone (1d) at 60 °C. Yield: 42%. dr: 95:5. $R_f = 0.22$ (1:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)$ -

241.2° (c = 0.6, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.47 (d, J = 4.4 Hz, 2H), 7.32 – 7.29 (m, 2H), 7.21 (t, J = 7.4 Hz, 1H), 7.14 (d, J = 7.6 Hz, 2H), 7.04 (d, J = 6.4 Hz, 2H), 2.26 (ddd, J = 9.1, 6.2, 4.5 Hz, 1H), 2.14 – 2.09 (m, 1H), 1.61 – 1.57 (m, 1H), 1.53 – 1.50 (m, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 152.44, 149.47, 141.32, 128.68, 126.42, 125.98, 121.00, 29.35, 27.33, 19.05. IR (neat, cm⁻¹): 3025.25, 2924.14, 1599.51, 1498.48, 1417.07, 1206.97, 901.54, 813.39. HPLC analysis (*E*)-isomer: ee = 89%. OD-H (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 32.54$ min, $t_{minor} = 25.61$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₄H₁₄N⁺: 196.11208, Found: 196.11284.

3-((1R,2R)-2-Phenylcyclopropyl)thiophene (3an) was obtained through the general



procedure using 1.5 equiv of olefin with 3thiophenecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (1e) at 22 °C. Yield: 71%. dr: 95:5. $R_f = 0.45$ (20:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-248.3^{\circ} (c = 1.2, \text{CHCl}_3)$. ¹H NMR (600 MHz, CDCl₃) δ 7.32 - 7.27 (m, 2H), 7.27 - 7.25 (m, 1H), 7.21 - 7.16 (m, 1H), 7.13 (dd, J = 8.2, 1.4 Hz, 2H), 6.94 (dd, J = 3.1, 1.3 Hz, 1H), 6.91 (dd, J = 5.0, 1.3 Hz, 1H), 2.22 (ddd, J = 8.6, 6.0, 4.5 Hz, 1H), 2.14 (ddd, J = 8.6, 5.9, 4.5 Hz, 1H), 1.43 – 1.37 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 143.84, 142.53, 128.53, 126.30, 125.89, 125.86, 125.70, 118.50, 27.37, 23.83, 18.23. IR (neat, cm⁻¹): 3025.30, 1604.20, 1537.90, 1499.71, 1459.39, 1073.66, 776.57. HPLC analysis (*E*)-isomer: ee = 93%. OJ-H (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 17.71$ min, $t_{minor} = 20.83$ min. HRMS (DART) ([M+H]⁺) Calcd. For C₁₃H₁₃S⁺: 201.07325, Found: 201.07289.

tert-Butyl 3-((1R,2R)-2-phenylcyclopropyl)-1H-indole-1-carboxylate (3ao) was



obtained through the general procedure using 1.5 equiv of olefin with N-Boc-1H-indole-3-carboxaldehyde 2,4,6triisopropylbenzenesulfonyl hydrazone (**1f**) at 22 °C. Yield:

98%. dr: 98:2. $R_f = 0.34$ (5:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-106.0^\circ$ (c = 1.2, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.12 (s, 1H), 7.61 (d, J = 8.7 Hz, 1H), 7.36 – 7.31 (m, 4H), 7.26 – 7.19 (m, 4H), 2.22 – 2.13 (m, 2H), 1.68 (s, 9H), 1.49 – 1.39 (m, 2H). ¹³C NMR (125



MHz, CDCl₃) δ 149.96, 142.68, 135.68, 131.12, 128.60, 126.03, 125.95, 124.64, 122.80, 122.62, 121.46, 119.40, 115.43, 83.64, 28.39, 25.57, 18.51, 16.11. IR (neat, cm⁻¹): 1730.69, 1452.06, 1374.29, 1255.18, 1157.34, 1088.11,

856.09, 746.20. HPLC analysis (*E*)-isomer: ee = 85%. OD-H (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 6.11$ min, $t_{minor} = 5.66$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₂H₂₄NO₂⁺: 334.18016, Found: 334.17963. The absolute configuration of the major (*E*)-isomer was assigned as (1*R*,2*R*) by X-ray crystallography.

3-((1R,2R)-2-Phenylcyclopropyl)quinoline (3ap) was obtained through the general



procedure using 1.5 equiv of olefin with 3quinolinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1g**) at 22 °C. Yield: 96%. dr: 93:7. $R_f = 0.30$ (3:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-414.2^{\circ} (c = 1.8, CHCl_3)$. ¹H NMR (500 MHz, CDCl_3) δ 8.82 (d, J = 2.3 Hz, 1H), 8.09 (d, J = 8.4 Hz, 1H), 7.81 (d, J = 2.3 Hz, 1H), 7.77 (d, J = 8.1 Hz, 1H), 7.65 (ddd, J = 8.4, 6.9, 1.5 Hz, 1H), 7.53 (ddd, J = 8.2, 6.9, 1.2 Hz, 1H), 7.37 – 7.30 (m, 2H), 7.25 – 7.17 (m, 3H), 2.36 (ddd, J = 8.3, 6.3, 4.5 Hz, 1H), 2.31 (ddd, J = 8.6, 6.5, 4.5 Hz, 1H), 1.65 – 1.58 (m, 2H). ¹³C NMR (125 MHz, CDCl_3) δ 150.55, 146.97, 141.82, 135.47, 131.11, 129.31, 128.71, 128.68, 128.20, 127.38, 126.93, 126.27, 125.97, 28.32, 25.71, 18.08. IR (neat, cm⁻¹): 3025.24, 1602.76, 1494.46, 1342.61, 1126.47, 1073.30, 912.98, 786.17. HPLC analysis (*E*)-isomer: ee = 99%. ID (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 29.44$ min, $t_{minor} = 26.78$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₁₆N⁺: 246.12773, Found: 246.12896.

3-((1S,2S)-2-Chloro-2-phenylcyclopropyl)quinoline (3aq) was obtained through the



general procedure using 1.5 equiv of olefin with 3quinolinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1g**) at 22 °C. Yield: 95%. dr: 99:1. $R_f = 0.38$ (3:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-248.3^{\circ} (c = 2.7, \text{CHCl}_3)$. ¹H NMR (500 MHz, CDCl₃) δ 8.98 (d, J = 2.3 Hz, 1H), 8.15 (d, J = 8.4 Hz, 1H), 8.06 (d, J = 2.2 Hz, 1H), 7.85 (d, J = 8.2 Hz, 1H), 7.71 (ddd, J = 8.4, 6.8, 1.5 Hz, 1H), 7.62 – 7.53 (m, 3H), 7.45 – 7.40 (m, 2H), 7.37 – 7.33 (m, 1H), 2.80 (dd, J = 9.8, 7.8 Hz, 1H), 2.08 (dd, J = 9.9, 6.8 Hz, 1H), 2.00 (dd, J = 7.8, 6.8 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) δ 152.34, 147.31, 142.21, 135.45, 129.96,

129.34, 129.33, 128.90, 128.27, 127.85, 127.77, 127.65, 126.96, 49.85, 30.04, 21.40. IR



(neat, cm⁻¹): 1600.39, 1569.59, 1494.29, 1449.17, 1161.62, 1040.54, 757.12. HPLC analysis (Z)-isomer: ee = 99%. IF (95% hexanes: 5% isopropanol, 0.8 mL/min): t_{major} = 30.71 min, t_{minor} = 27.88 min. HRMS (DART) ([M+H]⁺) Calcd. For C₁₈H₁₅NCl⁺:

280.08875, Found: 280.08893. The absolute configuration of the major (Z)-isomer was assigned as (1*S*,2*S*) by X-ray crystallography.





general procedure using 3.0 equiv of olefin with 3quinolinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1g**) at 22 °C. Yield: 70%. dr: 77:23. $R_f = 0.23$ (1:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-327.6^{\circ} (c = 0.9, CHCl_3)$. ¹H NMR (600 MHz, CDCl₃) δ 8.74 (d, J = 2.3 Hz, 1H), 8.08 (d, J = 8.5 Hz, 1H), 7.79 (d, J = 2.4 Hz, 1H), 7.75 (d, J = 8.7 Hz, 1H), 7.67 (ddd, J = 8.4, 6.8, 1.4 Hz, 1H), 7.56 – 7.51 (m, 1H), 2.72 (ddd, J = 10.3, 6.6, 4.1 Hz, 1H), 2.36 – 2.33 (m, 4H), 1.80 – 1.77 (m, 1H), 1.53 (ddd, J = 8.2, 6.6, 4.4 Hz, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 206.36, 150.19, 147.32, 133.27, 132.01, 129.42, 129.15, 127.92, 127.42, 127.14, 32.58, 31.10, 26.52, 18.91. IR (neat, cm⁻¹): 3004.58, 1697.41, 1494.12, 1354.81, 1178.26, 976.22, 753.39. HPLC analysis (*E*)-isomer: ee = 98%. IE (80% hexanes: 20% isopropanol, 0.8 mL/min): $t_{major} = 28.82$ min, $t_{minor} = 31.10$ min. HRMS (DART) ([M+H]⁺) Calcd. For C₁₄H₁₄NO⁺: 212.10699, Found: 212.10702.

Ethyl (1R,2R)-2-(quinolin-3-yl)cyclopropane-1-carboxylate (3as) was obtained through



the general procedure using 1.5 equiv of olefin with 3quinolinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1g**) at 22 °C. Yield: 90%. dr: 96:4. $R_f = 0.28$ (2:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-231.6^{\circ} (c = 2.5, CHCl_3)$. ¹H NMR (500 MHz, CDCl₃) δ 8.77 (s, 1H), 8.06 (d, J = 8.5 Hz, 1H), 7.78 (d, J = 2.2 Hz, 1H), 7.74 (d, J = 8.2 Hz, 1H), 7.66 (ddd, J = 8.4, 6.8, 1.5 Hz, 1H), 7.52 (ddd, J = 8.0, 6.9, 1.1 Hz, 1H), 4.21 (q, J = 7.6 Hz, 2H), 2.74 – 2.67 (m, 1H), 2.04 (ddd, J = 8.5, 5.3, 4.2 Hz, 1H), 1.78 – 1.69 (m, 1H), 1.45 (ddd, J = 8.6, 6.5, 4.7 Hz, 1H), 1.30 (t, J = 7.1 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 173.01, 150.37, 147.32, 133.11, 132.09, 129.40, 129.10, 127.96, 127.44, 127.07, 61.10, 24.09, 23.90, 16.87, 14.39. IR (neat, cm⁻¹): 2981.08, 1719.87, 1409.27, 1337.94, 1181.43, 1047.72, 752.75. HPLC analysis (*E*)-isomer: ee = 99%. IE (90% hexanes: 10% isopropanol, 0.8 mL/min): $t_{major} = 38.23$ min, $t_{minor} = 35.34$ min. HRMS (DART) ([M+H]⁺) Calcd. For C₁₅H₁₆NO₂⁺: 242.11756, Found: 242.11756.

3-((1S,2R)-2-(Phenylsulfonyl)cyclopropyl)quinoline (3at) was obtained through the



general procedure using 1.5 equiv of olefin with 3quinolinecarboxaldehyde 2,4,6-triisopropylbenzenesulfonyl hydrazone (**1g**) at 22 °C. Yield: 92%. dr: 99:1. $R_f = 0.31$ (1:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-65.0^\circ (c = 2.4, \text{CHCl}_3)$. ¹H NMR (500 MHz, CDCl}_3) δ 8.58 (d, J = 2.3 Hz, 1H), 8.04 (d, J = 8.5 Hz, 1H), 7.97 (d, J = 7.1 Hz, 2H), 7.75 (d, J = 2.3 Hz, 1H), 7.73 – 7.64 (m, 3H), 7.59 (dd, J = 8.4, 7.0 Hz, 2H), 7.52 (ddd, J = 8.2, 6.8, 1.2 Hz, 1H), 3.04 (ddd, J = 10.4, 6.6, 4.6 Hz, 1H), 2.81 – 2.77 (m, 1H), 2.05 – 2.01 (m, 1H), 1.66 – 1.62 (m, 1H). ¹³C NMR (125 MHz, CDCl_3) δ 150.07, 147.48, 140.35, 133.91, 132.98,

130.43, 129.61, 129.57, 129.39, 127.73, 127.66, 127.47, 127.31, 41.88, 21.77, 13.79. IR (neat, cm⁻¹): 1494.32, 1446.39, 1304.88, 1146.07, 1087.44, 968.80, 733.93. HPLC analysis (*E*)-isomer: ee = 99%. OD-H (80% hexanes: 20% isopropanol, 0.8 mL/min): t_{major} = 30.64 min, t_{minor} = 26.95 min. HRMS (DART) ([M+H]⁺) Calcd. For C₁₈H₁₆NO₂S⁺: 310.08963, Found: 310.09007.

(1R,2R)-1,2-Diphenylcyclopropane (3au) was obtained through the general procedure



using 1.5 equiv of olefin with benzaldehyde 2,4,6triisopropylbenzene sulfonyl hydrazone (**1h**) at 22°C. Yield: 97%. dr: 94:6. $R_f = 0.42$ (Hexanes only). $[\alpha]_D^{20} = (-)-364.8^\circ$ (c = 1.0,

CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 7.34 – 7.29 (m, 4H), 7.23 – 7.14 (m, 6H), 2.24 – 2.15 (m, 2H), 1.50 – 1.45 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 142.66, 128.53, 125.92, 125.88, 28.14, 18.32. IR (neat, cm⁻¹): 3028.19, 1603.28, 1498.87, 1450.32, 1030.32, 907.02, 735.59, 695.49. HPLC analysis (*E*)-isomer: ee = 97%. IA (99.5% hexanes: 0.5% isopropanol, 0.8 mL/min): $t_{major} = 6.75$ min, $t_{minor} = 6.01$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₅⁺: 195.11683, Found: 195.11621.

1-Methyl-4-((1R,2R)-2-phenylcyclopropyl)benzene (3av) was obtained through the



general procedure using 1.5 equiv of olefin with 4methylbenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (1i) at 22°C. Yield: 95%. dr: 93:7. $R_f = 0.38$ (Hexanes

only). [α]_D²⁰ = (-)-369.4° (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 7.32 – 7.29 (m, 2H), 7.23 – 7.11 (m, 5H), 7.06 (d, J = 7.9 Hz, 2H), 2.35 (s, 3H), 2.18 – 2.12 (m, 2H), 1.45 – 1.42 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 142.83, 139.59, 135.42, 129.21, 128.51, 125.89, 125.86, 125.80, 27.96, 27.85, 21.11, 18.13. IR (neat, cm⁻¹): 3024.14, 2920.67, 1603.03, 1516.20, 1497.54, 1211.03, 1031.24, 809.20, 750.29, 695.67. HPLC analysis (*E*)isomer: ee = 98%. IA (99.5% hexanes: 0.5% isopropanol, 0.8 mL/min): t_{major} = 6.15 min, t_{minor} = 5.24 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₇⁺: 209.13248, Found: 209.13275.

1-Methyl-3-((1R,2R)-2-phenylcyclopropyl)benzene (3aw) was obtained through the



general procedure using 1.5 equiv of olefin with 3methylbenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (1j) at 22°C. Yield: 92%. dr: 95:5. $R_f = 0.44$ (Hexanes

only). $[\alpha]_D^{20} = (-)-356.4^{\circ} (c = 1.0, CHCl_3)$. ¹H NMR (500 MHz, CDCl₃) δ 7.36 – 7.28 (m, 2H), 7.24 – 7.15 (m, 4H), 7.03 (d, J = 7.6 Hz, 1H), 7.01 – 6.95 (m, 2H), 2.37 (s, 3H), 2.21 – 2.13 (m, 2H), 1.50 – 1.43 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 142.75, 142.61, 138.11, 128.51, 128.45, 126.72, 126.67, 125.91, 125.84, 122.92, 28.11, 28.06, 21.57, 18.24. IR (neat, cm⁻¹): 3026.23, 2919.93, 1604.23, 1497.78, 1113.06, 1031.27, 923.33, 781.91, 695.03. HPLC analysis (*E*)-isomer: ee = 97%. IF (100% hexanes, 0.5 mL/min): $t_{major} = 16.53 \text{ min}, t_{minor} = 19.15 \text{ min}.$ HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₇⁺: 209.13248, Found: 209.13282.

1-Methyl-2-((1R,2R)-2-phenylcyclopropyl)benzene (3ax) was obtained through the



general procedure using 1.5 equiv of olefin with 2methylbenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (1k) at 22°C. Yield: 86%. dr: 95:5. $R_f = 0.41$ (Hexanes

only). [α]_D²⁰ = (-)-192.4° (*c* = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 7.34 – 7.32 (m, 2H), 7.24 – 7.11 (m, 7H), 2.37 (s, 3H), 2.24 – 2.21 (m, 1H), 2.09 – 2.06 (m, 1H), 1.49 (ddd, *J* = 8.7, 6.2, 5.0 Hz, 1H), 1.43 – 1.40 (m, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 143.09, 140.34, 138.08, 129.84, 128.54, 126.21, 126.00, 125.80, 125.69, 26.27, 26.07, 20.04, 16.49. IR (neat, cm⁻¹): 3062.67, 3024.39, 1603.22, 1491.19, 1458.43, 1113.67, 1030.88, 753.29, 696.20. HPLC analysis (*E*)-isomer: ee = 97%. IB (99.9% hexanes: 0.1% isopropanol, 0.8 mL/min): $t_{major} = 8.56$ min, $t_{minor} = 7.45$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₇⁺: 209.13248, Found: 209.13224.

1-Ethyl-2-((1R,2R)-2-phenylcyclopropyl)benzene (3ay) was obtained through the



general procedure using 1.5 equiv of olefin with 2ethylbenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (11) at 22°C. Yield: 57%. dr: 96:4. $R_f = 0.33$ (Hexanes only). $[\alpha]_D^{20}$

= (-)-188.4° (c = 0.5, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 7.33 – 7.30 (m, 2H), 7.23 – 7.15 (m, 6H), 7.13 – 7.07 (m, 1H), 2.75 (q, J = 7.6 Hz, 2H), 2.28 – 2.24 (m, 1H), 2.10 – 2.06 (m, 1H), 1.51 – 1.47 (m, 1H), 1.42 – 1.38 (m, 1H), 1.20 (t, J = 7.6 Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ 143.75, 142.99, 139.63, 128.56, 128.14, 126.36, 126.01, 125.81, 125.53, 26.46, 26.14, 25.56, 16.75, 14.95. IR (neat, cm⁻¹): 3028.71, 2956.21, 2930.71, 2872.49, 1602.83, 1489.77, 1457.98, 1116.52, 752.13, 696.18. HPLC analysis (*E*)-isomer: ee = 89%. IF (100% hexanes, 0.5 mL/min): $t_{major} = 13.21$ min, $t_{minor} = 14.41$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₁₉⁺: 223.14813, Found: 223.14781.

1-Methoxy-4-((1R,2R)-2-phenylcyclopropyl)benzene (3az) was obtained through the



general procedure using 1.5 equiv of olefin with 4methoxybenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (**1m**) at 22°C. Yield: 94%. dr: 94:6. $R_f = 0.26$ (20:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-323.4^\circ$ (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 7.33 -7.27 (m, 2H), 7.22 -7.17 (m, 1H), 7.16 -7.14 (m, 2H), 7.12 -7.07 (m, 2H), 6.88 -6.84

(m, 2H), 3.81 (s, 3H), 2.18 – 2.08 (m, 2H), 1.42 – 1.39 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 158.02, 142.88, 134.67, 128.51, 127.07, 125.86, 125.78, 114.02, 55.48, 27.63, 27.45, 17.92. IR (neat, cm⁻¹): 3001.07, 2833.70, 1603.92, 1514.04, 1462.78, 1245.89, 1035.44, 823.28, 697.15. HPLC analysis (*E*)-isomer: ee = 97%. IA (99.5% hexanes: 0.5% isopropanol, 0.8 mL/min): $t_{major} = 8.37$ min, $t_{minor} = 7.16$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₇O⁺: 225.12739, Found: 225.12767.

1-((1R,2R)-2-Phenylcyclopropyl)-4-(trifluoromethyl)benzene (3ba) was obtained



through the general procedure using 1.5 equiv of olefin with 4-(trifluoromethyl)benzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (**1n**) at 22°C. Yield: 96%. dr: 93:7. $R_f = 0.42$

(Hexanes only). $[\alpha]_D^{20} = (-)-299.2^{\circ} (c = 1.0, CHCl_3)$. ¹H NMR (500 MHz, CDCl₃) δ 7.56 (d, J = 8.0 Hz, 2H), 7.35 – 7.32 (m, 2H), 7.29 – 7.21 (m, 3H), 7.17 (d, J = 7.7 Hz, 2H), 2.25 – 2.22 (m, 2H), 1.59 – 1.49 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 146.93, 141.90, 128.65, 128.14 (q, J = 32.1 Hz), 126.23, 126.05, 125.96, 125.47 (q, J = 3.8 Hz), 124.51 (q, J = 272.2 Hz), 28.82, 27.89, 18.70. ¹⁹F NMR (470 MHz, CDCl₃) δ –62.25. IR (neat, cm⁻¹): 3028.88, 1618.95, 1497.31, 1324.51, 1163.44, 1117.66, 1067.93, 1015.72, 824.57. HPLC analysis (*E*)-isomer: ee = 99%. IA (99.5% hexanes: 0.5% isopropanol, 0.8 mL/min): $t_{major} = 6.70$ min, $t_{minor} = 5.76$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₆H₁₄F₃⁺: 263.10421, Found: 263.10378.

Methyl 4-((1R,2R)-2-phenylcyclopropyl)benzoate (3bb) was obtained through the



general procedure using 1.5 equiv of olefin with 4-(methoxycarbonyl)benzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (10) at 22°C. Yield: 94%. dr: 94:6. $R_f =$ 0.51 (8:1 Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-362.0^\circ$ (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 7.97 (d, J = 8.3 Hz, 2H), 7.32 – 7.29 (m, 2H), 7.22 – 7.14 (m, 5H), 3.91 (s, 3H), 2.26 – 2.19 (m, 2H), 1.57 – 1.50 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 167.17, 148.37, 141.96, 129.89, 128.60, 127.73, 126.16, 125.93, 125.64, 52.10, 29.04, 28.25, 18.98. IR (neat, cm⁻¹): 3025.65, 2950.24, 1716.33, 1608.87, 1434.59, 1277.00, 1181.01, 1111.59, 1017.91. HPLC analysis (E)-isomer: e = 97%. IA (95% hexanes: 5% isopropanol, 0.8 mL/min): $t_{major} = 14.45$ min, $t_{minor} = 8.49$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₁₇O₂⁺: 253.12231, Found: 253.12345.



general procedure using 1.5 equiv of olefin with 3nitrobenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (1p) at 22°C. Yield: 89%. dr: 98:2. $R_f = 0.54$ (8:1

Hexanes/EtOAc). $[\alpha]_D^{20} = (-)-338.2^\circ$ (*c* = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 8.04 (d, J = 7.8 Hz, 1H), 7.98 (s, 1H), 7.52 - 7.41 (m, 2H), 7.33 - 7.31 (m, 2H), 7.22 (t, J = 7.9 Hz)Hz, 1H), 7.16 (d, J = 7.6 Hz, 2H), 2.29 – 2.24 (m, 2H), 1.60 – 1.52 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) δ 148.64, 144.96, 141.48, 132.36, 129.34, 128.68, 126.36, 125.94, 120.92, 120.49, 28.79, 27.52, 18.68. IR (neat, cm⁻¹): 3028.40, 1603.50, 1526.14, 1348.86, 1075.34, 924.33, 867.15. HPLC analysis (E)-isomer: e = 96%. IA (99.5% hexanes: 0.5% isopropanol, 0.8 mL/min): $t_{major} = 13.17 \text{ min}, t_{minor} = 12.28 \text{ min}. \text{HRMS (DART) ([M+H]^+)}$ Calcd. for C₁₅H₁₄NO₂⁺: 240.10191, Found: 240.10197.

1-Bromo-2-((1R,2R)-2-phenylcyclopropyl)benzene (3bd) was obtained through the



general procedure using 1.5 equiv of olefin with 2bromobenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (**1q**) at 22°C. Yield: 89%. dr: 96:4. $R_f = 0.42$ (Hexanes only). $[\alpha]_D^{20} = (-)-125.8^{\circ}$ ($c = 1.0, CHCl_3$). ¹H NMR (600 MHz, CDCl_3) δ 7.57 (d, J = 7.9 Hz, 1H), 7.34 – 7.29 (m, 2H), 7.29 – 7.18 (m, 4H), 7.11 – 7.04 (m, 2H), 2.51 – 2.45 (m, 1H), 2.17 – 2.10 (m, 1H), 1.47 – 1.42 (m, 2H). ¹³C NMR (150 MHz, CDCl_3) δ 142.30, 141.50, 132.66, 128.49, 127.56, 127.49, 127.04, 126.24, 126.21, 126.01, 28.05, 27.01, 17.24. IR (neat, cm⁻¹): 3058.29, 3026.34, 1603.91, 1476.11, 1214.07, 1023.15, 749.96, 695.59. HPLC analysis (*E*)-isomer: ee = 93%. ADH (100% hexanes, 0.5 mL/min): $t_{major} = 14.57$ min, $t_{minor} = 17.12$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₅H₁₄Br⁺: 273.02734, Found: 273.02743.

4-((1R,2R)-2-phenylcyclopropyl)-1,1'-biphenyl (3be) was obtained through the general



procedure using 1.5 equiv of olefin with 4-phenylbenzaldehyde 2,4,6-triisopropylbenzene sulfonyl hydrazone (**1r**) at 22°C. Yield: 90%. dr: 95:5. $R_f = 0.21$ (Hexanes only). $[\alpha]_D^{20} = (-)-399.6^\circ$ (c =

1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 7.62 – 7.59 (m, 2H), 7.55 (d, *J* = 8.3 Hz, 2H), 7.47 – 7.44 (m, 2H), 7.37 – 7.31 (m, 3H), 7.27 – 7.17 (m, 5H), 2.26 – 2.21 (m, 2H), 1.54 – 1.50 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) δ 142.57, 141.87, 141.11, 138.89, 128.88, 128.56, 127.27, 127.19, 127.09, 126.30, 125.93, 125.92, 28.34, 27.95, 18.42. IR (neat, cm⁻¹): 3056.44, 3027.70, 1602.57, 1487.14, 1121.24, 823.97. HPLC analysis (*E*)-isomer: ee = 98%. IF (99.9% hexanes: 0.1% isopropanol, 0.8 mL/min): t_{major} = 22.98 min, t_{minor} = 19.92 min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₁₉⁺: 271.14813, Found: 271.14900.

2-((1R,2R)-2-phenylcyclopropyl)naphthalene (3bf) was obtained through the general



procedure using 1.5 equiv of olefin with 2-naphthaldehyde 2,4,6triisopropylbenzene sulfonyl hydrazone (**1s**) at 22°C. Yield: 91%. dr: 95:5. $R_f = 0.33$ (Hexanes only). $[\alpha]_D^{20} = (-)-336.0^\circ$ (c = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 7.85 – 7.77 (m, 3H), 7.62 (s, 1H), 7.50 – 7.47 (m, 1H), 7.45 – 7.43 (m, 1H), 7.36 – 7.33 (m, 2H), 7.31 (dd, *J* = 8.4, 1.8 Hz, 1H), 7.25 – 7.20 (m, 3H), 2.37 (ddd, *J* = 8.7, 5.9, 4.5 Hz, 1H), 2.30 (ddd, *J* = 8.7, 5.9, 4.5 Hz, 1H), 1.63 – 1.60 (m, 1H), 1.58 – 1.54 (m, 1H). ¹³C NMR (150 MHz, CDCl₃) δ 142.61, 140.11, 133.67, 132.17, 128.57, 128.14, 127.76, 127.46, 126.25, 125.94, 125.23, 124.83, 124.02, 28.42, 28.26, 18.28. IR (neat, cm⁻¹): 3054.46, 3024.95, 1632.09, 1601.30, 1497.28, 1219.67, 813.58, 751.88, 698.47. HPLC analysis (*E*)-isomer: ee = 97%. IA (99.5% hexanes: 0.5% isopropanol, 0.8 mL/min): *t_{major}* = 7.96 min, *t_{minor}* = 6.81 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₁₇⁺: 245.13248, Found: 245.13248.

2.4.5. Mechanistic Studies of Stepwise Radical Mechanism

2.4.5.1. General Procedure for TEMPO Trapping Experiments

An oven-dried Schlenk tube was charged with sulfonyl hydrazone 1 (0.10 mmol, 1.0 equiv), [Co(P1)] (2 mol %) and Cs_2CO_3 (0.20 mmol, 2.0 equiv). The Schlenk tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen 3 times. Under nitrogen atmosphere, TEMPO (0.30 mmol, 3.0 equiv) and anhydrous toluene (1.0 mL) were added. The Schlenk tube was then purged with nitrogen for 1 min and sealed with the Teflon screw cap. The reaction mixture was stirred at 22 °C for 16 h. Following completion of the reaction, the reaction mixture was filtered through a pad of silica gel, concentrated under vacuum and purified by flash column chromatography.

1,1'-((Pyridin-3-ylmethylene)bis(0xy))bis(2,2,6,6-tetramethylpiperidine) (5) Yield: 19%. $R_f = 0.33$ (15:1 Hexanes/EtOAc). ¹H NMR (400 MHz, CDCl₃) δ 8.69 (d, J = 2.0 Hz, 1H), 8.51 (dd, J = 5.0, 1.7 Hz, 1H), 7.84 (d, J = 7.8 Hz, 1H), 7.28 – 7.25 (m, 1H), 6.09 (s,



1H), 1.49 (s, 12H), 1.36 – 1.20 (m, 12H), 0.98 (s, 6H), 0.47 (s, 6H).
¹³C NMR (100 MHz, CDCl₃) δ 149.69, 148.52, 136.41, 136.16, 122.79, 107.20, 60.74, 60.25, 40.69, 40.06, 34.52,

33.32, 21.34, 20.81, 17.28. IR (neat, cm⁻¹): 2974.24, 2930.02, 1467.64, 1376.57, 1362.48, 1132.32, 1084.29, 945.22, 713.19. HRMS (DART) ([M+H]⁺) Calcd. for C₂₄H₄₂N₃O₂⁺: 404.32715, Found: 404.32774.

3-(Bis((2,2,6,6-tetramethylpiperidin-1-yl)oxyl)methyl)quinoline (6) Yield: 12%. R_f =



0.35 (15:1 Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ
9.10 (d, J = 2.0 Hz, 1H), 8.15 (s, 1H), 8.10 (d, J = 8.4 Hz, 1H),
7.86 (d, J = 8.1 Hz, 1H), 7.70 (t, J = 7.6 Hz, 1H), 7.54 (t, J =

7.5 Hz, 1H), 6.26 (s, 1H), 1.55 (s, 12H), 1.28 (s, 12H), 0.99 (s, 6H), 0.45 (s, 6H). ¹³C NMR (125 MHz, CDCl₃) δ 151.73, 147.75, 135.18, 133.50, 129.35, 129.30, 128.28, 127.51, 126.43, 107.86, 60.82, 60.32, 40.77, 40.11, 34.60, 33.55, 21.41, 20.87, 17.32. IR (neat, cm⁻¹): 2973.66, 2929.43, 2870.84, 1572.03, 1466.04, 1362.41, 1132.18, 949.79, 903.44. HRMS (DART) ([M+H]⁺) Calcd. for C₂₈H₄₄N₃O₂⁺: 454.34280, Found: 454.34263.

2.4.5.2. General Procedure for Cyclopropanation of (*E*)- and (*Z*)-β-Deuterostyrenes

An oven-dried Schlenk tube was charged with 2-pyridinecarboxaldehyde tosylhydrazone **1a** (0.1 mmol, 1.0 equiv), [Co(Por)] (2 mol %) and Cs₂CO₃ (0.2 mmol, 2.0 equiv). The Schlenk tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen 3 times. Under nitrogen atmosphere, (*E*) or (*Z*)- β -deuterostyrene **2a**_D (0.15 mmol, 1.5 equiv) was added, followed by the addition of toluene (1.0 mL) as solvent via a gas tight syringe. The Schlenk tube was then purged with nitrogen for 1 min and sealed with the Teflon screw cap. The mixture was stirred at 80 °C for 16 h. Following the completion of the reaction, the reaction mixture was filtrated through a short pad of silica, concentrated under vacuum and purified by preparative TLC.

2.4.6. X-Ray Crystallography

The X-ray diffraction data were collected using Bruker-AXS SMART-APEXII CCD diffractometer (CuK α , $\lambda = 1.54178$ Å). Indexing was performed using *APEX2*²⁷ (Difference Vectors method). Data integration and reduction were performed using SaintPlus.²⁸ Absorption correction was performed by multi-scan method implemented in SADABS.²⁹ Space groups were determined using XPREP implemented in APEX2.²⁰ The structure was solved using SHELXS-97 (direct methods) and refined using SHELXL97 contained in WinGX v1.70.01^{30,31,32} program.

²⁷ Bruker (2012). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.

²⁸ Bruker (**2012**). SAINT. Data Reduction Software.

²⁹ Sheldrick, G. M. (1996). SADABS. University of Gottingen, Germany.

³⁰ Farrugia, L. J. J. Appl. Cryst. **1999**, 32, 837–838.

³¹ Sheldrick, G. M. (**2012** Beta) SHELXL-97. Program for the Refinement of Crystal.

³² Sheldrick, G. M. Acta Cryst. 1990, A46, 467–473.





Table S1. Crystal data and structure refinement for (-)-3b

Identification code	C15H15NO	
Empirical formula	C15 H15 N O	
Formula weight	225.28	
Temperature	173(2) K	
Wavelength	1.54178 ≈	
Crystal system	Orthorhombic	
Space group	P212121	
Unit cell dimensions	$a = 5.7035(5) \approx$	$\alpha = 90\infty$.
	$b = 7.6545(7) \approx$	$\beta = 90\infty$.
	$c = 28.102(3) \approx$	$\gamma = 90\infty$.
Volume	1226.84(19) ≈ ³	
Ζ	4	
Density (calculated)	1.220 Mg/m ³	
Absorption coefficient	0.598 mm ⁻¹	
F(000)	480	
Crystal size	0.360 x 0.180 x 0.120 mm ³	
Theta range for data collection	3.145 to 66.709∞.	
Index ranges	-6<=h<=6, -8<=k<=9, -33<=l<=33	
Reflections collected	25515	
Independent reflections	2148 [R(int) = 0.0263]	
Completeness to theta = 66.709∞	98.8 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7528 and 0.6980	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2148 / 0 / 154	
Goodness-of-fit on F ²	1.064	
Final R indices [I>2sigma(I)]	R1 = 0.0279, wR2 = 0.0710	
R indices (all data)	R1 = 0.0284, wR2 = 0.0717	
Absolute structure parameter	0.05(5)	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.121 and -0.137 e. \approx^{-3}	





Identification code	C14H12BrN	
Empirical formula	C14 H12 Br N	
Formula weight	274.16	
Temperature	173(2) K	
Wavelength	1.54178 ≈	
Crystal system	Orthorhombic	
Space group	P212121	
Unit cell dimensions	$a = 7.3856(5) \approx$	$\alpha = 90\infty$.
	$b = 7.7795(6) \approx$	$\beta = 90\infty$.
	$c = 20.3419(15) \approx$	$\gamma = 90\infty$.
Volume	1168.77(15) ≈ ³	
Z	4	
Density (calculated)	1.558 Mg/m ³	
Absorption coefficient	4.531 mm ⁻¹	
F(000)	552	
Crystal size	0.420 x 0.180 x 0.100 mm ³	
Theta range for data collection	4.347 to 66.624∞.	
Index ranges	-8<=h<=8, -9<=k<=9, -23<=l<=24	
Reflections collected	18297	
Independent reflections	2018 [R(int) = 0.0229]	
Completeness to theta = 66.624∞	98.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7528 and 0.5921	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2018 / 0 / 145	
Goodness-of-fit on F ²	1.105	
Final R indices [I>2sigma(I)]	R1 = 0.0167, wR2 = 0.0467	
R indices (all data)	R1 = 0.0168, wR2 = 0.0469	
Absolute structure parameter	-0.028(5)	
Extinction coefficient	n/a	
Largest diff. peak and hole	$0.168 \text{ and } -0.465 \text{ e.} \approx^{-3}$	

Table 62 Constal data and structure refinement for () 2al





Table S3. Crystal data and structure refinement for (-)-3ao

Identification code	C22H23NO2	
Empirical formula	C22 H23 N O2	
Formula weight	333.41	
Temperature	173(2) K	
Wavelength	$1.54178 \approx$	
Crystal system	Orthorhombic	
Space group	P212121	
Unit cell dimensions	$a = 6.1640(7) \approx$	$\alpha = 90\infty$.
	$b = 11.4688(14) \approx$	$\beta = 90\infty$.
	$c = 25.469(3) \approx$	$\gamma = 90\infty$.
Volume	1800.5(4) ≈ ³	
Z	4	
Density (calculated)	1.230 Mg/m ³	
Absorption coefficient	0.616 mm ⁻¹	
F(000)	712	
Crystal size	0.480 x 0.260 x 0.120 mm ³	
Theta range for data collection	3.471 to 66.600∞.	
Index ranges	-7<=h<=7, -13<=k<=13, -30<=l<=30	
Reflections collected	37857	
Independent reflections	3145 [R(int) = 0.0324]	
Completeness to theta = 66.600∞	99.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7528 and 0.6790	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	3145 / 0 / 226	
Goodness-of-fit on F ²	1.025	
Final R indices [I>2sigma(I)]	R1 = 0.0284, wR2 = 0.0735	
R indices (all data)	R1 = 0.0288, $wR2 = 0.0740$	
Absolute structure parameter	0.08(4)	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.160 and -0.179 e. \approx^{-3}	





ſ	Identification code	C18H14CIN	
	Empirical formula	C18 H14 Cl N	
	Formula weight	279.75	
	Temperature	123(2) K	
	Wavelength	1.54178 ≈	
	Crystal system	Orthorhombic	
	Space group	P212121	
	Unit cell dimensions	$a = 5.8771(3) \approx$	$\alpha = 90\infty$.
		$b = 9.8218(5) \approx$	$\beta = 90\infty$.
		$c = 23.3707(11) \approx$	$\gamma = 90\infty$.
	Volume	1349.04(12) ≈ ³	
	Ζ	4 1.377 Mg/m ³ 2.384 mm ⁻¹ 584	
	Density (calculated)		
	Absorption coefficient		
	F(000)		
	Crystal size	0.520 x 0.300 x 0.140 mm ³	
	Theta range for data collection	3.783 to 66.563∞.	
	Index ranges	-6<=h<=6, -10<=k<=11, -27<=l<=27	
	Reflections collected	7681	
	Independent reflections	2353 [R(int) = 0.0259]	
	Completeness to theta = 66.563∞	99.6 %	
	Absorption correction	Semi-empirical from equivalents	
	Max. and min. transmission	0.7528 and 0.5692	
	Refinement method	Full-matrix least-squares on F ²	
	Data / restraints / parameters	2353 / 0 / 181	
	Goodness-of-fit on F ²	1.029	
	Final R indices [I>2sigma(I)]	R1 = 0.0259, wR2 = 0.0694	
	R indices (all data)	R1 = 0.0260, wR2 = 0.0696	
	Absolute structure parameter	blute structure parameter 0.010(5)	
I	Extinction coefficient	n/a	
	Largest diff. peak and hole	$0.237 \text{ and } -0.197 \text{ e.}^{-3}$	

2.4.7. DFT Calculations

Considering the cost of time and computing resources for the large system with [Co(P9)], the geometry optimizations were performed with the Gaussian 16 at the BP86/lanl2dz level of theory in the gas phase at room temperature. Gas-phase Hessian matrix calculations were applied to the characterization of all minima (without imaginary frequency) and transition states (with only one imaginary frequency).

Thermochemical parameters such as internal energy, enthalpy, entropy, Gibbs free energy and thermal corrections (entropy and enthalpy, 298.15 K, 1 Atm) were obtained from these calculations. To further improve the accuracy of energies, single point energies were carried out at the B3LYP/def2-tzvp level of theory along with Grimme's dispersion correction (D3BJ) and SMD solvation model (in toluene).

Independent Gradient Model (IGM)³³ analysis was performed with Multiwfn³⁴ software package using high quality grid option to generate files for further plotting. The visualization of IGM analysis results were presented with VMD³⁵ visualization software. As shown in Scheme S2, the 3D diagrams of optimized structures were generated with CYLview software.³⁶ The NCI (noncovalent interaction) visual representations of optimized structures were generated with VMD and rendered with Tachyon.³⁷

³³ Lefebvre, C.; Rubez, G.; Khartabil, H.; Boisson, J.-C.; Contreras-García, J.; Hénon, E. *Phys. Chem. Chem. Phys.* **2017**, *19*, 17928–17936.

³⁴ Lu, T.; Chen, F. J. Comput. Chem. **2012**, 33, 580–592.

³⁵ Stone, J. E. Master's Thesis, University of Missouri, **1998**.

Scheme S1. Calculated Energy Diagram for [Co(P9)]-Catalyzed Radical Cyclopropanation of Styrene with 2-Pyridyldiazomethane



Scheme S2. Optimized Structure Models and NCI Visual Representation of Intermediates and Transition States










Intermediate A

A_[Co(II)(P9)] Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.446803 Hartree H_corr: 1.724596 Hartree SCF: -6888.38 Hartree S: 584.663 Cal/Mol-Kelvin H: -6886.65 Hartree G: -6886.93 Hartree

С	0.45453300	0.59833800	2.92526100
Ν	0.54431300	-0.38619100	1.91803000
Н	0.70270100	0.58058900	5.17005600
С	0.84728900	-1.57778600	2.61219600
С	0.99378700	-1.31621100	4.03601300
Н	1.24931300	-2.07281900	4.77625900
С	0.71910800	0.02050600	4.23591100
С	0.88265900	-2.86590800	2.05348500
Ν	0.30133100	-2.10742500	-0.26681500
С	-0.02387500	-2.83312800	-1.43287000
С	0.00157500	-4.26642200	-1.17057400
Н	-0.22957100	-5.03257300	-1.90896200
С	0.38563800	-4.42702100	0.14283500
Н	0.52367900	-5.35342700	0.69818300
С	0.55992400	-3.09630500	0.70589000
С	0.18577400	1.96713800	2.71984200
Н	0.08238800	5.06271400	-0.75069900
С	0.10799800	4.13166800	-0.18663700
С	0.22260900	2.80945400	-0.78531500
Н	-0.07461600	4.71056700	1.95738900
Ν	0.23772100	1.81268500	0.21287200
С	0.12943800	2.52459400	1.42702100
С	0.02620700	3.95622700	1.17810200
С	-0.25691700	-2.28431200	-2.71035100
Н	0.06146800	1.80345000	-4.91702400
С	0.03537900	1.03586000	-4.14532400
С	-0.17387900	-0.31962800	-4.29306700
Η	-0.34010700	-0.88846900	-5.20761800
С	-0.12328300	-0.90406400	-2.95903600
N	0.10530000	0.09324200	-1.98581800
С	0.17405500	1.29711300	-2.72021500

С	0.24676100	2.58883400	-2.17217700
Co	0.30272300	-0.14744900	-0.03162600
С	0.22270900	3.78038300	-3.08707900
С	1.39852400	4.53231900	-3.36757300
С	-0.99382400	4.22311000	-3.67217200
С	1.36768500	5.67720100	-4.20054000
С	-1.05063100	5.35070100	-4.52694600
С	0.13786400	6.07050300	-4.77744800
Η	0.10914300	6.94955100	-5.43104900
С	1.15893900	-4.04829400	2.93957500
С	0.14256900	-4.60155400	3.76567800
С	2.42916100	-4.68958500	2.94798500
С	0.36957500	-5.74060000	4.57644300
С	2.68460800	-5.83132800	3.74667900
С	1.64611700	-6.34307700	4.55611000
Н	1.82973000	-7.23552900	5.16481200
С	-0.06214400	2.85747700	3.91328700
С	1.00586500	3.51477700	4.59714300
С	-1.40451300	3.07287700	4.36097500
С	0.74362800	4.34691200	5.71963700
С	-1.67070600	3.89834400	5.48620100
С	-0.58969700	4.51777300	6.14697900
Н	1.57500500	4.84533700	6.21976300
Н	-2.70354800	4.04741800	5.80508100
Н	-0.79164200	5.15692400	7.01415600
С	-0.63158200	-3.23199700	-3.82439000
С	-2.00575500	-3.54285900	-4.06420400
С	0.37078900	-3.90519600	-4.58989600
С	-2.36702700	-4.55273700	-4.99620600
С	0.01429800	-4.90950200	-5.52928500
С	-1.34994100	-5.22412400	-5.70567900
Η	-3.41989600	-4.80423100	-5.13206500
Η	0.80007200	-5.42644000	-6.08230600
Н	-1.62696100	-6.00787000	-6.41999600
N	2.32728800	3.30688300	4.09648400
Η	2.39226300	2.65149500	3.30969000
N	1.72425700	-3.51096600	-4.37410800
Η	1.84466200	-2.72110900	-3.73075800
N	-2.98532800	-2.77981800	-3.36236100
Η	-2.63235400	-1.93966000	-2.89204400
Ν	-2.44569400	2.44709300	3.61470500
Η	-2.12384300	1.85208500	2.84385400
С	3.51015900	3.91771300	4.51010900

С	2.88496800	-4.03863100	-4.93941600
С	-4.34793300	-3.05038600	-3.21146500
С	-3.82250900	2.66222400	3.69515700
0	-4.90658900	-4.12249900	-3.60287600
0	-4.38726000	3.38638800	4.57426100
0	2.91736600	-5.05932500	-5.69781500
0	3.58458900	4.77810500	5.44481600
С	-4.60116400	2.01629700	2.57442800
Н	-4.12655300	1.20693600	2.00836700
С	-5.11273600	-1.94553900	-2.51983200
Н	-4.72674800	-0.92479000	-2.62211200
С	-5.84374400	-2.24989400	-1.19592800
Н	-5.89916900	-1.44525900	-0.45662400
Н	-5.77401500	-3.26936400	-0.80504700
С	-6.64482400	-2.08951100	-2.50974700
Н	-7.03567400	-3.01296200	-2.94505400
С	-5.52947200	2.94927200	1.77500100
Н	-5.57795800	3.99352100	2.09469800
Н	-5.62558500	2.75802100	0.70288800
С	-6.13151400	1.91165200	2.75158900
Н	-6.54231000	2.29877000	3.68668700
С	-7.49491200	-0.87953500	-2.69158200
С	-6.84635100	0.76231900	2.12986100
0	-6.42026100	0.05957900	1.17663000
0	-7.12061200	0.32092300	-2.65619700
0	-8.10708000	0.60001800	2.71347100
0	-8.82421000	-1.26702700	-2.89873400
С	-9.07868800	-0.31649200	2.02755800
Η	-8.55420000	-0.82147900	1.19792800
Η	-9.38011000	-1.05805600	2.78785300
С	-10.26768900	0.53683200	1.55157800
Η	-10.77102700	0.97088100	2.43668200
Η	-9.87576400	1.38881700	0.96430900
С	-9.84635500	-0.17478200	-2.99195600
Η	-10.68669500	-0.66215300	-3.51467100
Η	-9.43222300	0.64399900	-3.60806300
С	-10.24894200	0.32924700	-1.59422800
Η	-9.37472600	0.83178200	-1.14096900
Η	-11.03047200	1.10659700	-1.73356500
С	-11.30276100	-0.25932100	0.70793700
Н	-11.68168800	-1.11629300	1.30292500
Η	-12.17535400	0.40261700	0.53060500
С	-10.78670000	-0.78547600	-0.66126800

Η	-9.99548800	-1.54549500	-0.51260900
Н	-11.62122100	-1.31176900	-1.16943300
С	4.74507700	3.48661500	3.75580400
Н	4.65432100	2.68717400	3.01304400
С	4.13696600	-3.27208100	-4.59453200
Н	4.03581200	-2.34764500	-4.01717500
С	5.28645500	-3.33332100	-5.60286200
Н	5.89930300	-2.43678300	-5.71730500
Н	5.10722600	-3.93837400	-6.49596200
С	6.07771100	3.61240100	4.49947600
Н	6.84844600	2.87359600	4.27089300
Н	6.01636700	3.97462800	5.52941700
С	5.74982300	4.64294100	3.40885400
Н	5.43593700	5.63641700	3.74001300
С	5.41860000	-4.11595100	-4.28338800
Н	5.30236200	-5.20147100	-4.33590800
С	6.48105500	4.64539900	2.10922800
С	6.36100900	-3.64743100	-3.22702400
0	6.99962800	3.38364500	1.78888300
0	6.43303300	-2.24719500	-3.19104300
0	7.03130500	-4.39308500	-2.46877700
0	6.65058800	5.65689100	1.38049400
С	7.89458500	3.32081500	0.58403800
Η	7.31822100	3.64907900	-0.30056300
Η	8.72479400	4.03209300	0.74951500
С	8.37522400	1.87069200	0.46950900
Η	9.26001800	1.85038700	-0.19991500
Η	8.72434300	1.53467900	1.46618600
С	7.39776300	-1.63313100	-2.22496800
С	7.29642800	0.90354100	-0.07570100
Η	6.34083800	1.07834100	0.45717900
Η	7.10626900	1.14275500	-1.14300900
С	7.68973400	-0.58934300	0.07361100
Η	7.59854700	-0.88255400	1.13722900
Η	8.76197700	-0.72329800	-0.18862600
С	6.81992000	-1.53624800	-0.80246500
Η	5.78314400	-1.14851900	-0.85224000
Η	6.77384100	-2.54656100	-0.36092800
Η	7.58504300	-0.63689100	-2.66056500
Η	8.32645600	-2.23354100	-2.23113100
Η	2.27919900	6.25130700	-4.38790700
Η	-1.99908300	5.64669200	-4.98452800
Η	3.66299600	-6.31841700	3.72037800

Η	-0.43868700	-6.15812700	5.18321700
0	2.56044900	4.07333600	-2.70659300
0	-2.11939700	3.40233100	-3.42673000
0	3.37743700	-4.13756900	2.05691100
0	-1.13015300	-3.99836000	3.63148000
С	3.83934400	4.65591000	-2.96651900
С	4.43915200	4.57439500	-4.24492100
С	4.52543700	5.20623100	-1.86240000
С	5.75078600	5.07823500	-4.41430500
Н	3.89726300	4.11765100	-5.08018300
С	5.84091600	5.69871700	-2.04102100
Н	4.03445600	5.23877200	-0.88488700
С	6.45182600	5.64162600	-3.31838900
Н	6.22561800	5.02224100	-5.40080200
Η	6.37251200	6.11780500	-1.17920000
Н	7.46725600	6.02943200	-3.45852800
С	-3.39490200	3.93438700	-3.07816400
С	-3.59883800	5.24598200	-2.58678700
С	-4.46195400	3.01122500	-3.16820100
С	-4.90558800	5.62966600	-2.19863900
Н	-2.76473400	5.94821300	-2.49193600
С	-5.75741700	3.40230100	-2.76291300
Η	-4.26488500	1.99996400	-3.53618500
С	-5.98531300	4.71780200	-2.28461700
Η	-5.07159600	6.64431800	-1.81737100
Η	-6.56704500	2.66670100	-2.81049600
Η	-6.99060100	5.02592600	-1.97527700
С	4.75907100	-4.49153200	2.13421500
С	5.50146600	-4.33867600	3.32938900
С	5.37757800	-4.88371500	0.92764300
С	6.88992000	-4.61019000	3.30763500
Η	5.00772500	-4.00610400	4.24880900
С	6.76998200	-5.13908400	0.91505900
Η	4.77451000	-4.97440200	0.01913300
С	7.52630400	-5.01160700	2.10617200
Η	7.47417400	-4.49742400	4.22845500
Η	7.24941900	-5.42201900	-0.02789700
Η	8.60309300	-5.21477700	2.09679600
С	-2.09531200	-4.04946500	4.67733200
С	-3.43978000	-4.19450800	4.26864200
С	-1.75743900	-3.85273600	6.03777600
С	-4.46393300	-4.15253800	5.24275900
Η	-3.66128600	-4.32899000	3.20523400

С	-2.79258400	-3.81820700	7.00096100
Н	-0.71306900	-3.71726900	6.33744200
С	-4.14583300	-3.96944000	6.61095800
Η	-5.50885500	-4.26257400	4.93124700
Н	-2.53909800	-3.66129500	8.05578600
Н	-4.94211600	-3.93689400	7.36269600

A_{1a},

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.067656 Hartree H_corr: 0.108155 Hartree SCF: -396.0115859 Hartree S: 85.238 Cal/Mol-Kelvin H: -395.9034309 Hartree G: -395.9439299 Hartree

Cartesian Coordinates:

С	2.57926300	0.43134200	0.00000700
С	1.56870200	1.42638700	0.00000300
Ν	-3.48625600	0.61552000	0.00004900
С	2.17819700	-0.92610900	-0.00013800
С	0.21418400	1.03738700	-0.00000700
Ν	-2.53428400	-0.09550900	-0.00005700
Η	3.64119200	0.69633600	0.00001400
Η	1.83528600	2.48929500	-0.00007500
С	-0.10042800	-0.35681900	0.00029700
Ν	0.87594700	-1.33194500	0.00005600
С	-1.46813200	-0.87918300	-0.00022700
Η	2.92233500	-1.73186400	0.00003600
Η	-0.58210000	1.79040200	-0.00018000
Η	-1.63528000	-1.95865900	0.00025200

Intermediate B

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.541374 Hartree H_corr: 1.834124 Hartree SCF: -7284.429555 Hartree S: 616.146 Cal/Mol-Kelvin H: -7282.595431 Hartree G: -7282.888181 Hartree

Ν	3.25355500	1.58381100	0.79717700
Ν	3.02586800	0.49842300	0.38144600
С	-0.17518400	2.29141200	1.97752700
Ν	-0.24465300	0.91635200	1.66122700
Η	-0.44785800	3.49031500	3.87100000
С	-0.54340100	0.29023500	2.89197100
С	-0.66453700	1.27533100	3.95551900
Η	-0.89284200	1.03743200	4.99309800
С	-0.43952700	2.51223200	3.39215300
С	-0.72068100	-1.08805800	3.09569000
Ν	-0.48025900	-1.69299100	0.67154900
С	-0.59905300	-2.92881800	0.00588200
С	-0.83262800	-4.00790700	0.95630200
Η	-0.94831600	-5.05510000	0.68095300
С	-0.86399700	-3.43811500	2.20976100
Н	-1.01102000	-3.92544600	3.17253700
С	-0.66761400	-2.00704000	2.03576200
С	0.11250600	3.33740100	1.07954900
Η	1.17846700	4.03757100	-3.38710000
С	0.90802300	3.56472900	-2.44468000
С	0.67845000	2.13747800	-2.27354800
Η	0.79158100	5.21479000	-0.95298400
Ν	0.35413600	1.84392600	-0.93143200
С	0.37744400	3.09980100	-0.28121700
С	0.71556300	4.16109700	-1.21889200
С	-0.50956100	-3.13899200	-1.38434600
Η	0.62993500	-0.88406000	-5.28457900
С	0.39292800	-1.11196000	-4.24770200
С	-0.00293500	-2.31676600	-3.70936700
Η	-0.14341100	-3.27280200	-4.21205100
С	-0.18930000	-2.10403500	-2.28078700
Ν	0.05701600	-0.75613300	-1.93880200
С	0.41586300	-0.14174300	-3.16191700
С	0.73662600	1.21580600	-3.33648100
Co	-0.05803000	0.06903600	-0.13125200
С	1.14551800	1.71516400	-4.69407600
С	2.48388900	2.16387900	-4.91840100
С	0.25935000	1.76833200	-5.81070200
С	2.91305900	2.66858400	-6.16854600
С	0.67377400	2.24981800	-7.07443900
С	1.99771100	2.70571500	-7.24417700
Η	2.32131500	3.09486800	-8.21575000

С	-1.05900400	-1.57669000	4.47729000
С	-2.39613200	-1.94051400	4.81323100
С	-0.09217200	-1.65939700	5.51565800
С	-2.75529500	-2.38057900	6.11017700
С	-0.42520600	-2.07610400	6.82848200
С	-1.75903900	-2.43986900	7.10910700
Η	-2.02826600	-2.76336400	8.12075600
С	0.10234800	4.75329200	1.60453500
С	1.28268300	5.36605500	2.13467500
С	-1.12317700	5.49547500	1.61536800
С	1.22397200	6.67687200	2.68712500
С	-1.18133300	6.79986700	2.17541000
С	-0.00477600	7.36537200	2.70440000
Н	2.13260400	7.12187800	3.09285500
Н	-2.13329000	7.33201400	2.19006200
Н	-0.04444800	8.37102100	3.13843800
С	-0.80666200	-4.52237200	-1.90449300
С	-2.17223500	-4.93612500	-2.01532000
С	0.22743600	-5.44577600	-2.25130700
С	-2.50703200	-6.26157600	-2.39403800
С	-0.10744500	-6.76984700	-2.65372100
С	-1.46306400	-7.15833600	-2.70332800
Η	-3.55611200	-6.56016700	-2.43670700
Η	0.69260600	-7.45963900	-2.92498000
Η	-1.71037100	-8.18386100	-3.00099000
N	2.50959700	4.63060900	2.08489300
Η	2.46508800	3.70155500	1.64535600
N	1.58222500	-4.99674300	-2.17785500
Η	1.75584500	-4.09730600	-1.69354900
N	-3.16547500	-3.93984500	-1.78776700
Η	-2.80696800	-2.98023000	-1.73683700
N	-2.27152300	4.88540500	1.02333400
Η	-2.09989700	3.98079500	0.57081600
С	3.75809400	5.04783000	2.55220300
С	2.69965600	-5.61732200	-2.74336400
С	-4.53743300	-4.08845800	-1.58347500
С	-3.56768900	5.39494300	0.93802000
0	-5.13341100	-5.20529500	-1.47034500
0	-3.95487500	6.47555000	1.48780400
0	2.68488500	-6.76102100	-3.30410800
0	3.97152900	6.16132000	3.13355300
С	-4.50869800	4.58338300	0.08199000
Η	-4.15395000	3.63787500	-0.34294900

С	-5.25271900	-2.76191500	-1.49677700
Н	-4.77486300	-1.91050600	-1.99533200
С	-6.06318900	-2.40917900	-0.23475500
Н	-6.04415700	-1.36375200	0.08567700
Н	-6.11962800	-3.15342600	0.56375900
С	-6.78881400	-2.76165400	-1.55344800
Н	-7.28094200	-3.73739600	-1.58107500
С	-5.51150700	5.38937200	-0.75582900
Н	-5.45896200	6.47694400	-0.65926500
Н	-5.78785600	4.98560800	-1.73344100
С	-6.01624800	4.62742700	0.48250200
Н	-6.26264300	5.21416300	1.36981900
С	-7.44554100	-1.66425100	-2.32518700
С	-6.83024200	3.40563100	0.23127800
0	-6.66886500	2.60715000	-0.73048900
0	-6.86266900	-0.67794700	-2.84452900
0	-7.83263200	3.25748500	1.19259600
0	-8.82459300	-1.86203100	-2.39223100
С	-8.79442300	2.12338700	0.97241700
Н	-8.23366100	1.26791100	0.55742800
Н	-9.16022900	1.89051100	1.98643200
С	-9.93094900	2.57179700	0.03778100
Η	-10.41413400	3.46671200	0.47569700
Η	-9.49521800	2.87906800	-0.93103000
С	-9.61757000	-0.81228700	-3.12026700
Η	-10.56637200	-1.33182000	-3.33701600
Η	-9.08711600	-0.56912400	-4.05887200
С	-9.83419500	0.45153200	-2.26927000
Η	-8.86430100	0.96358400	-2.13250600
Η	-10.47507900	1.13196900	-2.87083800
С	-11.00666600	1.47257500	-0.18912800
Η	-11.45881100	1.19555400	0.78627600
Η	-11.82304900	1.92090100	-0.79204300
С	-10.51421600	0.18150600	-0.90271400
Η	-9.81661300	-0.38103500	-0.25181400
Η	-11.38921700	-0.48416100	-1.05464200
С	4.90037300	4.08526300	2.33070500
Η	4.70222800	3.14339200	1.81449200
С	3.96610500	-4.80314100	-2.67683400
Η	3.92518900	-3.83228100	-2.17522700
С	4.94145100	-4.95930700	-3.84821300
Η	5.51095800	-4.07471000	-4.14123800
Η	4.63723200	-5.64763600	-4.64161600

С	6.02283200	4.11176500	3.37866500
Η	6.54559000	3.17249400	3.57191100
Н	5.87376000	4.78644500	4.22623300
С	6.27474100	4.78412100	2.02732400
Н	6.21360800	5.87565900	1.99741900
С	5.29594800	-5.61119200	-2.50120800
Н	5.18205600	-6.69636700	-2.43348700
С	7.21858500	4.21718300	1.02206100
С	6.38830300	-5.05488900	-1.65441500
0	7.56377300	2.88297500	1.28483600
0	6.50209300	-3.66221000	-1.81106500
0	7.14076100	-5.72399800	-0.90346000
0	7.69504000	4.85876500	0.04948700
С	8.62929200	2.30851300	0.39604300
Н	8.26619100	2.33548600	-0.64811700
Н	9.51692300	2.96273900	0.47736300
С	8.91042800	0.88125900	0.87462900
Н	9.85610300	0.54835900	0.39745000
Н	9.09394300	0.89402400	1.96742100
С	7.58696200	-2.98115400	-1.04233900
С	7.78652500	-0.12239100	0.52495900
Н	6.83003000	0.22963600	0.95943800
Н	7.64564200	-0.12362400	-0.57604600
С	8.07830400	-1.56285300	1.02154000
Η	7.97991700	-1.59539400	2.12465800
Η	9.13505700	-1.82780700	0.80030600
С	7.13566000	-2.62356200	0.38586700
Η	6.10292100	-2.22467500	0.34873300
Η	7.11156700	-3.54486300	0.99557700
Η	7.79425300	-2.08020700	-1.64371600
Η	8.47366600	-3.64129100	-1.02303200
Η	3.93892000	3.02498600	-6.29672900
Η	-0.05127300	2.25702600	-7.89318700
Η	0.33671100	-2.10098400	7.61221900
Η	-3.78914700	-2.65729100	6.33394600
0	3.33460900	2.09636700	-3.78674900
0	-1.03864400	1.20347700	-5.77110000
0	1.20936800	-1.20997700	5.17151600
0	-3.33145900	-1.82645200	3.75883700
С	4.74551900	2.28643800	-3.90891900
С	5.53561600	1.45262800	-4.73514600
С	5.32867300	3.26400400	-3.07343700
С	6.93969400	1.62752200	-4.73399600

Η	5.06372700	0.68297400	-5.35530500
С	6.73565200	3.42261200	-3.07191000
Η	4.68420700	3.87844100	-2.43688600
С	7.54146500	2.61032400	-3.90829500
Η	7.56146400	0.99034000	-5.37370100
Н	7.19248000	4.17403600	-2.41741200
Н	8.62947600	2.74213900	-3.91649100
С	-2.04657800	1.65677300	-4.86719100
С	-2.06921400	2.96353300	-4.33186500
С	-3.08884400	0.73600800	-4.61532100
С	-3.15383700	3.33661900	-3.50076200
Н	-1.26994900	3.67558300	-4.56284300
С	-4.16874000	1.12502800	-3.79152300
Н	-3.03853600	-0.26017600	-5.06732400
С	-4.20096300	2.42462600	-3.22338600
Н	-3.17669900	4.34948100	-3.07958900
Н	-4.98947000	0.42793400	-3.58909800
Η	-5.04023100	2.70266400	-2.57579100
С	2.34954500	-1.61006400	5.92128600
С	3.31570300	-0.60732600	6.16446500
С	2.57422400	-2.95445800	6.30750900
С	4.52404800	-0.95552500	6.81316000
Η	3.10734100	0.42009000	5.84902800
С	3.78346200	-3.28572000	6.96297400
Η	1.82498000	-3.72366900	6.09375500
С	4.76130500	-2.29215900	7.21864800
Η	5.27440100	-0.18076600	7.00742700
Η	3.96231900	-4.32345700	7.26736000
Η	5.69522700	-2.55688400	7.72638700
С	-4.70613300	-2.13549900	4.00746200
С	-5.19216800	-3.40738300	3.63387400
С	-5.55965400	-1.13778300	4.52854500
С	-6.57094400	-3.68708600	3.79447100
Η	-4.50228700	-4.14873100	3.21770400
С	-6.93556100	-1.43035300	4.68611900
Η	-5.15075200	-0.15649600	4.79151400
С	-7.44261400	-2.70295800	4.32291100
Η	-6.96064300	-4.66964500	3.50521900
Η	-7.60821900	-0.66429000	5.08848900
Η	-8.50881400	-2.92424400	4.44539700
С	2.99245400	-1.91574100	0.73082900
С	3.42946500	-1.85590600	2.08704800
С	3.69842300	-3.05411300	2.77479600

H3.54292400-0.891533002.59335700C3.05448100-4.265278000.77262600C3.51463200-4.289305002.10678400H4.03751200-3.025631003.81519800H2.87964700-5.199038000.22645600H3.71642300-5.244101002.60118900N2.79679300-3.115193000.07811500C2.73965800-0.71174100-0.07138400H2.44167400-0.78844300-1.12057300

Transition State TS1

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -251.8937 cm⁻¹ G_corr: 1.532815 Hartree H_corr: 1.83204 Hartree SCF: -7284.38968422 Hartree S: 629.772 Cal/Mol-Kelvin H: -7282.557644 Hartree G: -7282.856869 Hartree

С	0.13185400	-2.30271800	1.99444800
N	0.00686100	-1.81576200	0.67215200
Н	-0.13402700	-4.32252000	2.96338600
С	-0.34757400	-2.95377000	-0.08529500
С	-0.43441900	-4.13216400	0.76373800
Н	-0.69590700	-5.12823900	0.41072200
С	-0.15476200	-3.72834300	2.05082600
С	-0.66019000	-2.96705400	-1.45627500
N	-0.57926800	-0.46905200	-1.69054600
С	-0.88086400	0.37442600	-2.78030300
С	-1.22773300	-0.40882500	-3.95781800
Η	-1.49725900	0.01927000	-4.92209000
С	-1.13839400	-1.73520600	-3.59828400
Η	-1.32574900	-2.61629700	-4.20978500
С	-0.75630500	-1.77709300	-2.19549700
С	0.49308700	-1.55251700	3.12680300
Η	1.92826700	2.72201800	4.26560400
С	1.56248600	1.85915600	3.71110400
С	1.16243100	1.88958600	2.31302800
Η	1.47917400	0.17627600	5.17012200
N	0.72657000	0.61682300	1.88714900

С	0.82725200	-0.18903900	3.04079000
С	1.32952100	0.58032400	4.16979900
С	-0.85678400	1.78119600	-2.77339400
Н	0.52187200	5.32068300	-0.09803200
С	0.26558700	4.32538900	-0.45756300
С	-0.32996200	3.97030600	-1.64833700
Н	-0.65896000	4.61154600	-2.46484500
С	-0.42837300	2.51766100	-1.65539900
Ν	0.04547800	1.97856900	-0.44252600
С	0.50431600	3.09886100	0.28595900
С	1.09106700	3.07105800	1.55982800
Co	0.21101400	0.06610700	0.04393400
С	1.56622100	4.35783300	2.17248100
С	2.91611200	4.78279700	2.04780500
С	0.68005400	5.19659000	2.90257700
С	3.37227000	5.99988200	2.61154300
С	1.11085500	6.41098200	3.49024300
С	2.45888200	6.80109100	3.33139900
Н	2.80293100	7.74143100	3.77635400
С	-1.01792300	-4.27877200	-2.09971500
С	-2.37045100	-4.58593100	-2.43445600
С	-0.05218400	-5.29510300	-2.34756300
С	-2.74414500	-5.82605400	-3.00700300
С	-0.40145500	-6.55340200	-2.89585500
С	-1.74992600	-6.80247700	-3.22789500
Η	-2.02962800	-7.77058700	-3.65795700
С	0.45919700	-2.21243900	4.48138900
С	1.60129500	-2.86405500	5.03963400
С	-0.76348800	-2.18406300	5.22639900
С	1.51228100	-3.48128700	6.31988100
С	-0.84983000	-2.79105200	6.50654700
С	0.29372200	-3.43154700	7.02795300
Η	2.38954400	-3.98518800	6.72709200
Η	-1.79188700	-2.74878000	7.05599700
Η	0.23482700	-3.90568100	8.01446800
С	-1.25157300	2.50936000	-4.03020500
С	-2.63468200	2.66510000	-4.35765600
С	-0.25898600	3.03528500	-4.91744000
С	-3.02505300	3.25660000	-5.58858500
С	-0.65188300	3.64912300	-6.13923700
С	-2.02350200	3.73360800	-6.45971700
Η	-4.08529200	3.33088200	-5.83615500
Η	0.11586900	4.05111500	-6.80176000

Η	-2.31919700	4.19407600	-7.40947600
Ν	2.81140200	-2.85638400	4.28065800
Н	2.79790800	-2.32213000	3.39991600
Ν	1.11146000	2.92804900	-4.53099100
Н	1.33302200	2.30443400	-3.73334100
Ν	-3.58773900	2.26867700	-3.37191800
Η	-3.20314400	2.04702300	-2.44683600
N	-1.87474700	-1.53088100	4.61605300
Η	-1.69356700	-1.16517800	3.67536600
С	4.00997400	-3.49902700	4.59825200
С	2.19597600	3.62447200	-5.07114400
С	-4.97861500	2.18807500	-3.48382700
С	-3.15215800	-1.30434100	5.12628300
0	-5.60853000	2.28044500	-4.58141700
0	-3.54515000	-1.66757200	6.28015000
0	2.13989400	4.36833600	-6.10333400
0	4.18500800	-4.22434100	5.63052300
С	-4.05568500	-0.55543700	4.17785200
Н	-3.68876800	-0.34731700	3.16633800
С	-5.66682100	2.01792600	-2.14014700
Н	-5.56991200	2.87791100	-1.46431300
С	-5.79460200	0.65428300	-1.44190000
Н	-5.76259000	0.63566000	-0.34774500
Η	-5.39895300	-0.23497300	-1.94207900
С	-7.03036000	1.32111400	-2.11238800
Η	-7.38413900	0.88898700	-3.05130000
С	-5.01541100	0.48123900	4.77511400
Η	-4.98354400	0.61267200	5.86010000
Η	-5.22738400	1.36659500	4.16965800
С	-5.57623100	-0.86572700	4.27912000
Η	-5.87352500	-1.58756200	5.04279700
С	-8.08931400	1.89137600	-1.23312100
С	-6.36949300	-0.86611600	3.01685000
0	-6.10094600	-0.20758300	1.97886800
0	-7.91109400	2.67521000	-0.26665200
0	-7.48730200	-1.69551900	3.15248700
0	-9.35251700	1.42447000	-1.62537400
С	-8.49006900	-1.71401800	2.03665200
Η	-8.09365800	-1.11289200	1.20036100
Η	-8.57209700	-2.77295000	1.73329800
С	-9.82235700	-1.17693100	2.58502900
Н	-10.08544000	-1.75485300	3.49201200
Η	-9.68147800	-0.12765200	2.90697600

С	C -10.51624800	1.88393100	-0.79896800
Η	I -11.37782800	1.73175800	-1.47128600
Η	I -10.38424500	2.95832500	-0.57666200
С	2 -10.64144600	1.06402400	0.49737900
Η	I -9.75496900 1	.27048600	1.12441300
Η	I -11.52358000	1.44927600	1.05212000
С	c -10.98770300 -	1.27930500	1.56134000
Η	I-11.14214000 -	2.34617400	1.29552400
Η	I-11.91769800 -	0.94922900	2.06844100
С	C-10.80602800 -	0.45682900	0.25533300
Η	I -9.93406400 -(0.82854500	-0.31736600
Η	I-11.69172300 -	0.62631400	-0.39206800
С	5.13005800 -3	.28748900	3.60964800
Η	I 4.95934000 -2	2.61109400	2.76815100
С	3.48113100 3	.44723000	-4.30176700
Η	I 3.45122700 2	.82678100	-3.40106600
С	2 4.47752500 4	.60756100	-4.33481000
Η	I 5.06827400 4	.78550800	-3.43371300
Η	I 4.17822300 5	.48390200	-4.91670500
С	C 6.08513300 -4	.46873800	3.38581600
Η	I 6.51739400 -4	.57335400	2.38803900
Η	I 5.86037500 -5	5.38775800	3.93432100
С	C 6.56378200 -3	.28339500	4.23660100
Η	I 6.58315800 -3	3.43687300	5.31902600
С	c 4.79550500 3	.34600400	-5.15210400
Η	I 4.66472900 3	.41588200	-6.23544500
С	7.62068300 -2	.35290900	3.74851900
С	5.86193200 2	.39431600	-4.72946200
0	0 7.64129200 -2	2.23886700	2.34746500
0	0 6.02886900 2	.37930100	-3.33354800
0	0 6.54659200 1	.68573400	-5.50954400
0	8.43834300 -1	.75014900	4.48801700
С	8.76443500 -1	.41255300	1.79120000
Η	I 8.63837500 -0	0.37177600	2.14412200
Η	I 9.70945300 -1	.80856100	2.20633500
С	2 8.70804400 -1	.52155400	0.26440000
Η	I 9.67111800 -1	.13865300	-0.13453000
Η	I 8.65325100 -2	2.59251300	-0.01636800
С	2 7.05129400 1	.42711300	-2.79198600
С	2 7.54282700 -0	.73720300	-0.38571300
Η	I 6.58271300 -1	.05691500	0.06393300
Η	I 7.65860000 0	.33954900	-0.14297500
С	2 7.48172700 -0	.93856900	-1.92268300

Η	7.19909000	-1.98949100	-2.13293500
Н	8.49568200	-0.80110900	-2.35795500
С	6.47935200	0.00634400	-2.64300100
Н	5.53136800	0.05022700	-2.07271000
Н	6.24665300	-0.38599200	-3.64978600
Н	7.31691600	1.86830100	-1.81769300
Н	7.92463800	1.42977800	-3.47055500
Н	4.41160100	6.31428700	2.48309300
Н	0.41255200	7.02634300	4.06427200
Н	0.36067000	-7.32201100	-3.04836500
Н	-3.78824300	-6.02230600	-3.26509400
0	3.74079500	3.94621100	1.25657500
0	-0.62551400	4.68000100	3.07209000
0	1.27200500	-4.98985400	-1.93869500
0	-3.30551600	-3.55535500	-2.18262600
С	5.16150800	4.00502900	1.38639500
С	5.90695600	4.08710400	0.19051700
С	5.79612300	3.88066900	2.64526700
С	7.32072800	4.06810600	0.26165600
Η	5.38236300	4.15384200	-0.76745900
С	7.20896600	3.86172100	2.70074200
Η	5.19863400	3.79092400	3.55860600
С	7.97565900	3.96157700	1.51292700
Η	7.90670600	4.14986500	-0.66119100
Η	7.70798500	3.76045900	3.67116800
Η	9.07017900	3.95283500	1.56318100
С	-1.72766200	5.53901400	3.35920900
С	-2.63169100	5.08852400	4.34582400
С	-1.97191900	6.71972000	2.61858700
С	-3.80052500	5.84310400	4.60252600
Η	-2.41324600	4.16257400	4.88712500
С	-3.14197200	7.46628800	2.89000900
Η	-1.27197100	7.04080800	1.84001300
С	-4.05782500	7.03423500	3.88078000
Η	-4.50793500	5.49962200	5.36577400
Η	-3.34214900	8.37898700	2.31712600
Η	-4.96516100	7.61449800	4.08123300
С	2.37715000	-5.77028300	-2.39109100
С	3.29900800	-6.18639200	-1.40556900
С	2.60800500	-6.01473900	-3.76697700
С	4.47285100	-6.86809700	-1.80591000
Η	3.08985600	-5.97112200	-0.35290800
С	3.78218800	-6.70401700	-4.15119500

Η	1.88721900	-5.67390900	-4.51793200
С	4.71653600	-7.13252000	-3.17581000
Η	5.19234200	-7.19440600	-1.04657000
Η	3.96831700	-6.89834300	-5.21367000
Η	5.62493700	-7.66343000	-3.48085900
С	-4.70672600	-3.83410300	-2.24294500
С	-5.46896300	-3.13766700	-3.20524800
С	-5.31154100	-4.69719000	-1.30040200
С	-6.87310100	-3.31874000	-3.22912300
Η	-4.96426300	-2.47086100	-3.91180700
С	-6.71425600	-4.87442000	-1.33983000
Η	-4.69722800	-5.20711000	-0.55064800
С	-7.49705200	-4.18933000	-2.30272600
Η	-7.47456200	-2.78058700	-3.97023300
Η	-7.19363200	-5.54177200	-0.61441100
Η	-8.58355000	-4.32903200	-2.32770700
С	2.34272700	-0.42549200	-1.99812200
С	2.57506500	-1.81377200	-2.20551600
С	2.98567100	-2.26976000	-3.47544100
Η	2.44525000	-2.51487300	-1.37713000
С	2.82216100	0.01909600	-4.26498600
С	3.11816600	-1.33685600	-4.52916100
Η	3.19567300	-3.33193800	-3.63580900
Η	2.88587700	0.76588500	-5.06430500
Η	3.43730200	-1.64168100	-5.53086000
Ν	2.45525000	0.48604000	-3.03273500
N	3.38535500	-0.71716000	0.48442200
Ν	3.27481700	-1.31461600	1.48661700
С	2.01738800	0.12022700	-0.64678600
Η	2.40649300	1.14524100	-0.50248100

Intermediate C

C_[Co(III)(P9)] Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.532528 Hartree H_corr: 1.822339 Hartree SCF: -7174.84839066 Hartree S: 609.959 Cal/Mol-Kelvin H: -7173.026052 Hartree G: -7173.315863 Hartree

С	-0.26297000	2.46044900	1.74387600
Ν	-0.30021000	1.05400300	1.61977200
Η	-0.55461500	3.90038400	3.45766500
С	-0.53352500	0.59539300	2.93369700
С	-0.66271000	1.71095800	3.85591400
Η	-0.85444100	1.60806200	4.92286900
С	-0.51246900	2.86620200	3.11920700
С	-0.68689900	-0.74446000	3.32087500
Ν	-0.55225400	-1.63935900	0.98128100
С	-0.75015200	-2.93551900	0.47304100
С	-1.05888300	-3.87183600	1.54481400
Н	-1.26231700	-4.93107600	1.39641800
С	-1.01468500	-3.16130400	2.72276500
Η	-1.18916600	-3.51861000	3.73623500
С	-0.71629600	-1.78058400	2.37841800
С	0.01947700	3.37687000	0.71231800
Η	1.32572800	3.45639100	-3.74315400
С	1.01379400	3.11682400	-2.75729600
С	0.82274400	1.72258100	-2.38728700
Н	0.73249500	4.96317500	-1.54228900
N	0.41058400	1.61501900	-1.04353700
С	0.35620100	2.95304700	-0.58576900
С	0.71693100	3.87909000	-1.64975900
С	-0.61913200	-3.32966800	-0.87057800
Η	0.95328600	-1.69517400	-4.92151200
С	0.63187500	-1.76763000	-3.88509700
С	0.11847200	-2.86224100	-3.22842700
Η	-0.05612600	-3.86610600	-3.61270600
С	-0.15438500	-2.44773600	-1.85883900
Ν	0.15626600	-1.08438900	-1.67778500
С	0.62938500	-0.65868900	-2.94083600
C	0.96211700	0.65785900	-3.29660600
Co	0.06872600	-0.05209000	-0.00277200
C	1.38428900	0.96872000	-4.70569200
C	2.70170000	1.44304800	-4.98851100
C	0.50525000	0.826/4/00	-5.82140300
C	3.12062200	1.77184800	-6.30006500
C	0.90971600	1.13154200	-7.14127600
C	2.216/9/00	1.60888700	-7.37312200
H	2.53483400	1.86075800	-8.39071800
C	-0.89923300	-1.05836900	4.7/5347/00
C	-2.19693100	-1.31500200	5.29968/00
C	0.187/84400	-1.09243800	5.69159700

С	-2.41032700	-1.61115700	6.66817800
С	0.00403800	-1.37484200	7.06751300
С	-1.30014400	-1.63679300	7.54027400
Н	-1.45381000	-1.86060900	8.60182600
С	-0.02375600	4.85087200	1.04041200
С	1.14019700	5.53506500	1.51462100
С	-1.25246400	5.57806900	0.93726500
С	1.06677400	6.89510400	1.92188300
С	-1.32833600	6.93666800	1.34930200
С	-0.16870800	7.56813600	1.84064600
Н	1.96365600	7.38746000	2.29894100
Н	-2.28317000	7.45969400	1.28356400
Н	-0.22733000	8.61401400	2.16284900
С	-0.89457200	-4.77036800	-1.20944100
С	-2.24310200	-5.23769700	-1.29123900
С	0.17583700	-5.69842200	-1.39919400
С	-2.52847500	-6.61175000	-1.50137600
С	-0.10737600	-7.07482800	-1.62306600
С	-1.45022600	-7.50782300	-1.65898100
Н	-3.56710900	-6.94596800	-1.53232300
Η	0.72011900	-7.76945100	-1.77155000
Н	-1.66004600	-8.57132800	-1.82134500
Ν	2.37127300	4.80845500	1.54424100
Н	2.32892600	3.86520800	1.14365700
Ν	1.51004500	-5.18705700	-1.36397200
Н	1.62111300	-4.21488000	-1.02281000
N	-3.26801400	-4.25154200	-1.20233000
Н	-2.93094400	-3.28285900	-1.21800100
N	-2.38008000	4.89908900	0.38391500
Η	-2.19411700	3.95166100	0.03704400
С	3.60747900	5.24235500	2.02254400
С	2.65045300	-5.81903700	-1.86659800
С	-4.64482300	-4.41790400	-1.06014600
С	-3.67170400	5.39560700	0.19484000
0	-5.22803200	-5.53837800	-0.91420800
0	-4.07502600	6.52624500	0.61673200
0	2.68676400	-7.03445800	-2.24618300
0	3.80239000	6.36988900	2.58026100
С	-4.57791800	4.50314800	-0.61516300
Η	-4.20915400	3.52011500	-0.92820900
С	-5.39498100	-3.11061800	-1.08710300
Η	-4.86290500	-2.23579700	-1.47831300
С	-6.43423900	-2.81324400	0.01009000

Η	-6.50509500	-1.77978500	0.36030900
Н	-6.61041600	-3.59172100	0.75757100
С	-6.89636600	-3.14733200	-1.42481300
Н	-7.35050600	-4.13348300	-1.55415900
С	-5.53829300	5.22397300	-1.57324100
Н	-5.48231000	6.31556000	-1.58361600
Η	-5.77199800	4.72417800	-2.51704800
С	-6.10220400	4.58916000	-0.29097000
Н	-6.38444200	5.25819300	0.52449300
С	-7.42430300	-2.05504300	-2.29956300
С	-6.90786600	3.34714100	-0.45828400
0	-6.71754900	2.47113700	-1.34419400
0	-6.75796800	-1.09003700	-2.75424300
0	-7.93389100	3.27684800	0.48550200
0	-8.78167800	-2.23809600	-2.55423800
С	-8.87601100	2.11329300	0.34878200
Н	-8.29194800	1.23065800	0.03482600
Н	-9.26843300	1.97418800	1.36987800
С	-9.98763200	2.45204900	-0.65911300
Н	-10.50030200	3.37401200	-0.32258300
Н	-9.52321200	2.67684100	-1.63718700
С	-9.44904500	-1.21374700	-3.43112100
Н	-10.38754900	-1.71929300	-3.71500000
Н	-8.81181200	-1.05713800	-4.32072200
С	-9.71003700	0.11743800	-2.70376900
Н	-8.74747200	0.63346200	-2.53603500
Н	-10.29580600	0.74811200	-3.40724500
С	-11.03709600	1.31664400	-0.82126400
Н	-11.53341400	1.13220800	0.15439900
Η	-11.82791800	1.68732600	-1.50544900
С	-10.49344000	-0.03168500	-1.37440100
Η	-9.84700200	-0.52582000	-0.62269700
Η	-11.35406000	-0.71515400	-1.52801000
С	4.75569100	4.27865200	1.83467300
Η	4.57231400	3.33204900	1.31506200
С	3.86345600	-4.93352900	-2.00233300
Η	3.81289100	-3.91380200	-1.61004700
С	4.70380400	-5.15908700	-3.26919800
Η	5.18155800	-4.28037500	-3.70751800
Η	4.34957600	-5.93419300	-3.95423400
С	5.85404800	4.30398700	2.91129800
Η	6.37456400	3.36514600	3.11226200
Η	5.67837500	4.97030700	3.76041700

С	6.13510300	4.98246800	1.57110800
Η	6.05955100	6.07351300	1.54388200
С	5.24731700	-5.65429900	-1.91919300
Η	5.19727700	-6.73066000	-1.73390600
С	7.11406200	4.43856300	0.58349700
С	6.41664000	-4.97893600	-1.28841100
Ο	7.52189100	3.13061800	0.87588200
0	6.45840700	-3.61146300	-1.60750700
0	7.28912900	-5.54567800	-0.58426500
0	7.56581800	5.08750200	-0.39546100
С	8.62126500	2.59360800	0.00099900
Η	8.24463300	2.54534000	-1.03713700
Η	9.45709500	3.31617600	0.04129700
С	9.01381200	1.21499500	0.53744900
Η	9.96434300	0.92524900	0.04171200
Η	9.23305000	1.29345400	1.62103900
С	7.62035600	-2.83163300	-1.07953300
С	7.95167900	0.12127600	0.27809800
Η	6.99495100	0.41428800	0.75366800
Η	7.75346500	0.07114200	-0.81260800
С	8.37784000	-1.27318900	0.80461800
Н	8.43923200	-1.23826300	1.91098700
Н	9.40397800	-1.51056600	0.44859600
С	7.40109300	-2.40583300	0.38292800
Н	6.35831500	-2.05429900	0.50067100
Η	7.52601500	-3.29117800	1.03235500
Η	7.66708900	-1.96423800	-1.75873500
Η	8.53235700	-3.44879600	-1.18153600
Η	4.13173300	2.14907700	-6.47651400
Η	0.19124800	0.99002700	-7.95361500
Η	0.85847600	-1.37950400	7.74994000
Η	-3.41810700	-1.82007700	7.03768900
0	3.54171400	1.59633100	-3.85847000
0	-0.77491900	0.23314700	-5.69183200
0	1.44703500	-0.74091900	5.14786000
0	-3.24257700	-1.32113900	4.34556600
С	4.93771700	1.85279200	-4.01825500
С	5.78537800	0.92607100	-4.66951100
С	5.44988200	3.01526600	-3.40231500
С	7.17400300	1.19234200	-4.72068500
Η	5.36737700	0.01807000	-5.11711400
С	6.84254000	3.26597400	-3.44973100
Η	4.76375900	3.70000100	-2.89405900

С	7.70468200	2.36005400	-4.11653600
Н	7.83952400	0.48421100	-5.22802700
Н	7.24472400	4.16152700	-2.96257700
Н	8.78087900	2.56184900	-4.16463000
С	-1.82494200	0.84561300	-4.94290000
С	-1.87280500	2.23116100	-4.67239300
С	-2.87647300	-0.01897600	-4.56164200
С	-2.99739900	2.74890400	-3.98373200
Н	-1.06331000	2.89219100	-4.99962700
С	-3.99462000	0.51297000	-3.88138700
Н	-2.80335300	-1.08391200	-4.80624500
С	-4.05711500	1.89901200	-3.58426900
Н	-3.04178700	3.82444200	-3.77130800
Н	-4.82246800	-0.13940300	-3.58241400
Н	-4.93076500	2.29107900	-3.05112200
С	2.65404600	-1.12381700	5.80058700
С	3.63182500	-0.11795600	5.96382300
С	2.90945800	-2.46647300	6.16651100
С	4.88971700	-0.46361200	6.51281300
Н	3.39710000	0.90815900	5.66322700
С	4.16831500	-2.79655400	6.71981200
Н	2.14640600	-3.23561900	6.00785200
С	5.16108300	-1.80026000	6.89602400
Η	5.65195700	0.31245800	6.64757400
Η	4.37520100	-3.83464100	7.00438400
Η	6.13395800	-2.06345600	7.32595700
С	-4.59994200	-1.16610700	4.75804000
С	-5.52518200	-2.11922000	4.27978900
С	-5.01718800	-0.04725500	5.51696000
С	-6.89856000	-1.95331300	4.57974600
Η	-5.16270100	-2.96473200	3.68656100
С	-6.39192200	0.10395500	5.81207300
Η	-4.28429100	0.69197300	5.85715800
С	-7.33524400	-0.84673400	5.34842200
Η	-7.62358300	-2.69094100	4.21708800
Η	-6.72512500	0.96978400	6.39561900
Η	-8.39888800	-0.72352400	5.58095200
С	2.60089100	-1.40053400	1.00166000
С	3.67933500	-1.11888800	1.93038600
С	4.37728600	-2.16284800	2.55447100
Η	3.90175800	-0.07573800	2.18098800
С	3.01613300	-3.71573600	1.27665000
С	4.03272600	-3.50384300	2.23870500

H5.15955900-1.942892003.28942600H2.74484500-4.740601000.99200300H4.53795900-4.354915002.70499400N2.31444300-2.726974000.65225400C1.86001100-0.271539000.51946300H2.283575000.723685000.73563700

C_[N2]

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: -0.013594 Hartree H_corr: 0.008229 Hartree SCF: -109.560708554Hartree S: 45.93 Cal/Mol-Kelvin H: -109.5524796 Hartree G: -109.5743026 Hartree

Cartesian Coordinates:

Ν	0.00000000	0.00000000	0.57307000
Ν	0.00000000	0.00000000	-0.57307000

C_[2a]

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.099513 Hartree H_corr: 0.138461 Hartree SCF: -309.796492304 Hartree S: 81.974 Cal/Mol-Kelvin H: -309.6580313 Hartree G: -309.6969793 Hartree

С	-1.80367300	-1.06029300	0.00000000
С	-0.41103500	-1.30025800	0.00000000
С	0.52368900	-0.22415100	0.00000000
С	-2.29347400	0.26752500	0.00000000
С	0.01393200	1.10796100	0.00000000
С	-1.37608400	1.34941800	0.00000000
Н	-2.50370300	-1.90413500	0.00000000
Η	-3.37285400	0.45913400	0.00000000
Н	-0.03630000	-2.33249700	0.00000000
Η	0.70708900	1.95750300	0.00000000

H-1.749263002.380614000.00000000C1.97164800-0.539989000.00000000C3.012808000.339894000.00000000H2.20893300-1.615285000.00000000H2.869558001.42734700-0.0000100H4.04967000-0.013326000.00000000

Transition State TS2

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -172.433 cm⁻¹ G_corr: 1.656566 Hartree H_corr: 1.961559 Hartree SCF: -7484.66562 Hartree S: 641.911 Cal/Mol-Kelvin H: -7482.704061 Hartree G: -7483.009054 Hartree

Н	-3.75529200	5.08463900	-0.12924900
С	-4.42600800	4.21723200	-0.12975800
Н	-3.02800200	2.97638200	0.96942400
С	-4.00984600	3.01881100	0.48697200
Н	-6.02496000	5.24605300	-1.21324800
С	-5.70587600	4.31392000	-0.73398800
С	-4.86750300	1.87375000	0.51741800
С	-6.56750700	3.19005900	-0.71494400
С	-6.15042400	1.98796100	-0.10200900
С	-4.48855000	0.60693400	1.16345400
Н	-7.55708400	3.25693900	-1.18215500
Н	-5.20854300	-0.21748500	1.04728100
Н	-6.82217000	1.12035200	-0.09172500
С	0.10033100	1.50597500	-2.27684600
N	0.34222400	0.30318600	-1.57519800
Н	0.15224000	2.10134800	-4.45449700
С	0.66466800	-0.62820600	-2.58789300
С	0.63215900	0.00016000	-3.89849900
Н	0.85396800	-0.51264300	-4.83279700
С	0.27644900	1.31685300	-3.70955100
С	0.99987800	-1.97615100	-2.39006100
Ν	0.76362300	-1.87494200	0.10954600
С	0.91318500	-2.87029100	1.09293700
С	1.28202900	-4.14468200	0.49315500

Η	1.45636600	-5.06103300	1.05497800
С	1.35207100	-3.94118700	-0.86661400
Н	1.60002900	-4.65870900	-1.64713200
С	1.02841200	-2.54394600	-1.10819800
С	-0.19657800	2.75642000	-1.70508400
Н	-0.49540200	4.81731700	2.44832500
С	-0.40999800	4.06811200	1.66335600
С	-0.23864400	2.64389300	1.90062000
Н	-0.55955800	5.18389300	-0.25876200
N	-0.13218700	1.94345600	0.67719000
С	-0.26975900	2.94496100	-0.31175100
С	-0.44657000	4.25498700	0.29882800
С	0.73928600	-2.70350900	2.47716400
Н	-0.19497200	0.57821100	5.62181700
С	-0.02887500	0.05602200	4.68202600
С	0.23689000	-1.27885400	4.48527300
Н	0.33409600	-2.07032800	5.22681100
С	0.39098500	-1.47150600	3.04942500
Ν	0.20491000	-0.26076600	2.35419200
С	-0.03097100	0.69209100	3.37120400
С	-0.19978300	2.07424300	3.18705900
Co	0.13492800	-0.02225800	0.38827500
С	-0.31393100	2.96049000	4.39741200
С	-1.53068800	3.63235100	4.72467200
С	0.76864100	3.13158700	5.31135900
С	-1.66380300	4.42690700	5.88926400
С	0.66197500	3.93093400	6.47097300
С	-0.55993000	4.57665800	6.75584200
Η	-0.66090800	5.18549000	7.66084900
С	1.36630400	-2.82398500	-3.57799100
С	2.72239900	-3.16303500	-3.85219700
С	0.38772900	-3.31795000	-4.48573500
С	3.09048000	-3.97204400	-4.95512400
С	0.72829600	-4.11238000	-5.60777400
С	2.08338700	-4.43833800	-5.82650200
Η	2.35807900	-5.05845000	-6.68696700
С	-0.33600000	3.94483400	-2.62493000
С	-1.54268200	4.18242200	-3.34815400
С	0.78050300	4.81843100	-2.84879100
С	-1.64658400	5.27534300	-4.24791200
С	0.67603400	5.91059000	-3.75131700
С	-0.53883300	6.12492200	-4.43503100
Η	-2.57452600	5.43054700	-4.79994300

Η	1.54454400	6.55117700	-3.91144700
Н	-0.61863400	6.96760400	-5.13087200
С	0.98136600	-3.88536800	3.37396600
С	2.32322800	-4.26085700	3.69465200
С	-0.10635000	-4.63327400	3.91402800
С	2.58609300	-5.40313800	4.49394600
С	0.15350000	-5.77009400	4.73022800
С	1.49043300	-6.13996700	4.99445000
Н	3.61955000	-5.68296900	4.70887400
Н	-0.68431300	-6.32697700	5.15119700
Н	1.68213500	-7.02197000	5.61638500
Ν	-2.65860100	3.30058200	-3.16014900
Н	-2.78118200	2.88780200	-2.22903000
Ν	-1.43013500	-4.18962300	3.60284700
Н	-1.51728400	-3.48157200	2.84687900
N	3.35263100	-3.40304600	3.21121100
Н	3.02225800	-2.54518100	2.75530700
Ν	1.99849500	4.52756500	-2.16214400
Η	1.95509000	3.73580100	-1.51161500
С	-3.54839000	2.91552900	-4.16589400
С	-2.59580900	-4.52755100	4.29427100
С	4.73936800	-3.52487300	3.29514200
С	3.23860700	5.15497000	-2.29851500
0	5.35214300	-4.53366900	3.75701600
0	3.46605900	6.13663700	-3.07557300
0	-2.64866100	-5.38299500	5.23639300
0	-3.45669900	3.31123400	-5.37167400
С	4.33290400	4.57607800	-1.43591500
Н	4.07884300	3.77765000	-0.72965800
С	5.43372200	-2.25573300	2.81834600
Η	5.59601200	-1.51988500	3.61696100
С	5.19526600	-1.67538500	1.42409700
Η	5.16300000	-0.58697300	1.32200000
Η	4.57777700	-2.23435900	0.71354200
С	6.57125400	-2.31233400	1.78644700
Η	6.82518700	-3.27640200	1.34065200
С	5.47550600	5.51677000	-1.04336200
Η	5.39569800	6.54643300	-1.40205600
Η	5.94049400	5.36045900	-0.06644400
С	5.74108600	4.45142000	-2.11658100
Η	5.80025300	4.77907600	-3.15654500
С	7.70671300	-1.36794000	1.99301600
С	6.57146400	3.27269800	-1.73807500

0	6.66563100	2.78698500	-0.57919300
0	7.64640300	-0.29761500	2.65458400
0	7.24728000	2.75686100	-2.84342000
0	8.86337000	-1.80709100	1.34711300
С	8.12454700	1.56329300	-2.58759200
Η	7.61733800	0.91167200	-1.85526100
Н	8.17273500	1.06557000	-3.57026000
С	9.51506400	2.00563200	-2.10129100
Н	9.90445800	2.76839400	-2.80365800
Н	9.41641400	2.49125000	-1.11295000
С	10.07772200	-0.93213400	1.49927300
Н	10.91377700	-1.63928300	1.36282200
Н	10.07834100	-0.52445800	2.52591600
С	10.10845500	0.19948300	0.45777300
Н	9.22958400	0.85286200	0.61039800
Н	11.00556200	0.81542000	0.68448700
С	10.52628100	0.82706200	-2.02848800
Н	10.62616400	0.37708400	-3.03882600
Η	11.52220900	1.24691400	-1.77729300
С	10.18993700	-0.29655000	-1.00800100
Η	9.24189300	-0.79966900	-1.28078400
Η	10.98143800	-1.07231200	-1.07928300
С	-4.63609800	1.95365900	-3.75106300
Η	-4.75357700	1.71726900	-2.68857500
С	-3.82784100	-3.76423300	3.87439400
Η	-3.76212600	-3.11905500	2.99396000
С	-4.76943200	-3.31427000	5.00071900
Η	-5.29453100	-2.36719000	4.85731900
Η	-4.46969700	-3.56985900	6.02074300
С	-5.00540500	0.86964900	-4.78074200
Η	-5.34198200	-0.09307800	-4.38955600
Η	-4.41312700	0.85828400	-5.70034700
С	-5.94592800	2.06131900	-4.60575900
Η	-5.89538800	2.84598500	-5.36671600
С	-5.18661100	-4.49736100	4.11006400
Η	-5.09966900	-5.49215100	4.55556600
С	-7.29566200	1.93551700	-3.98170700
С	-6.30852000	-4.36555400	3.13973800
0	-7.50046000	0.69267400	-3.35816400
0	-6.37387000	-3.07729000	2.57757700
0	-7.13022800	-5.27350500	2.85929900
0	-8.18687100	2.82052700	-4.02628100
С	-8.87374500	0.47412900	-2.79506100

Η	-9.00548300	1.14946300	-1.92788900
Н	-9.60922700	0.75849700	-3.56927200
С	-8.98037100	-1.00257200	-2.40153300
Н	-10.03770700	-1.20101100	-2.12734700
Н	-8.75871100	-1.62688400	-3.28997100
С	-7.49124700	-2.83153600	1.61126400
С	-8.05511000	-1.40695800	-1.22920500
Н	-7.00770400	-1.16077100	-1.49386700
Н	-8.31572200	-0.79104700	-0.34250700
С	-8.15193700	-2.91343000	-0.87726300
Н	-7.96187900	-3.50449800	-1.79639500
Н	-9.18778300	-3.16071100	-0.56092900
С	-7.14276200	-3.36896300	0.21181000
Н	-6.12642700	-3.02418500	-0.06164900
Н	-7.11618300	-4.47304600	0.26111800
Н	-7.59816300	-1.73447800	1.62618100
Н	-8.40757800	-3.31077400	2.00300600
Н	-2.62197700	4.89854300	6.12380000
Н	1.52971800	4.01374400	7.13152200
Н	-0.04823000	-4.45897900	-6.29509000
Η	4.13927700	-4.22991300	-5.12491300
0	-2.62195700	3.36770400	3.86266100
0	1.96379400	2.38102400	5.17971900
0	-0.94180800	-2.89519800	-4.23929300
0	3.67245800	-2.68822100	-2.91422000
С	-3.78744800	4.19510900	3.88569000
С	-5.02982600	3.52826300	3.92458500
С	-3.70704300	5.60212100	3.76136900
С	-6.21932400	4.29006600	3.85126300
Η	-5.04845600	2.43622100	3.99306900
С	-4.90472700	6.35119900	3.69483100
Η	-2.73317300	6.10094800	3.71167800
С	-6.16221400	5.70061500	3.74249500
Η	-7.18826900	3.77902600	3.87390000
Η	-4.85282700	7.44176800	3.59705400
Η	-7.08636100	6.28638300	3.68536600
С	2.93684400	2.69145300	4.18644000
С	2.88057900	3.84394000	3.37190500
С	4.01283000	1.77794800	4.10159300
С	3.91892700	4.06027200	2.43327200
Η	2.05675400	4.55855500	3.46861700
С	5.04354200	2.01142600	3.16488800
Η	4.02500200	0.91066400	4.77006600

С	4.99542200	3.14878300	2.31761900
Н	3.88060200	4.95233600	1.79582800
Н	5.88805700	1.31499900	3.09665600
Н	5.78122000	3.30085300	1.57078100
С	-2.04115800	-3.60598600	-4.80019400
С	-3.02987100	-2.83417500	-5.44941600
С	-2.19721200	-5.00075600	-4.61749800
С	-4.19723400	-3.47440400	-5.93069100
Н	-2.87304500	-1.75739300	-5.56845700
С	-3.36520600	-5.62854900	-5.10935000
Н	-1.42801200	-5.57687700	-4.09269400
С	-4.36761400	-4.87113800	-5.76612900
Н	-4.96664100	-2.88289600	-6.43983200
Н	-3.49430800	-6.70827300	-4.97185400
Н	-5.26980600	-5.36419100	-6.14471400
С	5.05587900	-2.64305900	-3.26729200
С	5.96439400	-3.27346700	-2.39027400
С	5.50223400	-1.91344900	-4.39441100
С	7.35179900	-3.18216700	-2.65500300
Н	5.58041900	-3.82350600	-1.52540300
С	6.89029000	-1.83917800	-4.65382300
Н	4.77954200	-1.41337000	-5.04779700
С	7.81753600	-2.47397000	-3.78969400
Η	8.06362100	-3.66717200	-1.97810700
Η	7.24452900	-1.28255200	-5.52924200
Η	8.89180300	-2.41716900	-3.99819800
С	-2.27549900	-1.78335100	-0.13297300
С	-3.01611100	-1.98980800	-1.34519700
С	-3.58446900	-3.24522200	-1.63449900
Η	-3.10488300	-1.16333800	-2.05796200
С	-2.68721100	-4.02914700	0.47836600
С	-3.40763000	-4.30014900	-0.70831400
Η	-4.13421800	-3.40313200	-2.56843500
Н	-2.54016700	-4.82378600	1.22079300
Н	-3.81617700	-5.29933800	-0.88813600
N	-2.14029200	-2.81730500	0.78986700
С	-1.72280000	-0.44590200	0.11841900
Η	-2.10557000	0.32972800	-0.56943200
С	-3.32011800	0.35305800	1.85464300
Η	-2.60762700	1.14433400	2.10267100
Н	-3.17087100	-0.60570800	2.36013000

Intermediate D

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.661495 Hartree H_corr: 1.964892 Hartree SCF: -7484.71001440 Hartree S: 638.553 Cal/Mol-Kelvin H: -7482.745122 Hartree G: -7483.048519 Hartree

Н	4.28197000	4.61393800	0.23084300
С	4.81400500	3.67271700	0.04979700
Н	3.06491700	2.69452800	-0.77254600
С	4.12212300	2.58351200	-0.51075500
Н	6.72735400	4.41765200	0.81336000
С	6.19451200	3.57048300	0.36858300
С	4.79518300	1.33415100	-0.77452600
С	6.88067400	2.35266500	0.11078700
С	6.19811300	1.25705800	-0.44501200
С	4.13134700	0.19783000	-1.34774900
Н	7.94690200	2.27420900	0.35359800
Н	4.71530700	-0.72628400	-1.47020900
Н	6.73141200	0.31839500	-0.64123200
С	0.07449500	1.51639300	2.32836100
Ν	-0.20159500	0.33907600	1.59430400
Н	-0.00668700	2.07374300	4.51542700
С	-0.59618100	-0.59456200	2.58213000
С	-0.57418200	0.00943100	3.90536800
Η	-0.85570900	-0.50461700	4.82232300
С	-0.14265300	1.30779400	3.75328300
С	-0.96942600	-1.92885200	2.35157500
Ν	-0.71488000	-1.77807100	-0.14395600
С	-0.87451500	-2.75058100	-1.15028400
С	-1.21147100	-4.04620100	-0.57683600
Η	-1.37092200	-4.95322100	-1.15806400
С	-1.27737900	-3.87220500	0.78731800
Н	-1.49538300	-4.61260500	1.55571800
С	-0.97672100	-2.47416200	1.05765500
С	0.41132700	2.77296100	1.79039000
Η	0.57725700	4.97696200	-2.29584200
С	0.51217900	4.20018700	-1.53606100
С	0.26744800	2.79453200	-1.81535500
Н	0.81940200	5.23383400	0.41282000

Ν	0.21206100	2.04847100	-0.61432200
С	0.44226700	3.00709500	0.40125500
С	0.63765300	4.33215400	-0.17051100
С	-0.76740000	-2.53212100	-2.53438500
Н	-0.10361000	0.88547300	-5.60397300
С	-0.20058500	0.32684300	-4.67569200
С	-0.44082400	-1.01780900	-4.51308700
Н	-0.58105700	-1.78161800	-5.27654200
С	-0.48980800	-1.26765700	-3.07859900
N	-0.27354600	-0.08090400	-2.34964000
С	-0.10605600	0.91062100	-3.34429300
С	0.12455800	2.27799200	-3.11754900
Co	-0.09159900	0.08692900	-0.38822800
С	0.24703700	3.20422400	-4.29656900
С	1.49363600	3.81923600	-4.62587800
С	-0.84190500	3.47340400	-5.17889800
С	1.64692600	4.66147900	-5.75356900
С	-0.71335500	4.31891900	-6.30343100
С	0.53502700	4.91267800	-6.58565200
Η	0.64970800	5.55855200	-7.46291200
С	-1.43307100	-2.77672700	3.50487600
С	-2.80829800	-3.14017000	3.62403100
С	-0.56438300	-3.23715600	4.53428000
С	-3.29379000	-3.93536600	4.68983600
С	-1.02605500	-4.01561200	5.62432500
С	-2.39100900	-4.36428600	5.68591700
Η	-2.75735900	-4.96669600	6.52456400
С	0.57670100	3.93880200	2.73756900
С	1.80684600	4.18071900	3.42574700
С	-0.53919500	4.80015900	3.00613000
С	1.93239200	5.28221000	4.31653400
С	-0.41357000	5.89589600	3.90166000
С	0.82438000	6.12376100	4.53468400
Η	2.87505300	5.44099600	4.83974200
Η	-1.28212300	6.52677100	4.09475300
Η	0.92447200	6.97073800	5.22285000
С	-1.07576700	-3.67503600	-3.46280100
С	-2.44269400	-4.01020600	-3.72137300
С	-0.03870100	-4.42350100	-4.09482400
С	-2.77552000	-5.11514900	-4.54692300
С	-0.37059100	-5.51924900	-4.94110800
С	-1.72780000	-5.85204300	-5.14047000
Η	-3.82551000	-5.36653800	-4.71038300

Η	0.42961500	-6.07455800	-5.43178900
Η	-1.97465900	-6.70355900	-5.78507900
N	2.90342000	3.28670800	3.20314500
Н	2.82914900	2.67377500	2.38400000
Ν	1.31201300	-4.03294600	-3.83806100
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N	-3.42854200	-3.15176600	-3.15418000
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С	3.99906300	3.09264100	4.04832700
С	2.44449700	-4.42924300	-4.55204400
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Η	-3.92192100	3.75358800	1.01111700
С	-5.47100300	-1.97502300	-2.65237900
Η	-5.60899300	-1.21334600	-3.43141300
С	-5.21224900	-1.43277200	-1.24601700
Η	-5.14407600	-0.34808200	-1.12310900
Η	-4.61672800	-2.02719300	-0.54610500
С	-6.60720600	-2.01984400	-1.61902100
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Η	-5.19031300	6.52423200	1.76594100
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С	-5.53158100	4.42204500	2.46208300
Η	-5.55440600	4.73820400	3.50700300
С	-7.71663700	-1.04180700	-1.80302200
С	-6.39626800	3.26629300	2.09267300
0	-6.51943900	2.78364500	0.93541600
0	-7.62926300	0.04476800	-2.43501800
0	-7.07995400	2.77152700	3.20523200
0	-8.88716800	-1.46897100	-1.17308200
С	-8.02235100	1.63022500	2.94660400
Η	-7.55641300	0.95557200	2.20701300
Η	-8.09839000	1.13478300	3.92964000
С	-9.38775600	2.15227200	2.46747600
Η	-9.74951800	2.90394100	3.19583800

Η	-9.25680500	2.67077500	1.49997000
С	-10.07225300	-0.55144200	-1.28709300
Η	-10.93167600	-1.23696000	-1.19097600
Η	-10.05457300	-0.08958300	-2.29034400
С	-10.06992800	0.52293600	-0.18593300
Η	-9.16925400	1.15374300	-0.30288000
Н	-10.94642000	1.17938500	-0.37637400
С	-10.44928400	1.02428400	2.33550500
Н	-10.56792900	0.52299800	3.31957600
Η	-11.42626800	1.50102500	2.11356200
С	-10.16657900	-0.05225200	1.24977200
Η	-9.23978300	-0.61265700	1.48161700
Η	-10.99050000	-0.79605400	1.28226600
С	4.95387200	1.98756400	3.65654400
Η	4.89502700	1.56794700	2.64710200
С	3.72947600	-3.78170100	-4.09918200
Η	3.69183000	-3.11890000	-3.22920000
С	4.76851500	-3.46206700	-5.17941600
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Η	4.49348100	-3.70252100	-6.21012500
С	5.38345500	1.05154600	4.80742500
Η	5.59755000	0.01274800	4.54624800
Η	4.92050800	1.23466200	5.78128900
С	6.36688300	2.11200400	4.31488700
Η	6.46130700	3.01520800	4.92616900
С	5.00898900	-4.66907500	-4.25681500
Η	4.83253200	-5.65429600	-4.69710900
С	7.62820300	1.76274900	3.59650500
С	6.07842400	-4.65592200	-3.22199800
0	7.69897300	0.40831000	3.23298100
0	6.26187400	-3.37730800	-2.66273600
0	6.76539100	-5.65164000	-2.88271000
0	8.56950400	2.56699800	3.37923800
С	9.01407400	-0.03361300	2.66338000
Η	9.22406200	0.56248300	1.75528600
Η	9.79636900	0.19027500	3.41189900
С	8.90334700	-1.53118300	2.36201400
Η	9.92092000	-1.90246400	2.11890100
Η	8.57976700	-2.06212500	3.27923400
С	7.33199700	-3.26049900	-1.62235500
С	7.94106200	-1.84983000	1.19411900
Η	6.93978200	-1.43743100	1.42733100
Η	8.30320800	-1.31695900	0.28950300

С	7.81576800	-3.36585800	0.90113200
Н	7.47876800	-3.88117800	1.82334300
Н	8.81499100	-3.78680700	0.65873100
С	6.81468400	-3.69675100	-0.23967400
Н	5.84802100	-3.19394700	-0.03878300
Н	6.62228500	-4.78503000	-0.26731700
Н	7.59726800	-2.19107000	-1.65387700
Н	8.19521600	-3.87736100	-1.93363200
Н	2.62516900	5.09041300	-5.98774700
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Н	-0.33416900	-4.33217100	6.40928400
Н	-4.35361800	-4.19798800	4.74459400
0	2.59310100	3.44384500	-3.81497600
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С	5.00411000	3.53017000	-3.91241500
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С	6.21998300	4.24508300	-3.80742200
Н	4.98583800	2.44707000	-4.06756000
С	4.97869700	6.33007800	-3.47243900
Н	2.80007600	6.15279600	-3.46573900
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Н	7.17034000	3.70594000	-3.88839700
Н	4.96514800	7.41165900	-3.29458600
Н	7.15696200	6.19444400	-3.51255300
С	-2.98114600	3.02361700	-3.99664300
С	-2.86512000	4.11502900	-3.10791500
С	-4.06450900	2.11867500	-3.92028700
С	-3.84999700	4.27607200	-2.10314600
Η	-2.03705100	4.82562200	-3.19858000
С	-5.04201500	2.29654700	-2.91700700
Η	-4.12417700	1.30115500	-4.64636200
С	-4.93253900	3.37053300	-1.99632800
Η	-3.76553800	5.12037300	-1.40786900
Η	-5.89180200	1.60640400	-2.85285900
Η	-5.67832800	3.47669400	-1.20187400
С	1.80664500	-3.45849300	5.19497300
С	2.68184600	-2.62320300	5.92380500
С	2.01092300	-4.85836700	5.12634400
С	3.77840000	-3.20327100	6.60548800
Η	2.49603000	-1.54468400	5.94588100

С	3.10701900	-5.42472500	5.81826700
Η	1.32954700	-5.48693800	4.54311900
С	3.99224700	-4.60257800	6.55922300
Η	4.46009100	-2.56179700	7.17530000
Η	3.27159900	-6.50731500	5.77123600
Η	4.83987000	-5.04769900	7.09179800
С	-5.02793000	-3.04847100	2.58706600
С	-5.38473700	-4.15982500	1.79209000
С	-5.99563600	-2.29001700	3.28074000
С	-6.75011300	-4.51915100	1.69162400
Η	-4.60681200	-4.71631800	1.25896900
С	-7.35737500	-2.66262300	3.17543700
Η	-5.68077100	-1.42813300	3.87860600
С	-7.73662200	-3.77450000	2.38373300
Η	-7.04050100	-5.37435100	1.07110300
Η	-8.11986000	-2.08614900	3.71186900
Н	-8.79266800	-4.05253600	2.29788700
С	2.22050800	-1.78677000	-0.03274200
С	2.41639700	-2.15326500	1.33115400
С	2.77020300	-3.47461800	1.66608100
Η	2.29948600	-1.39589500	2.11268600
С	2.66251600	-4.01017000	-0.69458500
С	2.90035700	-4.43076100	0.63200100
Η	2.94230900	-3.75140400	2.71053900
Η	2.73701100	-4.72834400	-1.52009000
Η	3.17543500	-5.46984200	0.83884000
N	2.33793100	-2.72782000	-1.04150100
С	1.95114600	-0.35448300	-0.39019600
Η	2.18006300	0.30375100	0.46664600
С	2.66172200	0.13093700	-1.68329600
Η	2.27863500	1.11090800	-2.00068300
Η	2.48265400	-0.59478300	-2.49301100

<u>Intermediate E</u>

 $E_{[Co(II)(P6)]} = A_{[Co(II)(P9)]}$

E[3a]

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.186841 Hartree H_corr: 0.239936 Hartree SCF: -595.949895 Hartree S: 111.747 Cal/Mol-Kelvin

H: -596.0850148 Hartree

G: -596.1381098 Hartree

0.07866500	1.25055600	0.88539800
4.51601900	-1.99091800	-0.36545400
3.96410000	-1.04455800	-0.31259600
2.19284600	-1.96983700	0.53535000
2.64545800	-1.02892600	0.19580300
5.60062500	0.14260400	-1.14269700
4.57708700	0.15515600	-0.75024300
1.90776500	0.18605600	0.27764900
3.85519900	1.37051100	-0.67375100
2.53538800	1.38335000	-0.16691100
0.50575300	0.24444800	0.80017200
4.31945300	2.30619700	-1.00802400
1.98218100	2.32982600	-0.11176000
-1.93741700	-0.37065600	0.01844000
-2.66362100	-0.94840300	-1.06047400
-3.96799200	-0.48599300	-1.34297400
-2.20743500	-1.74393300	-1.66058900
-3.73188600	1.06968600	0.50707200
-4.52064900	0.54539400	-0.54495600
-4.54302400	-0.92028800	-2.16882800
-4.11623800	1.87101400	1.15055700
-5.52793500	0.93307000	-0.72919100
-2.47177400	0.63314100	0.79380200
-0.55703900	-0.82298800	0.35250500
-0.05458000	-0.76353300	1.80651000
0.59779400	-1.57002500	2.15679700
-0.76754600	-0.37462700	2.53933400
-0.16837400	-1.63289200	-0.27499000
	0.07866500 4.51601900 3.96410000 2.19284600 2.64545800 5.60062500 4.57708700 1.90776500 3.85519900 2.53538800 0.50575300 4.31945300 1.98218100 -1.93741700 -2.66362100 -3.96799200 -2.20743500 -3.73188600 -4.52064900 -4.54302400 -4.54302400 -4.54302400 -5.52793500 -2.47177400 -0.55703900 -0.05458000 0.59779400 -0.76754600 -0.16837400	0.078665001.250556004.51601900-1.990918003.96410000-1.044558002.19284600-1.969837002.64545800-1.028926005.600625000.142604004.577087000.155156001.907765000.186056003.855199001.370511002.535388001.383350000.505753000.244448004.319453002.306197001.982181002.32982600-1.93741700-0.37065600-2.66362100-0.94840300-3.96799200-0.48599300-2.20743500-1.74393300-3.731886001.06968600-4.54302400-0.92028800-4.116238001.87101400-5.527935000.93307000-2.471774000.63314100-0.55703900-0.82298800-0.05458000-0.763533000.59779400-1.57002500-0.76754600-0.37462700-0.16837400-1.63289200
2.5. NMR/HPLC SPECTRAL DATA















¹H NMR of **1a**, 500 MHz, CDCl₃



---2.40

$N \vee \vee 4$	らてつてての		
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0014	000040	4 - 0	P
0444	\circ	6 4 4	<u>-</u>
			0
$\langle \langle \rangle \rangle$	$12 \times 11 \times$		

¹³C NMR of **1a**, 125 MHz, $CDCI_3$





0823402	4 6 6 6 6 6 8 6	2002
		6,6,6,6
444444	ϕ ϕ ϕ ϕ ϕ ϕ	
		$\leq \sim$

¹H NMR of **1b**, 500 MHz, $CDCI_3$







¹H NMR of **1c**, 600 MHz, CDCl₃







---2.43

-1.57

¹H NMR of **1d**, 500 MHz, CDCl₃



— 150.33	→ 144.43 → 143.91 → 140.92 → 136.00	— 129.92 — 127.33	—120.80	39.94 39.56 39.24 39.24 39.24	-21.12
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¹³C NMR of **1d**, 150 MHz, DMSO- d_6





0 8 N 9 9 9 0 0	0 7 0 0 7 N M	- O Q 4
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4 4 4 4 4 4 4	~~~~~~	
		$\leq \sim \sim$

¹H NMR of **1e**, 600 MHz, $CDCI_3$











¹H NMR of **1g**, 500 MHz, CDCl₃

0=\$=0

N^{-NH}

10.0

1g





f1 (ppm)

0.0

0.5





¹H NMR of **1h**, 600 MHz, CDCl₃





¹³C NMR of **1h**, 150 MHz, CDCl₃





¹H NMR of **1**i, 500 MHz, $CDCl_3$





¹³C NMR of **1i**, 125 MHz, CDCl₃



151182233332473	222333322222222222222222222222222222222	93 92 92 93 93 93 93 32 93 32	25 25 24
	444444		\leftarrow \leftarrow \leftarrow
			\leq

¹H NMR of **1** \mathbf{j} , 600 MHz, CDCl₃







¹H NMR of **1k**, 600 MHz, $CDCI_3$







¹H NMR of **1**I, 600 MHz, CDCl₃





f1 (ppm)

Ò





¹H NMR of **1m**, 600 MHz, CDCl₃



-161.46	~153.52 ~151.50 ~146.76	-131.51 -129.01 -126.25 -124.00	- 114.19	77.37 - 77.16 - 77.16	-55.49	- 34.32 - 30.20 - 30.20 25.03 - 23.67
	12	$\langle \rangle$				

13 C NMR of **1m**, 150 MHz, CDCl₃



~8.13 7.78 7.76 7.65 7.59 7.59 7.59 7.58 4.29 4.28 4.27 4.26 4.25 4.25 2.94 2.93 2.92 2.92 2.89 2.89 2.88 32 31 25 25 25

¹H NMR of **1n**, 600 MHz, CDCl₃





--62.91

$^{19}\mathsf{F}$ NMR of $\mathbf{1n},\,564$ MHz, CDCI_3



			· · · ·	·	· · ·	· · ·			· · · ·	·	· · ·	· · ·	·	· · ·	· · · ·		·	· · · ·			
-15	-20	-25	-30	-35	-40	-45	-50	-55	-60	-65	-70	-75	-80	-85	-90	-95	-100	-105	-110	-115	-120
										f1	(ppm)										~
$\begin{array}{c} 8.14 \\ \hline 8.00 \\ \hline 7.99 \\ \hline 7.77 \\ \hline 7.60 \\ \hline 7.60 \end{array}$ ~7.19 ~7.19 4.20 4.27 4.25 4.25 4.25 4.25 4.25 3.91 2.93 2.91 2.91 2.91 2.89 2.89 25 25 25 25

¹H NMR of **1o**, 600 MHz, CDCl₃









¹H NMR of **1p**, 600 MHz, CDCl₃







¹H NMR of **1q**, 600 MHz, CDCl₃





¹³C NMR of **1q**, 150 MHz, CDCl₃





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<u> </u>	0 0 0 0 0 0 0 0	

¹H NMR of 1r, 600 MHz, CDCl₃









¹³C NMR of 1r, 150 MHz, CDCl₃





¹H NMR of **1s**, 600 MHz, $CDCI_3$









¹³C NMR of **1s**, 150 MHz, $CDCI_3$





¹H NMR of **3a**, 600 MHz, CDCl₃





f1 (ppm)



Peak Table

PDA Ch1 220nm Peak# Ret. Time Area% Area 9205035 49.936 6.925 1 9228706 2 8.147 50.064 Total 18433741 100.000

н

H

(±)-3a



PDA Ch1 220nm

ͺH

(–)-3a

н

	Dir oni zzonin						
Peak#	Ret. Time	Area	Area%				
1	6.943	86133	0.601				
2	8.153	14247173	99.399				
Total		14333305	100.000				



161.46 158.04	149.52	135.95 134.42	127.14 122.04 120.65	113.99	77.37 77.16 76.95	55.47	-29.12 -27.61	18.70	
		57	$ \leq $				57		

¹³C NMR of **3b**, 150 MHz, $CDCI_3$







-					
	~ ~	-	10		
		ĸ	1.24		
-	- ci.	<u>n</u> .	T CI	ω,	

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	11.387	2449783	50.029
2	14.019	2446986	49.971
Total		4896769	100.000

н

	Sample Information
Sample Name	: XXW-1040-ID-5%-0.8mL
Sample ID	: XXW-1040-ID-5%-0.8mL
Data File	: XXW-1040-ID-5%-0.8mL001.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram

mAU



UV Spectrum Retention time = 11.388





UV Spectrum Retention time = 13.884



Peak Table

PDA Ch1 220nm

Peak#	Ret. Time	Area	Area%		
1	11.388	10880	0.201		
2	13.884	5401596	99.799		
Total		5412476	100.000		

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¹H NMR of **3c**, 600 MHz, $CDCI_3$











PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	5.654	5386382	50.028
2	6.337	5380364	49.972
Total		10766746	100.000

ͺH

(±)-3c

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mAU



UV Spectrum Retention time = 5.635





UV Spectrum Retention time = 6.320



Peak Table

PDA Ch1 254nm

Peak#	Ret. Time	Area	Area%			
1	5.635	13046	0.561			
2	6.320	2312257	99.439			
Total		2325304	100.000			



¹H NMR of **3d**, 500 MHz, CDCl₃







¹³C NMR of **3d**, 150 MHz, CDCl₃





mAU













Peak Table

PDA Ch1 220nm

I DIL CHI			
Peak#	Ret. Time	Area	Area%
1	6.970	7238830	49.898
2	7.588	7268338	50.102
Total		14507169	100.000







UV Spectrum Retention time = 7.570



Peak Table

PDA Ch1 220nm

Peak#	Ret. Time	Area	Area%
1	6.964	66186	1.415
2	7.570	4610794	98.585
Total		4676980	100.000

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				577	

¹³C NMR of **3e**, 150 MHz, $CDCI_3$



—31.24

¹¹B NMR of **3e**, 160 MHz, $CDCI_3$

















PDA Ch1 220nm

Peak#	Ret. Time	Area	Area%
1	6.753	12959531	50.097
2	7.330	12909283	49.903
Total		25868814	100.000





DD 4 C1 1 220

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(–)-3e

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Peak Table

	PDA Chi	220nm		
	Peak#	Ret. Time	Area	Area%
	1	6.779	135561	1.456
	2	7.222	9177073	98.544
1	Total		9312635	100.000



¹H NMR of **3f**, 500 MHz, CDCl₃





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			\searrow	57	

¹³C NMR of **3f**, 125 MHz, CDCl₃







Peak Table

PDA Ch1 220nm

Н

(±)-3f

Peak#	Ret. Time	Area	Area%	
1	12.927	9811405	49.900	
2	15.170	9850900	50.100	
Total		19662305	100.000	







H CHO N (-)-3f





Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	12.968	136510	1.831
2	15.184	7319877	98.169
Total		7456386	100.000


¹H NMR of **3g**, 500 MHz, CDCl₃







f1 (ppm)















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Peak#	Ret. Time	Area	Area%
1	6.927	861886	49.856
2	7.363	866881	50.144
Total		1728767	100.000















Peak Table

PDA Ch1 220nm

-				
	Peak#	Ret. Time	Area	Area%
	1	6.871	4722920	98.926
ſ	2	7.311	51283	1.074
	Total		4774203	100.000





¹H NMR of **3h**, 500 MHz, CDCl₃





7.24	9.19 1 22	4.22 6.17	8.80 8.74 8.26 2.08		.33	63	.85
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¹³C NMR of **3h**, 150 MHz, CDCl₃









Peak Table

PDA Chi	220nm		20
Peak#	Ret. Time	Area	Area%
1	8.865	10205296	50.507
2	16.040	10000384	49.493
Total		20205680	100.000

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	Sample Information
Sample Name	: XXW-1042-ID-5%-0.8mL
Sample ID	: XXW-1042-ID-5%-0.8mL
Data File	: XXW-1042-ID-5%-0.8mL001.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram









UV Spectrum Retention time = 16.211



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.983	149588	1.607
2	16.211	9161389	98.393
Total		9310977	100.000



¹H NMR of **3i**, 500 MHz, CDCl₃







	Sample Information
Sample Name	: XXW-0959-IA-5%-0.8mL
Sample ID	: XXW-0959-IA-5%-0.8mL
Data File	: XXW-0959-IA-5%-0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 8.183









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PDA	(n	1/20nm
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Р	eak	Table	

Peak#	Ret. Time	Area	Area%
1	8.183	4454549	49.102
2	11.582	4617443	50.898
Total		9071993	100.000

	Sample Information
Sample Name	: XXW-1052-IA-5%-0.8mL
Sample ID	: XXW-1052-IA-5%-0.8mL
Data File	: XXW-1052-IA-5%-0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram





UV Spectrum Retention time = 8.176











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PDA	Ch1	220)nm
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Peak#	Ret. Time	Area	Area%
1	8.176	147035	0.821
2	11.560	17754835	99.179
Total		17901870	100.000



¹H NMR of 3j, 600 MHz, CDCl₃







¹³C NMR of **3j**, 150 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-0892-ID-5%0.8mL
Sample ID	: XXW-0892-ID-5%0.8mL
Data File	: XXW-0892-ID-5%0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram













Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.252	2304869	50.321
2	9.559	2275441	49.679
Total		4580311	100.000





UV Spectrum Retention time = 8.539









Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.539	47863	1.135
2	9.822	4169169	98.865
Total		4217032	100.000



¹H NMR of **3k**, 500 MHz, CDCl₃





¹³C NMR of **3k**, 150 MHz, CDCl₃

















Peak Table

PDA Ch1	220nm		2
Peak#	Ret. Time	Area	Area%
1	6.187	2088519	49.785
2	8.139	2106594	50.215
Total		4195113	100.000













UV Spectrum Retention time = 8.060



Peak Table

PDA Ch1 220nm

-	Dirent			
	Peak#	Ret. Time	Area	Area%
	1	6.173	192058	3.561
	2	8.060	5201880	96.439
	Total		5393938	100.000



¹H NMR of **3I**, 600 MHz, $CDCI_3$













UV Spectrum Retention time = 7.582



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	7.150	1308983	49.385
2	7.582	1341604	50.615
Total		2650587	100.000











UV Spectrum Retention time = 7.577



Peak Table

PDA Ch1	220nm	A 25 Y 1	
Peak#	Ret. Time	Area	Area%
1	7.129	14047	0.828
2	7.577	1681990	99.172
Total		1696037	100.000





¹H NMR of 3m, 600 MHz, CDCl₃





1.12	9.52	5.98	2.56 0.83		.39	.67
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			57			

13 C NMR of **3m**, 125 MHz, CDCl₃







UV Spectrum Retention time = 5.622









Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	5.622	4966083	49.435
2	7.011	5079643	50.565
Total		10045726	100.000

	Sample Information
Sample Name	: XXW-1058-ADH-30%-0.8mL
Sample ID	: XXW-1058-ADH-30%-0.8mL
Data File	: XXW-1058-ADH-30%-0.8mL.lcd
Method File	: XXW-30%.0.8.mL.lcm
	Chromatogram













PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	5.636	22806	0.435
2	7.030	5216556	99.565
Total		5239362	100.000





¹H NMR of **3n**, 600 MHz, CDCl₃





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¹³C NMR of **3n**, 125 MHz, $CDCl_3$





Sample Information : XXW-1071-ODH-30%-0.8mL : XXW-1071-ODH-30%-0.8mL : XXW-1071-ODH-30%-0.8mL001.lcd : XXW-30%.0.8.mL.lcm Chromatogram











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Peak#	Ret. Time	Area	Area%			
1	7.556	698282	49.984			
2	8.471	698734	50.016			
Total		1397015	100.000			









Peak Table

PDA Ch1	254nm
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Peak#	Ret. Time	Area	Area%			
1	7.563	26403	0.748			
2	8.428	3502542	99.252			
Total		3528945	100.000			





¹H NMR of **3o**, 500 MHz, CDCl₃





0.02	61.89 9.74 9.67	6.15	22.32 11.20	4 1 2 5	.19	.65
-16	15 14 14 15 14 15	-13			—30 —27	- 19

¹³C NMR of **30**, 125 MHz, $CDCl_3$









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(±)-30







Peak Table

PDA Ch1 254nm

Peak#	Ret. Time	Area	Area%
1	7.629	2241011	50.686
2	11.166	2180382	49.314
Total		4421393	100.000



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(–)-30



UV Spectrum Retention time = 11.325



Peak Table

PDA Ch1 254nm

	Peak#	Ret. Time	Area	Area%
	1	7.905	15271	0.340
	2	11.325	4473401	99.660
	Total		4488672	100.000
845755555555555555555555555555555555555	75 75 75 75 75 75 75 75 73	44444 4444 4444 4444 4444 4444 4444 4444		
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¹H NMR of **3p**, 600 MHz, $CDCI_3$





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160	146	136	122122	- 12 - 12 - 12 - 12 - 12 - 12 - 12 - 12	30.	23.
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¹³C NMR of **3p**, 150 MHz, $CDCI_3$





Peak Table

PDA Ch1 220nm

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(±)-3p

Peak#	Ret. Time	Area	Area%
1	6.135	13894627	49.999
2	6.737	13895237	50.001
Total		27789864	100.000













UV Spectrum Retention time = 6.749



Peak Table

PDA Ch1 220nm

Peak#	Ret. Time	Area	Area%
1	6.151	45485	0.851
2	6.749	5299273	99.149
Total		5344758	100.000



¹H NMR of 3q, 600 MHz, CDCl₃



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¹³C NMR of 3q, 150 MHz, CDCl₃



















Peak Table

PDA	Ch	1254	nm
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Peak#	Ret. Time	Area	Area%
1	6.726	2089095	49.356
2	7.435	2143591	50.644
Total		4232687	100.000

	Sample Information
Sample Name	: XXW-1063-ID-10%-0.8mL
Sample ID	: XXW-1063-ID-10%-0.8mL
Data File	: XXW-1063-ID-10%-0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram

mAU















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Peak#	Ret. Time	Area	Area%
1	6.760	42711	0.679
2	7.475	6248775	99.321
Total		6291486	100.000



¹H NMR of **3r**, 600 MHz, $CDCl_3$







¹³C NMR of **3r**, 150 MHz, $CDCl_3$



	Sample Information
Sample Name	: XXW-1077-ID-10-%0.8mL
Sample ID	: XXW-1077-ID-10-%0.8mL
Data File	: XXW-1077-ID-10-%0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
- ATI	Chromatogram

mAU



UV Spectrum Retention time = 6.601











P	DA Ch1	220nm		
	Peak#	Ret. Time	Area	Area%
	1	6.601	3784539	49.899
	2	8.267	3799875	50.101
	Total	- A	7584414	100.000

	Sample Information
Sample Name	: XXW-1078-ID-10-%0.8mL
Sample ID	: XXW-1078-ID-10-%0.8mL
Data File	: XXW-1078-ID-10-%0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram















Peak Table

PDA Ch1 220nm

I DI L CHI			
Peak#	Ret. Time	Area	Area%
1	6.671	137884	1.222
2	8.403	11149098	98.778
Total		11286982	100.000





	Sample Information
Sample Name	: XXW-1089-ODH-2-%0.5mL
Sample ID	: XXW-1089-ODH-2-%0.5mL
Data File	: XXW-1089-ODH-2-%0.5mL.lcd
Method File	: XXW-2%.0.5.mL.lcm
	Chromatogram















Peak Table

PDA Ch1	220nm	1000	
Peak#	Ret. Time	Area	Area%
1	12.677	12589114	49.792
2	13.541	12694305	50.208
Total		25283419	100.000

	Sample Information
Sample Name	: XXW-1090-ODH-2-%0.5mL
Sample ID	: XXW-1090-ODH-2-%0.5mL
Data File	: XXW-1090-ODH-2-%0.5mL.lcd
Method File	: XXW-2%.0.5.mL.lcm
	Chromatogram













DD I CI I AAA



Peak Table

PDA Chi	220nm		
Peak#	Ret. Time	Area	Area%
1	12.767	127494	0.713
2	13.563	17765923	99.287
Total		17893417	100.000



¹H NMR of **3t**, 600 MHz, $CDCI_3$















Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	17.135	951266	49.772
2	20.920	959987	50.228
Total		1911253	100.000



mAU

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(–)-3t

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UV Spectrum Retention time = 17.346





Retention time = 20.974



Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	17.346	9379	1.018
2	20.974	911905	98.982
Total		921284	100.000



¹H NMR of 3u, 600 MHz, CDCl₃



95	37	49	24	42 42	N (0.10	- 10 0	ŝ	
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50	15	4	13	44	202	37 31 29	19	8.0
				57	\checkmark	155		

¹³C NMR of 3u, 150 MHz, CDCl₃







Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	9.986	7526371	49.653
2	10.721	7631620	50.347
Total		15157991	100.000

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0

(±)-3u



Chromatogram

mAU



UV Spectrum Retention time = 9.892





UV Spectrum Retention time = 10.669



Peak Table

PDA Ch1	220nm		2
Peak#	Ret. Time	Area	Area%
1	9.892	59588	1.002
2	10.669	5887987	98.998
Total		5947575	100.000





¹H NMR of **3v**, 500 MHz, CDCl₃



-173.96	-158.89	- 149.55	-136.15	-122.65 -121.45	-77.47 -77.16 -76.84	-52.01	-27.40 -24.24	-17.43
				57				

¹³C NMR of 3v, 100 MHz, CDCl₃











Sample Information

Chromatogram

: XXW-0967-IE-5%-0.8mL

: XXW-0967-IE-5%-0.8mL

: XXW-5%.0.8.mL.lcm

: XXW-0967-IE-5%-0.8mL.lcd











Peak Table

PDA Ch1	254nm		I van Inoiv
Peak#	Ret. Time	Area	Area%
1	13.514	1214750	49.820
2	15.275	1223534	50.180
Total	1	2438284	100.000



mAU



UV Spectrum Retention time = 13.583









Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	13.583	1633261	97.626
2	15.451	39711	2.374
Total		1672972	100.000





¹H NMR of **3w**, 600 MHz, CDCl₃







¹³C NMR of **3w**, 125 MHz, CDCl₃







100-

50-



200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390









PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.448	1520686	50.026
2	10.763	1519119	49.974
Total		3039804	100.000



0-200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390 mm

Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.490	71499	0.792
2	10.629	8957428	99.208
Total		9028927	100.000

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(–)-3w

NH₂

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¹H NMR of **3x**, 600 MHz, CDCl₃



—172.12	—160.09	—149.48	—136.10	~121.17	77.37 77.16	~37.49 ~35.97	~26.58 ~23.06 ~17.59
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¹³C NMR of 3x, 150 MHz, CDCl₃



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200	100	180	170	160	150	1/0	130	120	110	100	۵n	80	70	60	50	10	30	20	10	Δ	_10
200	190	100	170	100	100	140	130	120	110	100	30	00	10	00	50	40	50	20	10	0	-10
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																					312



Retention time = 32.319

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⊖ (±)-3x Me

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Peak Table

PDA Ch1	220nm		I Cak Table
Peak#	Ret. Time	Area	Area%
1	15.389	11890586	49.790
2	32.319	11991076	50.210
Total		23881662	100.000



mAU













Peak Table

PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	15.833	25940	0.373
2	33.045	6928239	99.627
Total		6954178	100.000


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¹H NMR of 3y, 600 MHz, CDCl₃





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# ¹³C NMR of 3y, 150 MHz, CDCl₃

















Peak Table

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]	PDA Ch1	254nm		
	Peak#	Ret. Time	Area	Area?
	1	11.164	1460645	50
ſ	2	13.213	1457808	49
	Total		2918453	100

	Sample Information
Sample Name	: XXW-1057-ODH-5%-0.8mL
Sample ID	: XXW-1057-ODH-5%-0.8mL
Data File	: XXW-1057-ODH-5%-0.8mL001.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram





(–)-3y





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PDA Chl	254nm		
Peak#	Ret. Time	Area	Area%
1	11.212	68385	1.416
2	13.266	4762678	98.584
Total		4831063	100.000





## ¹H NMR of **3z**, 500 MHz, CDCl₃





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<u> </u>	<u> </u>	<u> </u>			(i)	
	1	1		1 AF		

## ¹³C NMR of 3z, 125 MHz, CDCl₃











Peak Table

PDA Ch1	220nm		I Cak Table
Peak#	Ret. Time	Area	Area%
1	9.254	1969724	50.081
2	10.791	1963318	49.919
Total		3933042	100.000

	Sample Information
Sample Name	: XXW-1274ODH-5%0.8mL
Sample ID	: XXW-1274ODH-5%0.8mL
Data File	: XXW-1274ODH-5%0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 9.303





UV Spectrum Retention time = 10.834



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	9.303	43373	1.112
2	10.834	3855380	98.888
Total		3898752	100.000





¹H NMR of **3aa**, 600 MHz,  $CDCI_3$ 







## ¹³C NMR of **3aa**, 150 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-1064IF-20%-0.8mL
Sample ID	: XXW-1064IF-20%-0.8mL
Data File	: XXW-1064IF-20%-0.8mL001.lcd
Method File	: XXW-20%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 8.249









Peak#	Ret. Time	Area	Area%
1	8.249	3263979	49.905
2	8.883	3276359	50.095
Total		6540338	100.000













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<b>T</b> 1				
Pegl	-	0	h l	0
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_		_	_	

PDA ChI	220nm		
Peak#	Ret. Time	Area	Area%
1	8.315	52965	1.003
2	8.967	5227203	98.997
Total		5280168	100.000





¹H NMR of **3ab**, 600 MHz, CDCl₃





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## ¹³C NMR of **3ab**, 150 MHz, CDCl₃







UV Spectrum Retention time = 5.737





UV Spectrum Retention time = 6.483



Peak Table

PDA Ch1	254nm	224	
Peak#	Ret. Time	Area	Area%
1	5.737	1608473	49.338
2	6.483	1651624	50.662
Total		3260097	100.000

	Sample Information
Sample Name	: XXW-1079-ID-10%-0.8mL
Sample ID	: XXW-1079-ID-10%-0.8mL
Data File	: XXW-1079-ID-10%-0.8mL001.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram











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Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	5.726	47907	4.368
2	6.466	1048805	95.632
Total		1096711	100.000



¹H NMR of **3ac**, 600 MHz, CDCl₃



• belongs to another set of diastereomer





### ¹³C NMR of **3ac**, 150 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-0897-ODH-2%-0.8mL
Sample ID	: XXW-0897-ODH-2%-0.8mL
Data File	: XXW-0897-ODH-2%-0.8mL001.lcd
Method File	: XXW-2%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 9.049



mAU 500 200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390 mm

UV Spectrum Retention time = 10.129



<b>D</b> 1			
Dool	I	ab	10
FEAL			
I Cu			

PDA	Ch1	254nm
-----	-----	-------

Peak#	Ret. Time	Area	Area%
1	9.049	1212243	22.315
2	10.129	1367649	25.175
3	10.390	1353103	24.908
4	17.872	1499517	27.603
Total		5432512	100.000

	Sample Information
Sample Name	: XXW-1081-ODH-2%-0.8mL
Sample ID	: XXW-1081-ODH-2%-0.8mL
Data File	: XXW-1081-ODH-2%-0.8mL.lcd
Method File	: XXW-2%.0.8.mL.lcm
	Chromatogram













F	Pea.	k '	a	b	le

1	PDA Ch1	254nm		
I	Peak#	Ret. Time	Area	Area%
I	1	8.499	218784	22.645
I	2	9.567	37294	3.860
	3	9.941	644757	66.736
I	4	17.784	65291	6.758
I	Total		966127	100.000



¹H NMR of **3ad**, 600 MHz, CDCl₃



belongs to another set of diastereomer









### ¹³C NMR of **3ad**, 125 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-1001-1%-IC-0.5mL
Sample ID	: XXW-1001-1%-IC-0.5mL
Data File	: XXW-1001-1%-IC-0.5mL.lcd
Method File	: XXW-1%.0.5.mL.lem
	Chromatogram











nm

Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	12.911	3588140	51.362
2	13.441	3397896	48.638
Total		6986036	100.000

	Sample Information
Sample Name	: XXW-1118-IC-1-%-0.5mL
Sample ID	: XXW-1118-IC-1-%-0.5mL
Data File	: XXW-1118-IC-1-%-0.5mL.led
Method File	: XXW-1%.0.5.mL.lem
	Chromatogram











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Peak Table

PDA Ch1	254nm	1012	1000 000 000 004 004
Peak#	Ret. Time	Area	Area%
1	12.841	1291644	96.879
2	13.403	41618	3.121
Total		1333261	100.000



#### ¹H NMR of *exo*-**3ae**, 500 MHz, CDCl₃









	Sample Information
Sample Name	: XXW-0979-n1-ID-5%-0.8mL
Sample ID	: XXW-0979-n1-ID-5%-0.8mL
Data File	: XXW-0979-n1-ID-5%-0.8mL001.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram













<b>D</b>				
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T CON			Ο,	

PDA	Ch1	220nm

Peak#	Ret. Time	Area	Area%	
1	7.292	1553872	49.551	
2	7.847	1582019	50.449	
Total		3135891	100.000	

	Sample Information
Sample Name	: XXW-0997-n1-ID-5%-0.8mL
Sample ID	: XXW-0997-n1-ID-5%-0.8mL
Data File	: XXW-0997-n1-ID-5%-0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram















Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	7.297	384322	9.296
2	7.843	3749799	90.704
Total		4134121	100.000





¹H NMR of *endo*-**3ae**, 600 MHz, CDCl₃

















UV Spectrum Retention time = 8.632



Peak Table

PDA Chl	254nm		
Peak#	Ret. Time	Area	Area%
1	8.214	94861	50.359
2	8.632	93509	49.641
Total		188370	100.000













Pea	kТ	a	bl	e

PDA Ch1 254nm

Peak#	Ret. Time	Area	Area%
1	8.168	10443	3.365
2	8.545	299928	96.635
Total		310371	100.000

nm



¹H NMR of **3af**, 600 MHz, CDCl₃







### ¹³C NMR of *exo*-3af, 150 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-1135-n1-ID-10%-0.8mL
Sample ID	: XXW-1135-n1-ID-10%-0.8mL
Data File	: XXW-1135-n1-ID-10%-0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 7.011









Peak Table

PDA Ch1	220nm	N	
Peak#	Ret. Time	Area	Area%
1	7.011	3647322	50.033
2	7.454	3642545	49.967
Total	Sector Contraction of the Sector Contraction	7289867	100.000

	Sample Information
Sample Name	: XXW-1167-1-ID-10%-0.8mL
Sample ID	: XXW-1167-1-ID-10%-0.8mL
Data File	: XXW-1167-1-ID-10%-0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram















Peak Table

PDA	Ch1	220	nm
-----	-----	-----	----

-				
	Peak#	Ret. Time	Area	Area%
	1	7.008	2544105	98.734
	2	7.456	32630	1.266
	Total		2576735	100.000


¹H NMR of **2ag**, 600 MHz, CDCl₃



 <ul> <li>✓ 139.66</li> <li>✓ 136.70</li> <li>✓ 135.35</li> <li>✓ 125.67</li> <li>✓ 123.74</li> <li>─ 113.32</li> </ul>	77.37 77.16 76.95	

¹³C NMR of **2g**, 150 MHz,  $CDCI_3$ 





-221.01	-161.31 -149.46	7 139.87 137.51 137.55 137.55 136.02 126.77 126.74 120.70	~77.37 -77.16 ~76.95	「13.26 13.36 13.36 13.36 13.36 13.36 13.36 13.36 13.36 13.36 13.36
			$\checkmark$	

¹³C NMR of **3ag**, 150 MHz, CDCl₃





¹H NMR of **3ah**, 500 MHz, CDCl₃



belongs to another set of diastereomer







--27.96 --25.37 --17.98

#### ¹³C NMR of **3ah**, 125 MHz, CDCl₃







UV Spectrum Retention time = 24.730









Peak Table

PDA	Ch1	220nm		
Pea	k#	Ret. Time	Area	Area%
	1	24.730	10564697	49.963
	2	28.520	10580295	50.037
	Fotal		21144992	100.000





UV Spectrum Retention time = 24.419









<b>D</b> 1				
Dag	<b>Z</b>		h	0
FCA	K 1	<b>a</b>		
-			-	

	Chi	1 220	100000
PDA		440	шп

Peak#	Ret. Time	Area	Area%
1	24.419	7952276	99.370
2	28.734	50392	0.630
Total		8002668	100.000





¹H NMR of **3ai**, 600 MHz, CDCl₃





24	3337 = 233	N (0.10	<del>, +</del>	-	~	
0 8 6	0,0,0,0,0,0,0	6,00,00	5	ò	N.	
044	้ดดังดังดัง	N N 9	~	ø	~	
			4	$\sim$	$\sim$	
17 \	>	$\checkmark$				

### ¹³C NMR of **3ai**, 150 MHz, CDCl₃









UV Spectrum Retention time = 27.628



Peak Table

DDA Ch1	22000		Peak Table
Peak#	Ret. Time	Area	Area%
1	21.224	6168831	49.913
2	27.628	6190289	50.087
Total	And shake a shear of the spin	12359120	100.000

	Sample Information
Sample Name	: XXW-1282-IE-10%0.8mL
Sample ID	: XXW-1282-IE-10%0.8mL
Data File	: XXW-1282-IE-10%0.8mL001.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 21.556





UV Spectrum Retention time = 27.735



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	21.556	5719	0.112
2	27.735	5102140	99.888
Total		5107860	100.000



¹H NMR of **3aj**, 500 MHz, CDCl₃







Peak Table

PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	11.978	2883411	49.182
2	14.129	2979325	50.818
Total		5862737	100.000

H

(±)-3aj



ͺH

(–)-3aj





Peak Table

PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	11.946	32571	1.066
2	13.843	3023249	98.934
Total		3055820	100.000











44	55 94	74	47	- 0 <u>-</u>	თ	0 0 0
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~	44	n n	2	~ ~ 9 ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	2	e 33
<del>~</del>	~ ~		<u>_</u>		2	7 00
	$\mathbf{Y}$	11				$\vee$

# ¹³C NMR of **3ak**, 125 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-0816-ASH-10%-0.8mL
Sample ID	: XXW-0816-ASH-10%-0.8mL
Data File	: XXW-0816-ASH-10%-0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 14.111





UV Spectrum Retention time = 20.158



Peak Table

PDA Ch1	220nm	and the second sec	
Peak#	Ret. Time	Area	Area%
1	14.111	3211098	50.082
2	20.158	3200552	49.918
Total		6411650	100.000

	Sample Information
Sample Name	: XXW-0859-ASH-10%-0.8mL
Sample ID	: XXW-0859-ASH-10%-0.8mL
Data File	: XXW-0859-ASH-10%-0.8mL001.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram















Peak Table

PE	DA Ch1	220nm		
I	Peak#	Ret. Time	Area	Area%
	1	14.243	5950037	98.496
	2	20.238	90830	1.504
	Total		6040867	100.000





¹H NMR of **3al**, 600 MHz, CDCl₃











UV Spectrum Retention time = 11.640









Peak Table

PDA Ch1 254nm				
Peak#	Ret. Time	Area	Area%	
1	11.640	691467	50.042	
2	16.750	690315	49.958	
Total		1381782	100.000	



Sample ID

Method File

Data File









Peak Table

PDA Ch1 254nm

Peak#	Ret. Time	Area	Area%
1	11.647	4294	0.719
2	16.750	592974	99.281
Total		597268	100.000





#### ¹H NMR of **3am**, 600 MHz, $CDCI_3$





44	32	68 98 00	5 0 <del>7</del>	33 55	5
152 149	141	128.128.121.121.121.121.121.121.121.121.	5.17 77.13 76.5	29.3	19.C
	Ì			11	Ì

## ¹³C NMR of **3am**, 150 MHz, CDCl₃



	Sample Information
Sample Name	: XXW-0901-rac-ODH5%-0.8mL
Sample ID	: XXW-0901-rac-ODH5%-0.8mL
Data File	: XXW-0901-rac-ODH5%-0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram













Peak Table

PDA Ch1			
Peak#	Ret. Time	Area	Area%
1	24.507	5135162	49.946
2	32.554	5146268	50.054
Total		10281430	100.000

	Sample Information
Sample Name	: XXW-0864-ODH5%-0.8mL
Sample ID	: XXW-0864-ODH5%-0.8mL
Data File	: XXW-0864-ODH5%-0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram







UV Spectrum Retention time = 32.539

mAU 500 0 200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390 mm

Peak Table

PDA Ch1	220nm		I can Iuoic
Peak#	Ret. Time	Area	Area%
1	25.608	1797606	5.560
2	32.539	30536272	94.440
Total		32333878	100.000

230 230 230 230 230 230 230 230 230 230	119 119 119 119 119	13 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	95 95 95 96 95 95 97 92 92
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#### ¹H NMR of **3an**, 600 MHz, CDCl₃



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## ¹³C NMR of **3an**, 125 MHz, $CDCI_3$



	Sample Information
Sample Name	: XXW-1334OJH5%-0.8mL001
Sample ID	: XXW-1334OJH5%-0.8mL001
Data File	: XXW-1334OJH5%-0.8mL001.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram



UV Spectrum Retention time = 17.658









Peak Table

PDA Ch1	220nm		r cuir ruore
Peak#	Ret. Time	Area	Area%
1	17.658	7963820	49.990
2	20.737	7966978	50.010
Total		15930798	100.000

	Sample Information
Sample Name	: XXW-1332OJH5%-0.8mL
Sample ID	: XXW-1332OJH5%-0.8mL
Data File	: XXW-1332OJH5%-0.8mL.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram













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DDA	C1-1	220	
PDA	Cni	$220 \mathrm{nm}$	

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2						
	Peak#	Ret. Time	Area	Area%		
	1	17.707	1974865	96.571		
	2	20.829	70131	3.429		
	Total		2044996	100.000		
ľ						



# ¹H NMR of **3ao**, 600 MHz, CDCl₃



0.93-]

8.0

8.5

9.0

0.99⊸

7.5

3.86 ₄ 4.10 ₄

7.0









UV Spectrum











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PDA Ch1 220nm

I DI I CHII			
Peak#	Ret. Time	Area	Area%
1	5.660	1601673	50.479
2	6.123	1571307	49.521
Total		3172980	100.000





UV Spectrum Retention time = 5.655





UV Spectrum Retention time = 6.114



Peak Table

PDA Ch1	220mm		I that Inclu
Peak#	Ret. Time	Area	Area%
1	5.655	707705	7.458
2	6.114	8780960	92.542
Total		9488664	100.000


¹H NMR of **3ap**, 500 MHz, CDCl₃



















Peak Table

DDA CL1	220		
PDA Chi	220nm		
Peak#	Ret. Time	Area	Area%
1	26.697	9808145	50.009
2	29.768	9804518	49.991
Total	3	19612663	100.000















Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	26.776	124178	0.223
2	29.438	55656271	99.777
Total		55780449	100.000



¹H NMR of **3aq**, 500 MHz, CDCl₃









nm

Peak Table

PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	27.904	6764417	49.895
2	30.609	6793009	50.105
Total		13557426	100.000

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(±)-3aq

	Sample Information
Sample Name	: XXW-1330-IF-5%-0.8mL
Sample ID	: XXW-1330-IF-5%-0.8mL
Data File	: XXW-1330-IF-5%-0.8mL001.lcd
Method File	: XXW-5%.0.8.mL.lcm
	Chromatogram

H.



UV Spectrum Retention time = 27.883



UV Spectrum Retention time = 30.709



Peak Table

PDA Ch1 220nm

Peak#	Ret. Time	Area	Area%
1	27.883	582	0.008
2	30.709	6943317	99.992
Total		6943900	100.000



## ¹H NMR of **3ar**, 600 MHz, $CDCI_3$



	-150.19 $-147.32$ $133.27$ $123.42$ $127.92$ $127.42$ $127.42$	77.37 77.16 76.95	<ul> <li>32.58</li> <li>31.10</li> <li>26.52</li> <li>18.91</li> </ul>
--	----------------------------------------------------------------	-------------------------	------------------------------------------------------------------------

¹³C NMR of **3ar**, 150 MHz, CDCl₃













UV Spectrum Retention time = 30.656



Peak Table

PDA Ch1	220nm	1.61	
Peak#	Ret. Time	Area	Area%
1	28.487	1211416	50.016
2	30.656	1210651	49.984
Total		2422068	100.000





UV Spectrum Retention time = 28.816







Peak Table

PDA Ch1 220nm

Peak#	Ret. Time	Area	Area%
1	28.816	2833027	99.236
2	31.103	21823	0.764
Total		2854850	100.000

nm



¹H NMR of **3as**, 500 MHz, CDCl₃











#### Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	34.935	6709239	49.637
2	38.242	6807392	50.363
Total		13516630	100.000



UV Spectrum Retention time = 38.233



PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	35.336	54117	0.362
2	38.233	14908033	99.638
Total		14962150	100.000



¹H NMR of **3at**, 500 MHz, CDCl₃







	Sample Information
Sample Name	: XXW-0543-20%-ODH0.8mL
Sample ID	: XXW-0543-20%-ODH0.8mL
Data File	: XXW-0543-20%-ODH0.8mL001.lcd
Method File	: XXW-20%.0.8.mL.lcm
	Chromatogram









^{(Н}О

UV Spectrum Retention time = 31.054



Peak Table

PDA Ch1	220nm	103	
Peak#	Ret. Time	Area	Area%
1	26.918	4788624	50.285
2	31.054	4734386	49.715
Total		9523010	100.000

	Sample Information
Sample Name	: XXW-1329-ODH-20%-0.8mL
Sample ID	: XXW-1329-ODH-20%-0.8mL
Data File	: XXW-1329-ODH-20%-0.8mL001.lcd
Method File	: XXW-20%.0.8.mL.lcm
	Chromatogram



N 0 (-)-3at

^{.Н}О

UV Spectrum Retention time = 30.636



Peak Table

PDA Ch1	220nm		a management of the
Peak#	Ret. Time	Area	Area%
1	26.954	40610	0.377
2	30.636	10741180	99.623
Total		10781790	100.000





¹H NMR of **3au**, 500 MHz, CDCl₃





408

Ò



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	5.958	5970983	49.998
2	6.664	5971534	50.002
Total		11942518	100.000

(±)-3au





<b>D</b> 1		1 1
Daa	la i Lia	bla
FEA	K 12	11 21 5

Peak#	Ret. Time	Area	Area%
1	6.010	60687	1.504
2	6.754	3973479	98.496
Total		4034167	100.000





## ¹H NMR of **3av**, 500 MHz, $CDCI_3$









### ¹³C NMR of **3av**, 125 MHz, CDCl₃







Me



UV Spectrum Retention time = 5.238



UV Spectrum Retention time = 6.144

~ . . . .



Peak Table

PDA ChI	220nm		
Peak#	Ret. Time	Area	Area%
1	5.238	2738669	50.110
2	6.144	2726664	49.890
Total		5465333	100.000



UV Spectrum Retention time = 6.150

н

(–)-3av

Me



Peak Table

PDA Ch1	220nm		NACE AND ADDRESS
Peak#	Ret. Time	Area	Area%
1	5.239	48107	0.888
2	6.150	5370510	99.112
Total		5418617	100.000





## ¹H NMR of **3aw**, 500 MHz, $CDCI_3$



75 61 11	547 145 145 145 145 145 145 145 145 145 145
142. 142. 138.	128. 126. 125.
ΥÌ	



28.11 28.06 -21.57 -18.24

### ¹³C NMR of **3aw**, 125 MHz, CDCl₃





Peak Table

PDA Ch1	220nm	200	
Peak#	Ret. Time	Area	Area%
1	16.595	5843564	50.111
2	18.737	5817743	49.889
Total		11661307	100.000

н

Me



Peak Table

PDA ChI	220nm	1.00	
Peak#	Ret. Time	Area	Area%
1	16.531	15165667	98.574
2	19.146	219367	1.426
Total		15385034	100.000

01

н

Me









 belongs to another set of diastereomer



34 08 08	84 54 80 80 80 80 80 80 80 80 80 80 80 80 80
143. 140. 138.	126.129.129.129.125.125.125.125.125.125.125.125.125.125
111	



 $<^{26.27}_{26.07}$ -20.04 -16.49

# $^{13}\text{C}$ NMR of **3ax**, 150 MHz, CDCl_3









	Peak Table
DDA CL1	220

PDA Chi 220nm				
Peak#	Ret. Time	Area%		
1	7.539	49.983		
2	8.669	50.017		
Total		100.000		

XXW--1596-chiral-IB-0.1%0.8mL XXW-0.1%-0.8mL.lcm







1.555 98.445

100.000

Dool	Tabl	0
Peak	1401	e

DA Ch1	220nm	
Peak#	Ret. Time	Area%
1	7.447	1.5
2	8.560	98.4

DI .

Total




#### ¹H NMR of **3ay**, 500 MHz, CDCl₃







∠26.46 25.56 25.56 25.56 216.75 214.95

#### ¹³C NMR of **3ay**, 125 MHz, CDCl₃





mAU

Sample ID

Data File



UV Spectrum Retention time = 13.655







Peak Table

PDA Ch1	220nm	1017	
Peak#	Ret. Time	Area	Area%
1	13.655	3596346	49.997
2	14.958	3596797	50.003
Total		7193144	100.000



nm

Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	13.206	5142912	94.402
2	14.409	304946	5.598
Total		5447858	100.000

Н

`Et (-)-3ay

H

828888888888888888888888888888888888888	15 117
	<u> </u>



¹H NMR of **3az**, 500 MHz, CDCl₃





#### ¹³C NMR of **3az**, 125 MHz, CDCl₃





Peak Table

PDA Ch1 235nm

MeO

Peak#	Ret. Time	Area	Area%
1	7.427	9240579	50.174
2	8.734	9176390	49.826
Total		18416969	100.000



mAU

MeO







UV Spectrum Retention time = 8.367



Peak Table

PDA Ch1	235nm		
Peak#	Ret. Time	Area	Area%
1	7.162	52997	1.340
2	8.367	3900841	98.660
Total		3953838	100.000





#### ¹H NMR of **3ba**, 500 MHz, CDCl₃



 belongs to another set of diastereomer





¹⁹F NMR of **3ba**, 470 MHz, CDCl₃



15 10 5 0 -5 -10 -15 -20 -25 -30 -35 -40 -45 -50 -55 -60 -65 -70 -75 -80 -85 -90 -95 -100 -105 -110 -115 -120 f1 (ppm)



Peak Table

PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	5.739	7671092	50.039
2	6.663	7659156	49.961
Total		15330249	100.000

-----

н

F₃C



Peak Table

STATES AND			I cuit Inoie
PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	5.763	39966	0.684
2	6.699	5807209	99.316
Total		5847175	100.000

F₃C





#### ¹H NMR of **3bb**, 500 MHz, CDCl₃

ͺH



н









#### ¹³C NMR of **3bb**, 125 MHz, CDCl₃





Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	8.494	1705435	49.778
2	14.469	1720681	50.222
Total		3426116	100.000

MeO₂C



m

Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	8.491	41813	1.358
2	14.449	3037232	98.642
Total		3079045	100.000

ͺH

(–)-3bb

MeO₂C





#### ¹H NMR of **3bc**, 600 MHz, CDCl₃





# — 18.68

#### ¹³C NMR of **3bc**, 150 MHz, CDCl₃









UV Spectrum Retention time = 13.496



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	12.594	6668980	49.943
2	13.496	6684175	50.057
Total		13353156	100.000



mAU



UV Spectrum Retention time = 12.283



UV Spectrum Retention time = 13.173



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	12.283	91039	2.085
2	13.173	4275851	97.915
Total		4366890	100.000





#### ¹H NMR of **3bd**, 600 MHz, CDCl₃







~28.05 ~27.01 —17.24

## ¹³C NMR of **3bd**, 150 MHz, CDCl₃





mAU

н



UV Spectrum Retention time = 14.439







Peak Table

			I Can Table
PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	14.439	20353407	49.857
2	16.537	20470032	50.143
Total		40823439	100.000





UV Spectrum Retention time = 14,574



#### UV Spectrum Retention time = 17.121



Peak Table

PDA Ch1	220nm		a Madubian Nordera
Peak#	Ret. Time	Area	Area%
1	14.574	2880627	96.646
2	17.121	99983	3.354
Total		2980610	100.000





¹H NMR of **3be**, 600 MHz, CDCl₃



 belongs to another set of diastereomer











Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	19.474	3782776	50.107
2	23.244	3766621	49.893
Total		7549397	100.000

Н

Ph







H

(–)-3be

н

Ph



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	19.922	461588	1.102
2	22.977	41436145	98.898
Total		41897732	100.000

nm



#### ¹H NMR of **3bf**, 600 MHz, CDCl₃







Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	7.114	794998	50.130
2	8.307	790866	49.870
Total		1585864	100.000

454



#### Peak Table

PDA Ch1 220nm			
Peak#	Ret. Time	Area	Area%
1	6.814	46015	1.582
2	7.958	2862933	98.418
Total		2908948	100.000

----

Н

ͺH

(–)-3bf



--6.09

1.49
1.25
1.24
1.24
0.98
0.47

¹H NMR of **5**, 400 MHz,  $CDCI_3$ 









1.55
 1.28
 1.28
 0.99
 0.99
 0.45

### ¹H NMR of **6**, 500 MHz, CDCl₃






#### **CHAPTER 3**

## ENANTIOSELECTIVE METALLORADICAL 1,6-C–H ALKYLATION OF IN SITU-GENERATED ALKYLDIAZOMETHANES FOR SYNTHESIS OF CHIRAL PIPERIDINES

#### **3.1. INTRODUCTION**

The past decades have witnessed renaissance in radical reactions owing to their rich reactivities and unique attributes.¹ In particular, hydrogen atom abstraction (HAA) by free alkyl radicals has been demonstrated with the potential for direct functionalization of ubiquitous C(sp³)–H bond in organic molecules, offering a new strategy for C–C bond formation.² Despite significant advancements, development of HAA-mediated C–H alkylation have been hampered by several longstanding challenges that are associated with controlling reactivity and stereoselectivity of highly reactive radical intermediates. To address this enduring issue, metalloradical catalysis (MRC) offers a conceptually new approach for governing reactivity and controlling selectivity of radical processes through

¹ (a) Zard, S. Z. *Radical Reactions in Organic Synthesis*; Oxford University Press, 2003. (b) Chatgilialoglu, C.; Studer, A., *Encyclopedia of Radicals in Chemistry, Biology, and Materials*; John Wiley & Sons, 2012. For selected reviews, see: (c) Zard, S. Z. *Chem. Soc. Rev.* **2008**, *37*, 1603–1618. (d) Narayanam, J. M.; Stephenson, C. R. *Chem. Soc. Rev.* **2011**, *40*, 102–113. (e) Quiclet-Sire, B.; Zard, S. Z. *Pure Appl. Chem.* **2011**, *83*, 519–551. (f) Prier, C. K.; Rankic, D. A.; MacMillan, D. W. *Chem. Rev.* **2013**, *113*, 5322–5363. (g) Studer, A.; Curran, D. P. *Angew. Chem., Int. Ed.* **2016**, *55*, 58–102.

² (a) Choi, G. J.; Zhu, Q.; Miller, D. C.; Gu, C. J.; Knowles, R. R. *Nature* 2016, *539*, 268–271. (b) Chu, J. C. K.; Rovis, T. *Nature* 2016, *539*, 272–275. (c) Zhang, W.; Wang, F.; McCann, S. D.; Wang, D.; Chen, P.; Stahl, S. S.; Liu, G. *Science* 2016, *353*, 1014–1018. (d) Hu, X.-Q.; Chen, J.-R.; Xiao, W.-J. *Angew. Chem., Int. Ed.* 2017, *56*, 1960–1962. (e) Lu, Q.; Glorius, F. *Angew. Chem., Int. Ed.* 2017, *56*, 49–51. (f) Burg, F.; Gicquel, M.; Breitenlechner, S.; Pöthig, A.; Bach, T. *Angew. Chem., Int. Ed.* 2018, *57*, 2953–2957. (g) Stateman, L. M.; Nakafuku, K. M.; Nagib, D. A. *Synthesis* 2018, *50*, 1569–1586. (h) Nakafuku, K. M.; Zhang, Z.; Wappes, E. A.; Stateman, L. M.; Chen, A. D.; Nagib, D. A. *Nat. Chem.* 2020, *12*, 697–704. (i) Sarkar, S.; Cheung, K. P. S.; Gevorgyan, V. *Chem. Sci.* 2020, *11*, 12974–12993. (j) Zhang, C.; Li, Z. L.; Gu, Q. S.; Liu, X. Y. *Nat. Commun.* 2021, *12*, 475.

catalytic initiation and regulation of metal-stabilized organic radicals.^{3,4,5} As stable 15e metalloradicals, Co(II) complexes of  $D_2$ -symmetric chiral amidoporphyrins [Co( $D_2$ -Por^{*})] exhibit unique capability of homolytically activating diazo compounds as radial precursors to generate  $\alpha$ -Co(III)-alkyl radicals.⁶ These Co-stabilized C-centered radicals can serve as key catalytic intermediates to engage in radical addition and hydrogen atom abstraction as well as subsequent radical substitution, leading to the development of novel catalytic

³ For selected reviews and highlights on Co(II)-based MRC, see: (a) Lu, H. J.; Zhang, X. P. *Chem. Soc. Rev.* **2011**, *40*, 1899–1909. (b) Pellissier, H.; Clavier, H. *Chem. Rev.* **2014**, *114*, 2775–2823. (c) Demarteau, J.; Debuigne, A.; Detrembleur, C. *Chem. Rev.* **2019**, *119*, 6906–6955. (d) Huang, H.-M.; Garduño-Castro, M. H.; Morrill, C.; Procter, D. J. *Chem. Soc. Rev.* **2019**, *48*, 4626–4638. (e) Singh, R.; Mukherjee, A. *ACS Catal.* **2019**, *9*, 3604–3617.

⁴ For selected examples of Ti(III)-based radical processes, see: (a) Nugent, W. A.; RajanBabu, T. V. *J. Am. Chem. Soc.* **1988**, *110*, 8561–8562. (b) Rajanbabu, T. V.; Nugent, W. A. *J. Am. Chem. Soc.* **1994**, *116*, 986–997. (c) Gansäuer, A.; Hildebrandt, S.; Michelmann, A.; Dahmen, T.; von Laufenberg, D.; Kube, C.; Fianu, G. D.; Flowers II, R. A. *Angew. Chem., Int. Ed.* **2015**, *54*, 7003–7006. (d) Funken, N.; Mühlhaus, F.; Gansäuer, A. *Angew. Chem., Int. Ed.* **2016**, *55*, 12030–12034. (e) Hao, W.; Wu, X.; Sun, J. Z.; Siu, J. C.; MacMillan, S. N.; Lin, S. *J. Am. Chem. Soc.* **2017**, *139*, 12141–12144. (f) Yao, C. B.; Dahmen, T.; Gansäuer, A.; Norton, J. *Science* **2019**, *364*, 764–767. (g) Ye, K. Y.; McCallum, T.; Lin, S. *J. Am. Chem. Soc.* **2019**, *141*, 9548–9554. (f) Roy, S.; Das, S. K.; Khatua, H.; Das, S.; Chattopadhyay, B. Acc. Chem. Res. **2021**, *54*, 4395–4409.

⁵ For selected examples of metalloradical-mediated radical processes, see: (a) Wayland, B. B.; Poszmik, G.; Mukerjee, S. L.; Fryd, M. J. Am. Chem. Soc. **1994**, *116*, 7943–7944. (b) Zhang, X.-X.; Wayland, B. B. J. Am. Chem. Soc. **1994**, *116*, 7897–7898. (c) Chan, K. S.; Li, X. Z.; Dzik, W. I.; de Bruin, B. J. Am. Chem. Soc. **2008**, *130*, 2051–2061. (d) Chan, Y. W.; Chan, K. S. J. Am. Chem. Soc. **2010**, *132*, 6920–6922. (e) Li, G.; Han, A.; Pulling, M. E.; Estes, D. P.; Norton, J. R. J. Am. Chem. Soc. **2012**, *134*, 14662–14665. (f) Kuo, J. L.; Hartung, J.; Han, A.; Norton, J. R. J. Am. Chem. Soc. **2015**, *137*, 1036–1039. (g) Roy, S.; Khatua, H.; Das, S. K.; Chattopadhyay, B. Angew. Chem., Int. Ed. **2019**, *58*, 11439–11443. (h) Das, S. K.; Roy, S.; Khatua, H.; Chattopadhyay, B. J. Am. Chem. Soc. **2020**, *142*, 16211–16217. (i) Zhang, Z.; Gevorgyan, V. Org. Lett. **2020**, *22*, 8500–8504. (j) Roy, S.; Das, S. K.; Khatua, H.; Das, S.; Singh, K. N.; Chattopadhyay, B. Angew. Chem., Int. Ed. **2021**, *160*, 8772–8780.

⁶ (a) Dzik, W. I.; Xu, X.; Zhang, X. P.; Reek, J. N. H.; de Bruin, B. J. Am. Chem. Soc. 2010, 132, 10891–10902. (b) Belof, J. L.; Cioce, C. R.; Xu, X.; Zhang, X. P.; Space, B.; Woodcock, H. L. Organometallics 2011, 30, 2739–2746. (c) Lu, H.; Dzik, W. I.; Xu, X.; Wojtas, L.; de Bruin, B.; Zhang, X. P. J. Am. Chem. Soc. 2011, 133, 8518–8521. (d) Wang, Y.; Wen, X.; Cui, X.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2017, 139, 1049–1052. (e) Lee, W.-C. C.; Wang, D.-S.; Zhang, C.; Xie, J.; Li, B.; Zhang, X. P. J. Am. Chem 2021, 7, 1588–1601. (f) Wang, X.; Ke, J.; Zhu, Y.; Deb, A.; Xu, Y.; Zhang, X. P. J. Am. Chem. Soc. 2021, 143, 11121–11129. (g) Xie, J.; Xu, P.; Zhu, Y.; Wang, J.; Lee, W.-C. C.; Zhang, X. P. J. Am. Chem. Soc. 2021, 143, 11670–11678. (h) Zhang, C.; Wang, D.-S.; Lee, W.-C. C.; McKillop, A. M.; Zhang, X. P. J. Am. Chem. Soc. 2021, 143, 11130–11140. (i) Zhou, M.; Wolzak, L. A.; Li, Z.; de Zwart, F. J.; Mathew, S.; de Bruin, B. J. Am. Chem. Soc. 2021, 143, 20501–20512.

processes for stereoselective radical transformations.^{6d-h,7} Specifically, Co(II)-based MRC has been demonstrated as a potentially general strategy for stereoselective synthesis of cyclic molecules via new radical cyclization pathway that is based on asymmetric intramolecular radical C–H alkylation of diazo compounds (Scheme 3.1).^{6g,8}

Scheme 3.1. Enantioselective Radical C–H Alkylation via Co(II)-MRC



To harness the potential of this new radical cyclization strategy beyond synthesis of 5membered⁸ and 4-membered^{6g} cyclic compounds (Scheme 3.2), we sought to explore the possibility of constructing 6-membered ring structures, such as the important *N*heterocyclic compound piperidines **2**, by asymmetric 1,6-C–H alkylation of linear  $\alpha$ alkyldiazomethanes **1**' through in situ-generation from the corresponding 4-aminobutanalderived hydrazones **1** in the presence of base (Scheme 3.3).

⁷ (a) Chen, Y.; Fields, K. B.; Zhang, X. P. J. Am. Chem. Soc. 2004, 126, 14718–14719. (b) Caselli, A.; Gallo, E.; Ragaini, F.; Ricatto, F.; Abbiati, G.; Cenini, S. Inorg. Chim. Acta. 2006, 359, 2924–2932. (c) Chen, Y.; Ruppel, J. V.; Zhang, X. P. J. Am. Chem. Soc. 2007, 129, 12074–12075. (d) Fantauzzi, S.; Gallo, E.; Rose, E.; Raoul, N.; Caselli, A.; Issa, S.; Ragaini, F.; Cenini, S. Organometallics 2008, 27, 6143–6151. (e) Zhu, S.; Ruppel, J. V.; Lu, H.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2008, 130, 5042–5043. (f) Zhu, S.; Xu, X.; Perman, J. A.; Zhang, X. P. J. Am. Chem. Soc. 2010, 132, 12796–12799. (g) Xu, X.; Lu, H. J.; Ruppel, J. V.; Cui, X.; de Mesa, S. L.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2011, 133, 15292–15295. (h) Paul, N. D.; Mandal, S.; Otte, M.; Cui, X.; Zhang, X. P.; de Bruin, B. J. Am. Chem. Soc. 2014, 136, 1090–1096. (i) Reddy, A. R.; Hao, F.; Wu, K.; Zhou, C. Y.; Che, C. M. Angew. Chem., Int. Ed. 2016, 55, 1810–1815. (j) Chirila, A.; Gopal Das, B.; Paul, N. D.; de Bruin, B. ChemCatChem 2017, 9, 1413–1421. (k) Roy, S.; Das, S. K.; Chattopadhyay, B. Angew. Chem., Int. Ed. 2018, 57, 2238–2243.

⁸ (a) Cui, X.; Xu, X.; Jin, L.-M.; Wojtas, L.; Zhang, X. P. *Chem. Sci.* **2015**, *6*, 1219–1224. (b) Wang, Y.; Wen, X.; Cui, X.; Zhang, X. P. J. Am. Chem. Soc. **2018**, *140*, 4792–4796. (c) Wen, X.; Wang, Y.; Zhang, X. P. *Chem. Sci.* **2018**, *9*, 5082–5086.

$\sim \sim $	Scheme 3.2. Co	(II)	-Based	Meta	lloradi	cal 1	.4-	and	1.5	5-C-	-H	Alk	vlatio
---------------------------------------------------------------------------------------------------	----------------	------	--------	------	---------	-------	-----	-----	-----	------	----	-----	--------



However, the intended synthesis of piperidines 2 by Co(II)-based catalytic system presented several potential challenges. In addition to the prerequisite for matching the rates between metalloradical activation of alkyldiazomethanes 1' and their in situ-generation from hydrazones 1, it was uncertain whether the resulting  $\alpha$ -Co(III)-alkyl radicals I with C-H bonds at various positions could undergo competitive 1.6-HAA to generate the desired  $\zeta$ -Co(III)-aminoalkyl radical II over potential 1,2, 1,3- and 1,4-HAA that lead to side reactions. This uncertainty came from the consideration that the 7-membred cyclic transition state associated with 1,6-HAA is entropically less favorable. Additional challenges could arise in the subsequent step of radical substitution of  $\zeta$ -Co(III)-alkyl radicals II for formation of piperidines 2. If the 6-exo-tet radical cyclization of  $\zeta$ -Co(III)alkyl radicals II fails to compete with typically facile 1,5-HAA, it would result in the generation of  $\beta$ -Co(III)-alkyl radicals III, which would lead to formation of olefin 3 as side products via radical β-scission. Besides the concerns with controlling of reactivity, whether the ensuing C-C bond formation during radical cyclization could be rendered enantioselective is another unanswered question.

NH₂NHSO₂R³  $[Co(D_2-Por)]$ C Ŕ base NNHSO₂R" 1 4-aminobutanals Н base 1 Co Ν I R'  $N_2$ 1' metalloradical activation Co(II)-Metalloradicals l R substitution radical. 2 SCISS  $N_2$ radical 'n R 3 R' R Н н 1,5 H-atom abstraction н Co Co β-Co(III)-Alkyl [Co] ζ-Co(III)α-Co(III)-Radicals (III) Aminoalkyl Alkyl Radicals (II) Radicals (I) 1.6-H-atom abstraction

Scheme 3.3. Working Proposal for Stereoselective Synthesis of Piperidines by Metalloradical 1,6-C–H Alkylation

In view of the conformational flexibility associated with the linear methylene chain in  $\zeta$ -Co(III)-alkyl radicals **II**, what elements could be exploited to differentiate the two prochiral radical faces at the remote  $\zeta$ -position in order to induce high asymmetry during the course of 6-*exo-tet* radical cyclization? We reasoned that the key to address these and related challenges is to identify a suitable  $D_2$ - symmetric chiral amidoporphyrin ligand ( $D_2$ -Por^{*}) with proper electronic, steric, and chiral environments that could govern the stereochemical course of the Co(II)-based radical processes as proposed. If realized, it would lead to the development of a new catalytic method for enantioselective synthesis of

piperidines,⁹ which exist ubiquitously in natural products and bioactive compounds (Figure 3.1), from readily accessible 4-aminobutanal derivatives.

Figure 3.1. Selected Examples of Bioactive Compounds Containing Piperidines



Catalytic asymmetric intramolecular 1,6-C–H alkylation of diazo compounds represents an attractive approach for stereoselective construction of chiral 6-membered cyclic compounds.¹⁰ Despite considerable advancements, there have been only a few reports on

⁹ (a) Fujita, K.-i.; Fujii, T.; Yamaguchi, R. *Org. Lett.* 2004, *6*, 3525–3528. (b) Pastine, S. J.; Gribkov, D. V.; Sames, D. *J. Am. Chem. Soc.* 2006, *128*, 14220–14221. (c) Coldham, I.; Leonori, D. *Org. Lett.* 2008, *10*, 3923–3925. (d) Beng, T. K.; Gawley, R. E. *J. Am. Chem. Soc.* 2010, *132*, 12216–12217. (e) Stead, D.; Carbone, G.; O'Brien, P.; Campos, K. R.; Coldham, I.; Sanderson, A. *J. Am. Chem. Soc.* 2010, *132*, 7260–7261. (f) Yoshikai, N.; Mieczkowski, A.; Matsumoto, A.; Ilies, L.; Nakamura, E. *J. Am. Chem. Soc.* 2010, *132*, 5568–5569. (g) Cui, Z.; Yu, H.-J.; Yang, R.-F.; Gao, W.-Y.; Feng, C.-G.; Lin, G.-Q. *J. Am. Chem. Soc.* 2011, *133*, 12394–12397. (h) McNally, A.; Prier, C. K.; MacMillan, D. W. C. *Science* 2011, *334*, 1114–1117. (i) Qu, B.; Mangunuru, H. P. R.; Tcyrulnikov, S.; Rivalti, D.; Zatolochnaya, O. V.; Kurouski, D.; Radomkit, S.; Biswas, S.; Karyakarte, S.; Fandrick, K. R.; Sieber, J. D.; Rodriguez, S.; Desrosiers, J.-N.; Haddad, N.; McKellop, K.; Pennino, S.; Lee, H.; Yee, N. K.; Song, J. J.; Kozlowski, M. C.; Senanayake, C. H. *Org. Lett.* 2018, *20*, 1333–1337. (j) Boddy, A. J.; Affron, D. P.; Cordier, C. J.; Rivers, E. L.; Spivey, A. C.; Bull, J. A. *Angew. Chem. Int. Ed.* 2019, *58*, 1458–1462.

¹⁰ (a) Lee, E.; Choi, I.; Song, S. Y. *J. Chem. Soc., Chem. Commun.* **1995**, 321–322. (b) Rosales, A.; Rodríguez-García, I.; López-Sánchez, C.; Álvarez-Corral, M.; Muñoz-Dorado, M. *Tetrahedron* **2011**, *67*, 3071–3075. (c) Taber, D. F.; Paquette, C. M.; Gu, P.; Tian, W. *J. Org. Chem.* **2013**, *78*, 9772–9780. (d) Bergstrom, B. D.; Nickerson, L. A.; Shaw, J. T.; Souza, L. W. Angew. Chem. Int. Ed. **2021**, *60*, 6864–6878.

# Scheme 3.4. Rh₂-Catalyzed Asymmetric 1,6-C–H Alkylation for Synthesis of Tetrahydroisoquinolines

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metal-based catalytic systems for asymmetric 1,6-C–H alkylation. ¹¹ In addition to moderate enantioselectivity, the existing catalytic systems has been largely limited to the formation of oxygen-containing ring structures. To date, Shaw and coworkers reported the only notable example of asymmetric 1,6-C–H alkylation to form nitrogen-containing 6-membered heterocycles based on a Rh₂-catalytic system involving donor/donor-substituted diazo compounds (Scheme 3.4).¹²

¹¹ (a) McCarthy, N.; McKervey, M. A.; Ye, T.; McCann, M.; Murphy, E.; Doyle, M. P. *Tetrahedron Lett.* **1992**, *33*, 5983–5986. (b) McKervey, M. A.; Ye, T. *J. Chem. Soc., Chem. Commun.* **1992**, 823–824. (c) Ye, T.; García, C. F.; McKervey, M. A. *J. Chem. Soc., Perkin Trans. I* **1995**, 1373–1379. (d) Slattery, C. N.; Clarke, L.-A.; O'Neill, S.; Ring, A.; Ford, A.; Maguire, A. R. *Synlett* **2012**, *23*, 765–767. (e) Ito, M.; Kondo, Y.; Nambu, H.; Anada, M.; Takeda, K.; Hashimoto, S. *Tetrahedron Lett.* **2015**, *56*, 1397–1400. (f) Anada, M.; Hashimoto, S.; Ito, M.; Kondo, Y.; Namie, R.; Natori, Y.; Takeda, K.; Nambu, H.; Yamamoto, Y. *Heterocycles* **2021**, *103*, 1078–1098.

¹² Nickerson, L. A.; Bergstrom, B. D.; Gao, M.; Shiue, Y.-S.; Laconsay, C. J.; Culberson, M. R.; Knauss, W. A.; Fettinger, J. C.; Tantillo, D. J.; Shaw, J. T. *Chem. Sci.* **2020**, *11*, 494–498.

#### Scheme 3.5. Potential Challenges of Intramolecular 1,6-C–H Alkylation Involving

#### **Electrophilic Metallocarbenes**

Kinetic Preference of 5-Membered Ring Insertion



This underdevelopment is presumably attributed to the kinetic preference for 1,5-C–H alkylation and inherent propensity of electrophilic metallocarbene intermediate toward the formation of cyclic ylide by nucleophilic attack of the Lewis basic nitrogen atom (Scheme 3.5).^{12,13,14} Evidently, a new catalytic system that is fundamentally different from the existing ionic systems involving electrophilic metallocarbene intermediates would be highly appealing to allow selective C–H alkylation without complication by the heteroatoms. To circumvent the problem, de Bruin and coworkers recently disclosed a [Co(TPP)]-catalyzed (TPP = 5,10,15,20-tetraphenylporphyrin) metalloradical alkylation of  $\alpha$ -alkyldiazomethanes for synthesis of 6-membered *N*-heterocyclic piperidines (Scheme 3.6).¹⁵ However, the enantioselective control of this transformation remains to be addressed.

¹³ Estevan, F.; Herbst, K.; Lahuerta, P.; Barberis, M.; Pérez-Prieto, J. Organometallics 2001, 20, 950–957.

¹⁴ Doyle, M. P.; Ene, D. G.; Forbes, D. C.; Tedrow, J. S. *Tetrahedron Lett.* **1997**, *38*, 4367–4370.

¹⁵ Lankelma, M.; Olivares, A. M.; de Bruin, B. Chem. Eur. J. 2019, 25, 5658–5663.

#### Scheme 3.6. [Co(TPP)]-Catalyzed Formation of Substituted Piperidines





As an exciting application of Co(II)-MRC, we herein report the development of the first catalytic system that is highly efficient for asymmetric 1,6-C–H alkylation of  $\alpha$ -alkyldiazomethanes to construct chiral piperidines. Supported by an optimal  $D_2$ -symmetric chiral amidoporphyrin ligand, the Co(II)-based metalloradical system can activate a wide array of  $\alpha$ -alkyldiazomethanes with varied steric and electronic properties for intramolecular radical 1,6-C–H alkylation, enabling the synthesis of  $\alpha$ -substituted piperidines in good to high yields with excellent enantioselectivities. Furthermore, the radical alkylation process features high degree of functional group tolerance as well as compatibility of heteroaryl units. We also present detailed computational and experimental studies that have shed light on the underlying stepwise radical mechanism that consists of facile 1,6-HAA and 6-*exo-tet* radical substitution (RS) as the key steps.

#### **3.2. RESULTS AND DISCUSSION**

### 3.2.1. Condition Optimization for Co(Π)-Catalyzed 1,6-C–H Alkylation of α-Alkyldiazomethanes

At the outset of this project, 4-aminobutanal-derived trishydrazone (1a) was chosen as the model substrate to examine the feasibility of the proposed radical 1,6-C–H alkylation process by Co(II)-based metalloradical catalysts [Co(Por)] (Figure 3.2). It was gratifying to find that first-generation chiral metalloradical catalyst [Co(P3)] (P3 = 3,5-Di'Bu-

ChenPhyrin)^{7a} could effectively catalyze the C-H alkylation reaction, even at room temperature, affording the desired 2-phenylpiperidine (2a) in high yield (84%) with a low but significant level of asymmetric induction (9% ee) (Figure 3.2; entry 1). Switching to second-generation metalloradical catalyst [Co(P6)] (P6 = 3.5-Di⁷Bu-QingPhyrin)^{7g} containing phenyl groups in the chiral amide units resulted in some improvement in enantioselectivity (24% ee) but led to considerable decrease in reactivity (57% yield) (entry 2). Subsequent employment of naphthyl-substituted metalloradical catalyst [Co(P7)] (P7) = 3.5-Di^{*i*}Bu-Xu(2'-Naph)Phyrin)¹⁶ gave rise to higher yield (64%) without improving enantioselectivity (21% ee) (entry 3). When we turned our attention to [Co(P15)] (P15 = 3,5-Di'Bu-ZhuPhyrin)¹⁷ bearing rigid tetrahydrofurancarboxamide units, further increase in the yield (67%) of **2a** was attained albeit with lower enantioselectivity (17% ee) (entry 4). Considering the relative flexibility of aliphatic chain in **1a**, we decided to increase the steric bulkiness of the ligand environment by fine-tuning the nonchiral meso-aryl substituents of ZhuPhyrin ligand from 3,5-di-tert-butylphenyl to more sterically encumbered 2,6-dimethoxyphenyl groups. To our delight, the analogous catalyst [Co(P5)]  $(P5 = 2,6-DiMeO-ZhuPhyrin)^{17}$  could catalyze the alkylation reaction with substantially enhanced enantioselectivity (79% ee) in higher yield (69%) (entry 5). This ligand buttressing effect promoted us to evaluate [Co(P16)] (P16 = 2,4,6-TriMe-ZhuPhyrin)^{8b} containing even more sterically-demanding mesityl groups at achiral *meso*-aryl positions. Excitingly, the use of [Co(P6)] led to further improvement in both reactivity and enantioselectivity, allowing for the enantioselective construction of 6-membered Nheterocycle 2a in high yield (78%) with excellent enantioselectivity (92% ee) (entry 6).

¹⁶ Jin, L.-M.; Xu, X.; Lu, H.; Cui, X.; Wojtas, L.; Zhang, X. P. Angew. Chem. Int. Ed. **2013**, *52*, 5309–5313.

¹⁷ Zhu, S.; Ruppel, J. V.; Lu, H.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. **2008**, 130, 5042–5043.

Ph H Cbz	H [C Cs ₂ CO NNHTris 1a	<u>co(Por)] (5 mol %)</u> g; benzene; 22 °C; 48 h <b>2a</b>	$  + N_2  $
entry	[Co(Por)]	yield (%)	ee (%)
1	[Co( <b>P3</b> )]	84	9
2	[Co( <b>P6</b> )]	57	24
3	[Co( <b>P7</b> )]	64	21
4	[Co( <b>P15</b> )]	66	17
5	[Co( <b>P5</b> )]	69	79
6	[Co( <b>P16</b> )]	78	92
Me N-H N-H Me Me Me	Me Me Me Me NH N-H N-H N-H N-H N-H N-H N-H N-H N-H	H = 0 $H = 0$ $H =$	$ \begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$
[Co( <b>P3</b> )] ( <b>P3</b> = 3,5-Di ^t Bu-Ch	nenPhyrin) ( <b>P6</b> = 3	[Co( <b>P6</b> )] ,5-Di ^f Bu-QingPhyrin) ( <b>P7</b> = 3	[Co( <b>P7</b> )] 5,5-Di ¹ Bu-Xu(2'-Naph)Phyrin)

Figure 3.2. Ligand Effect on Co(II)-Based Catalytic System for Enantioselective Construction of  $\alpha$ -Piperidines^{*a*}

^aCarried out with **1a** (0.10 mmol) using [Co(Por)] (5 mol %) in the presence of  $Cs_2CO_3$  (0.20 mmol) in benzene (2.0 mL) at 22 °C for 48 h; Isolated yields; Determined by chiral HPLC analysis; Tris = (2,4,6-triisopropyl)benzene sulfonyl.

[Co(**P5**)]

(P5 = 2,6-DiMeO-ZhuPhyrin)

[Co(**P16**)]

(**P16** = 2,4,6-TriMe-ZhuPhyrin)

OMe

H MeO

[Co(**P15**)]

 $(P15 = 3,5-Di^tBu-ZhuPhyrin)$ 

The dramatic difference in catalytic performance between [Co(P5)] and [Co(P16)] revealed the immense power of judicious tuning of ligand environment in controlling reactivity and enantioselectivity of Co(II)-based metalloradical systems.

During the course of our investigation, we also conducted the C–H alkylation reactions with 4-aminobutanal-derived tosylhydrazone (1aa) (Table 3.1). At elevated temperature of 80 °C, the alkylation reaction afforded 2a in high yield (86%) but with diminished enantioselectivity (78% ee) (Table 3.1; entry 1). When the reaction was carried out 60 °C, a slight improvement in enantioselectivity (82% ee) was observed without affecting the yield (84%) (entry 2). Further lowering the reaction temperature to 40 °C led to significant decrease in yield due to incomplete decomposition of tosylhydrazone 1aa, forming alkylation product 2a in 26% yield with 86% ee (entry 3). Subsequent use of the corresponding trishydrazone 1a allowed the reaction to proceed even at room temperature, resulting in the formation of 2a in moderate yield (51%) with improved enantioselectivity (92%). When the reaction time was extended to 48 h with 5 mol % of [Co(P16)], the desired product 2a was isolated in 78% yield while maintaining the same high level of enantioselectivity (92%) (entry 5). Along with 2-phenylpiperidine (2a), we also observed the formation of linear alkene 3a as a side product in small amounts (Table 3.1).



Table 3.1. Effect of Different Sulfonyl Hydrazones on Enantioselective Radical 1,6-



^{*a*}Carried out with **1a** (0.10 mmol) using [Co(**P16**)] (2 mol %) in the presence of Cs₂CO₃ (0.20 mmol) in benzene (2.0 mL) for 24 h; Determined by ¹H NMR yield; Determined by chiral HPLC analysis. ^{*b*}Using [Co(**P16**)] (5 mol %) for 48 h. ^{*c*}Isolated yield.

## 3.2.2. Enantioselective Metalloradical 1,6-C–H Alkylation for Synthesis of α-Piperidines

Under the optimized conditions, the scope of the [Co(P16)]-catalyzed intramolecular 1,6-C–H alkylation was evaluated by employing 4-aminobutanal-derived trishydrazone 1 containing different types of C–H bonds (Table 3.2). Like 1a (Table 3.2; entry 1), its derivatives containing various electron-donating substituents at different positions, including *m*-Me (1b), *p*-'Bu (1c), *m*-OMe (1d), *p*-OMe (1e), and 1',3'-dioxolane-3,4-fused

Table 3.2. Enantioselective Metalloradical 1,6-C-H Alkylation of Different Benzylic

#### C-H Bonds^a



^aCarried out with **1** (0.10 mmol) using [Co(**P16**)] (5 mol %) in the presence of  $Cs_2CO_3$  (0.20 mmol) in benzene (2.0 mL) at 22 °C for 48 h; Isolated yields; Enantiomeric excess (ee) determined by chiral HPLC; Tris = (2,4,6-triisopropyl)benzenesulfonyl. ^bPerformed on 1.0 mmol scale. ^cAbsolute configuration determined by X-ray crystallography.

(1f) could be efficiently alkylated by [Co(P16)] at benzylic C–H bonds, affording the corresponding piperidines 2b-2f in moderate to high yields with excellent enantioselectivities (entries 2–6). Similarly, the Co(II)-catalyzed 1,6-C–H alkylation system was shown to be compatible with substrates containing electron withdrawing aryl

substituents, such as *m*-NO₂ (**1g**), *p*-CN (**1h**), *p*-CO₂Me (**1i**), and *p*-CF₃ (**1j**), allowing for the effective formation of  $\alpha$ -piperidines **2g**–**2j** in similarly high enantioselectivities (entries 7–10). Furthermore, halogenated substrates, including *m*-F (**1k**), 3,4-*di*-F (**11**), *p*-Br (**1m**), and 3,4-*di*-Br (**1n**), could also successfully undergo the radical alkylation as exemplified by the enantioselective formation of **2k**–**2n** in varied yields (entries 11–14). In addition, the Co(II)-based alkylation was highlighted by its applicability to extended aromatic ring system, delivering napthyl-substituted piperidine **2o** in high yield with high enantioselectivity (entry 15). Gratifyingly, the Co(II)-based alkylation system could be readily scaled up under the same reaction condition as demonstrated with the similarly high-yielding synthesis of phenylpiperidine **2a** on 1.0 mmol scale with the same high level of enantioselectivity (entry 1).

Notably, the catalytic radical alkylation system was equally effective for substrates containing various heteroarenes, leading to the enantioselective construction of a wide array of  $\alpha$ -heteroarylpiperidines. For example, furan and thiophene-based precursors 1p-1r could be effectively activated by [Co(P16)] for alkylation to furnish the corresponding piperidines 2p-2r with high enantioselectivities despite lower yields (Table 3.3, entries 1–3). Likewise, pyridine-based precursors 1s-1u could serve as suitable C–H substrates as well, enabling convenient access to highly-enantioenriched  $\alpha$ -pyridyl piperidine derivatives 2s-2u (entries 4–6). It is noteworthy that the C–H alkylation could be applied to the enantioselective synthesis of naturally-occurring anabasine (2u) albeit in a considerably lower yield (entry 6). Moreover, chiral piperidines containing  $\alpha$ -benzothiophene (2v) and  $\alpha$ -quinoline (2w) could be effectively constructed in moderate to high yields with excellent enantioselectivities (entries 7–8).

# Table 3.3. Enantioselective Metalloradical 1,6-C–H Alkylation of Different Heterobenzylic C–H Bonds^a



^{*a*}Carried out with **1** (0.10 mmol) using [Co(**P16**)] (5 mol %) in the presence of Cs₂CO₃ (0.20 mmol) in benzene (2.0 mL) at 22 °C for 48 h; Isolated yields; Enantiomeric excess (ee) determined by chiral HPLC; Tris = (2,4,6-triisopropyl)benzenesulfonyl.

### 3.2.3. Mechanistic Studies on Co(II)-Catalyzed 1,6-C–H Alkylation of α-Alkyldiazomethanes

To gain insights into the proposed stepwise radical mechanism (Scheme 3.3), both computational and experimental studies was conducted. First, density functional theory (DFT) calculations were performed with collaboration of Jing Ke to elucidate the details of the catalytic pathway as well as the origin of asymmetric induction for 1,6-C–H alkylation reaction with alkyldiazomethane (1a') by [Co(P16)] (Scheme 3.6; see Experimental Section for details). The geometry optimizations were performed with the

Gaussian 16¹⁸ at the BP86^{19,20}/lanl2dz^{21,22} level of theory in the gas phase at room temperature. To further improve the accuracy of energies, single point energies were carried out at the B3LYP²³/def2-tzvp^{21,22} level of theory along with Grimme's dispersion correction²⁴(D3BJ) and SMD²⁵ solvation model (in benzene).



Scheme 3.6. DFT Study on Catalytic Pathway for 1,6-C-H Alkylation by [Co(P6)]^a

^aApplied BP86/LANL2DZ for geometry optimization and B3LYP/def2-tzvp for calculations of single point energies (kcal/mol) along with Grimme's dispersion correction and SMD (benzene) solvation model.

The DFT calculations revealed the formation of intermediate **B** between the catalyst and the diazomethane through multiple noncovalent attractive interactions, including Hbonding and  $\pi$ -stacking interactions. The noncovalent complexation, which is exergonic

¹⁸ Frisch, M. J. et al., Gaussian 16, Revision A.03, Gaussian, Inc., Wallingford CT, 2016.

¹⁹ Schultz, N. E.; Zhao, Yan; Truhlar, D. G. J. Phys. Chem. A 2005, 109, 11127.

²⁰ Suarez, A. I. O.; Jiang, H.-L.; Zhang, X. P.; de Bruin, B. Dalton Trans. 2011, 40, 5697–5705.

²¹ Lang, K.; Torker, S.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. **2019**, 141, 12388–12396.

²² Eichkorn, K.; Weigend, F.; Treutler, O.; Ahlrichs, R. Theor. Chem. Acc. 1997, 97, 119–124.

²³ Furche, F.; Perdew, J. P. J. Chem. Phys. 2006, 124, 044103.

²⁴ Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. J. Chem. Phys. 2010, 132, 154104.

²⁵ Marenich, A. V.; Cramer, C. J.; Truhlar, D. G. J. Phys. Chem. B 2009, 113, 6378–6396.

by 2.1 kcal/mol, positions the  $\alpha$ -carbon atom of diazo **1a'** in close proximity to the Co(II)metalloradical center of [Co(**P16**)] (C---Co: ~2.45 Å) for further interactions. The ensuing metalloradical activation, which is slightly endergonic by 0.5 kcal/mol, is found to be associated with a relatively high but accessible activation barrier ( $\Delta G^{\ddagger}_{TS1} = 17.5$  kcal/mol), affording  $\alpha$ -Co(III)-alkyl radical intermediate **C** with release of dinitrogen as the byproduct. Subsequent intramolecular 1,6-HAA of radical intermediate **C**, which is highly exergonic by 11.8 kcal/mol, has a lower activation barrier ( $\Delta G^{\ddagger}_{TS2} = 13.0$  kcal/mol), delivering  $\zeta$ -Co(III)-aminoalkyl radical intermediate **D**. The final step of 6-*exo-tet* cyclization of radical intermediate **D** via radical substitution exhibits a relatively lower activation barrier ( $\Delta G^{\ddagger}_{TS3} = 7.4$  kcal/mol), leading to the formation of piperidine product **2a** with regeneration of metalloradical catalyst [Co(**P16**)].

According to the DFT calculations, the H-atom abstraction (HAA) is an enantiodifferentiative process, where abstraction of the *pro-(R)* benzylic hydrogen of intermediate **C** is kinetically less favorable, proceeding through a higher activation barrier ( $\Delta G^{\ddagger}_{TS2}$  = 19.1 kcal/mol) than that of the *pro-(S)* benzylic hydrogen ( $\Delta G^{\ddagger}_{TS2}$  = 13.0 kcal/mol). Furthermore, the resulting *pro-(R)*  $\zeta$ -Co(III)-aminoalkyl radical intermediate **D**' also undergoes a higher activation barrier in the final step of 6-*exo-tet* cyclization ( $\Delta G^{\ddagger}_{TS3}$  = 13.3 kcal/mol vs.  $\Delta G^{\ddagger}_{TS3}$  = 7.4 kcal/mol), implying that radical substitution is likely the stereodetermining step. In addition, the energetics associated with competitive 1,5-HAA of  $\zeta$ -Co(III)-aminoalkyl radical **D** was also calculated to elucidate the formation of linear alkene **3a**, which reveals a higher activation barrier ( $\Delta G^{\ddagger}_{TS4}$  = 9.2 kcal/mol) than that of the desired cyclization pathway ( $\Delta G^{\ddagger}_{TS3}$  = 7.4 kcal/mol).

To experimentally detect Co(III)-supported alkyl radical intermediates, the reaction solution of [Co(TPP)] with hydrazone **1a** was monitored by electron paramagnetic resonance (EPR) spectroscopy at room temperature (Scheme 3.7, see Experimental Section for details).



Scheme 3.7. Detection of Co(III)-Alkyl Radical Intermediates by EPR and HRMS

The isotropic EPR spectrum displays diagnostic signals at g-value of ~2.00 that are characteristic of  $\alpha$ -Co(III)-alkyl radicals and related radical species. Consistent with the proposed stepwise radical mechanism involving the key step of 1,6-H-atom abstraction, the observed broad signals (in black) could be fittingly simulated (in red) as a mixture of  $\alpha$ -Co(III)-alkyl radical and  $\zeta$ -Co(III)-aminoalkyl radical on the basis of couplings by ⁵⁹Co

(I = 7/2), ¹⁴N (I = 1) and ¹H (I = 1/2): 23% of C-centered radical at  $\alpha$ -position  $\mathbf{I}_{[Co(TPP)]/1a}$  $(g = 2.00030; A_{(Co)} = 72.5 \text{ MHz}; A_{(H)} = 25.1 \text{ MHz})$  and 77% of C-centered radical at  $\zeta$ -position  $\mathbf{I}_{[Co(TPP)]/1a}$   $(g = 2.01109; A_{(N)} = 4.9 \text{ MHz}; A_{(H)} = 7.2 \text{ MHz})$ .

Moreover, the corresponding Co(II)-supported alkyl radicals from the reaction solution of hydrazone **1a** with [Co(TPP)] could be directly detected by high resolution mass spectrometry (HRMS) with electrospray ionization (ESI) in the absence of any additives as electron carriers (Scheme 3.7). The observed spectrum evidently reveals a signal corresponding to  $[C_{63}H_{49}CoN_5O_2]^+$  (m/z = 966.31720), which resulted from neutral  $\alpha$ -Co(III)-alkyl radical intermediate  $I_{[Co(TPP)]/1a}$  or  $\zeta$ -Co(III)-aminoalkyl radical intermediate  $II_{[Co(TPP)]/1a}$  by the loss of one electron. Both the exact mass and the pattern of isotope distribution determined by ESI-HRMS matches almost perfectly with those calculated as a mixture of three species with the formula of  $[C_{63}H_{48}CoN_5O_2]^+$ ,  $[C_{63}H_{49}CoN_5O_2]^+$ , and  $[C_{63}H_{50}CoN_5O_2]^+$ .

In addition, spin trapping reagent PBN (*N-tert*-butyl- $\alpha$ -phenylnitrone) was employed to trap I_{[Co(TPP)]/1a} or II_{[Co(TPP)]/1a} for experimental detection by EPR spectroscopy (Scheme 3.8, see Experimental Section for details). The isotropic EPR spectrum exhibits the characteristic triplet of doublet signals at *g*-value of ~2.00, which was taken as evidence for formation of III_{[Co(TPP)]/1a} and IV_{[Co(TPP)]/1a} resulting from PBN trapping of the initially generated I_{[Co(TPP)]/1a} and II_{[Co(TPP)]/1a}.^{8b} The observed spectrum (in black) could be fittingly simulated (in red) as a mixture of  $\alpha$ -Co(III)-alkyl radical and  $\zeta$ -Co(III)-aminoalkyl radical on the basis of the hyperfine couplings by ¹⁴N (*I* = 1) and ¹H (*I* = 1/2): 71% of O-centered radical III_{[Co(TPP)]/1a} from  $\alpha$ -Co(III)-alkyl radical I_{[Co(TPP)]/1a}: *g* = 2.00630; A_(N) = 41.9 MHz;  $A_{(H)} = 8.0 \text{ MHz}$  and 29% O-centered radical  $IV_{[Co(TPP)]/1a}$  from  $\zeta$ -Co(III)-aminoalkyl radical  $II_{[Co(TPP)]/1a}$ : g = 2.00638;  $A_{(N)} = 40.2 \text{ MHz}$ ;  $A_{(H)} = 8.3 \text{ MHz}$ .

Scheme 3.8. Trapping by Spin Trap PBN for EPR Observation



#### **3.3. CONCLUSIONS AND OUTLOOK**

In summary, we have demonstrated an effective Co(II)-based alkylation system for asymmetric 1,6-C–H alkylation of  $\alpha$ -alkyldiazomethanes, enabling the enantioselective construction of six-membered *N*-heterocyclic compounds. Supported by an optimal  $D_2$ symmetric chiral amidoporphyrin 2,4,6-TriMe-ZhuPhyrin, the Co(II)-catalyzed alkylation system can effectively activate a wide array of  $\alpha$ -alkyldiazomethanes containing C(sp³)–H bonds with varied steric and electronic properties at room temperature, providing access to chiral  $\alpha$ -substituted piperidines in good to high yields with high enantioselectivities. In addition, the metalloradical system is highlighted by its tolerance to different functional groups as well as compatibility with heteroaryl units. Furthermore, our combined computational and experimental studies revealed an underlying stepwise radical mechanism involving 1,6-HAA and 6-*exo-tet* radical substitution (RS) as the key steps. Considering the synthetic utility and biological importance of enantioenriched  $\alpha$ piperidines, we envision that the Co(II)-catalyzed asymmetric radical alkylation process incorporating 1,6-HAA will find wide applications in organic synthesis related to drug discovery.

#### **3.4. EXPERIMENTAL SECTION**

#### 3.4.1. General Considerations

All alkylation reactions were performed in anhydrous solvents under N₂ atmosphere in an oven-dried glassware following standard Schlenk techniques. Gas tight syringes were used to transfer liquid reagents and solvents in catalytic reactions. Solvent was freshly distilled/degassed prior to use unless otherwise noted. Thin layer chromatography was performed on Merck TLC plates (silica gel 60 F254). Flash column chromatography was performed with ICN silica gel (60 Å, 230-400 mesh, 32-63 µm). ¹H NMR spectra were acquired using Varian INOVA 400 (400 MHz), Bruker 500 (500 MHz), or Varian INOVA 600 (600 MHz) spectrometer. Chemical shifts were internally referenced to residual solvent peak (CHCl₃  $\delta$  = 7.26 ppm). Data were reported as follows: chemical shift (ppm), integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, p = pentet, hept = heptet, br = broad, m = multiplet), and coupling constants J (Hz). ¹³C NMR spectra were acquired using Varian INOVA 400 (100 MHz), Bruker 500 (125 MHz), or INOVA 600 (150 MHz) spectrometer with complete proton decoupling. Chemical shifts were reported in ppm with residual solvent peak (CDCl₃ $\delta$  = 77.16 ppm) as the internal standard. ¹⁹F NMR spectrum was acquired using Varian Bruker 500 (470 MHz) or INOVA 600 (564 MHz) spectrometer. Infrared spectra were measured with a Nicolet Avatar 320 spectrometer with a Smart Miracle accessory. Optical rotations were measured on a Rudolph Research Analytical AUTOPOL® IV digital polarimeter. HPLC measurements were carried out on a Shimadzu HPLC system with Chiralcel AD-H, IA, IC, ID, IE, and IF columns. Highresolution mass spectrometry (DART and ESI) was performed at the Mass Spectrometry Facility, Boston College, Chestnut Hill, MA. The X-ray diffraction data were collected using Bruker-AXS SMART-APEXII CCD diffractometer. All reagents were purchased either from Aldrich, Alfa Aesar, Acros, Ak Sci, Oakwood Chemicals, Strem Chemicals or TCI and were used without further purification.

#### 3.4.2. Synthesis and Characterization of Substrates

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#### 3.4.2.1. Experimental Procedure for Preparation of s1

$$R \stackrel{H}{\longrightarrow} O + H_2 N \stackrel{OH}{\longrightarrow} OH = \frac{1. \text{ MeOH; r.t.; overnight}}{2. \text{ NaBH}_4 (1.5 \text{ equiv}); 0 ^{\circ}C \text{ to r.t.}} R \stackrel{N}{\longrightarrow} H \stackrel{OH}{\longrightarrow} H$$

To a solution of aldehyde (10.0 mmol) in anhydrous methanol (20.0 mL) was added 4amino-1- propanol (1.0 equiv) dropwise at room temperature. The reaction was stirred at room temperature overnight. After that, the reaction mixture was cooled down to 0 °C in an ice bath. Sodium borohydride (1.5 equiv) was added portionwise. After the bubbling stopped, the solvent was removed under reduced pressure. The resulting residue was then partitioned between H₂O (30.0 mL) and EtOAc (30.0 mL). The aqueous layer was extracted with EtOAc (2 x 20.0 mL). The combined organic extracts were then dried and concentrated under vacuum. The desired product **s1** was used for next step without further purification.

#### 3.4.2.2. Experimental Procedure for Preparation of s2

**General Procedure for Synthesis of Cbz-Protected Amines**: To a solution of **s1** (5.0 mmol) and triethylamine (1.5 equiv) in DCM (30.0 mL), benzyl chloroformate (1.1 equiv) was added portionwise at 0 °C. The reaction was then stirred at room temperature and monitored by TLC. After the reaction was completed, the solvent was then evaporated and

concentrated under vacuum. The obtained crude mixture was directly used for next step without further purification.

General Procedure for Swern Oxidation: To a solution of oxalyl chloride (1.5 equiv) in DCM (30.0 mL) at -78 °C was added dimethyl sulfoxide (3.0 equiv) dropwise via syringe. After stirring for 20 min at -78 °C, a solution of Cbz-protected **s1** (crude product from previous step) in DCM (5.0 mL) was added dropwise. The solution was then stirred at -78 °C for 30 min, followed by dropwise addition of triethylamine (4.5 equiv). After 10 min, the reaction mixture was warmed up to room temperature, poured into brine (30.0 mL) and stirred for another 10 min. The layers were then separated and the aqueous layer was extracted with DCM (3 x 20.0 mL). The combined organic layers were dried over anhydrous Na₂SO₄, filtered and concentrated under vacuum. The aldehyde product **s2** was further purified by flash chromatography.

Benzyl benzyl(4-oxobutyl)carbamate s2-a Yield: 45% over three steps. Hexanes/ethyl



acetate = 1:1,  $R_f$  = 0.40. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.65 and 9.54 (s, 1H), 7.47 – 7.05 (m, 10H),

5.19 and 5.17 (s, 2H), 4.50 and 4.48 (s, 2H), 3.35 – 3.15 (m, 2H), 2.49 – 2.15 (m, 2H), 1.92 – 1.68 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers δ 201.21, 200.97, 156.33, 156.05, 137.51, 136.48, 128.37, 128.25, 127.78, 127.67, 127.18, 127.10, 67.04, 50.27, 50.01, 46.07, 45.12, 40.68, 40.48, 20.31, 19.99. IR (neat, cm⁻¹): 2942.34, 1691.94, 1496.09, 1453.21, 1420.11, 1227.12, 1122.33, 1068.72, 734.08. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₂NO₃⁺: 312.15942, Found: 312.15854.

Benzyl (3-methylbenzyl)(4-oxobutyl)carbamate s2-b Yield: 44% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f = 0.38$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.72 and 9.60 (s, 1H),

7.59 – 6.85 (m, 9H), 5.22 and 5.20 (s, 2H), 4.50 and 4.48 (s, 2H), 3.39 - 3.20 (m, 2H), 2.56 – 2.23 (m, 5H), 1.97 – 1.77 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.43, 201.17, 156.50, 156.27, 138.18, 137.51, 136.64, 128.41, 127.93, 124.92, 124.34, 67.20, 50.34, 50.07, 46.14, 45.16, 40.90, 40.70, 21.31, 20.43, 20.13. IR (neat, cm⁻¹): 2943.14, 1692.23, 1417.82, 1224.47, 1121.54, 1070.15, 768.22, 697.86. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₄NO₃⁺: 326.17507, Found: 326.17535.

Benzyl (4-(tert-butyl)benzyl)(4-oxobutyl)carbamate s2-c Yield: 54% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f = 0.45$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.70 and 9.58 (s, 1H),

7.47 – 7.11 (m, 9H), 5.22 and 5.20 (br, 2H), 4.51 and 4.49 (br, 2H), 3.35 - 3.26 (m, 2H), 2.47 – 2.31 (m, 2H), 1.95 - 1.79 (m, 2H), 1.34 (s, 9H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.23, 201.00, 156.34, 156.15, 150.14, 136.58, 134.49, 128.31, 127.81, 127.69, 127.53, 126.95, 125.31, 67.05, 49.90, 49.69, 46.02, 45.06, 40.78, 40.58, 34.30, 31.21, 20.36, 20.06. IR (neat, cm⁻¹): 2959.90, 1693.86, 1410.91, 1225.58, 1125.93, 1070.00, 767.39. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₃₀NO₃⁺: 368.22202, Found: 368.22204.

Benzyl (3-methoxybenzyl)(4-oxobutyl)carbamate s2-d Yield: 66% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.35. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.59 (s, 1H),

7.49 – 7.12 (m, 6H), 6.94 – 6.62 (m, 3H), 5.18 and 5.16 (br, 2H), 4.47 and 4.45 (br, 2H), 3.76 and 3.70 (s, 3H), 3.36 - 3.18 (m, 2H), 2.47 - 2.27 (m, 2H), 1.97 - 1.68 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.63, 201.34, 159.96, 156.68, 156.39, 139.35, 136.70, 129.69, 128.57, 128.10, 128.01, 120.26, 119.68, 113.46, 113.00, 112.91, 67.41, 55.23, 50.54, 50.30, 46.36, 45.41, 41.09, 40.87, 20.59, 20.27. IR (neat, cm⁻¹): 2941.07, 1692.87, 1600.32, 1418.65, 1260.52, 1121.99, 1048.59, 768.57. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₄NO⁺: 342.16998, Found: 342.16974.

Benzyl (4-methoxybenzyl)(4-oxobutyl)carbamate s2-e Yield: 56% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.30. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.69 (s, 1H),

7.51 – 7.00 (m, 7H), 6.85 (d, J = 10.3 Hz, 2H), 5.18 (s, 2H), 4.43 (s, 2H), 3.78 (s, 3H), 3.31 – 3.21 (m, 2H), 2.47 – 2.28 (m, 2H), 1.88 – 1.75 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.65, 201.38, 159.07, 156.60, 156.33, 136.73, 129.72, 129.41, 128.81, 128.55, 128.03, 114.02, 67.34, 55.31, 49.97, 49.69, 46.00, 45.11, 41.05, 40.85, 20.57, 20.23. IR (neat, cm⁻¹): 2935.22, 1694.38, 1611.85, 1512.80, 1417.38, 1246.05, 1032.79. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₄NO₄⁺: 342.16998, Found: 342.17007.

Benzyl (benzo[d][1,3]dioxol-5-ylmethyl)(4-oxobutyl)carbamate s2-f Yield: 55% over



three steps. Hexanes/ethyl acetate = 1:1,  $R_f = 0.31$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.60

(s, 1H), 7.45 - 7.21 (m, 5H), 6.88 - 6.56 (m, 3H), 5.92 (s, 2H), 5.17 (s, 2H), 4.40 and 4.38 (br, 2H), 3.32 - 3.14 (m, 2H), 2.43 - 2.32 (m, 2H), 1.92 - 1.68 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.57, 201.29, 156.55, 156.20, 147.97, 147.01, 136.62, 131.50, 128.54, 128.11, 128.00, 121.33, 120.81, 108.49, 108.17, 107.98, 101.07, 67.38, 50.31, 50.01, 45.96, 45.11, 41.01, 40.81, 20.53, 20.17. IR (neat, cm⁻¹): 2893.04, 1693.50, 1489.39, 1444.37, 1422.23, 1245.49, 1124.73, 1037.88, 926.79, 699.40. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₂NO₅⁺: 356.14925, Found: 356.15003.

Benzyl (3-nitrobenzyl)(4-oxobutyl)carbamate s2-g Yield: 40 % yield over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.40. ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.61 (s, 1H),

8.17 – 7.98 (m, 2H), 7.67 – 7.18 (m, 7H), 5.18 and 5.13 (s, 2H), 4.57 and 4.55 (br, 2H), 3.46 – 3.20 (m, 2H), 2.56 – 2.30 (m, 2H), 1.87 – 1.82 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.21, 200.99, 156.48, 155.96, 148.35, 140.08, 136.24, 133.75, 133.19, 129.55, 128.48, 128.14, 127.97, 122.38, 122.07, 67.53, 49.95, 49.77, 46.78, 45.96, 40.73, 40.59, 20.50, 20.18. IR (neat, cm⁻¹): 2943.83, 1695.41, 1528.09, 1471.91, 1419.10, 1348.18, 1226.18, 1125.53, 731.66. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₁N₂O₅⁺: 357.14450, Found: 357.14310.

Benzyl (4-cyanobenzyl)(4-oxobutyl)carbamate s2-h Yield: 64% over three steps.



Hexanes/ethyl acetate = 1:2,  $R_f = 0.40$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.65 and 9.56 (s, 1H),

7.61 – 7.12 (m, 9H), 5.13 and 5.08 (s, 2H), 4.48 (s, 2H), 3.31 - 3.20 (m, 2H), 2.40 – 2.25 (m, 2H), 1.91 - 1.71 (m, 2H). ¹³C NMR (125 MHz, CDl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.26, 201.00, 156.51, 156.03, 143.34, 136.24, 132.36, 128.49, 128.19, 128.01, 127.69, 118.62, 111.20, 67.55, 50.33, 50.14, 46.84, 45.96, 40.78, 40.61, 20.50, 20.19. IR (neat, cm⁻¹): 2945.39, 2228.23, 1696.98, 1473.81, 1412.82, 1230.71, 1125.64, 698.69. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₁N₂O₃⁺: 337.15467, Found: 337.15474.

Methyl 4-((((benzyloxy)carbonyl)(4-oxobutyl)amino)methyl)benzoate s2-i Yield: 28%



over three steps. Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.25. ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide

rotamers  $\delta$  9.68 and 9.57 (s, 1H), 8.00 – 7.94 (m, 2H), 7.35 – 7.20 (m, 7H), 5.17 and 5.13 (s, 2H), 4.53 and 4.50 (br, 2H), 3.86 (s, 3H), 3.37 – 3.17 (m, 2H), 2.37 (m, 2H), 1.92 – 1.70 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.27, 201.00, 166.67, 156.51, 156.15, 143.00, 136.40, 129.85, 129.29, 128.45, 128.08, 127.93, 127.61, 127.03, 67.40, 52.00, 50.33, 50.13, 46.62, 45.69, 40.83, 40.64, 20.49, 20.20. IR (neat, cm⁻¹): 2950.99, 1693.42, 1612.09, 1411.90, 1276.02, 1109.46, 1018.72, 913.18, 731.51. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₂₄NO₅⁺: 370.16490, Found: 370.16563.

Benzyl (4-oxobutyl)(4-(trifluoromethyl)benzyl)carbamate s2-j Yield: 56% over three



steps. Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.45. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.73 and 9.62 (s,

1H), 7.69 – 7.11 (m, 9H), 5.18 and 5.14 (s, 2H), 4.54 and 4.52 (br, 2H), 3.31 - 3.24 (m, 2H), 2.51 - 2.34 (m, 2H), 1.94 - 1.78 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.39, 201.10, 156.66, 156.26, 141.98, 136.43, 129.76 (q, J = 31.5 Hz), 128.57, 128.25, 128.08, 127.46, 125.62 (q, J = 4.0 Hz), 124.18 (q, J = 270.0 Hz), 67.56, 50.26, 50.07, 46.77, 45.79, 40.97, 40.75, 20.57, 20.28. ¹⁹F NMR (564 MHz, CDCl₃)  $\delta$  –62.49. IR (neat, cm⁻¹): 2942.24, 1697.97, 1415.37, 1325.05, 1163.84, 1122.11, 1066.21, 1017.90, 698.87. HRMS (ESI) ([M+Na]⁺) Calcd. for C₂₀H₂₀O₃NF₃Na⁺: 402.12930, Found: 402.12970.

Benzyl (3-fluorobenzyl)(4-oxobutyl)carbamate s2-k Yield: 46% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f = 0.42$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.73 and 9.61 (s, 1H),

7.74 – 7.18 (m, 6H), 7.09 – 6.79 (m, 3H), 5.19 and 5.16 (s, 2H), 4.49 and 4.47 (br, 2H), 3.41 – 3.18 (m, 2H), 2.47 – 2.34 (m, 2H), 1.86 – 1.80 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.45, 201.18, 163.10 (d, *J* = 245.0 Hz) 156.65, 156.29, 140.46 (d, *J* = 7.5 Hz), 136.55, 130.20 (d, *J* = 8.2 Hz), 128.61, 128.21, 128.07, 123.43, 122.87, 114.78, 114.43 (d, *J* = 21.8 Hz), 114.14, 67.56, 50.20, 49.97, 46.58, 45.66, 41.01, 40.83, 20.60, 20.29. ¹⁹F NMR (564 MHz, CDCl₃)  $\delta$  – 112.71. IR (neat, cm⁻¹): 2944.97, 1694.22, 1590.25, 1420.49, 1228.16, 1134.46, 769.94. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₁NO₃F⁺: 330.1500, Found: 330.15040.

Benzyl (3,4-difluorobenzyl)(4-oxobutyl)carbamate s2-l Yield: 54% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f = 0.47$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.68 and 9.58 (s, 1H),

7.35 – 7.26 (m, 5H), 7.10 – 6.87 (m, 3H), 5.16 and 5.14 (br, 2H), 4.42 and 4.40 (br, 2H), 3.39 – 3.15 (m, 2H), 2.52 – 2.24 (m, 2H), 1.89 – 1.72 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.22, 200.98, 156.37, 155.91, 150.17 (dd, *J* = 248.6, 12.8 Hz), 149.47 (d, *J* = 248.6 Hz), 136.32, 134.87 (dd, *J* = 4.5, 3.0 Hz), 128.40, 128.03, 127.88, 123.71, 123.16, 117.14 (d, *J* = 17.1 Hz), 116.61 (d, *J* = 18.0 Hz), 116.11 (d, *J* = 15.0 Hz), 67.33, 49.52, 49.25, 46.31, 45.48, 40.68, 40.52, 20.37, 20.03. ¹⁹F NMR (470 MHz, CDCl₃)  $\delta$  –137.38, –139.78. IR (neat, cm⁻¹): 2944.28, 1698.37, 1473.99, 1379.57, 1276.69, 1169.19, 1125.96, 903.40, 699.95. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₀NO₃F₂⁺: 348.14058, Found: 348.14185.

Benzyl (4-bromobenzyl)(4-oxobutyl)carbamate s2-m Yield: 49% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f = 0.49$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.72 and 9.61 (s, 1H),

7.74 – 6.79 (m, 9H), 5.18 and 5.16 (br, 2H), 4.43 (s, 2H), 3.25 (dd, J = 23.1, 7.8 Hz, 2H), 2.46 – 2.33 (m, 2H), 1.82 (dd, J = 23.1, 8.2 Hz, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.40, 201.14, 156.57, 156.21, 136.81, 136.51, 131.73, 129.67, 129.05, 128.56, 128.18, 128.04, 121.33, 67.48, 50.02, 49.81, 46.46, 45.52, 40.94, 40.77, 20.54, 20.25. IR (neat, cm⁻¹): 2942.06, 1691.46, 1487.26, 1404.15, 1216.03, 1122.63, 1069.73, 1010.63, 697.50. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₁NO₃Br⁺: 390.06993, Found: 390.06954.

Benzyl (3,5-dibromobenzyl)(4-oxobutyl)carbamate s2-n Yield: 25% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f = 0.44$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.60 (s, 1H),

7.53 (s, 1H), 7.36 – 7.24 (m, 7H), 5.18 and 5.13 (s, 2H), 4.41 and 4.39 (br, 2H), 3.37 - 3.16 (m, 2H), 2.53 - 2.30 (m, 2H), 1.94 - 1.72 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.14, 200.87, 156.37, 155.88, 141.84, 136.21, 133.00, 129.37, 128.93, 128.54, 128.15, 127.94, 123.08, 67.54, 49.49, 49.27, 46.62, 45.68, 40.75, 40.57, 20.39, 20.09. IR (neat, cm⁻¹): 2942.68, 1693.41, 1584.74, 1556.59, 1415.39, 128.94, 1124.46, 739.85. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₀NO₃Br₂⁺: 467.98045, Found: 467.98095.

Benzyl (naphthalen-2-ylmethyl)(4-oxobutyl)carbamate s2-o Yield: 61% over three



steps. Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.48. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.72 and 9.59 (s,

1H), 7.89 – 7.55 (m, 4H), 7.52 – 7.22 (m, 8H), 5.26 and 5.24 (s, 2H), 4.70 and 4.67 (s, 2H), 3.45 – 3.23 (m, 2H), 2.54 – 2.30 (m, 2H), 1.98 – 1.78 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers δ 201.53, 201.25, 156.71, 156.41, 136.66, 135.12, 133.32, 132.83, 128.52, 128.05, 127.76, 127.71, 126.67, 126.29, 126.00, 125.94, 125.42, 67.43, 50.70, 50.44, 46.33, 45.28, 41.01, 40.79, 20.54, 20.26. IR

(neat, cm⁻¹): 2942.23, 1693.93, 1470.78, 1420.14, 1367.16, 1231.17, 1120.19, 957.17, 751.63. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₂₄NO₃⁺: 362.17507, Found: 362.17594. **Benzyl (furan-3-ylmethyl)(4-oxobutyl)carbamate s2-p** Yield: 51% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.35. ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.68 and 9.57 (s, 1H), 7.40 – 7.22 (m, 6H),

6.28 - 6.12 (m, 2H), 5.14 (s, 2H), 4.45 and 4.41 (s, 2H), 3.33 - 3.29 (m, 2H), 2.40 - 2.27 (m, 2H), 1.83 - 1.74 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.43, 201.19, 155.96, 151.08, 142.06, 136.51, 128.38, 127.92, 127.73, 110.32, 108.42, 107.95, 67.22, 46.39, 45.61, 43.53, 43.44, 40.77, 40.58, 20.52, 20.21. IR (neat, cm⁻¹): 2944.77, 2724.76, 1693.34, 1418.37, 1264.24, 1195.60, 1120.85, 1072.14, 914.85, 734.19, 697.60. HRMS (ESI) ([M+Na]⁺) Calcd. for C₁₇H₁₉O₄NNa⁺: 324.12118, Found: 324.12149.

Benzyl (4-oxobutyl)(thiophen-2-ylmethyl)carbamate s2-q Yield: 47% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.40. ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.60 (s, 1H), 7.49 – 7.13 (m, 6H),

6.98 – 6.90 (m, 2H), 5.19 (s, 2H), 4.63 and 4.61 (br, 2H), 3.38 – 3.27 (m, 2H), 2.49 – 2.27 (m, 2H), 1.94 – 1.72 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers δ 201.49, 201.24, 156.14, 155.86, 140.45, 140.29, 136.51, 128.50, 128.09, 128.03, 126.62, 126.29, 125.51, 125.39, 67.42, 46.14, 45.40, 40.93, 40.75, 20.65, 20.32. IR (neat, cm⁻¹): 2940.73, 1694.38, 1470.32, 1420.33, 1261.69,

1215.64, 976.18, 698.50. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₂₀NO₃S⁺: 318.11584, Found: 318.11763.

Benzyl (4-oxobutyl)(thiophen-3-ylmethyl)carbamate s2-r Yield: 41% over three steps.



Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.38. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.71 and 9.6 0 (s, 1H), 7.46 – 6.88 (m,

8H), 5.18 (s, 2H), 4.48 and 4.46 (br, 2H), 3.36 – 3.21 (m, 2H), 2.44 – 2.31 (m, 2H), 1.90 – 1.69 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers δ 201.61, 201.34, 156.35, 156.14, 138.60, 136.63, 128.56, 128.12, 128.01, 127.72, 127.19, 126.31, 122.86, 122.36, 67.36, 46.25, 45.86, 45.77, 45.40, 40.98, 40.78, 20.63, 20.29. IR (neat, cm⁻¹): 2942.73, 1693.05, 1423.27, 1244.59, 1212.49, 1120.22, 767.91. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₂₀NO₃S⁺: 318.11584, Found: 318.11531.

Benzyl (4-oxobutyl)(pyridin-2-ylmethyl)carbamate s2-s Yield: 57% over three steps.



100% ethyl acetate,  $R_f = 0.34$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.64 and 9.52 (s, 1H), 8.45 (d, J = 4.9 Hz, 1H),

7.57 – 7.50 (m, 1H), 7.38 – 6.96 (m, 7H), 5.12 and 5.06 (s, 2H), 4.55 and 4.52 (s, 2H), 3.36 – 3.29 (m, 2H), 2.40 – 2.27 (m, 2H), 1.83 – 1.75 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers δ 201.40, 201.16, 157.63, 157.54, 156.41, 156.20, 149.11, 149.04, 136.72, 136.62, 136.41, 128.40, 128.27, 128.00, 127.90, 127.83, 127.69, 122.30, 122.16, 121.93, 121.01, 67.30, 67.13, 52.52, 52.22, 47.12, 46.29, 40.84, 40.63, 20.60, 20.31. IR (neat, cm⁻¹): 2942.47, 1695.83, 1592.95, 1474.92,

1417.48, 1214.98, 1121.23, 994.27, 752.88. HRMS (DART) ( $[M+H]^+$ ) Calcd. for  $C_{18}H_{21}N_2O_3^+$ : 313.15467, Found: 313.15518.

Benzyl ((6-bromopyridin-2-yl)methyl)(4-oxobutyl)carbamate s2-t Yield: 65% over



three steps. 100% ethyl acetate,  $R_f = 0.35$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.69 and 9.58 (s, 1H),

7.44 – 6.99 (m, 8H), 5.13 and 5.07 (s, 2H), 4.51 and 4.49 (br, 2H), 3.46 – 3.27 (m, 2H), 2.46 – 2.33 (m, 2H), 1.86 - 1.79 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.35, 201.13, 159.35, 159.20, 156.31, 156.03, 141.50, 141.36, 139.03, 138.91, 136.31, 136.26, 128.44, 128.32, 128.05, 127.87, 127.70, 126.56, 126.46, 120.56, 119.67, 67.36, 67.20, 52.27, 51.88, 47.39, 46.71, 40.82, 40.63, 20.72, 20.39. IR (neat, cm⁻¹): 2942.05, 1693.33, 1582.01, 1555.61, 1404.30, 1216.40, 1115.11, 984.68, 768.54, 697.69. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₂₀N₂O₃Br⁺: 391.106518, Found: 391.06425.

Benzyl (4-oxobutyl)(pyridin-3-ylmethyl)carbamate s2-u Yield: 48% over three steps.



100% ethyl acetate,  $R_f = 0.25$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.68 and 9.57 (s, 1H), 8.54 – 8.37 (m, 2H), 7.59

-7.17 (m, 7H), 5.13 and 5.12 (br, 2H), 4.46 and 4.44 (br, 2H), 3.26 - 3.20 (m, 2H), 2.41 -2.32 (m, 2H), 1.82 - 1.77 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.28, 201.01, 156.56, 156.27, 149.25, 149.01, 136.38, 135.77, 135.09, 133.32, 128.59, 128.24, 128.09, 123.59, 67.60, 48.22, 48.05, 46.53, 45.67, 40.88, 40.72, 20.56, 20.24. IR (neat, cm⁻¹): 2944.82, 1690.91, 1577.29,
1472.30, 1416.95, 1214.11, 1122.42, 1027.75, 965.12, 760.52. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₂₁N₂O₃⁺: 313.15467, Found: 313.15558.

Benzyl (benzo[b]thiophen-3-ylmethyl)(4-oxobutyl)carbamate s2-v Yield: 45% over



three steps. Hexanes/ethyl acetate = 1:1,  $R_f = 0.36$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.69 and 9.56

(s, 1H), 7.95 - 7.61 (m, 2H), 7.46 - 7.14 (m, 8H), 5.24 (s, 2H), 4.79 and 4.74 (s, 2H), 3.33 - 3.23 (m, 2H), 2.41 - 2.28 (m, 2H), 1.84 - 1.74 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.57, 201.30, 156.39, 156.14, 140.66, 138.04, 137.80, 136.62, 132.41, 132.24, 128.61, 128.21, 128.07, 125.17, 124.65, 124.39, 124.29, 124.17, 122.92, 122.31, 121.68, 67.53, 45.92, 44.72, 44.60, 44.49, 40.98, 40.76, 20.38, 20.15. IR (neat, cm⁻¹): 2944.49, 1693.34, 1421.38, 1362.23, 1263.32, 1230.66, 1121.79, 768.27. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₂₂NO₃S⁺: 368.13149, Found: 368.13128.

Benzyl (4-oxobutyl)(quinolin-2-ylmethyl)carbamate s2-w Yield: 62% over three steps.



100% ethyl acetate,  $R_f = 0.30$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  9.62 and 9.48 (s, 1H), 8.01 – 7.95

(m, 2H), 7.69 - 7.67 (m, 1H), 7.61 - 7.59 (m, 1H), 7.42 - 7.26 (m, 4H), 7.18 - 7.13 (m, 3H), 5.17 and 5.10 (s, 2H), 4.75 and 4.70 (s, 2H), 3.46 - 3.24 (m, 2H), 2.38 - 2.24 (m, 2H), 1.85 - 1.76 (m, 2H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  201.17, 200.91, 157.88, 156.38, 156.08, 147.36, 136.57, 136.52, 136.35, 136.27, 129.38, 129.30, 128.74, 128.27, 128.06, 127.85, 127.72, 127.61,

127.45, 127.33, 127.07, 126.96, 126.05, 119.64, 118.77, 77.36, 67.15, 66.95, 53.18, 52.85, 47.17, 46.18, 40.70, 40.47, 20.49, 20.23. IR (neat, cm⁻¹): 2942.93, 1693.57, 1599.69, 1504.96, 1416.05, 1229.34, 1124.48, 1073.62, 822.13. HRMS (DART) ([M+H]⁺) Calcd. for C₂₂H₂₃N₂₃O₃⁺: 363.17032, Found: 363.16992.

#### 3.4.2.3. Experimental Procedure for Preparation of Sulfonyl Hydrazones



To a stirred solution of 2,4,6-triisopropylbenzenesulfonyl hydrazide (1.0 mmol) in THF (10.0 mL) at room temperature, aldehyde s2 (1.0 equiv) was added dropwise. After the reaction was stirred overnight, the solvent was removed directly under reduced pressure, and the crude mixture was further purified by flash chromatography.

#### Benzyl benzyl(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl hydrazone



(1a) Yield: 84%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.35. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  8.51

- 8.07 (m, 1H), 7.56 - 6.44 (m, 13H), 5.18 and 5.17 (br, 2H), 4.56 - 4.42 (m, 2H), 4.27 - 4.22 (m, 2H), 3.36 - 3.15 (m, 2H), 2.96 - 2.90 (m, 1H), 2.13 (m, 2H), 1.79 - 1.59 (m, 2H), 1.29 (d, J = 6.9 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.68, 156.26, 153.22, 151.33, 149.66, 149.45, 137.66, 136.66, 131.48, 128.66, 128.59, 128.48, 128.08, 127.99, 127.88, 127.39, 127.30, 123.80, 67.49, 67.43, 67.29, 50.58, 50.33, 46.49, 45.47, 34.18, 29.91, 29.88, 29.69, 29.52, 24.90, 24.84, 24.54, 24.18, 23.59. IR (neat, cm⁻¹): 2957.59, 1674.62, 1599.24, 1423.21, 1362.42, 1316.34, 1152.83, 882.15, 751.64. HRMS (DART) ([M+H]⁺) Calcd. for C₃₄H₄₆N₃O₄S⁺:

#### 592.32035, Found: 592.31736.

# Benzyl (3-methylbenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1b) Yield: 69%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.42$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the

presence of amide rotamers  $\delta$  8.47 – 8.01 (m, 1H), 7.58 – 6.47 (m, 12H), 5.19 and 5.17 (br, 2H), 4.44 and 4.42 (s, 2H), 4.31 – 4.18 (m, 2H), 3.31 – 3.12 (m, 2H), 2.95 – 2.90 (m, 1H), 2.33 and 2.30 (s, 3H), 2.21 – 2.04 (m, 2H), 1.78 – 1.60 (m, 2H), 1.29 (d, *J* = 6.7 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.69, 156.31, 153.23, 151.34, 149.74, 149.52, 138.27, 137.57, 136.72, 131.48, 128.55, 128.48, 128.12, 128.03, 127.96, 124.95, 124.40, 123.82, 67.48, 67.42, 67.27, 50.45, 50.20, 46.43, 45.37, 34.21, 34.19, 29.93, 29.90, 29.73, 29.55, 24.90, 24.85, 24.54, 24.19, 23.60, 21.42. IR (neat, cm⁻¹): 2958.21, 2868.54, 1695.43, 1600.30, 1461.93, 1421.48, 1228.82, 1153.25, 1008.82. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₅H₄₈N₃O₄S⁺: 606.33655, Found: 606.33740.

#### Benzyl

#### (4-(*tert*-butyl)benzyl)(4-oxobutyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1c) Yield: 79%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.46. ¹H NMR (600 MHz, CDCl₃) some signals

exist as a pair due to the presence of amide rotamers  $\delta$  7.89 – 6.55 (m, 13H), 5.16 and 5.14 (br, 2H), 4.46 – 4.39 (m, 2H), 4.20 – 4.17 (m, 2H), 3.31 – 3.15 (m, 2H), 2.89 (hept, J = 7.2 Hz, 1H), 2.23 – 1.97 (m, 2H), 1.78 – 1.57 (m, 2H), 1.30 (s, 9H), 1.25 (d, J = 7.0 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.61, 156.33, 153.26, 151.38, 150.38, 149.84, 149.63, 136.77, 134.62, 131.47, 128.61, 128.56, 128.15, 128.03, 127.93, 127.67, 127.11, 125.59, 125.57, 125.52, 123.85, 67.50, 67.40, 67.28, 50.14, 49.94, 46.38, 45.33, 34.54, 34.22, 31.44, 29.96, 29.94, 29.75, 29.56, 24.93, 24.88, 24.56, 24.22, 23.62. IR (neat, cm⁻¹): 2958.89, 1676.29, 1599.36, 1463.57, 1424.97, 1164.31, 1038.92, 882.04, 752.76. HRMS (DART) ([M+H]⁺) Calcd. for C₃₈H₅₄N₃O₄S⁺: 648.38295, Found: 648.38412.

Benzyl (3-methoxybenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1d) Yield: 77%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.35$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the

presence of amide rotamers  $\delta$  7.70 – 7.52 (m, 1H), 7.42 – 6.46 (m, 12H), 5.17 and 5.14 (br, 2H), 4.42 and 4.40 (br, 2H), 4.26 – 4.11 (m, 2H), 3.77 and 3.71 (s, 3H), 3.37 – 3.10 (m, 2H), 2.90 (hept, J = 7.0 Hz, 1H), 2.21 – 2.00 (m, 2H), 1.75 – 1.58 (m, 2H), 1.25 (d, J = 6.8 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  159.91, 156.26, 153.28, 151.36, 149.80, 149.59, 139.35, 136.72, 131.46, 129.72, 129.65, 128.58, 128.16, 128.07, 127.98, 120.19, 119.64, 67.55, 67.46, 67.32, 55.22, 50.55, 50.34, 46.52, 45.49, 34.21, 29.95, 29.92, 29.71, 29.54, 24.91, 24.86, 24.59, 24.21, 23.61. IR (neat, cm⁻¹): 2958.54, 1677.85, 1599.51, 1456.17, 1422.72, 1261.60, 1152.46, 1039.35, 907.33, 727.92. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₅H₄₈O₅N₃S⁺: 622.33147, Found: 622.32867.





hydrazone (1e) Yield: 75%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.33$ . ¹H NMR (500 MHz,

CDCl₃)  $\delta$  8.12 – 8.04 (m, 1H), 7.44 – 7.23 (m, 5H), 7.22 – 6.39 (m, 7H), 5.16 (s, 2H), 4.45 – 4.35 (m, 2H), 4.25 – 4.19 (m, 2H), 3.78 (s, 3H), 3.26 – 3.09 (m, 2H), 2.91 (hept, *J* = 7.0 Hz, 1H), 2.22 – 2.03 (m, 2H), 1.76 – 1.58 (m, 2H), 1.27 (d, *J* = 6.9 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  159.04, 156.22, 153.27, 151.35, 149.81, 149.61, 136.74, 131.48, 129.72, 129.33, 128.78, 128.58, 128.08, 128.03, 123.83, 114.08, 114.01, 67.49, 67.33, 55.30, 50.00, 49.75, 46.17, 45.19, 34.20, 29.94, 29.91, 29.69, 29.55, 24.91, 24.85, 24.57, 24.19, 23.60. IR (neat, cm⁻¹): 2958.76, 2868.93, 1693.21, 1611.57, 1512.88, 1462.59, 1245.64, 1008.78, 753.25. HRMS (DART) ([M+NH₄]⁺) Calcd. for C₃₅H₅₁N₄O₅S⁺: 639.35294, Found: 639.35425.

#### Benzyl (be

# (benzo[d][1,3]dioxol-5-ylmethyl)(4-oxobutyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1f) Yield: 70%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.35. ¹H NMR (600 MHz, CDCl₃) some signals

exist as a pair due to the presence of amide rotamers  $\delta$  8.12 – 7.95 (m, 1H), 7.45 – 6.93 (m, 8H), 6.79 – 6.56 (m, 3H), 5.92 (s, 2H), 5.15 (s, 2H), 4.35 and 4.32 (br, 2H), 4.28 – 4.15 (m, 2H), 3.26 – 3.07 (m, 2H), 2.91 (hept, J = 7.1 Hz, 1H), 2.23 – 2.01 (m, 2H), 1.75 – 1.57 (m, 2H), 1.26 (d, J = 6.8 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.64, 156.15, 153.31, 151.36, 149.80, 149.58, 147.99, 147.03, 136.67, 131.54, 131.44, 128.61, 128.14, 128.08, 123.86, 121.33, 120.83, 108.48, 108.20, 107.98, 101.09, 67.46, 50.38, 50.10, 46.17, 45.23, 34.21, 29.96, 29.94, 29.71, 29.54, 24.92, 24.86, 24.55, 24.14, 23.61. IR (neat, cm⁻¹): 2957.99, 1675.79, 1599.25, 1489.32, 1423.83, 1363.70, 1244.47, 1164.06, 1037.55, 927.08, 752.84. HRMS (DART) ([M+H]⁺) Calcd. for C₃₅H₄₆N₃O₆S⁺: 636.3108, Found: 636.30900.

Benzyl (3-nitrobenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl

1g

NNHTris NNHTris Cbz H acetate

 $O_2N$ 

hydrazone (1g) Yield: 79%. Hexanes/ethyl acetate = 1:1,  $R_f = 0.35$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the

presence of amide rotamers  $\delta$  8.33 – 7.95 (m, 3H), 7.63 – 7.27 (m, 6H), 7.24 – 7.05 (m, 4H), 5.20 and 5.13 (br, 2H), 4.58 and 4.50 (br, 2H), 4.23 – 4.17 (m, 2H), 3.39 – 3.16 (m, 2H), 2.98 – 2.83 (m, 1H), 2.19 – 2.12 (m, 2H), 1.76 – 1.70 (m, 2H), 1.25 (d, *J* = 7.0 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.67, 156.00, 153.35, 151.29, 149.39, 149.17, 148.46, 140.18, 140.04, 136.38, 133.87, 131.38, 129.66, 128.62, 128.28, 128.12, 123.84, 122.48, 122.32, 122.14, 67.67, 50.18, 50.00, 47.11, 46.17, 34.17, 29.92, 29.55, 29.42, 24.87, 24.81, 24.57, 24.19, 23.57. IR (neat, cm⁻¹): 2958.70, 2868.63, 1680.13, 1599.20, 1530.21, 1423.32, 1348.07, 1164.74, 755.45, 731.25. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₄H₄₄O₆N₄SNa⁺: 659.28793, Found: 659.28851.

Benzyl (4-cyanobenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1h) Yield: 76%. Hexanes/ethyl acetate = 2:1,  $R_f = 0.40$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the

presence of amide rotamers  $\delta$  8.05 – 6.85 (m, 13H), 5.13 and 5.09 (br, 2H), 4.47 and 4.45 (br, 2H), 4.19 – 4.14 (m, 2H), 3.34 – 3.11 (m, 2H), 2.96 – 2.82 (m, 1H), 2.17 – 2.03 (m, 2H), 1.77 – 1.66 (m, 2H), 1.23 (d, J = 6.8 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.71, 153.48, 151.36, 149.46, 149.20, 143.46, 136.32, 132.51, 131.37, 128.68, 128.39, 128.22, 127.81, 123.93,

118.80, 111.32, 67.88, 67.79, 67.63, 50.61, 50.41, 47.14, 46.10, 34.25, 30.00, 29.98, 29.57, 29.37, 24.93, 24.88, 24.58, 24.22, 23.64. IR (neat, cm⁻¹): 2958.96, 2228.99, 1681.21, 1599.38, 1463.69, 1425.14, 1216.66, 1164.29, 750.53. HRMS (DART) ([M+H]⁺) Calcd. for C₃₅H₄₅N₄O₄S⁺: 617.31560, Found: 617.31541.

# Methyl 4-((((benzyloxy)carbonyl)(4-oxobutyl)amino)methyl)benzoate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1i) Yield: 74%. Hexanes/ethyl acetate = 1:1,  $R_f$  = 0.35. ¹H NMR (600 MHz, CDCl₃) some

signals exist as a pair due to the presence of amide rotamers  $\delta$  8.06 – 7.60 (m, 3H), 7.45 – 6.83 (m, 10H), 5.14 and 5.11 (br, 2H), 4.48 and 4.45 (br, 2H), 4.19 – 4.14 (m, 2H), 3.90 (s, 3H), 3.28 – 3.14 (m, 2H), 2.98 – 2.85 (m, 1H), 2.25 – 2.00 (m, 2H), 1.78 – 1.57 (m, 2H), 1.23 (d, *J* = 6.8 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  166.94, 156.72, 153.40, 151.39, 149.64, 149.41, 143.12, 136.50, 131.40, 130.08, 130.02, 129.40, 128.71, 128.61, 128.25, 128.16, 128.08, 127.73, 127.13, 123.91, 67.77, 67.67, 67.52, 52.22, 50.54, 50.36, 46.92, 45.84, 34.25, 30.00, 29.68, 29.46, 24.94, 24.89, 24.59, 24.27, 23.64. IR (neat, cm⁻¹): 2957.28, 2868.81, 1699.10, 1599.56, 1462.50, 1277.67, 1153.42, 1107.80, 752.05. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₆H₄₇O₆N₃SNa⁺: 672.30833, Found: 672.30878.

# Benzyl (4-oxobutyl)(4-(trifluoromethyl)benzyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1j) Yield: 76%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.32. ¹H NMR (600 MHz, CDCl₃) some signals exist as

a pair due to the presence of amide rotamers  $\delta$  8.22 – 8.08 (m, 1H), 7.54 – 7.52 (m, 2H),

7.36 – 7.18 (m, 10H), 5.17 and 5.13 (br, 2H), 4.50 and 4.48 (br, 2H), 4.23 – 4.18 (m, 2H), 3.37 – 3.13 (m, 2H), 2.93 – 2.88 (m, 1H), 2.19 – 2.11 (m, 2H), 1.81 – 1.63 (m, 2H), 1.26 (d, J = 6.8 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.73, 156.20, 153.41, 151.41, 151.37, 149.53, 149.30, 141.99, 136.42, 131.41, 129.72 (q, J = 33.0 Hz), 128.68, 128.59, 128.21, 128.13, 128.08, 127.45, 125.62 (q, J = 4.5 Hz), 124.21 (q, J = 270.0 Hz), 123.90, 67.74, 67.68, 67.54, 50.39, 50.22, 46.98, 45.90, 34.23, 29.98, 29.63, 29.47, 24.91, 24.86, 24.57, 24.23, 23.61. ¹⁹F NMR (564 MHz, CDCl₃)  $\delta$  –62.48. IR (neat, cm⁻¹): 2959.34, 2869.47, 1678.02, 1599.40, 1416.09, 1323.45, 1162.70, 1122.76, 1065.84, 1017.69, 754.40. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₅H₄₄O₄N₃SF₃Na⁺: 682.29023, Found: 682.29071.

Benzyl (3-fluorobenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



**hydrazone (1k)** Yield: 77%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.30. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide

rotamers δ 8.22 – 8.08 (m, 1H), 7.41 – 7.14 (m, 9H), 6.88 (m, 3H), 5.15 and 5.13 (br, 2H), 4.43 and 4.40 (br, 2H), 4.31 – 4.14 (m, 2H), 3.34 – 3.10 (m, 2H), 2.90 (hept, J = 6.9 Hz, 1H), 2.16 – 2.07 (m, 2H), 1.72 – 1.62 (m, 2H), 1.25 (d, J = 7.0 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers δ 163.06 (d, J = 246.6 Hz), 156.69, 156.18, 153.32, 151.36, 149.59, 149.36, 140.45, 140.32 (d, J =7.5 Hz), 131.44, 130.17 (d, J = 9.0 Hz), 128.60, 128.40, 128.17, 128.08, 128.01, 123.86, 123.36, 122.82, 114.66, 114.33 (d, J = 21.0 Hz), 114.07, 67.67, 67.60, 67.47, 50.24, 50.02, 46.76, 45.76, 34.21, 29.95, 29.66, 29.50, 24.91, 24.85, 24.56, 24.19, 23.61. ¹⁹F NMR (470 MHz, CDCl₃) δ –112.78. IR (neat, cm⁻¹): 2959.08, 1678.22, 1597.05, 1422.68, 1152.82, 908.21, 729.99. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₄H₄₄O₄N₃SFNa⁺: 632.29343, Found: 632.29327.

### Benzyl (3,4-difluorobenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



**hydrazone (11)** Yield: 74%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.31$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide

rotamers  $\delta$  8.44 – 7.70 (m, 1H), 7.52 – 6.77 (m, 11H), 5.15 and 5.14 (br, 2H), 4.38 and 4.36 (br, 2H), 4.19 (hept, J = 6.5 Hz, 2H), 3.30 – 3.12 (m, 2H), 2.97 – 2.85 (m, 1H), 2.21 – 2.06 (m, 2H), 1.76 – 1.60 (m, 2H), 1.25 (d, J = 6.9 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.66, 156.09, 153.44, 151.42, 151.39, 149.74 (d, J = 250.0 Hz), 136.49, 134.96 (dd, J = 3.8, 3.0 Hz), 131.41, 128.68, 128.38, 128.34, 128.20, 123.91, 123.30, 117.44 (d, J = 17.5 Hz), 117.38 (d, J = 16.3 Hz), 116.80 (d, J = 17.5 Hz), 116.30 (d, J = 18.8 Hz), 67.79, 67.64, 49.91, 49.66, 46.74, 45.77, 34.25, 29.99, 29.62, 29.48, 24.92, 24.57, 24.20, 23.62. ¹⁹F NMR (470 MHz, CDCl₃)  $\delta$  –137.27, –139.70. IR (neat, cm⁻¹): 2959.41, 2869.29, 1694.28, 1599.33, 1518.31, 1281.79, 1212.40, 1114.06, 881.54, 753.33. 2959.39, 2869.41, 1678.69, 1599.67, 1518.54, 1421.93, 1281.89, 1164.52, 1153.16, 908.85, 822.85, 731.12. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₄H₄₃O₄N₃SF₂Na⁺: 650.28401, Found: 650.28412.

#### Benzyl (4-bromobenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1m) Yield: 76%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.36$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the

presence of amide rotamers  $\delta$  8.47 – 8.33 (m, 1H), 7.48 – 6.97 (m, 12H), 5.16 – 5.13 (m,

2H), 4.40 and 4.38 (br, 2H), 4.27 – 4.21 (m, 2H), 3.35 - 3.10 (m, 2H), 2.94 - 2.89 (m, 1H), 2.27 – 2.00 (m, 2H), 1.75 - 1.60 (m, 2H), 1.27 (d, J = 6.7 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.64, 153.25, 151.29, 149.47, 149.23, 136.79, 136.48, 131.66, 131.45, 129.59, 129.02, 128.54, 128.17, 128.01, 123.81, 122.65, 121.30, 121.18, 67.57, 67.51, 67.40, 50.09, 49.88, 46.65, 45.64, 34.16, 29.90, 29.87, 29.59, 29.46, 24.88, 24.82, 24.51, 24.13, 23.58. IR (neat, cm⁻¹): 2958.34, 1676.46, 1598.94, 1463.10, 1424.60, 1316.95, 1153.03, 1011,31, 882.44, 753.78. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₄H₄₅O₄N₃SBr⁺: 670.23142, Found: 670.23138.

Benzyl (3,5-dibromobenzyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1n) Yield: 71%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.38. ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide

rotamers  $\delta$  7.99 – 7.74 (m, 1H), 7.54 (s, 1H), 7.45 – 6.51 (m, 10H), 5.14 and 5.11 (s, 2H), 4.34 and 4.32 (s, 2H), 4.20 – 4.15 (m, 2H), 3.35 – 3.09 (m, 2H), 2.89 (hept, J = 7.1 Hz, 1H), 2.29 – 2.03 (m, 2H), 1.82 – 1.57 (m, 2H), 1.24 (d, J = 6.7 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.61, 156.00, 153.42, 151.38, 149.50, 149.27, 142.00, 136.32, 133.20, 131.35, 129.48, 129.03, 128.75, 128.37, 128.18, 123.94, 123.91, 123.31, 123.25, 77.36, 67.90, 67.73, 49.77, 49.58, 47.03, 45.95, 34.24, 29.99, 29.62, 29.45, 24.94, 24.90, 24.51, 24.19, 23.64. IR (neat, cm⁻¹): 2958.23, 2868.04, 1681.90, 1585.53, 1463.21, 1418.71, 1315.71, 1217.70, 1153.14, 882.71, 741.03. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₄H₄₄O₄N₃SBr₂⁺: 748.14193, Found: 748.14630.



Benzyl

triisopropylbenzenesulfonyl hydrazone (10) Yield: 72%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.40. ¹H NMR (600 MHz, CDCl₃) some signals

2,4,6-

exist as a pair due to the presence of amide rotamers  $\delta$  8.35 – 8.20 (m, 1H), 7.88 – 6.95 (m, 15H), 5.24 and 5.21 (br, 2H), 4.65 and 4.62 (s, 2H), 4.30 – 4.26 (m, 2H), 3.37 – 3.17 (m, 2H), 2.93 (hept, J = 6.6 Hz, 1H), 2.19 – 2.08 (m, 2H), 1.77 – 1.66 (m, 2H), 1.30 (d, J = 6.9 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.32, 153.25, 151.33, 149.68, 149.46, 136.67, 135.13, 133.34, 132.82, 131.48, 128.61, 128.49, 128.12, 128.06, 127.99, 127.81, 127.70, 126.61, 126.28, 125.92, 125.44, 123.83, 67.59, 67.51, 67.38, 50.75, 50.52, 46.50, 45.38, 34.18, 29.94, 29.91, 29.68, 29.50, 24.91, 24.85, 24.55, 24.20, 23.60. IR (neat, cm⁻¹): 2958.51, 1681.93, 1599.51, 1462.44, 1425.01, 1215.77, 1153.11, 750.96, 664.40. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₈H₄₈O₄N₃S⁺: 642.33655, Found: 642.33630.

#### Benzyl (furan-3-ylmethyl)(4-oxobutyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1p) Yield: 70%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.30$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide

rotamers  $\delta$  8.06 – 7.70 (m, 1H), 7.49 – 6.91 (m, 9H), 6.40 – 5.99 (m, 2H), 5.17 and 5.12 (br, 2H), 4.48 – 4.35 (m, 2H), 4.25 – 4.16 (m, 2H), 3.42 – 3.16 (m, 2H), 2.96 – 2.84 (m, 1H), 2.21 – 1.96 (m, 2H), 1.74 – 1.51 (m, 2H), 1.26 (d, *J* = 6.9 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.22, 153.33, 151.39, 151.20, 149.96, 149.69, 142.24, 136.68, 131.44, 128.57, 128.23, 128.09,

127.95, 123.87, 123.74, 110.45, 67.43, 46.67, 45.81, 43.70, 34.24, 29.98, 29.63, 29.47, 24.93, 24.88, 24.66, 24.34, 23.63. IR (neat, cm⁻¹): 2958.11, 2868.83, 1678.47, 1599.30, 1423.39, 1316.93, 1152.03, 1011.91, 910.23, 730.80. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₂H₄₃O₅N₃SNa⁺: 604.28211, Found: 604.28229.

# Benzyl (4-oxobutyl)(thiophen-2-ylmethyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1q) Yield: 71%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.33$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to

the presence of amide rotamers  $\delta$  7.51 – 6.48 (m, 12H), 5.18 and 5.15 (br, 2H), 4.56 and 4.53 (s, 2H), 4.21 – 4.16 (m, 2H), 3.34 – 3.16 (m, 2H), 2.95 – 2.82 (m, 1H), 2.22 – 2.04 (m, 2H), 1.77 – 1.61 (m, 2H), 1.26 – 1.24 (m, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.27, 155.83, 153.30, 151.36, 149.77, 149.53, 140.47, 140.32, 136.56, 131.43, 128.59, 128.12, 126.76, 126.66, 126.34, 125.50, 125.37, 123.86, 77.36, 67.49, 46.28, 45.64, 45.48, 34.21, 29.95, 29.92, 29.64, 29.48, 24.93, 24.87, 24.67, 24.27, 23.61. IR (neat, cm⁻¹): 2958.83, 2868.65, 1678.15, 1599.27, 1462.58, 1423.55, 1216.61, 1163.92, 1152.52, 907.97, 729.42. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₂H₄₃O₄N₃S₂Na⁺: 620.25927, Found: 620.25983.

#### Benzyl

# (4-oxobutyl)(thiophen-3-ylmethyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1r) Yield: 72%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.33$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to

the presence of amide rotamers  $\delta$  8.29 – 8.13 (m, 1H), 7.40 – 6.85 (m, 11H), 5.19 and 5.15 (br, 2H), 4.42 and 4.40 (br, 2H), 4.22 (hept, J = 7.1 Hz, 2H), 3.41 – 3.10 (m, 2H), 2.95 –

2.83 (m, 1H), 2.16 – 2.06 (m, 2H), 1.69 – 1.62 (m, 2H), 1.27 (d, J = 6.8 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.10, 153.29, 151.36, 149.76, 149.54, 148.43, 138.60, 136.67, 131.49, 128.60, 128.16, 128.06, 127.72, 127.21, 126.48, 126.23, 123.85, 122.83, 122.68, 122.38, 67.56, 67.47, 67.39, 46.39, 45.94, 45.48, 34.21, 29.94, 29.60, 29.46, 24.93, 24.87, 24.63, 24.24, 23.61. IR (neat, cm⁻¹): 2958.18, 2868.40, 1690.79, 1599.19 1461.52, 1426.61, 1214.68, 1083.13, 1009.48, 754.22. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₂H₄₃O₄N₃S₂Na⁺: 620.25927, Found: 620.26032.

Benzyl (4-oxobutyl)(pyridin-2-ylmethyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1s) Yield: 75%. 100% ethyl acetate,  $R_f = 0.35$ . ¹H NMR (500 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  8.56

-8.45 (m, 1H), 8.33 - 8.10 (m, 1H), 7.65 - 7.52 (m, 1H), 7.40 - 6.96 (m, 10H), 5.14 and 5.10 (s, 2H), 4.56 and 4.53 (s, 2H), 4.29 - 4.11 (m, 2H), 3.41 - 3.27 (m, 2H), 2.89 (hept, J = 6.9 Hz, 1H), 2.22 - 2.02 (m, 2H), 1.80 - 1.59 (m, 2H), 1.23 (d, J = 7.0 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  157.84, 157.75, 156.62, 156.28, 153.20, 151.34, 149.75, 149.58, 149.24, 149.12, 136.96, 136.82, 136.59, 131.48, 128.63, 128.47, 128.23, 128.16, 128.04, 127.88, 123.84, 122.48, 122.34, 122.15, 121.20, 67.58, 67.54, 67.32, 52.69, 52.47, 47.50, 46.54, 34.19, 29.92, 29.90, 29.77, 29.55, 24.89, 24.50, 23.60. IR (neat, cm⁻¹): 2958.18, 2868.20, 1694.43, 1597.52, 1463.37, 1422.45, 1215.38, 1152.74, 750.48. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₃H₄₅O₄N₄S⁺: 593.31615, Found: 593.31671.



Benzyl

triisopropylbenzenesulfonyl hydrazone (1t) Yield: 78%. 100% ethyl acetate,  $R_f = 0.37$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a

pair due to the presence of amide rotamers  $\delta$  8.38 – 7.87 (m, 1H), 7.58 – 6.54 (m, 11H), 5.12 and 5.08 (br, 2H), 4.49 and 4.46 (br, 2H), 4.29 – 4.11 (m, 2H), 3.44 – 3.25 (m, 2H), 2.89 (hept, J = 6.9 Hz, 1H), 2.24 – 2.03 (m, 2H), 1.80 – 1.63 (m, 2H), 1.23 (d, J = 7.1 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  159.55, 159.40, 156.55, 156.13, 153.24, 151.33, 149.67, 149.50, 141.66, 141.50, 139.19, 139.06, 136.49, 136.41, 131.38, 128.66, 128.51, 128.28, 128.15, 128.09, 128.03, 127.90, 126.74, 126.63, 123.83, 120.76, 119.81, 67.65, 67.61, 67.39, 52.43, 52.11, 47.71, 46.90, 34.19, 29.93, 29.70, 29.49, 24.90, 24.51, 23.61. IR (neat, cm⁻¹): 3172.55, 2958.63, 2868.27, 1681.45, 1557.02, 1425.67, 1316.71, 1153.42, 1119.25, 753.70. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₃H₄₄O₄N₄SBr⁺: 671.22666, Found: 671.22681.

#### Benzyl (4-oxobutyl)(pyridin-3-ylmethyl)carbamate 2,4,6-triisopropylbenzenesulfonyl



hydrazone (1u) Yield: 84%. 100% ethyl acetate,  $R_f = 0.31$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  8.58

- 8.02 (m, 3H), 7.71 – 6.96 (m, 10H), 5.14 and 5.13 (br, 2H), 4.45 and 4.41 (br, 2H), 4.21 – 4.16 (m, 2H), 3.27 - 3.17 (m, 2H), 2.92 - 2.82 (m, 1H), 2.17 - 2.09 (m, 2H), 1.72 - 1.65 (m, 2H), 1.24 (d, J = 6.8 Hz, 18H). ¹³C NMR (125 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  156.67, 153.32, 151.34, 149.13, 149.01, 148.85, 136.46, 136.08, 135.32, 133.54, 131.53, 128.68, 128.32, 128.19, 123.87, 123.75,

67.84, 67.64, 48.38, 46.78, 45.80, 34.22, 29.95, 29.62, 29.44, 24.93, 24.87, 24.61, 24.24, 23.63. IR (neat, cm⁻¹): 2958.11, 2868.47, 1695.99, 1598.90, 1464.24, 1419.75, 1320.01, 1164.43, 1153.44, 909.16, 730.30. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₃H₄₅O₄N₄S⁺: 593.31615, Found: 593.31677.

# Benzyl (benzo[b]thiophen-3-ylmethyl)(4-oxobutyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1v) Yield: 86%. Hexanes/ethyl acetate = 3:1,  $R_f$  = 0.36. ¹H NMR (600 MHz, CDCl₃) some signals exist as a

pair due to the presence of amide rotamers  $\delta$  8.07 – 6.69 (m, 14H), 5.23 and 5.19 (br, 2H), 4.72 and 4.68 (br, 2H), 4.21 – 4.15 (m, 2H), 3.33 – 3.11 (m, 2H), 2.93 – 2.86 (m, 1H), 2.13 – 2.03 (m, 2H), 1.67 – 1.57 (m, 2H), 1.26 (d, *J* = 6.8 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  153.44, 151.41, 149.91, 149.69, 140.70, 138.11, 137.87, 136.74, 132.51, 132.31, 131.36, 128.73, 128.70, 128.38, 128.31, 128.27, 125.21, 124.85, 124.69, 124.49, 123.94, 122.92, 122.39, 121.76, 67.59, 46.09, 45.13, 44.77, 34.27, 30.02, 29.65, 29.42, 24.97, 24.91, 24.39, 24.13, 23.66. IR (neat, cm⁻¹): 2958.01, 2868.37, 1677.23, 1599.06, 1459.39, 1423.21, 1314.54, 1163.89, 907.36, 728.00. HRMS (ESI) ([M+Na]⁺) Calcd. for C₃₆H₄₅O₄N₃S₂Na⁺: 670.27492, Found: 670.27521.

#### Benzyl

#### (4-oxobutyl)(quinolin-2-ylmethyl)carbamate 2,4,6-



triisopropylbenzenesulfonyl hydrazone (1w) Yield: 81%. 100% ethyl acetate,  $R_f = 0.34$ . ¹H NMR (600 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers.  $\delta$  8.26 – 8.00 (m, 2H), 7.80 – 7.69 (m, 3H), 7.60 – 6.71 (m, 10H), 5.18 and 5.13 (s, 2H), 4.75 and 4.71 (s, 2H), 4.24 – 4.16 (m, 2H), 3.43 – 3.30 (m, 2H), 2.89 (hept, J = 7.1 Hz, 1H), 2.21 – 2.09 (m, 2H), 1.76 – 1.66 (m, 2H), 1.24 (d, J = 7.3 Hz, 18H). ¹³C NMR (150 MHz, CDCl₃) some signals exist as a pair due to the presence of amide rotamers  $\delta$  158.30, 156.84, 156.39, 153.34, 151.40, 150.00, 149.78, 147.72, 137.05, 136.68, 136.59, 131.37, 129.84, 129.09, 128.75, 128.49, 128.38, 128.30, 128.07, 127.93, 127.72, 127.53, 127.40, 126.52, 123.90, 120.11, 119.14, 67.70, 67.43, 53.51, 53.29, 47.60, 46.45, 34.28, 34.26, 30.00, 29.82, 29.52, 24.94, 24.81, 24.58, 23.65. IR (neat, cm⁻¹): 2958.51, 2868.74, 1681.93, 1599.51, 1462.44, 1425.01, 1215.77, 1153.11, 750.96. HRMS (ESI) ([M+H]⁺) Calcd. for C₃₇H₄₇O₄N₄S⁺: 643.33180, Found: 643.33289.

# **3.4.3. Synthesis and Characterization of Piperidines**

# **3.4.3.1. Experimental Procedure for [Co(Por)]-Catalyzed Enantioselective Radical** Cyclization

A 10 mL oven-dried Schlenk tube was charged with *N*-sulfonyl hydrazone (0.10 mmol), [Co(Por)] (2 mol %) and Cs₂CO₃ (0.20 mmol, 2.0 equiv). The Schlenk tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen 3 times. Under nitrogen atmosphere, anhydrous benzene (2.0 mL) was added. The Schlenk tube was then purged with nitrogen for 1 min and sealed with the Teflon screw cap. The reaction mixture was stirred at 22 °C for 48 h. Following completion of the reaction, the reaction mixture was filtered through a pad of silica gel, concentrated under vacuum and purified by flash column chromatography.

Benzyl (S)-2-phenylpiperidine-1-carboxylate (2a) Following general procedure using 1a



as the starting material. Yield: 78%. Hexanes/ethyl acetate = 8:1,  $R_f$ = 0.36.  $[\alpha]_D^{20}$  = (-)-70.4° (c = 0.5, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.37 – 7.28 (m, 7H), 7.26 – 7.22 (m, 3H), 5.53 (s, 1H),

5.20 (s, 2H), 4.14 (d, J = 10.9 Hz, 1H), 2.85 (ddd, J = 13.6, 11.9, 3.9 Hz, 1H), 2.34 (d, J = 15.6 Hz, 1H), 1.95 - 1.88 (m, 1H), 1.66 - 1.46 (m, 4H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$ 156.29, 139.95, 137.07, 128.74, 128.57, 128.02, 127.89, 126.67, 126.65, 67.29, 53.67, 40.63, 28.19, 25.59, 19.45. IR (neat, cm⁻¹): 2934.29, 2860.66, 1694.40, 1421.73, 1258.07, 1162.18, 1028.74. HPLC analysis: ee = 92%. ID(95% hexanes: 5% isopropanol, 0.8 mL/min):  $t_{major} = 23.66 \text{ min}, t_{minor} = 26.31 \text{ min}. \text{ HRMS (DART) ([M+H]^+) Calcd. for}$ C₁₉H₂₂NO₂⁺: 296.16451, Found: 296.16263.

Benzyl (S)-2-(*m*-tolyl)piperidine-1-carboxylate (2b) Following general procedure using



**1b** as the starting material. Yield: 74%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.48. \ [\alpha]_D^{20} = (-)-92.0^{\circ} (c = 1.0, CHCl_3).$  ¹H NMR (500 MHz,  $CDCl_3$ )  $\delta$  7.43 – 7.21 (m, 6H), 7.10 – 7.00 (m, 3H), 5.51 (s, 1H), 5.22 (s, 2H), 4.15 (d, J = 12.7 Hz, 1H), 2.92 – 2.83 (m, 1H), 2.35 – 2.33 (m, 4H), 1.95 – 1.87 (m, 1H), 1.59 – 1.47 (m, 4H). ¹³C NMR (125 MHz, CDCl₃) δ 156.26, 139.91, 138.32, 137.13, 128.61, 128.56, 127.98, 127.85, 127.39, 127.35, 123.67, 67.22, 53.62, 40.63, 28.20, 25.59, 21.75, 19.48. IR (neat, cm⁻¹): 2937.30, 2861.67, 1694.88, 1421.48, 1259.04, 1114.44, 793.52. HPLC analysis: ee = 94%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 12.86 \text{ min}, t_{minor} = 15.84 \text{ min}. \text{ HRMS (DART) ([M+H]^+) Calcd. for}$ C₂₀H₂₄NO₂⁺: 310.18016, Found: 310.18091.

Benzyl (S)-2-(4-(tert-butyl)phenyl)piperidine-1-carboxylate (2c) Following general



procedure using 1c as the starting material. Yield: 80%. Hexanes/ethyl acetate = 8:1,  $R_f = 0.48$ .  $[\alpha]_D^{20} = (-)-87.4^\circ$  (c = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.42 – 7.27 (m, 7H), 7.16

(d, J = 8.4 Hz, 2H), 5.51 (s, 1H), 5.20 (s, 2H), 4.14 (d, J = 12.4 Hz, 1H), 2.92 – 2.82 (m, 1H), 2.34 (d, J = 13.2 Hz, 1H), 1.93 – 1.88 (m, 1H), 1.66 – 1.50 (m, 4H), 1.33 (s, 9H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$  156.26, 149.44, 137.12, 136.78, 128.55, 127.96, 127.86, 126.36, 125.59, 67.21, 53.40, 40.55, 34.48, 31.49, 28.13, 25.63, 19.47. IR (neat, cm⁻¹): 2949.16, 2865.44, 1696.01, 1421.63, 1254.91, 1163.46, 1104.88, 1029.28, 697.00. HPLC analysis: ee = 92%. IA(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 11.76$  min,  $t_{minor} = 10.55$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₃H₃₀NO₂⁺: 352.22711, Found: 352.22837.

Benzyl (S)-2-(3-methoxyphenyl)piperidine-1-carboxylate (2d) Following general



procedure using **1d** as the starting material. Yield: 51%. ee: 90%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.38$ .  $[\alpha]_D^{20} = (-)-78.0^\circ$  (*c* = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.43 – 7.19 (m, 6H), 6.85

- 6.72 (m, 3H), 5.48 (s, 1H), 5.18 (s, 2H), 4.12 (d, J = 12.8 Hz, 1H), 3.75 (s, 3H), 2.85 (td, J = 12.8, 3.6 Hz, 1H), 2.29 (d, J = 14.1 Hz, 1H), 1.88 (ddd, J = 13.2, 8.8, 4.8 Hz, 1H), 1.63 - 1.46 (m, 4H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$  160.15, 156.23, 141.83, 137.09, 129.72, 128.58, 128.02, 127.91, 119.01, 112.88, 111.64, 67.27, 55.31, 53.66, 40.72, 28.28, 25.53, 19.51. IR (neat, cm⁻¹): 2938.04, 1695.51, 1489.28, 1422.84, 1256.89, 1170.28, 1140.90, 1034.14. HPLC analysis: ee = 90%. IC(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major}$ 

= 17.17 min,  $t_{minor}$  = 19.37 min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₄NO₃⁺: 326.17507, Found: 326.17589.

Benzyl (S)-2-(4-methoxyphenyl)piperidine-1-carboxylate (2e) Yield: 81 %. Following



general procedure using **1e** as the starting material. Hexanes/ethyl acetate = 5:1,  $R_f = 0.31$ .  $[\alpha]_D^{20} = (-)-91.0^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.39 – 7.27 (m, 5H), 7.16 – 7.12 (m,

2H), 6.89 - 6.85 (m, 2H), 5.48 (d, J = 4.5 Hz, 1H), 5.19 (s, 2H), 4.11 (d, J = 13.6 Hz, 1H), 3.80 (s, 3H), 2.82 (ddd, J = 13.6, 11.8, 3.8 Hz, 1H), 2.30 (d, J = 14.1 Hz, 1H), 1.94 - 1.83(m, 1H), 1.64 - 1.46 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  158.36, 156.23, 137.11, 131.83, 128.58, 128.01, 127.88, 127.82, 114.10, 67.25, 55.40, 53.11, 40.43, 28.15, 25.66, 19.41. IR (neat, cm⁻¹): 2937.20, 1693.66, 1610.36, 1511.68, 1422.83, 1248.66, 1162.48, 1035.26. HPLC analysis: ee = 91%. IF(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major}$ = 15.60 min,  $t_{minor}$  = 19.82 min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₄NO₃⁺: 326.17507, Found: 326.17450.

Benzyl (S)-2-(benzo[d][1,3]dioxol-5-yl)piperidine-1-carboxylate (2f) Following general



procedure using **1f** as the starting material. Yield: 70%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.30$ .  $[\alpha]_D^{20} = (-)-89.8^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.45 – 7.20 (m, 5H), 6.83

- 6.65 (m, 3H), 5.94 (s, 2H), 5.43 (d, *J* = 4.3 Hz, 1H), 5.19 (s, 2H), 4.11 (d, *J* = 14.2 Hz, 1H), 2.90 – 2.76 (m, 1H), 2.30 – 2.20 (m, 1H), 1.91 – 1.84 (m, 1H), 1.66 – 1.46 (m, 4H). ¹³C NMR (125 MHz, CDCl₃) δ 156.15, 148.28, 146.27, 137.03, 133.95, 128.59, 128.04, 127.89, 119.70, 108.34, 107.38, 101.12, 67.30, 53.45, 40.50, 28.34, 25.53, 19.39. IR (neat, cm⁻¹): 2938.67, 1691.57, 1489.38, 1421.66, 1352.25, 1236.59, 1167.60, 1035.75, 930.46. HPLC analysis: ee = 84%. IE(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major}$  = 22.54 min,  $t_{minor}$  = 23.94 min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₂NO₄⁺: 340.15433, Found: 340.15405.

Benzyl (S)-2-(3-nitrophenyl)piperidine-1-carboxylate (2g) Following general procedure



using **1g** as the starting material. Yield: 48%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.33$ .  $[\alpha]_D^{20} = (-)-103.2^\circ$  (c = 0.5, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  8.11 – 8.10 (m, 2H), 7.55 – 7.49 (m,

2H), 7.37 – 7.30 (m, 5H), 5.56 (s, 1H), 5.20 (d, J = 2.9 Hz, 2H), 4.19 (d, J = 13.6 Hz, 1H), 2.82 (td, J = 12.8, 3.9 Hz, 1H), 2.34 (d, J = 14.3 Hz, 1H), 2.03 – 1.96 (m, 1H), 1.72 – 1.53 (m, 3H), 1.45 – 1.36 (m, 1H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  156.16, 148.92, 142.82, 136.72, 132.97, 129.77, 128.68, 128.25, 128.01, 121.99, 121.74, 67.64, 53.49, 40.79, 28.21, 25.27, 19.34. IR (neat, cm⁻¹): 2941.57, 2863.22, 1695.33, 1528.06, 1419.74, 1348.26, 1257.80, 1162.29, 1034.63. HPLC analysis: ee = 95%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 28.30$  min,  $t_{minor} = 37.22$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₁N₂O₄⁺: 341.14958, Found: 341.14983.

**Benzyl** (S)-2-(4-cyanophenyl)piperidine-1-carboxylate (2h) Following general



procedure using **1h** as the starting material. Yield: 83%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.50$ .  $[\alpha]_D^{20} = (-)-110.0^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.62 (d, J = 8.4 Hz,

2H), 7.42 – 7.24 (m, 7H), 5.51 (s, 1H), 5.18 (d, J = 3.8 Hz, 2H), 4.16 (d, J = 14.3 Hz, 1H), 2.79 (ddd, J = 13.5, 12.2, 3.8 Hz, 1H), 2.29 (dd, J = 14.3, 3.8 Hz, 1H), 1.99 – 1.92 (m, 1H), 1.70 – 1.53 (m, 3H), 1.40 – 1.31 (m, 1H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  156.16, 146.10, 136.71, 132.57, 128.65, 128.25, 128.03, 127.48, 118.90, 110.73, 67.59, 53.83, 40.87, 28.19, 25.24, 19.38. IR (neat, cm⁻¹): 2939.15, 2227.71, 1693.85, 1607.69, 1420.47, 1258.31, 1160.44, 1110.70, 1029.40. HPLC analysis: ee = 94%. IC(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major}$  = 32.13 min,  $t_{minor}$  = 40.78 min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₁N₂O₂⁺: 321.15975, Found: 321.15924.

Benzyl (S)-2-(4-(methoxycarbonyl)phenyl)piperidine-1-carboxylate (2i) Following



general procedure using **1i** as the starting material. Yield: 70%. Hexanes/ethyl acetate = 5:1,  $R_f$  = 0.24.  $[\alpha]_D^{20} = (-)-96.6^\circ$  (*c* = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  8.00 (d, *J* = 8.2 Hz, 2H), 7.39 – 7.24 (m, 7H), 5.53 (s, 1H), 5.19 (s, 2H), 4.16 (d, *J* 

= 13.5 Hz, 1H), 3.91 (s, 3H), 2.82 (td, J = 12.9, 3.7 Hz, 1H), 2.33 (d, J = 14.2 Hz, 1H), 1.97 – 1.91 (m, 1H), 1.69 – 1.52 (m, 3H), 1.45 – 1.36 (m, 1H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$  167.01, 156.20, 145.62, 136.87, 130.05, 128.69, 128.61, 128.12, 127.92, 126.69, 67.43, 53.82, 52.19, 40.80, 28.28, 25.40, 19.44. IR (neat, cm⁻¹): 2943.50, 2862.76, 1720.05, 1691.61, 1610.57, 1419.04, 1276.64, 1251.88, 1104.99, 1017.73, 860.77. HPLC analysis: ee = 86%. IA(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major} = 9.98$  min,  $t_{minor} = 8.70$ min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₂₄NO₄⁺: 354.16998, Found: 354.16887. The absolute configuration was assigned as (*S*) by X-ray crystallography.

Benzyl (S)-2-(4-(trifluoromethyl)phenyl)piperidine-1-carboxylate (2j) Following



general procedure using **1j** as the starting material. Yield: 73%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.48$ .  $[\alpha]_D^{20} = (-)-77.8^\circ$  (c = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.57 (d, J = 8.2 Hz, 2H),

7.41 – 7.19 (m, 7H), 5.53 (s, 1H), 5.18 (s, 2H), 4.16 (d, *J* = 14.8 Hz, 1H), 2.82 (td, *J* = 13.0,

3.7 Hz, 1H), 2.31 (d, J = 12.7 Hz, 1H), 1.98 – 1.91 (m, 1H), 1.70 – 1.52 (m, 3H), 1.42 – 1.35 (m, 1H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$  156.21, 144.48, 136.83, 129.08 (q, J = 31.5 Hz), 128.63, 128.18, 127.98, 127.03, 125.70 (q, J = 3.7 Hz), 124.31 (q, J = 270.0 Hz), 67.49, 53.67, 40.78, 28.29, 25.35, 19.36. ¹⁹F NMR (564 MHz, CDCl₃)  $\delta$  –62.48. IR (neat, cm⁻¹): 2939.63, 1697.62, 1619.26, 1420.52, 1327.08, 1257.37, 1162.41, 1120.77, 1068.45. HPLC analysis: ee = 94%. ID(95% hexanes: 5% isopropanol, 0.8 mL/min):  $t_{major} = 10.62$  min,  $t_{minor} = 10.16$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₀H₂₁NO₂F₃⁺: 364.15189, Found: 364.15117.





procedure using **1k** as the starting material. Yield: 69%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.42$ .  $[\alpha]_D^{20} = (-)-82.2^\circ$  (c = 1.0,

CHCl₃). ¹H NMR (500 MHz, CDCl₃) δ 7.38 – 7.26 (m, 6H), 7.00

(d, J = 7.9 Hz, 1H), 6.97 – 6.90 (m, 2H), 5.49 (s, 1H), 5.19 (s, 2H), 4.14 (d, J = 13.6 Hz, 1H), 2.83 (td, J = 12.9, 3.6 Hz, 1H), 2.28 (d, J = 14.3 Hz, 1H), 1.97 – 1.86 (m, 1H), 1.64 – 1.41 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  163.49 (d, J = 245.6 Hz), 156.18, 143.02 (d, J = 6.5 Hz), 136.93, 130.22 (d, J = 8.2 Hz), 128.62, 128.11, 127.93, 122.24 (d, J = 2.8 Hz), 113.82 (d, J = 22.3 Hz), 113.60 (d, J = 21.0 Hz), 67.43, 53.47, 40.70, 28.25, 25.45, 19.42. ¹⁹F NMR (470 MHz, CDCl₃)  $\delta$  –112.96. IR (neat, cm⁻¹): 2938.49, 1694.30, 1614.17, 1588.45, 1420.29, 1256.44, 1172.63, 1112.17, 1029.46. HPLC analysis: ee = 92%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 14.46$  min,  $t_{minor} = 16.48$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₁NO₂F⁺: 314.15508, Found: 314.15596.

Benzyl (S)-2-(3,4-difluorophenyl)piperidine-1-carboxylate (21) Following general



procedure using **11** as the starting material. Yield: 67%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.50$ .  $[\alpha]_D^{20} = (-)-81.8^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.42 – 7.25 (m, 5H), 7.13

- 7.08 (m, 1H), 7.06 – 6.98 (m, 1H), 6.95 – 6.89 (m, 1H), 5.45 (s, 1H), 5.19 (s, 2H), 4.13 (d, J = 13.8 Hz, 1H), 2.79 (td, J = 13.0, 4.7 Hz, 1H), 2.23 (d, J = 14.4 Hz, 1H), 1.95 – 1.87 (m, 1H), 1.70 – 1.41 (m, 4H). ¹³C NMR (150 MHz, CDCl₃) δ 156.11, 150.73 (dd, J = 243.8, 12.6 Hz), 149.19 (dd, J = 246.1, 12.6 Hz), 137.26 (dd, J = 4.3, 3.8 Hz), 136.81, 128.64, 128.19, 127.98, 122.57 (dd, J = 6.2, 3.5 Hz), 117.42 (d, J = 17.2 Hz), 115.88 (d, J = 17.9 Hz), 67.50, 53.00, 40.57, 28.14, 25.39, 19.30. ¹⁹F NMR (470 MHz, CDCl₃) δ –137.46, – 141.03. IR (neat, cm⁻¹): 2941.00, 1696.86, 1607.88, 1518.25, 1423.18, 1272.39, 1171.43, 1105.00, 1036.15. HPLC analysis: ee = 94%. ID(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major} = 9.46$  min,  $t_{minor} = 10.57$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₀NO₂F₂⁺: 332.14566, Found: 332.14627.

Benzyl (S)-2-(4-bromophenyl)piperidine-1-carboxylate (2m) Following general



procedure using **1m** as the starting material. Yield: 66%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.46$ .  $[\alpha]_D^{20} = (-)-90.0^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.45 (d, J = 8.5 Hz, 2H),

7.39 – 7.28 (m, 5H), 7.09 (d, J = 8.7 Hz, 2H), 5.45 (s, 1H), 5.18 (d, J = 2.2 Hz, 2H), 4.13 (d, J = 16.5 Hz, 1H), 2.90 – 2.70 (m, 1H), 2.27 (d, J = 12.5 Hz, 1H), 1.95 – 1.86 (m, 1H), 1.66 – 1.41 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  156.19, 139.18, 136.91, 131.82, 128.62, 128.53, 128.13, 127.94, 120.60, 67.41, 53.35, 40.62, 28.14, 25.46, 19.35. IR (neat, cm⁻¹): 2938.34, 1696.36, 1487.97, 1421.06, 1257.63, 1161.38, 1075.21, 1008.36. HPLC

analysis: ee = 91%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major}$  = 13.09 min,  $t_{minor}$  = 14.15 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₁NO₂Br⁺: 374.07502, Found: 374.07570.

Benzyl (S)-2-(3,5-dibromophenyl)piperidine-1-carboxylate (2n) Following general



procedure using **1n** as the starting material. Yield: 53%. Hexanes/ethyl acetate = 8:1,  $R_f = 0.36$ .  $[\alpha]_D^{20} = (-)-74.0^\circ$  (c = 0.5, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.52 (s, 1H), 7.38 – 7.21

(m, 7H), 5.42 (s, 1H), 5.21 – 5.13 (m, 2H), 4.13 (d, J = 12.3 Hz, 1H), 2.79 (td, J = 13.1, 3.4 Hz, 1H), 2.18 (d, J = 17.5 Hz, 1H), 1.91 – 1.85 (m, 1H), 1.64 – 1.50 (m, 3H), 1.44 – 1.33 (m, 1H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$  156.04, 144.68, 136.76, 132.54, 128.72, 128.67, 128.22, 127.92, 123.54, 67.59, 53.23, 40.77, 28.20, 25.25, 19.35. IR (neat, cm⁻¹): 2938.22, 2861.49, 1697.13, 1553.30, 1419.67, 1259.82, 1164.53, 740.54. HPLC analysis: ee = 93%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 10.64$  min,  $t_{minor} = 11.32$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₉H₂₀NO₂Br₂⁺: 451.98553, Found: 451.98584.



procedure using **10** as the starting material. Yield: 78%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.50$ .  $[\alpha]_D^{20} = (-)-139.8^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.86 – 7.76 (m, 3H),

7.66 (s, 1H), 7.51 – 7.45 (m, 2H), 7.40 – 7.25 (m, 6H), 5.69 (s, 1H), 5.24 (s, 2H), 4.20 (d, J = 13.7 Hz, 1H), 3.00 – 2.86 (m, 1H), 2.48 (d, J = 17.6 Hz, 1H), 2.05 – 1.96 (m, 1H), 1.71 – 1.52 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  156.33, 137.52, 137.06, 133.55, 132.37, 128.58, 128.51, 128.02, 127.88, 127.62, 126.20, 125.86, 125.35, 125.02, 67.33, 53.85,

40.78, 28.21, 25.60, 19.51. IR (neat, cm⁻¹): 2935.08, 1693.76, 1421.32, 1316.03, 1257.07, 1175.43, 1108.52, 1029.30. HPLC analysis: ee = 91%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 14.91 \text{ min}, t_{minor} = 17.19 \text{ min}. \text{ HRMS (DART) ([M+H]^+) Calcd. for}$ C₂₃H₂₄NO₂⁺: 346.18016, Found: 346.18022.

Benzyl (S)-2-(furan-3-yl)piperidine-1-carboxylate (2p) Following general procedure



using 1p as the starting material. Yield: 30%. Hexanes/ethyl acetate = 8:1,  $R_f = 0.32$ .  $[\alpha]_D^{20} = (-)-60.0^\circ$  (c = 0.5, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.67 – 7.08 (m, 8H), 6.21 (s, 1H), 5.42 (s, 1H), 5.18

(s, 2H), 4.06 (d, J = 11.0 Hz, 1H), 2.83 (t, J = 13.3 Hz, 1H), 2.00 (d, J = 13.8 Hz, 1H), 1.88 -1.78 (m, 1H), 1.67 - 1.48 (m, 4H). ¹³C NMR (125 MHz, CDCl₃) δ 155.75, 143.39, 139.74, 137.04, 128.63, 128.10, 127.96, 125.02, 110.04, 67.28, 47.85, 40.22, 28.61, 25.63, 19.69. IR (neat, cm⁻¹): 2939.37, 1694.72, 1422.43, 1324.70, 1263.71, 1170.34, 1117.41, 1026.33. HPLC analysis: ee = 91%. IE(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 16.80$ min,  $t_{minor} = 15.49$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₂₀NO₃⁺: 286.14377, Found: 286.14437.

Benzyl (S)-2-(thiophen-2-yl)piperidine-1-carboxylate (2q) Following general procedure



using 1q as the starting material. Yield: 53%. Hexanes/ethyl acetate = 8:1,  $R_f = 0.40$ .  $[\alpha]_{D}^{20} = (-)-74.4$  (c = 0.5, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.41 – 7.29 (m, 5H), 7.21 (d, J = 5.1 Hz, 1H), 6.96 (dd, J = 5.1, 3.5 Hz, 1H), 6.83 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 4.10 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 4.10 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 4.10 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 4.10 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 4.10 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 4.10 (d, J = 3.5 Hz, 1H), 5.70 (s, 1H), 5.20 (s, 2H), 5

13.5 Hz, 1H), 2.95 (t, J = 13.1 Hz, 1H), 2.19 (d, J = 14.0 Hz, 1H), 1.93 (ddd, J = 13.7, 11.5, 12.55.6 Hz, 1H), 1.68 – 1.49 (m, 4H). ¹³C NMR (150 MHz, CDCl₃) δ 155.68, 144.85, 136.93, 128.61, 128.09, 127.98, 127.04, 124.68, 124.48, 67.42, 51.22, 40.32, 29.85, 25.44, 19.63. IR (neat, cm⁻¹): 2935.71, 2854.39, 1698.10, 1418.89, 1319.37, 1259.04, 1231.40, 1109.45, 1028.21. HPLC analysis: ee = 92%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major}$ = 15.12 min,  $t_{minor}$  =16.64 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₂₀NO₂S⁺: 302.12093, Found: 302.12113.

Benzyl (S)-2-(thiophen-3-yl)piperidine-1-carboxylate (2r) Following general procedure



using **1r** as the starting material. Yield: 77%. Hexanes/ethyl acetate = 8:1,  $R_f = 0.38$ .  $[\alpha]_D^{20} = (-)-100.0^\circ$  (c = 1.0, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  7.48 – 7.17 (m, 6H), 6.97 (s, 1H), 6.89 (d, J = 5.0

Hz, 1H), 5.50 (s, 1H), 5.18 (s, 2H), 4.09 (d, J = 13.5 Hz, 1H), 2.82 (t, J = 12.8 Hz, 1H), 2.20 (d, J = 13.9 Hz, 1H), 1.90 – 1.86 (m, 1H), 1.67 – 1.48 (m, 4H). ¹³C NMR (150 MHz, CDC1₃)  $\delta$  155.84, 141.77, 137.04, 128.61, 128.07, 127.93, 127.04, 126.07, 121.11, 67.29, 51.42, 40.47, 28.90, 25.64, 19.67. IR (neat, cm⁻¹): 2938.86, 2861.25, 1693.38, 1421.95, 1258.51, 1113.00, 1036.04. HPLC analysis: ee = 93%. IF(95% hexanes: 5% isopropanol, 0.8 mL/min):  $t_{major} = 17.85$  min,  $t_{minor} = 19.78$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₇H₂₀NO₂S⁺: 302.12093, Found: 302.12179.

Benzyl (S)-2-(pyridin-2-yl)piperidine-1-carboxylate (2s) Following general procedure



using **1s** as the starting material. Yield: 87%. Hexanes/ethyl acetate = 3:1,  $R_f = 0.33$ .  $[\alpha]_D^{20} = (-)-96.4^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  8.58 (d, J = 4.3 Hz, 1H), 7.61 (td, J = 7.7, 1.8 Hz,

1H), 7.45 - 7.09 (m, 7H), 5.50 (s, 1H), 5.18 (s, 2H), 4.18 (d, J = 13.2 Hz, 1H), 3.02 - 2.90 (m, 1H), 2.68 - 2.61 (m, 1H), 1.89 - 1.82 (m, 1H), 1.65 - 1.35 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  160.06, 156.48, 149.31, 137.01, 136.64, 128.55, 128.00, 127.86, 121.62, 121.30, 67.29, 55.81, 41.28, 27.82, 25.42, 19.75. IR (neat, cm⁻¹): 2937.24, 1694.15,

1588.09, 1419.97, 1257.74, 1166.21, 1118.83, 1035.88. HPLC analysis: ee = 92%. IC(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major}$  = 25.81 min,  $t_{minor}$  = 15.35 min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₂₁N₂O₂⁺: 297.15975, Found: 297.15999.

Benzyl (S)-2-(6-bromopyridin-2-yl)piperidine-1-carboxylate (2t) Following general



procedure using **1t** as the starting material. Yield: 72%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.43$ .  $[\alpha]_D^{20} = (-)-80.6^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.46 (t, J = 7.8 Hz, 1H),

7.40 – 7.24 (m, 6H), 7.10 (d, J = 7.7 Hz, 1H), 5.46 (s, 1H), 5.17 (s, 2H), 4.18 (d, J = 12.3 Hz, 1H), 2.96 (t, J = 13.2 Hz, 1H), 2.59 (d, J = 12.8 Hz, 1H), 1.87 – 1.80 (m, 1H), 1.66 – 1.34 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  161.97, 156.34, 141.83, 138.92, 136.87, 128.60, 128.09, 127.89, 126.03, 120.14, 67.39, 55.51, 41.27, 27.85, 25.26, 19.65. IR (neat, cm⁻¹): 2938.02, 2858.74, 1695.33, 1578.24, 1553.80, 1410.23, 1260.03, 1170.74, 1112.70. HPLC analysis: ee = 96%. IA(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major} = 7.93$  min,  $t_{minor} = 7.46$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₂₀N₂O₂Br⁺: 375.07027, Found: 375.07083.

Benzyl (S)-2-(pyridin-3-yl)piperidine-1-carboxylate (2u) Following general procedure



using **1u** as the starting material. Yield: 21%. Hexanes/ethyl acetate = 1:2,  $R_f = 0.32$ .  $[\alpha]_D^{20} = (-)-64.5^\circ$  (c = 0.4, CHCl₃). ¹H NMR (600 MHz, CDCl₃)  $\delta$  8.52 (s, 1H), 8.49 (d, J = 4.9 Hz, 1H), 7.51 (d, J =

8.0 Hz, 1H), 7.45 – 7.16 (m, 6H), 5.55 (s, 1H), 5.18 (s, 2H), 4.15 (d, J = 13.7 Hz, 1H), 2.79 (td, J = 13.1, 3.5 Hz, 1H), 2.30 (d, J = 12.3 Hz, 1H), 2.01 – 1.94 (m, 1H), 1.74 – 1.37 (m, 4H). ¹³C NMR (150 MHz, CDCl₃)  $\delta$  156.12, 148.47, 148.06, 136.77, 135.57, 134.74, 128.66, 128.20, 128.00, 123.60, 67.55, 52.06, 40.64, 27.90, 25.40, 19.33. IR (neat, cm⁻¹):

2940.19, 1694.51, 1420.58, 1259.05, 1162.90, 1115.66, 1024.29. HPLC analysis: ee = 90%. ADH(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major} = 17.36 \text{ min}, t_{minor} = 11.78 \text{ min}.$  HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₂₁N₂O₂⁺: 297.15975, Found: 297.16051.

Benzyl (S)-2-(benzo[b]thiophen-3-yl)piperidine-1-carboxylate (2v) Following general



procedure using 1v as the starting material. Yield: 54%. Hexanes/ethyl acetate = 8:1,  $R_f = 0.32$ .  $[\alpha]_D^{20} = (-)-76.6^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.85 (d, J = 7.9 Hz, 1H),

7.80 (d, J = 8.0 Hz, 1H), 7.42 – 7.16 (m, 8H), 5.82 (d, J = 5.9 Hz, 1H), 5.31 – 5.16 (m, 2H), 4.09 (d, J = 13.6 Hz, 1H), 2.92 (td, J = 13.7, 13.2, 3.1 Hz, 1H), 2.30 (d, J = 13.9 Hz, 1H), 2.07 – 1.99 (m, 1H), 1.85 – 1.56 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  155.80, 140.46, 138.36, 137.02, 135.82, 128.59, 128.04, 127.90, 124.48, 124.26, 123.12, 122.80, 67.35, 50.59, 41.23, 28.40, 25.67, 20.15. IR (neat, cm⁻¹): 2937.98, 1689.81, 1419.41, 1311.05, 1249.83, 1168.72, 1026.68, 735.00. HPLC analysis: ee = 94%. ID(90% hexanes: 10% isopropanol, 0.8 mL/min):  $t_{major} = 21.94$  min,  $t_{minor} = 15.66$  min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₁H₂₂NO₂S⁺: 352.13658, Found: 352.13679.

Benzyl (S)-2-(quinolin-2-yl)piperidine-1-carboxylate (2w) Following general procedure



using **1w** as the starting material. Yield: 75%. Hexanes/ethyl acetate = 5:1,  $R_f = 0.47$ .  $[\alpha]_D^{20} = (-)-145.4^\circ$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  8.07 (dd, J = 13.0, 8.5 Hz, 2H), 7.79

(d, J = 8.1 Hz, 1H), 7.69 (t, J = 7.7 Hz, 1H), 7.51 (t, J = 7.5 Hz, 1H), 7.33 – 7.26 (m, 6H), 5.65 (s, 1H), 5.21 (s, 2H), 4.22 (d, J = 13.4 Hz, 1H), 3.07 (td, J = 12.5, 3.8 Hz, 1H), 2.79 (d, J = 13.5 Hz, 1H), 1.97 – 1.89 (m, 1H), 1.73 – 1.48 (m, 4H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  160.58, 156.61, 147.88, 137.06, 136.58, 129.59, 129.39, 128.53, 127.96, 127.80,

127.53, 126.95, 126.25, 119.33, 67.27, 56.37, 41.70, 27.83, 25.44, 19.82. IR (neat, cm⁻¹): 2936.71, 1693.99, 1600.17, 1502.91, 1421.97, 1253.41, 1168.25, 1120.18, 1072.11. HPLC analysis: ee = 93%. IE(80% hexanes: 20% isopropanol, 0.8 mL/min):  $t_{major}$  = 14.99 min,  $t_{minor}$  = 13.12 min. HRMS (DART) ([M+H]⁺) Calcd. for C₂₂H₂₃N₂O₂⁺: 347.17540, Found: 347.17515.

# 3.4.4. Mechanistic Studies of Proposed Stepwise Radical Mechanism

#### 3.4.4.1. General Procedure of HRMS Experiment

To an over-dried Schlenk tube, sulfonyl hydrazone **1a** (0.10 mmol), [Co(TPP)] (5 mol %), and Cs₂CO₃ (2.0 equiv) were added. The Schlenk tube was then evacuated and backfilled with nitrogen 3 times. The Teflon screw cap was replaced with a rubber septum, and benzene (2.0 mL) was added via a gas-tight syringe. The mixture was stirred at 22 °C for 1 h. The resulting solution was then passed through a short pad of silica gel (to get rid of base and salt) under the flow of nitrogen, and the filtrate was collected in a HPLC vial (degassed and backfilled with argon). The sample was further diluted with CH₃CN and immediately injected into HRMS instrument. The HRMS experiment was carried out in the absence of any additives such as formic acid, commonly act as electron carriers for ionization, allowing the detection of the molecular ion signals corresponding to Co(III)alkyl radical [C₆₃H₄₉CoN₅O₂]⁺ (m/z = 966.31720) by the loss of one electron.

#### **3.4.4.2. General Procedure of EPR Experiment**



To an oven-dried Schlenk tube, sulfonyl hydrazone **1a** (0.20 mmol, 1.0 equiv) and [Co(TPP)] (5 mol %) were added. The Schlenk tube was then evacuated and backfilled with nitrogen for 3 times. The Teflon screw cap was replaced with a rubber septum, Et₃N (2.0 equiv) and benzene (2.0 mL) was added via a gastight syringe. The mixture was then stirred at 22 °C for 1.5 h. The reaction mixture was then transferred into a degassed EPR tube (filled with argon) through a syringe. The sample was then carried out for EPR experiment at room temperature (EPR settings: T = 298 K; microwave frequency: 9.37762 GHz; power: 20 mW; modulation amplitude: 1.0 G).

The resulting notable EPR signal (in red) has been simulated (in blue) with 23% C-centered radical at  $\alpha$ -position  $I_{[Co(TPP)]/1a}$ : g = 2.00030,  $A_{(Co)} = 72.5$  MHz,  $A_{(H)} = 25.1$  MHz and 77% C-centered at  $\zeta$ -position  $II_{[Co(TPP)]/1a}$ : g = 2.01109,  $A_{(N)} = 4.9$  MHz,  $A_{(H)} = 7.2$  MHz, which is assigned to Co(III)-supported alkyl radical intermediates. [The simulation of the EPR spectrum was performed by iteration of the isotropic g-values and line widths using the EPR simulation program SpinFit Xenon]



# **EPR Simulation Details:**



g = 2.00030

 $A_{(C_0)} = 25.8887 \text{ x } 2.00030 \text{ x } 1.39961145 = 72.5 \text{ MHz}$  $A_{(H)} = 8.97801 \text{ x } 2.00030 \text{ x } 1.39961145 = 25.1 \text{ MHz}$ 



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g = 2.01109

 $A_{(N)} = 1.7349 \text{ x } 2.01109 \text{ x } 1.39961145 = 4.9 \text{ MHz}$  $A_{(H)} = 2.5621 \text{ x } 2.01109 \text{ x } 1.39961145 = 7.2 \text{ MHz}$ 

3.4.4.3. Experimental Procedure for Radical Trapping by EPR



To an oven-dried Schlenk tube, sulfonyl hydrazone **1a** (0.20 mmol, 1.0 equiv), *N-tert*butyl- $\alpha$ -phenylnitrone (0.5 equiv) and [Co(TPP)] (5 mol %) were added. The Schlenk tube

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was then evacuated and backfilled with nitrogen for 3 times. The Teflon screw cap was replaced with a rubber septum, Et₃N (2.0 equiv) and benzene (2.0 mL) was added via a gastight syringe. The mixture was then stirred at 22 °C for 1.5 h. The reaction mixture was then transferred into a degassed EPR tube (filled with argon) through a syringe. The sample was then carried out for EPR experiment at room temperature (EPR settings: T = 298 K; microwave frequency: 9.37762 GHz; power: 20 mW; modulation amplitude: 1.0 G).

The resulting notable EPR signal (in red) has been simulated (in blue) with 71% O-centered radical  $III_{[C0(TPP)]/1a}$ : g = 2.00630,  $A_{(N)}$  = 41.9 MHz,  $A_{(H)}$  = 8.0 MHz and 29% O-centered  $IV_{[C0(TPP)]/1a}$ : g = 2.00638,  $A_{(N)}$  = 40.2 MHz,  $A_{(H)}$  = 8.3 MHz, which is assigned to PBN-trapped Co(III)-supported alkyl radical intermediates. [The simulation of the EPR spectrum was performed by iteration of the isotropic g-values and line widths using the EPR simulation program SpinFit Xenon]



# **EPR Simulation Details:**



g = 2.00630

 $A_{(N)} = 14.93760 \text{ x } 2.00630 \text{ x } 1.39961145 = 41.9 \text{ MHz}$ 

 $A_{(H)} = 2.86027 \text{ x } 2.00630 \text{ x } 1.39961145 = 8.0 \text{ MHz}$ 



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g = 2.00638

 $A_{(N)} = 14.31220 \text{ x } 2.00638 \text{ x } 1.39961145 = 40.2 \text{ MHz}$  $A_{(H)} = 2.95343 \text{ x } 2.00638 \text{ x } 1.39961145 = 8.3 \text{ MHz}$ 

# 3.4.5. X-Ray Crystallography

The X-ray diffraction data were collected using Bruker-AXS SMART-APEXII CCD diffractometer (CuK $\alpha$ ,  $\lambda = 1.54178$  Å). Indexing was performed using *APEX2*²⁶ (Difference Vectors method). Data integration and reduction were performed using SaintPlus.²⁷ Absorption correction was performed by multi-scan method implemented in

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²⁶ Bruker (2012). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.

²⁷ Bruker (**2012**). SAINT. Data Reduction Software.

SADABS.²⁸ Space groups were determined using XPREP implemented in APEX2.¹⁸ The structure was solved using SHELXS-97 (direct methods) and refined using SHELXL97 contained in WinGX v1.70.01^{29,30,31} program.

²⁸ Sheldrick, G. M. (1996). SADABS. Program for Empirical Absorption Correction. University of Gottingen, Germany.

²⁹ Farrugia, L. J. J. Appl. Cryst. **1999**, 32, 837–838.

³⁰ Sheldrick, G. M. (**2012** Beta) SHELXL-97. Program for the Refinement of Crystal.

³¹ Sheldrick, G. M. Acta Cryst. 1990, A46, 467–473.





	Table S1. Crystal data and structure refin	nement for (–)-21			
I	Identification code	C21H23NO4			
Empirical formula		C21 H23 N O4			
Formula weight		353.40			
	Temperature	173(2) K			
	Wavelength	1.54178 ≈			
	Crystal system	Orthorhombic			
	Space group	P212121			
	Unit cell dimensions	$a = 6.1219(3) \approx$	$\alpha = 90 \infty$ .		
		$b = 11.7799(6) \approx$	$\beta = 90 \infty$ .		
		$c = 24.8215(12) \approx$	$\gamma = 90  \infty$ .		
	Volume	$1790.01(15) \approx^3$			
	Z	4			
	Density (calculated)	1.311 Mg/m ³			
	Absorption coefficient	0.736 mm ⁻¹			
	F(000)	752			
	Crystal size	0.240 x 0.180 x 0.080 mm ³			
Theta range for data collection		$3.561$ to $66.611 \infty$ .			
Index ranges		-7<=h<=7, -13<=k<=13, -28<=l<=29			
Reflections collected		17941			
Independent reflections		3116 [R(int) = 0.0233]			
Completeness to theta = $66.611 \infty$		99.4 %			
Absorption correction		Semi-empirical from equivalents			
Max. and min. transmission		0.7528 and 0.7076			
	Refinement method	Full-matrix least-squares on F ²			
Data / restraints / parameters		3116 / 0 / 236			
Goodness-of-fit on F ²		1.090			
Final R indices [I>2sigma(I)]		R1 = 0.0273, wR2 = 0.0667			
R indices (all data)		R1 = 0.0287, wR2 = 0.0680			
Absolute structure parameter		-0.04(5)			
I	Extinction coefficient	n/a			
	Largest diff. peak and hole	0.097 and -0.155 e. $\approx^{-3}$			

#### _ _ . _

### **3.4.6. DFT Calculations**

Considering the cost of time and computing resources for the large system with [Co(P6)], the geometry optimizations were performed with the Gaussian 16 at the BP86/lanl2dz level of theory in the gas phase at room temperature. Gas-phase Hessian matrix calculations were applied to the characterization of all minima (without imaginary frequency) and transition states (with only one imaginary frequency).

Thermochemical parameters such as internal energy, enthalpy, entropy, Gibbs free energy and thermal corrections (entropy and enthalpy, 298.15 K, 1 Atm) were obtained from these calculations. To further improve the accuracy of energies, single point energies were carried out at the B3LYP/def2-tzvplevel of theory along with Grimme's dispersion correction(D3BJ) and SMD solvation model (in toluene).

Independent Gradient Model (IGM)³² analysis was performed with Multiwfn³³ software package using high quality grid option to generate files for further plotting. The visualization of IGM analysis results were presented with VMD³⁴ visualization software. As shown in Scheme S2, the 3D diagrams of optimized structures were generated with CYLview software. ³⁵ The NCI (noncovalent interaction) visual representations of optimized structures were generated with VMD and rendered with Tachyon.³⁶

³² Lefebvre, C.; Rubez, G.; Khartabil, H.; Boisson, J.-C.; Contreras-García, J.; Hénon, E. *Phys. Chem. Chem. Phys.* **2017**, *19*, 17928–17936.

³³ Lu, T.; Chen, F. J. Comput. Chem. 2012, 33, 580–592.

³⁴ Humphrey, W.; Dalke, A.; Schulten, K. J. Mol. Graph. 1996, 14, 33–38.

³⁵ Legault, C. Y. CYLview, 1.0b (Université de Sherbrooke, Québec, Montreal, Canada, **2009**.

³⁶ Stone, J. E. Master's Thesis, University of Missouri, **1998**.
Scheme S1. Calculated Energy Diagram for [Co(P16)]-Catalyzed Radical 1,6-C-H Alkylation





Scheme S2. Optimized Structure Models, NCI Visual Representation and Spin Density Representation of Intermediates and Transition States











# Intermediate A

 $A_{[Co(II)(P16)]}$ Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.102584 Hartree H_corr: 1.310561 Hartree SCF: -5133.04 Hartree S: 437.725 Cal/Mol-Kelvin H: -5131.726235 Hartree G: -5131.934212 Hartree

С	-1.38634100	-2.70086500	-0.39155300
С	-2.71235900	-3.30415000	-0.36413600
Η	-2.90734000	-4.36309800	-0.52992200
С	-3.60934400	-2.30067000	-0.05657400
Η	-4.68882800	-2.36850100	0.07376700
С	-2.84531800	-1.06457400	0.04884000
С	-3.43544200	0.19789200	0.25710400
С	-2.69420900	1.38873500	0.12507000
С	-3.30160800	2.71310700	0.10422900
Η	-4.36128200	2.90586700	0.26746000
С	-2.29768300	3.61184000	-0.19450900
Η	-2.36628200	4.69235100	-0.31498600
С	-1.06029700	2.85169200	-0.30789300
С	0.27707800	4.92940500	-0.77483500
С	0.17529500	5.40975200	-2.11531200
С	0.25242000	6.80633700	-2.35606300
С	0.42256100	7.73826500	-1.30332100
С	0.51403200	7.23835000	0.02151500
С	0.44395000	5.85219800	0.30486900
С	-4.90669100	0.28320000	0.57634400
С	-5.33489600	0.32142400	1.93985600
С	-6.71541300	0.41770700	2.26451600
Η	-7.01390000	0.45223700	3.31443500
С	-7.66048900	0.46996400	1.21799100
Η	-8.72591500	0.54270800	1.46524600
С	-7.27279500	0.43201300	-0.13836900
Η	-8.00394700	0.47344300	-0.94837200
С	-5.89017300	0.34149800	-0.45819800
С	0.19870900	3.44687400	-0.50925800
Co	-0.00000600	0.00003400	-0.13395800
Ν	-1.47241900	-1.31067000	-0.15865700
Ν	-1.30604700	1.47568000	-0.11023900
С	1.38630600	2.70110300	-0.39008500
С	2.71232900	3.30435600	-0.36248500

Η	2.90729600	4.36339200	-0.52772200
С	3.60934500	2.30069500	-0.05559700
Η	4.68884600	2.36843700	0.07464100
С	2.84532200	1.06455000	0.04921700
С	3.43547500	-0.19803200	0.25668900
Ċ	2.69421900	-1.38880100	0.12411500
Ċ	3.30161300	-2.71316700	0.10252000
Н	4.36130500	-2.90600800	0.26553600
C	2.29765700	-3.61174200	-0.19658000
Н	2.36623800	-4.69218700	-0.31765400
C	1.06025700	-2.85153100	-0.30943500
C	-0.27717700	-4.92897800	-0.77746800
C	-0.17592600	-5.40859700	-2.11824800
C	-0 25307400	-6 80504900	-2 35971500
C	-0 42281700	-7 73755200	-1 30740600
C	-0 51383600	-7 23835300	0.01772300
C	-0.44366300	-5 85234600	0.30179500
C	4 90678100	-0 28353000	0.57560500
$\frac{c}{c}$	5 33523700	-0.32267500	1 93901100
C	6 71581500	-0.41916900	2 26335300
Н	7 01449300	-0 45440800	3 31319300
C	7 66069700	-0 47072400	1 21661800
Н	8 72616800	-0 54364300	1 46363000
C	7 27275200	-0.43186500	-0 13964500
Н	8.00375500	-0 47275000	-0.94981000
C	5 89007100	-0 34113200	-0 45915800
C	-0.19875400	-3.44658800	-0.51111600
N	1.47240400	1.31076600	-0.15797200
N	1.30603000	-1.47562300	-0.11109000
N	5.42822500	-0.31067200	-1.80342500
С	6.15863500	-0.33297300	-2.97804700
Н	4.40886100	-0.26996200	-1.97148500
0	7.42416400	-0.37891100	-3.08450400
C	5.26515100	-0.32101800	-4.24869300
Ō	3.83040700	-0.14057900	-3.88013200
С	5.61548000	0.84646900	-5.20598100
Н	5.36560100	-1.30812700	-4.73627800
С	3.37623400	1.20716200	-4.36089000
C	4.66389500	1.97853800	-4.72718500
Н	5.38534700	0.56108300	-6.24875900
Н	6.68354900	1.10854100	-5.13732000
H	2.79533900	1.66667000	-3.54303200
H	2.72230900	1.05025400	-5.24040800
H	4.48619100	2.73996300	-5.50600600
H	5.08263500	2.48250100	-3.83717400
N	-5.42858300	0.31196600	-1.80257200

С	-6.15922700	0.33517500	-2.97702900
Η	-4.40925000	0.27138800	-1.97085900
0	-7.42478000	0.38120600	-3.08319500
С	-5.26601600	0.32422900	-4.24786900
0	-3.83119100	0.14364500	-3.87972700
С	-5.61645900	-0.84258200	-5.20596500
Н	-5.36668600	1.31168500	-4.73470900
С	-3.37687300	-1.20351800	-4.36195800
С	-4.66450900	-1.97485700	-4.72839200
Н	-5.38672500	-0.55633400	-6.24859500
Н	-6.68445900	-1.10488800	-5.13715700
Н	-2.79547800	-1.66361200	-3.54479100
Н	-2.72339400	-1.04563200	-5.24163600
Н	-4.48694200	-2.73564500	-5.50786700
Н	-5.08283300	-2.47958500	-3.83861900
N	-4.32772800	0.26007900	2.94095400
C	-4.44870300	0.36867100	4.31462100
Н	-3.34767700	0.12448900	2.64125400
0	-5.52361200	0.53093300	4.97240000
Č	-3.08403800	0.25428600	5.05096400
0	-1.95872600	0.16011200	4.07842800
C	-2.78359000	1.49581600	5.92643600
Н	-3.09798400	-0.68256800	5.63725500
C	-1.21192700	1.47328800	4.07046500
C	-2.05271800	2.44293300	4.93308100
Н	-2.11475800	1.21338000	6.75992400
Н	-3.70976700	1.92416800	6.34249200
Н	-1.11514600	1.78713900	3.01652700
Н	-0.20658900	1.28022800	4.48549600
Н	-1.42291200	3.18779100	5.44931100
Н	-2.78924200	2.98390800	4.31142000
Ν	4.32823400	-0.26204100	2.94031800
С	4.44942800	-0.37162300	4.31388300
Н	3.34812500	-0.12634000	2.64086400
0	5.52445100	-0.53427400	4.97138800
С	3.08486400	-0.25792400	5.05051400
0	1.95939600	-0.16329100	4.07820100
С	2.78475300	-1.50010100	5.92518300
Η	3.09874200	0.67853500	5.63743600
С	1.21287200	-1.47659500	4.06940800
С	2.05394000	-2.44667800	4.93127400
Н	2.11598000	-1.21835700	6.75895200
Н	3.71106800	-1.92855600	6.34082600
Н	1.11602600	-1.78975000	3.01526800
Н	0.20753600	-1.28403800	4.48467100
Н	1.42432100	-3.19199900	5.44706400

Η	2.79048900	-2.98709500	4.30916000
Н	0.17570200	7.17098600	-3.38899600
Η	0.64134700	7.94363800	0.85398000
С	-0.01966800	4.44861800	-3.28054100
Η	0.77141100	3.67507700	-3.30414000
Η	-0.00431800	4.98646300	-4.24463900
Η	-0.98453800	3.91152600	-3.20505200
С	0.53643200	5.36660800	1.74544600
Η	-0.36468000	4.79235900	2.03446900
Η	0.63856800	6.21611100	2.44313300
Η	1.40369000	4.69556800	1.89409200
С	0.50441700	9.23342300	-1.57837400
Η	0.39868500	9.45294500	-2.65529900
Η	1.47290500	9.65124400	-1.24202400
Η	-0.29114400	9.78486100	-1.04140400
Η	-0.17669300	-7.16916000	-3.39286300
Η	-0.64086500	-7.94407700	0.84986100
С	-0.50473900	-9.23255800	-1.58326100
Η	-0.39918500	-9.45149000	-2.66032300
Η	-1.47318500	-9.65054400	-1.24698900
Η	0.29089200	-9.78430900	-1.04671800
С	-0.53574700	-5.36752100	1.74265400
Η	0.36560500	-4.79374800	2.03187100
Η	-0.63808500	-6.21737000	2.43988900
Η	-1.40274800	-4.69625200	1.89178600
С	0.01838500	-4.44678700	-3.28302600
Η	-0.77410100	-3.67468000	-3.30708400
Н	0.00476700	-4.98437000	-4.24729500
Η	0.98220500	-3.90797500	-3.20656100

### A_{1a},

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.298852 Hartree H_corr: 0.384763 Hartree SCF: -1051.816529 Hartree S: 180.814 Cal/Mol-Kelvin H: -1051.431766 Hartree G: -1051.517677 Hartree

Cartesian Coordinates:

H0.083499000.966332000.12942600H0.658468002.44208800-1.83092100C-0.691978001.21126400-0.61876100C-0.321474002.54786900-1.32566600C-0.278130003.73422800-0.37229000

Ν	0.86026800	4.27074200	0.01753600
Ν	1.89570000	4.75796700	0.35638800
Η	-1.65268500	1.32348100	-0.08226400
Η	-1.06722900	2.76410400	-2.11890300
Η	-1.18659800	4.14376600	0.07870300
С	-0.79575600	0.04438000	-1.63611300
Ν	-1.14248400	-1.25871400	-1.01574400
Η	-1.58190400	0.26890100	-2.38370700
Η	0.16529800	-0.06210700	-2.17111600
С	-2.54926300	-1.74083700	-1.06342700
С	-3.50920300	-0.95216700	-0.17134100
Η	-2.52095100	-2.79954200	-0.74991600
Η	-2.89126300	-1.69813700	-2.11615400
С	-4.54092300	-0.15938100	-0.73490400
С	-5.43311700	0.56324500	0.09545700
С	-5.29951400	0.49765700	1.50246900
С	-3.38573800	-1.01715200	1.24326900
С	-4.27312700	-0.29662400	2.07407400
Η	-2.59712800	-1.63960100	1.68410600
Η	-4.17199400	-0.35764000	3.16419100
Η	-5.98972100	1.05365200	2.14794700
Η	-6.22848700	1.17052000	-0.35289900
Η	-4.65248500	-0.11121900	-1.82659600
С	-0.20837400	-2.03645700	-0.35711400
0	1.07840900	-1.45476400	-0.43327200
0	-0.44245600	-3.12914900	0.23700800
С	2.16755000	-2.26031500	0.22025700
С	3.38379900	-1.36873200	0.33477900
Η	1.78830400	-2.60238700	1.20018400
Η	2.36165400	-3.14851900	-0.40758900
С	3.40397900	-0.29541700	1.26544000
С	4.53855900	0.53988100	1.37423400
С	5.67402100	0.30496000	0.55893600
С	4.52336500	-1.59503400	-0.47876800
С	5.66536300	-0.76579500	-0.36638600
Η	2.52682600	-0.11382600	1.89903800
Н	4.54096500	1.36941100	2.09066900
Н	6.55555700	0.95100500	0.64525800
Н	6.54124200	-0.95237000	-0.99871000
Η	4.51638100	-2.42001900	-1.20276300

Intermediate B Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.429333 Hartree H_corr: 1.698588 Hartree

SCF: -6184.884607 Hartree S: 566.694 Cal/Mol-Kelvin H: -6183.186019 Hartree G: -6183.455274 Hartree

Н	11.19246000	-0.63280200	3.72759300
Η	12.15235500	-1.28491400	1.49723300
С	10.54899300	-1.08562900	2.96394700
С	11.08963600	-1.45287400	1.70760100
Η	8.75535500	-1.02813600	4.20526100
С	9.17564200	-1.30870100	3.23231100
С	10.25840900	-2.04052200	0.72512600
С	8.34608500	-1.89817300	2.25048600
Η	10.67886000	-2.32352400	-0.24866300
С	8.88116900	-2.26672100	0.98605600
Η	7.28087100	-2.06754900	2.44772400
0	1.37036400	-7.17235600	-2.01647800
Η	3.52162400	-6.94571900	-0.56320600
С	7.99194600	-2.89552500	-0.06870900
Η	2.96966500	-4.84814300	-2.28680500
С	3.55637700	-5.85635500	-0.40034100
Η	4.57721500	-5.49144700	-0.61371700
С	1.26806600	-5.98356200	-1.57897500
Ν	1.70543100	1.27731000	1.93274300
С	2.53711900	-5.12732200	-1.30805300
Η	8.57495000	-3.45009300	-0.82168200
0	7.28839300	-1.84826500	-0.89362400
Ν	1.45108200	0.53610200	1.04819800
Ν	0.09551700	-5.29825700	-1.31940300
С	-1.25125600	-5.73086700	-1.45808100
Η	3.21785000	-0.39899800	-2.72813000
С	3.11181000	-5.42582600	1.02512400
0	2.18032100	-3.86409900	-0.59706300
Η	2.25891600	-6.04279600	1.36294300
С	6.01182500	-1.46233500	-0.42476200
Η	3.92452300	-5.52005500	1.76542800
Η	2.95060700	-1.62724100	-1.45779700
0	5.44455700	-2.01317000	0.56713300
С	2.99730400	-0.53786300	-1.64888400
С	1.03133400	-0.22180800	0.02094600
С	2.68808600	-3.95395700	0.82481300
Н	0.26385000	-4.34479400	-0.95856900
Н	3.54030000	-3.25688500	0.91223900
С	-2.27480200	-4.77676700	-1.17443700
Η	-1.82005200	-5.09599100	1.45992500

С	1.60457500	0.09852400	-1.36446500
Ν	5.47961500	-0.43412100	-1.17223600
Η	0.88342000	-1.27681600	0.28774600
С	4.13721800	0.08901100	-0.80582400
Н	0.87831400	-0.29404500	-2.09478500
Н	-2.22091800	-3.71611800	-3.61839300
С	-1.67453000	-4.02722300	1.61344700
Н	1.85850100	-3.64322700	1.48453700
Н	3.98300700	-0.12464700	0.26480800
С	-1.93832800	-3.35468800	-0.80136700
Н	1.64213200	1.19637200	-1.49676700
С	6.15535400	0.13876800	-2.36179100
Н	-1.32059700	-3.75183100	3.79683000
C	-2.03204000	-2.71375600	-3.23601200
Ċ	-1.42214200	-3.35063000	2.78912700
Н	4.15599100	1.18870500	-0.93411100
C	-1.72023900	-3.03020300	0.55121300
Н	2.35335000	-1.32048600	5.28318500
Н	7 01540500	-0 51038500	-2 59416200
C	-1.88065800	-2.38741900	-1.82392900
Н	-1 27515400	-3 37863700	9 32435400
C	-1 87980400	-1 53736500	-3 94022300
Н	1 28535600	-1 86197100	7 30337900
н	-1 92669200	-1 37913900	-5 01709500
Н	0.32702300	-2 58977800	9 39529600
C	1.44819100	-1.12280600	4.68278600
C	0.28852600	-1.78372200	6.84945000
C	-1.34909900	-1.93289700	2.46000200
C	-0.72829500	-2.44173100	9.10599300
Ň	-1.51161200	-1.73493900	1.07023900
С	6.63007200	1.58225700	-2.16867500
С	0.18606100	-1.41107500	5.48361700
Н	1.50043900	-1.74749000	3.77049800
С	-0.85531500	-2.05430000	7.63914900
Ν	-1.64877300	-1.00515100	-1.65068900
Н	5.85792800	2.26557300	-4.09066300
Н	7.52360500	1.22075000	-0.21391500
Н	1.48511900	-0.06751300	4.35132300
Н	0.95201700	-0.20339200	-4.68491100
С	-1.65397000	-0.47942400	-2.96476900
С	-1.10596800	-1.31062400	4.88534900
С	6.40856900	2.54845500	-3.18318900
С	7.34594900	1.96143700	-1.00260800
С	-2.12952600	-1.94565700	7.02476800
С	-1.23954000	-0.92201200	3.43299800
С	-2.27536200	-1.57855300	5.66443500

Η	-1.15016100	-1.65787200	9.76489600
Η	-4.44474600	-1.70744200	5.80060400
Η	-3.03114100	-2.15082100	7.61780600
С	0.95036600	0.85950100	-4.99540300
Co	-1.40988500	-0.02036700	0.06304600
Η	1.19578000	1.45710700	-4.09759600
С	-3.66646000	-1.46902800	5.05461700
С	-0.39757200	1.26153600	-5.57828200
С	-1.51814900	0.87615900	-3.31699500
С	-0.48569900	1.63493500	-6.94485200
Η	0.42730500	1.63199000	-7.55517200
Η	-2.40951900	1.67505800	-9.58276400
С	-1.57712300	1.26546900	-4.77522900
Η	-3.79068200	-2.15738700	4.19727900
С	6.90166800	3.86906800	-3.04409500
С	7.83298200	3.28064400	-0.85764200
Η	-4.69493400	-0.66409000	-1.62670000
Η	-4.30957800	0.63610100	-4.10269600
С	-1.71495000	2.00717600	-7.54063400
С	-1.33259300	0.43957800	3.09251200
Η	1.75810900	0.99811300	-5.73555500
Η	-0.79422600	2.42818200	-9.47564000
С	-2.83075400	1.63563000	-5.35856500
С	-1.79504900	2.39667300	-9.01013500
С	-2.87681000	1.99955700	-6.72676300
С	-4.11102100	1.63578900	-4.53426200
Ν	-1.44145500	0.96773700	1.78588800
С	7.61508700	4.23913100	-1.87904100
Η	6.73329800	4.60279100	-3.84157600
Η	8.38846100	3.55988700	0.04543900
Η	-4.04909500	2.34279400	-3.68548600
Η	-3.85782100	-0.44733100	4.67389500
С	-1.44046300	1.88541400	-2.33901400
Η	-3.84260400	2.28250400	-7.16699700
Ν	-1.46421100	1.69401800	-0.93911500
Η	-4.42807500	0.85077200	2.11214500
Η	-6.60540200	1.00873100	3.89307600
С	-1.37097500	1.50701200	4.08321300
Η	-6.05788600	0.26720200	1.61740800
Η	-1.30030200	1.34802500	5.15855500
С	-5.48669800	1.13669000	1.99114900
Н	1.80140400	4.08337900	-2.10078400
H	-2.26034800	3.39259900	-9.13805600
H	3.48174200	3.41948700	-1.97195600
Н	8.00045400	5.25960900	-1.76886400
С	-6.11828000	1.76519400	3.25395400

Η	-4.97938400	1.92026500	-5.15393800
С	-1.53691100	2.36160000	1.97650300
С	2.75847700	4.14503900	-1.55310400
С	-1.44379000	3.30640100	-2.66352700
С	-1.52712600	2.99809900	-0.40575000
Н	-5.35285600	2.28805400	3.85598000
С	-1.51283500	2.69330200	3.39572300
Н	-1.42659600	3.70353300	-3.67789700
0	2.49462700	3.73847100	-0.12765400
Č	-1.60116200	3.33372000	0.95929700
Ō	-5.53869200	2.20859100	0.93109200
Н	4.07515000	5.74865300	-2.31126200
C	-1 51754100	3 99438200	-1 46994400
Н	-8.06061000	2 27671000	2 35136700
C	3 35244700	5 57204700	-1 49667800
C	-7 13259300	2 78741100	2 66767800
н	-3 98275800	3 4 5 4 8 5 9 0 0	1 07811100
н	0.66151500	4 20018200	0.70626000
н	-157417400	3 70584900	3 79226400
Н	2 55520300	6 33465300	-1 56453900
C	-6 38339200	3 33319100	1 42579500
н	5.00338300	5 11359400	-0 11499300
н	-7 39380000	3 59963700	3 36514700
C	3 02841300	4 80841500	0 76883300
C	4 02073500	5 61774600	-0.09403600
N	-4 17357900	4 39061500	1 47424000
C	-1 77304000	4 77567200	1 36473200
Н	-1.57205100	5.07006900	-1.30483800
N	0.62887300	5.12303000	1.16680200
C	-5.51250900	4.57559500	1.76613000
Н	3.51095300	4.30068800	1.62326600
Н	-7.07195400	3.62063800	0.61041200
С	-3.07895300	5.27838600	1.65825600
С	1.87656300	5.70241400	1.31312500
С	-0.64882000	5.64291500	1.51581000
Н	4.14096200	6.64046600	0.29798400
0	-6.04679300	5.62185900	2.25089300
0	2.12536000	6.83162800	1.84045100
С	-3.25650800	6.61162700	2.11788100
С	-0.81953500	6.97399900	1.98586600
Н	-4.26435300	6.96711400	2.34284400
С	-2.12086700	7.43350800	2.28156500
Н	0.05888700	7.60888100	2.11791000
Н	-2.25338100	8.45906600	2.64533500
Η	5.45960100	0.09186600	-3.22325700
Н	7.21791900	-3.54158600	0.38012900

С	-3.64271600	-5.17838800	-1.28034000
С	-3.98106600	-6.50027200	-1.67906600
Η	-5.03408700	-6.77792400	-1.76121800
С	-2.94395500	-7.41397600	-1.96488600
Η	-3.20252400	-8.43288900	-2.27570600
С	-1.58334700	-7.05402900	-1.85851400
Η	-0.78044700	-7.76015500	-2.08044900
Ν	-4.63230500	-4.20488200	-0.97461100
С	-6.00810300	-4.28710200	-1.08786000
Η	-4.32005500	-3.28543700	-0.61976300
0	-6.67893800	-5.29148700	-1.48309600
С	-6.73094600	-2.97951700	-0.65758400
0	-5.74728800	-1.92087400	-0.29137700
С	-7.59029800	-2.38340300	-1.80110000
Η	-7.32851400	-3.20920100	0.24334500
С	-5.74851900	-0.86303400	-1.36600400
С	-6.58520600	-1.44570700	-2.52780400
Η	-8.42805400	-1.80080400	-1.37594400
Η	-8.00047900	-3.17627600	-2.44723300
Η	-6.19585500	0.05172200	-0.93580700
Η	-7.09175900	-0.65629400	-3.10936500
Η	-5.94705400	-2.02923800	-3.21624600

## **Transition State TS1**

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -219.5313 cm-1 G_corr: 1.426108 Hartree H_corr: 1.696261 Hartree SCF: -6184.853523 Hartree S: 568.585 Cal/Mol-Kelvin H: -6183.157262 Hartree G: -6183.427415 Hartree

Η	11.21791200 -0.77947800	3.61022400
Η	12.10973200 -1.31832600	1.32199900
С	10.55375400 -1.20338500	2.84762500
С	11.05605000 -1.50697300	1.55884300
Η	8.80106200 -1.22098800	4.14733000
С	9.19180700 -1.45241700	3.14939700
С	10.19811400 -2.05732000	0.57765700
С	8.33563300 -2.00461200	2.16873400
Η	10.58873800 -2.29048500	-0.42144400
С	8.83224600 -2.30949100	0.87201200
Η	7.27872800 -2.19335800	2.39134600

0	1.33291200	-7.27578400	-1.96226400
Η	3.45376100	-7.07564500	-0.42804200
С	7.91484200	-2.90002100	-0.18119600
Η	3.02066800	-5.02443400	-2.24082200
С	3.51749100	-5.98255500	-0.30268200
Η	4.55525200	-5.65714700	-0.49735600
С	1.26167700	-6.07749300	-1.54573600
Ν	1.47054100	1.45498400	2.18478600
С	2.55003300	-5.25398500	-1.26686200
Н	8.47954900	-3.40789300	-0.97983400
0	7.16652700	-1.82760800	-0.92826300
N	1.73446600	0.61023500	1.42688600
N	0.10646500	-5.35212100	-1.31130300
С	-1.24962100	-5.75013200	-1.45160800
Н	3.10350200	-0.34319800	-2.58898700
С	3.04530200	-5.48648200	1.09249700
Ō	2.20979700	-3.95898500	-0.61152700
Н	2.16459600	-6.06542100	1.42686500
C	5.90067500	-1.48584800	-0.39365200
Н	3.83332700	-5.57796000	1.85946300
Н	2.73732800	-1.59135100	-1.36944300
0	5.38720400	-2.09070600	0.59558700
Ċ	2.83424900	-0.49888100	-1.52253200
C	0.65892300	-0.27059100	-0.07422500
C	2.67373900	-4.01055400	0.82768400
Н	0.30695500	-4.39746100	-0.96773000
Н	3.54207900	-3.33436600	0.91426500
С	-2.25277700	-4.77095300	-1.17872500
Н	-1.76502900	-5.08903400	1.45130400
С	1.45763900	0.19577600	-1.28243600
Ν	5.32248200	-0.43881500	-1.07677200
Н	0.91540600	-1.29978300	0.25291200
С	3.98525400	0.05176500	-0.64088300
Н	0.82502100	-0.01593400	-2.16695400
Н	-2.16767300	-3.71481700	-3.62351000
С	-1.62117800	-4.01997800	1.60381700
Н	1.83563000	-3.64896000	1.44904300
Н	3.84505200	-0.24677400	0.40855500
С	-1.88952900	-3.35425100	-0.80993700
Н	1.57301800	1.29878800	-1.25766800
С	5.95586500	0.20038900	-2.25344000
Н	-1.28642500	-3.73687700	3.78884100
С	-1.96493700	-2.71463200	-3.24282700
С	-1.37912900	-3.33990300	2.77876400
Н	4.00220300	1.15863200	-0.67782600
С	-1.66250800	-3.02571700	0.53923900

Η	2.25544500	-1.32302400	5.42842400
Η	6.78688000	-0.44893400	-2.57546800
С	-1.82030800	-2.38889700	-1.83146300
Η	-1.72356200	-3.22989100	9.35662400
С	-1.78013900	-1.54383600	-3.94861300
Н	1.09430500	-1.82623300	7.41309500
Н	-1.81047000	-1.38744100	-5.02620900
Н	0.00909100	-2.78826100	9.39881800
С	1.37764100	-1.05409800	4.81514600
С	0.12025300	-1.76028000	6.91025400
С	-1.31131600	-1.92236700	2.44676900
Ċ	-0.99695800	-2.43295000	9.11317400
N	-1.46105000	-1.72869300	1.05590300
С	6.48295800	1.61579900	-1.99330400
С	0.08185700	-1.37919700	5.54384300
Н	1.45278400	-1.59083200	3.85126500
С	-1.05749600	-2.05407700	7.64002600
Ν	-1.57300900	-1.00865200	-1.65920100
Н	5.83066300	2.37924200	-3.92997700
Н	7.26983000	1.16887500	-0.01147600
Н	1.44558000	0.02554700	4.58164000
Н	1.11996600	-0.19794200	-4.65100500
С	-1.54580500	-0.48773200	-2.97507100
С	-1.17993900	-1.29992800	4.88039200
С	6.35405000	2.61426500	-2.99308800
С	7.16358300	1.93390700	-0.78932700
С	-2.29988100	-1.96603900	6.96168900
С	-1.25172500	-0.91154900	3.42331400
С	-2.38267300	-1.59385200	5.59702900
Н	-1.23684700	-1.56380200	9.75739500
Н	-4.55293200	-1.76286000	5.62568300
Η	-3.22652800	-2.19305000	7.50594400
С	1.11806500	0.85888500	-4.98148500
Co	-1.27274000	-0.03092400	0.04357100
Η	1.36011700	1.47447100	-4.09488400
С	-3.74359000	-1.51028400	4.91850000
С	-0.22628400	1.24915200	-5.58031100
С	-1.37803800	0.86329300	-3.33289100
С	-0.30054400	1.61716700	-6.94926800
Η	0.61991800	1.61714600	-7.54810800
Η	-2.28333000	1.72304300	-9.59219500
С	-1.41656800	1.24732700	-4.79320200
Н	-3.81360700	-2.20160400	4.05735700
С	6.90519900	3.90473300	-2.80221000
С	7.70789700	3.22406300	-0.59229200
Н	-4.57814500	-0.64091700	-1.70111100

Η	-4.16379600	0.59842100	-4.16758200
С	-1.52417200	1.98032600	-7.56174600
С	-1.36345000	0.44568000	3.07599000
Н	1.92973800	0.98353400	-5.71966900
Н	-0.59536700	2.30206500	-9.51179300
С	-2.66537500	1.60751200	-5.39330900
С	-1.58657500	2.37450800	-9.03074100
С	-2.69642000	1.96679600	-6.76295900
Ċ	-3.95627400	1.60206500	-4.58568500
Ň	-1.42516300	0.96739200	1.76394100
С	7.58387000	4.21390100	-1.59890300
Н	6.80847800	4.66264900	-3.58903200
Н	8.23553000	3.45503300	0.34068900
Н	-3 90443800	2 30041600	-3 72910900
Н	-3 93568000	-0.49253900	4 52770200
C	-1 30296700	1 87415200	-2 35759100
н	-3 65848800	2 24143000	-7 21630500
N	-1 35476200	1 68414400	-0.95837700
н	-4 38211100	0.83435800	2 05479300
Н	-6 65525900	0.03435000	3 72322100
C	-1 47615200	1 51213900	4 06109700
н	-5.96223000	0.17812000	1 /0737300
н	-3.90223000 -1.45754500	1 35/7/700	5 13878600
C	-5.44406600	1.076/0500	1 87877700
н	1 95229800	4 11865200	-2 03984100
Н	-1 94553700	3 41476000	-9 15220000
н	3 64866100	3 52468400	-1 81467000
н	8 01455800	5 21092400	-1 44856700
C	-6 15431700	1 70898900	3 09746800
Н	-4 81555900	1 89494300	-5 21390500
C	-1 54707900	2 36058500	1 94796300
C	2 87930800	4 23120500	-1 45012700
C	-1 29660600	3 29356100	-2 68569500
C	-1 43226600	2 98957500	-0.42965800
н	-5 43362400	2 25919300	3 72958400
C	-1 60584500	2.23919300	3 36504900
н	-1 25984600	3 68791500	-3 70051300
<b>0</b>	2 56119600	3 85069800	-0.02938800
C	-1 56840300	3 33068200	0.02990000
$\hat{0}$	-5.48661100	2 12197900	0.72045400
н	4 15985600	5 87020900	-2 19318300
C	-1 39726000	3 98317700	-1 /0550800
Н	-8 06302800	2 16175200	2 09583700
C	3 41486100	5 68107300	-1 40176900
C	_7 16220700	2 60700200	2 44660300
Ч	-7.10223700	2.07799300	2.77009300 0 96906100
11	5.7575757000	5.7007/000	0.70770100

Η	0.69983600	4.25249400	0.75127200
Η	-1.70586200	3.70620900	3.75494800
Η	2.59206100	6.41095800	-1.51327000
С	-6.36748200	3.23995900	1.23259400
Н	5.02941000	5.30954400	0.04546600
Н	-7.47326700	3.51532700	3.11725600
С	3.03807500	4.94558800	0.86638300
С	4.02916300	5.77759500	0.02271000
Ν	-4.17915900	4.33652800	1.35083300
С	-1.78160400	4.77145200	1.31941600
Н	-1.45894400	5.05880100	-1.33327900
Ν	0.61934400	5.17992900	1.19738800
С	-5.53097500	4.49964800	1.59491800
Н	3.51389900	4.46520600	1.74048500
Н	-7.02305300	3.50640300	0.38356700
С	-3.10823400	5.24694300	1.56522700
С	1.84294500	5.80495900	1.36919900
С	-0.68262300	5.66582600	1.50019600
Н	4.09830700	6.81070800	0.39962600
0	-6.10059100	5.53928500	2.05249300
0	2.03761700	6.94826400	1.88857900
С	-3.33141800	6.57871500	2.00835900
С	-0.90124600	6.99584700	1.95459900
Н	-4.35413400	6.91246700	2.19608400
С	-2.22119100	7.42750700	2.20366600
Н	-0.04246900	7.65178600	2.11022400
Н	-2.38887200	8.45222000	2.55500800
Н	5.21863400	0.22979100	-3.08032000
Н	7.16795500	-3.58062500	0.26274100
С	-3.62957400	-5.14171500	-1.28942100
С	-3.99635000	-6.45788500	-1.68128800
Η	-5.05505800	-6.71157700	-1.76699900
С	-2.97944100	-7.39728500	-1.95548400
Η	-3.25992400	-8.41208400	-2.26061600
С	-1.61184000	-7.06795100	-1.84475100
Η	-0.82417400	-7.79387200	-2.05723200
Ν	-4.60053800	-4.14566700	-0.99495000
С	-5.97735600	-4.20028700	-1.11811000
Η	-4.27349200	-3.23063100	-0.64245800
0	-6.66594500	-5.19228300	-1.51344900
С	-6.67601000	-2.87624300	-0.69829400
0	-5.67400800	-1.83282000	-0.33967200
С	-7.52291000	-2.27297600	-1.84660300
Н	-7.27784400	-3.08743900	0.20430200
С	-5.63811800	-0.79582000	-1.43597800
С	-6.49595700	-1.37013100	-2.58686900

H -8.34449500 -1.66448100 -1.42632700 H -7.95428300 -3.06300500 -2.48232500 H -6.04847300 0.14227100 -1.02039100 H -6.98422800 -0.57419800 -3.17514400 H -5.87757600 -1.97840000 -3.27201100

### **Intermediate** C

 $\begin{array}{l} C_{[Co(III)(P16)]} \\ Temperature: 298.15 \ Kelvin \\ Pressure: 1.0 \ Atm \\ Imaginary \ Frequency: -13.1627 \ cm-1 \\ G_{corr:} 1.42143 \ Hartree \\ H_{corr:} 1.685488 \ Hartree \\ SCF: -6075.301459 \ Hartree \\ S: 555.757 \ Cal/Mol-Kelvin \\ H: -6073.615971 \ Hartree \\ G: -6073.880029 \ Hartree \end{array}$ 

Η	8.50365000	1.85854800	-5.74262200
Η	6.57686400	0.28756400	-5.36120100
С	8.39191700	1.34892500	-4.77807100
С	7.30616600	0.46486100	-4.56192200
Η	10.18191300	2.24750700	-3.91131600
С	9.33746500	1.56792500	-3.74644200
С	7.16526000	-0.19840900	-3.32068400
С	9.19642600	0.90253300	-2.50620100
Η	6.32263900	-0.87681500	-3.14234300
С	8.10960700	0.01657500	-2.28094400
Η	9.93187300	1.07019200	-1.70854900
0	3.52438600	-6.15223600	-0.41826400
Η	5.28016800	-5.30393500	1.24678400
С	7.96216000	-0.69867500	-0.95200600
Η	4.23710400	-3.31833600	-0.38976100
С	4.88891600	-4.30244500	1.48870600
Η	5.72044100	-3.57522300	1.48360200
С	2.97187300	-5.07330200	-0.03811400
С	3.81228000	-3.86688400	0.46996100
Η	7.46847100	-1.67865200	-1.06947300
0	7.15343500	0.10743900	0.03013000
Ν	1.61098500	-4.80905200	-0.03225100
С	0.52606000	-5.65282300	-0.39276000
Η	3.16781000	2.58935400	0.23354000
С	4.11870500	-4.25211000	2.83864600
0	2.93069400	-2.90108000	1.19579300
Η	3.51966400	-5.17215900	2.96900200

С	5.75295700	-0.13775700	0.00527900
Η	4.79113800	-4.15085900	3.70801700
Η	3.01466400	1.28694200	-0.98069100
0	5.23239400	-1.06102100	-0.68831700
С	2.81348100	1.55024800	0.07641800
С	0.66397800	0.18573900	0.02641200
С	3.20184600	-3.01858700	2.66775500
Н	1.39614100	-3.86909500	0.33955700
Н	3.70383100	-2.08565400	2.98827400
С	-0.76998000	-5.05676900	-0.48081000
Н	-0.90993100	-4.86013400	2.21576200
С	1.27697200	1.52484000	0.34104500
N	5.07488000	0.72433600	0.83723800
С	3.60116300	0.57654300	0.99019600
Н	0.81480700	2.33545000	-0.26403400
Н	-0.41841900	-4.38203200	-3.02146400
C	-0.97240300	-3.77295100	2.17789300
Н	2.22739000	-3.11099400	3.17814800
Н	3 33213900	-0 47181300	0 77132100
C	-0.93543700	-3.56632700	-0.33313000
Н	1.07216200	1.80929000	1.39679300
C	5 74076800	1 82938200	1 57675900
н	-0 90427900	-3 06468400	4 29041900
C	-0 55523100	-3 32188800	-2 81288800
C	-0.96681400	-2 86909500	3 22038500
Н	3.35643400	0.77782400	2.05219400
C	-1.05963500	-2.99665100	0.94664200
Н	2.34443000	0.30009000	5.20538900
Н	6.72594600	1.99238700	1.11094700
C	-0.85697300	-2.77501300	-1.49593300
Н	-0.64065100	-1.72546600	9.60037400
С	-0.45634700	-2.25418800	-3.68028700
Н	1.35046600	0.00680400	7.33982800
Н	-0.23359900	-2.26243800	-4.74657100
Н	0.46925600	-0.32740800	9.56186100
С	1.40678300	0.28353500	4.62297500
С	0.37272300	-0.22175700	6.89532100
Ċ	-1.09777300	-1.54025100	2.63561000
Ċ	-0.52851000	-0.68148500	9.24816600
N	-1.17223300	-1.62180400	1.23024900
С	5.90022200	1.55391100	3.07476300
Ċ	0.22599500	-0.13886900	5.48604100
H	1.54187700	-0.39286600	3.75913900
Ċ	-0.70436500	-0.58756600	7.73879000
N	-0.98681200	-1.37224800	-1.55695000
Η	4.70320200	3.26533700	3.70001600

Η	7.14575600	-0.20826800	2.78483800
Η	1.25046300	1.29841600	4.20714400
Η	2.15267200	-0.13247900	-3.82517700
С	-0.73270000	-1.05014900	-2.90960900
С	-1.04014400	-0.44072500	4.90006600
Ċ	5.30119800	2.40748000	4.03566700
C	6.67260500	0.44893000	3.52461600
C	-1 95707900	-0.87096300	7 13547400
C	-1 20052700	-0.36530300	3 40233100
C	-2 14425200	-0.80515300	5 73285000
Н	-1 28648700	-0.07604100	9 78058400
н	-4 25096100	-1 32010600	5 92893400
н	-2 80867300	-1.1/896700	7 77122200
C	1 05035000	0.83604000	1 32152200
	1.95055900	0.11564700	-4.32132200
С0 Ц	-1.21932100	-0.11304/00	-0.03943100
П	2 51160200	1.11109500	-5.51964100
C	-3.31109300	-1.11108300	5.15599400
C	0.833/9100	0.70904000	-3.34898000
C	-0.8182/300	0.22981/00	-3.4831/600
C	1.11688/00	0.8845/600	-6./2/36400
H	2.14552100	1.12101400	-/.03125100
H	0.73593400	-0.05239600	-9.64453100
C	-0.50344400	0.40814800	-4.94/05100
Н	-3.47463200	-1.98906700	4.46319500
С	5.46980600	2.16791300	5.42197700
С	6.84102700	0.20580900	4.90660500
Η	-4.07205500	-2.02886500	-2.16506900
Η	-3.62105900	-0.10665500	-6.41408200
С	0.11224200	0.75970800	-7.71948400
С	-1.53458300	0.86385200	2.80664500
Η	2.88853000	1.17682000	-4.79336600
Η	1.28864600	1.61597000	-9.34913900
С	-1.53307700	0.27783300	-5.92927200
С	0.44510400	0.91865300	-9.19648800
С	-1.20726000	0.45871400	-7.29736100
С	-2.96468400	-0.05233500	-5.52803000
Ν	-1.69620400	1.09744400	1.42357700
С	6.24026500	1.06543600	5.86087900
Η	5.00360100	2.83850100	6.15364500
Η	7.44500200	-0.64579500	5.24245100
Η	-3.01922600	-1.02462400	-5.00157000
Η	-3.88612300	-0.26177900	4.53315900
С	-1.27765200	1.32923300	-2.73625400
Н	-2.00243400	0.36144900	-8.04865800
Ν	-1.56563800	1.32739800	-1.35417900
Н	-4.33311600	-0.21236200	1.32510400

Η	-6.63598900	-0.67505600	2.86699300
С	-1.91198100	2.04376300	3.57337200
Η	-5.58975000	-1.36954700	0.74852100
Η	-1.86831800	2.10887200	4.66000800
С	-5.39988000	-0.32885900	1.06904000
Н	1.02056600	4.32533700	-2.46892100
Н	-0.42084100	1.29762600	-9.76875800
Н	2.74804400	4.59615600	-2.01026200
Н	6.37562200	0.87903200	6.93288700
C	-6.36770400	0.14297600	2.17644600
Н	-3.38063900	0.70704200	-4.83899600
С	-2.20136100	2.41238500	1.34237800
Ċ	1.71660200	4.98707300	-1.92509400
Ċ	-1.64492500	2.60453800	-3.33553200
Ċ	-2.12078000	2.60117100	-1.10505500
Н	-5.91987900	0.96415300	2.76592800
С	-2.35648500	2.98877700	2.67106200
Н	-1 52189700	2 84335700	-4 39136600
0	1.33688000	4.94172100	-0.46915700
C	-2.48295300	3.11463200	0.15449800
0	-5.66913400	0.57173000	-0.11206800
Н	2.33257100	6.72105000	-3.15090400
C	-2.20340700	3.37594300	-2.33565900
Н	-8.23054700	-0.17784400	1.04977000
C	1.62089800	6.47164900	-2.34537000
C	-7.59145600	0.66458200	1.37201400
Н	-4.72033100	2.31593900	0.00941700
Н	-0.65695100	4.78294300	-0.01866800
Н	-2.73924900	3.98977600	2.86648900
Η	0.60145300	6.71501600	-2.69618600
С	-6.93054900	1.32638900	0.13517000
Н	3.01694600	7.26327700	-0.83650500
Н	-8.21113700	1.38695900	1.92764800
С	1.23736600	6.34060200	0.03878300
С	1.92813400	7.22887400	-1.02319400
Ν	-5.29481600	3.14810600	0.22557200
С	-3.21956700	4.42724600	0.25288100
Н	-2.62691300	4.37774200	-2.40237900
Ν	-1.12425300	5.67971700	0.19015500
С	-6.63644900	2.83625300	0.35636600
Н	1.75008000	6.37435700	1.01768200
Н	-7.55955200	1.24298400	-0.77021500
С	-4.64911100	4.40941500	0.33205200
С	-0.24585100	6.74718300	0.26167200
С	-2.54300400	5.68338100	0.31589900
Η	1.52772600	8.25505800	-1.00105600

0	-7.58329700	3.64432300	0.61148900
0	-0.54578600	7.95909300	0.49985200
С	-5.38255000	5.61454100	0.50380700
С	-3.27383800	6.89125900	0.48967600
Η	-6.47137600	5.56827200	0.57433600
С	-4.68017400	6.83497500	0.58623300
Η	-2.73123200	7.83643600	0.55233100
Η	-5.24208900	7.76592400	0.72441400
Η	5.13890200	2.74759600	1.43261500
Η	8.92989700	-0.80954600	-0.43595300
С	-1.89523700	-5.88112400	-0.78678500
С	-1.73021300	-7.27239800	-1.02639300
Η	-2.60576500	-7.87975600	-1.26547700
С	-0.43256900	-7.82518000	-0.96240700
Η	-0.29989100	-8.89547000	-1.15891600
С	0.69830700	-7.04098600	-0.64776700
Η	1.70245100	-7.46831700	-0.60671200
Ν	-3.16536800	-5.24507400	-0.84670600
С	-4.39493100	-5.77040700	-1.19826300
Η	-3.21895000	-4.24046600	-0.60838800
Ο	-4.64090400	-6.97242500	-1.52914500
С	-5.53910300	-4.71876000	-1.14255200
Ο	-5.00363800	-3.36584400	-0.81466600
С	-6.26238900	-4.56938600	-2.50405800
Η	-6.22659700	-5.01300600	-0.32878500
С	-5.05977300	-2.50787300	-2.05481300
С	-5.42815100	-3.46434400	-3.21158500
Η	-7.30159400	-4.22931800	-2.34187600
Η	-6.28202000	-5.52539000	-3.05180100
Η	-5.82319800	-1.72762600	-1.88364700
Н	-5.99696600	-2.95181800	-4.00641200
Н	-4.51934900	-3.90251500	-3.66292700
Н	1.20874200	-0.50017300	-0.64883500

# **C**[N2]

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: -0.013594 Hartree H_corr: 0.008229 Hartree SCF: -109.5607783 Hartree S: 45.93 Cal/Mol-Kelvin H: -109.5525493 Hartree G: -109.5743723 Hartree

Cartesian Coordinates: N 0.0000000 0.0000000 0.57307000

# N 0.0000000 0.0000000 -0.57307000

# **Transition State TS2**

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -1200.1007 cm-1 G_corr: 1.428625 Hartree H_corr: 1.680565 Hartree SCF: -6075.288033 Hartree S: 530.253 Cal/Mol-Kelvin H: -6073.607468 Hartree G: -6073.859408 Hartree

Η	7.31121000	-6.31563900	-0.66179100
Η	5.60401100	-6.74221100	1.13569300
С	7.05257300	-5.52294000	0.05046000
С	6.09202300	-5.76300100	1.06513300
Η	8.43575600	-4.07051500	-0.81009600
С	7.68643200	-4.25955200	-0.03233200
С	5.76665600	-4.74792300	1.99378400
С	7.35893800	-3.24366000	0.89717100
Η	5.02345600	-4.92045900	2.78077100
С	6.39755000	-3.47584900	1.91601000
Η	7.86516800	-2.27059600	0.84064900
0	-4.66177100	-6.12691400	2.56459200
Η	-4.66062700	-5.65778400	5.06463300
С	6.08565600	-2.39624400	2.93508200
Η	-2.24212500	-5.20133700	3.79092900
С	-3.96356700	-4.80761700	5.13564400
Η	-3.25892400	-4.98770000	5.96771200
С	-4.03119200	-5.02389900	2.57489200
С	-3.18319000	-4.62122900	3.81300500
Η	6.00213200	-2.81937900	3.95139300
0	4.80025500	-1.65910000	2.64939100
Ν	-3.99089800	-4.08784000	1.55363500
С	-4.70601000	-4.09088500	0.32059700
Η	0.79260900	-3.64368500	2.91806600
С	-4.67459700	-3.43547100	5.30388300
0	-2.81836400	-3.17609500	3.75179600
Η	-5.58741100	-3.40037200	4.68150400
С	3.66879900	-2.17351200	3.32362900
Η	-4.96116000	-3.23294900	6.35007200
Н	-0.60632000	-2.65520100	3.41653100
0	3.68675300	-3.26052700	3.97485700
С	0.40373600	-2.60722000	2.96593700

С	0.14583200	-0.54824000	1.36611400
С	-3.61789500	-2.43368900	4.78600900
Η	-3.42695800	-3.25593300	1.78830100
Η	-2.91895000	-2.12036200	5.58507700
С	-4.40810800	-3.07324900	-0.63975800
Η	-5.46607200	-1.32343900	1.18139100
С	0.29196300	-2.06893100	1.50159400
Ν	2.57101100	-1.34208500	3.21981900
С	1.31198100	-1.77192600	3.89388600
Н	-0.58318400	-2.56811500	1.04618800
Н	-2.50012900	-4.47171900	-1.89216900
С	-4.56259000	-0.71392000	1.18517500
Н	-4.04289100	-1.54342200	4.29272900
Н	1.57621600	-2.35494300	4.79260200
C	-3 25823300	-2 10908200	-0.46585300
Н	1.16939400	-2.40032300	0.91249200
C	2 53094000	-0.06964700	2 51227300
Н	-4 90431900	0.90031900	2 68569700
C	-1 78106100	-3 65801600	-1 80590900
C	-4 27850400	0 40448000	1 94424100
н	0 79668400	-0.84952700	4 21674600
C	-3 39742800	-0.96808600	0.34819400
н	-1 34019400	1 66886800	5 86509600
н	3 30242900	0.00934000	1 72726900
C	-2 04607100	-2 39352700	-1 13184000
Н	-6.00888600	4.53324400	6.62695000
C	-0.47917200	-3.60835100	-2.26124500
Н	-2.94074300	3.26577600	6.51138500
Н	0.07984200	-4.36843300	-2.80595000
Н	-4.42921100	5.07451400	7.25391500
С	-1.48244900	1.75074900	4.77336300
C	-3.28836100	3.39995200	5.47835200
C	-2.96565200	0.88128800	1.53121300
C	-5.05929900	5.03164900	6.34776300
N	-2.42184300	0.03455900	0.54111400
C	2.49269900	1.14543500	3.40309200
Ċ	-2.64121700	2.68017100	4.44092500
Н	-1.65189300	0.73443800	4.36891200
C	-4.36158900	4.28712400	5.21794300
Ň	-0.91613600	-1.54851200	-1.19872800
Н	2.06480400	2.52084900	1.76554500
Н	2.93855100	0.07837000	5.25538700
Н	-0.53286800	2.11531000	4.33708800
Н	2.51004400	-3.85190900	-0.44656900
C	0.04726200	-2.29554900	-1.91319400
Ċ	-3.08459300	2.85285300	3.09434400
			•

С	2.26365200	2.43161600	2.83862200
С	2.73350000	1.05651900	4.80526400
С	-4.77921700	4.44977900	3.87236600
С	-2.40403800	2.09198100	1.98321500
С	-4.16135500	3.74890700	2.80674200
Н	-5.30901000	6.06826500	6.05556700
Н	-5.46739600	4.70267000	1.34915800
Н	-5.60287300	5.14004000	3.64524800
C	3 37415300	-3 39172900	-0.96477200
$\tilde{C}_{0}$	-0.64926100	0 19374300	-0 31044600
Н	3 43239600	-2 34387900	-0.61708400
C	-4 64192900	3 97006900	1 37865400
C	3 21233000	-3 47855800	-2 /7/99100
C	1 20188200	1 81033300	2 36466300
C	1.29188200	-1.81933300	2.30400300
с ц	4.00982300	4.31978200	-3.22/92200
	4.83037900	-4.88318000	-2.09973000
П	4.49/03000	-0.43002000	-3.30203300
	2.19883300	-2./3293000	-3.13191000
H	-5.00198600	3.02965300	0.91964100
C	2.28388200	3.59102900	3.64399900
C	2./5198600	2.21/36600	5.61186200
H	-1.8910/900	-0.268/2/00	-3.6855/200
Н	1.03898700	-2.26494800	-6.40411500
С	3.94501500	-4.45062500	-4.63388000
С	-1.29789900	2.66788500	1.32944400
Η	4.28393000	-3.91616200	-0.62735400
Η	5.88829500	-5.37177900	-5.01734100
С	2.05676100	-2.84614100	-4.56935500
С	4.85769500	-5.38398300	-5.41723900
С	2.93299400	-3.70178700	-5.28480000
С	0.98768100	-2.06322300	-5.31982800
Ν	-0.55679600	2.07378700	0.28517300
С	2.52872700	3.49214800	5.03661500
Η	2.10813000	4.57016900	3.18411300
Η	2.95189200	2.12767200	6.68667200
Η	-0.02713500	-2.33260000	-4.96999500
Η	-3.82537700	4.34835600	0.73449100
$\mathbf{C}$	1.64689100	-0.46477200	-2.21199000
Η	2.82145800	-3.78268100	-6.37436600
Ν	0.93532700	0.49802500	-1.46737200
Η	-2.68520400	2.89646100	-1.69076600
Η	-4.66414500	4.83902300	-2.17843800
С	-0.88217600	4.04938900	1.53046400
Н	-3.90423100	2.63374100	-2.98949600
Н	-1.30085800	4.71286400	2.28623400
С	-3.12294900	3.31671300	-2.61227900

Η	5.10351600	0.10758400	-1.21513800
Η	4.90153800	-5.10614800	-6.48551300
Η	5.75868500	-0.59536200	0.31470500
Н	2.54827000	4.39347300	5.66020500
С	-3.60142400	4.77885800	-2.47006300
Н	1.10028300	-0.97268200	-5.16804700
С	0.29528000	3.09895600	-0.17902000
Ċ	5.62830600	0.33247000	-0.27150800
Ċ	2.73611800	0.15934300	-2.95264700
Ċ	1.57536600	1.72196900	-1.76257600
Н	-3.00195600	5.31680300	-1.71264300
C	0.07477400	4.32663200	0.57541900
Н	3 41836100	-0 37965700	-3 60930700
0	4 73378800	1 23849400	0 52478200
C	1 26661300	2 97426400	-1 19623900
$\hat{0}$	-2 02366400	3 34904300	-3 64726300
н	7 81875100	0.43578400	-0 56363600
C	2 66662700	1 51603600	-2 70680800
н	-4 14897800	5 09177800	-4 57935500
C	6 95094800	1 10959600	-0.45690900
C	-3 34017200	5 37562900	-3 88146000
н	-0.26731600	3 82170600	-2 86826200
н	3 57363000	2 62323800	-0.30885600
н	0.60415400	5 26160900	0.39533100
н	6 90408300	1 75500300	-1 35325000
C	-2 01766200	4 68665600	-4 30171800
н	7 40404700	1 39185100	1 68179600
н	-3 23668700	6 47285500	-3 88509300
C	5 54823100	2 36746600	1 06534900
C	7 02655600	1 98276900	0.82760200
N	0.09207600	4 77120300	-3 06481600
C	1 99523400	4 22032400	-1 64192700
Н	3 28529400	2 31478000	-3 11551800
N	3 96123100	3 58017600	-0 34317600
C	-0.76201900	5 49612900	-3 87720200
Н	5 29811900	2 47033200	2 13692500
Н	-1 95593400	4 52935100	-5 39437500
C	1 35154200	5 14534900	-2 52509100
C	5 15606900	3 68943700	0.34891100
C	3 29832900	4 54514700	-1 14928500
Н	7 65220200	2 88244000	0 71075000
$\hat{0}$	-0 59395500	6 68619500	-4 28987700
0	5 88408300	4 72568400	0 44457900
C	1 96390100	6 38461800	-2 85715900
C	3 90966700	5 78791400	_1 <u>47100900</u>
Ч	1 43887/00	7 07385600	-3 52180700
11	1.7300/700	1.01505000	5.52100/00

С	3.22801600	6.69137300	-2.31243400
Н	4.89056600	6.02025900	-1.05145200
Η	3.69717200	7.65118400	-2.55766800
Η	1.42422300	-0.09185800	1.75626500
Η	6.83823600	-1.59071200	2.91367500
С	-5.19158500	-3.01849100	-1.83849000
С	-6.19897200	-3.98637300	-2.09954500
Η	-6.76648000	-3.92311100	-3.03033700
С	-6.42953600	-5.00473500	-1.15260900
Η	-7.19657800	-5.76095800	-1.35543900
С	-5.70637200	-5.06709700	0.05592500
Η	-5.88662500	-5.85384800	0.79062000
Ν	-4.93497900	-1.95567700	-2.74509500
С	-5.49478300	-1.71095200	-3.98716500
Η	-4.25982300	-1.22201800	-2.47057700
0	-6.35483000	-2.42615700	-4.58998500
С	-4.98081400	-0.39263400	-4.62738900
0	-3.94279000	0.23739300	-3.76589200
С	-4.31159700	-0.62239200	-6.00562100
Η	-5.83864800	0.30218600	-4.68829400
С	-2.59937600	0.10546400	-4.4441100
С	-2.82451300	-0.86583200	-5.62478200
Η	-4.41048100	0.28693000	-6.62622500
Η	-4.77552000	-1.46838400	-6.53827400
Η	-2.29556600	1.11866100	-4.76100500
Н	-2.13685800	-0.66243300	-6.46363200
Н	-2.68098500	-1.91419900	-5.30421800
Η	-0.45155100	-0.09585300	2.19102700

## Intermediate D

 Temperature: 298.15 Kelvin

 Pressure: 1.0 Atm

 G_corr: 1.427402 Hartree

 H_corr: 1.685916 Hartree

 SCF: -6075.326268 Hartree

 S: 544.09 Cal/Mol-Kelvin

 H: -6073.640352 Hartree

 G: -6073.898866 Hartree

Cartesian Coordinates:

H -8.87346500 -4.45222000 0.20428300 H -6.90892500 -4.89864400 -1.30031700 C -8.54322700 -3.68851600 -0.50979900 C -7.43523100 -3.93859800 -1.35818000 H -10.09096600 -2.25655900 0.05326400

С	-9.22799700	-2.45204400	-0.59404500
С	-7.01517400	-2.96046600	-2.28859400
С	-8.80316600	-1.47079700	-1.52206500
Н	-6.16414800	-3.14686100	-2.95467700
С	-7.69548300	-1.71477700	-2.37591800
Н	-9.34103100	-0.51608900	-1.59160000
0	4.01155100	-2.67052200	-6.43646300
Н	2 77226000	-0.93182100	-7 90433900
C	-7 26580100	-0 67413000	-3 39144900
н	1 20759800	-2 36324100	-6 12685800
C	1 98666700	-0 58967600	-7 21172500
Н	1.01816200	-0 57728500	-7 74387700
C	3 28497000	-2 08646600	-5 57296700
C	1 89733300	-1 51323200	-5 97428700
н	-7 03281300	-1 135///200	-1 36732600
$\hat{0}$	-6.04190800	0.09017700	-2 95140200
N	3 58181600	-1.90105900	-1 23/19800
$\hat{\mathbf{C}}$	<i>A</i> 74678600	-2 28678100	-3.51/2/200
н	-2 53103800	-1 88808000	-2 00508700
C	2 28374000	0.79671200	-6 57454200
$\hat{\mathbf{O}}$	1 33256700	-0.67319900	-4 87700600
ч	3 3 5 9 7 3 8 0 0	0.88536100	6 338/6800
C	<i>A</i> 82106100	0.36656100	3 /0528000
н	2 00/08800	1 63/88700	-7 23586300
н	-0.90/11300	-1 38025700	-3 /1076000
<b>0</b>	-4 70806100	-1.33023700	-4 16634300
$\frac{0}{C}$	-1 81419500	-1 04611100	-2 88115200
$\frac{C}{C}$	-0 17585500	0.14928100	-1 31179500
$\frac{C}{C}$	1 43898000	0.77293900	-5 27978200
Н	2 84066500	-1 39383600	-3 72430100
Н	0.41090100	1.14594900	-5.44669700
C	4.77647900	-2.01771700	-2.11117100
Н	5.26146600	0.59960100	-2.56138100
C	-1.43143700	-0.72195700	-1.40121800
Ň	-3.78603300	0.52084400	-3.22730800
C	-2.41850500	0.12020400	-3.70308400
Н	-1.25132000	-1.69224100	-0.90600200
Н	3.40860000	-4.25067300	-1.58385600
C	4.38055200	0.91012500	-2.00038600
Н	1.90093600	1.31840600	-4.43875500
Н	-2.49830600	-0.18761500	-4.76176600
C	3.59167900	-1.40691200	-1.40252500
H	-2.27898500	-0.24552500	-0.87161800
C	-3.97647800	1.69299100	-2.43646200
H	4.27969400	3.13099300	-2.16874200
С	2.63807200	-3.71017400	-1.03549600
			•

С	3.88485400	2.18321600	-1.80452700
Η	-1.77995800	1.01687000	-3.63778200
С	3.47885200	-0.00619600	-1.31489500
Н	0.30463500	4.86135700	-3.76203300
Н	-4.46207700	1.55333400	-1.46289400
С	2.61646700	-2.26424400	-0.85818400
Н	4.52062300	8.68616200	-2.42832300
C	1.52140300	-4.21059000	-0.39970200
Н	1 54592300	6 83970200	-3 40473800
Н	1 19574100	-5 24637300	-0 31242900
Н	2 97720800	8 79471300	-3 32388600
C	0 59413900	4 34736600	-2 82956600
$\frac{C}{C}$	2 05427200	6 41118800	-2 53079300
$\frac{C}{C}$	2 70322500	2 05760200	-0.96196400
C	3 42441400	8 56632100	-2 34034600
N	2 45882800	0.70330900	-0 64437400
$\hat{\mathbf{C}}$	-3 58664900	2 99796800	-2 88143600
C	1 68516700	5 11189300	-2.00145000
н	0.01857200	3 32078/00	-3.08328100
C	3 05391200	7 16638300	-1 87062000
N	1 /0/00800	-1 86827500	-0.09577600
H	-3 80730000	3 89629500	-0.00000000000000000000000000000000000
н	3 22351300	2 48240700	4 98587700
н	-3.22331300	<i>2.</i> 48240700	2 21162100
и П	1 58824000	4.23041300	-2.21102100
C	-1.38824900	-4.13272800	-1.44/24000
C	2 24422000	4 54670200	0.19838900
C	2.54455000	4.00802800	1 0/038500
C	3 21644400	3 20025000	-1.94938500
C	3 69625300	6 58384000	-0.74760400
C	1 96909600	3 16272300	-0.49323600
C	3 3 5 9 6 4 8 0 0	5 28907500	-0.49323000
н	3 06950800	9 33433500	-1 62527300
н	<i>A</i> 82549700	5 /20/9/00	1 32285300
н	4 47786000	7 14941200	-0 22245600
C	-2 38629800	-4 41019600	-0.222+3000
$C_{0}$	0.93068900	-0.03385600	0 36402300
со н	-2 85513800	-3.45771600	-0.41348000
C	4 07870300	<i>A</i> 70918600	0.02870300
C	-1 83280000	-5 1770/700	0.72877500
C	-0.29118000	-3 22564800	1 0/266900
C	-2 31882400	-6.48007800	0.74836200
Ч	-3 00005000	-6 90860200	0.00553500
н	-1 65437300	-0.70009200	1 6997//00
$\Gamma$	-1.03+32300 -0.83068500	-/ 610/2800	1 31107100
с µ	4 60204700	3 76705200	0.67707700
11	<b>¬.</b> 00∠04700	5.10105200	0.07707700

С	-3.32332100	5.40787200	-2.35955700
С	-2.91982500	4.60663300	-4.64364000
Η	3.31525400	-2.26450500	2.41977300
Η	1.03196100	-5.52093900	4.11351800
С	-1.83526300	-7.24126700	1.84063600
С	0.95627700	3.01010800	0.46997200
Η	-3.14703800	-5.00461500	-1.26056600
Η	-3.33476000	-8.81669800	1.63891100
С	-0.32542000	-5.36285600	2.41904200
С	-2.34889500	-8.64907200	2.10807900
С	-0.83842600	-6.66074400	2.66510300
С	0.75571200	-4.79500700	3.32890700
Ν	0.50446100	1.78248600	1.00992500
С	-2.96837800	5.67678200	-3.71016500
Η	-3.36485400	6.22791800	-1.63283000
Η	-2.66612200	4.81115900	-5.69121200
Η	1.66948700	-4.54477300	2.75644200
Η	3.36770600	4.47345300	1.74356900
С	-0.82084200	-2.12839100	1.74713100
Η	-0.44805900	-7.23120800	3.51865700
Ν	-0.40269600	-0.78353500	1.62698800
Η	2.87893700	1.69816900	2.71183800
Η	4.80161200	3.16092600	4.16042600
С	0.33428800	4.13408400	1.15591300
Η	4.36759100	0.88468900	3.32075400
Η	0.51618400	5.17942200	0.90923800
С	3.48127300	1.47740100	3.60946700
Η	-4.25077200	-1.15841300	0.95394000
Η	-2.44734000	-8.84515100	3.19157100
Η	-5.42310200	-0.80594600	-0.38449200
Η	-2.74348000	6.70097000	-4.02906500
С	3.83093000	2.72440900	4.45231500
Η	0.42399700	-3.86344500	3.82537700
С	-0.36744700	2.15913300	2.05486300
С	-5.12758700	-0.56130400	0.65171400
С	-1.81107100	-2.26938300	2.80652600
С	-1.12109300	-0.10306200	2.63658700
Η	3.05331200	3.50308100	4.34533500
С	-0.45805200	3.60948500	2.15594900
Η	-2.28174300	-3.21187100	3.08407900
Ο	-4.70325800	0.88590200	0.67037200
С	-1.08502300	1.28035400	2.89048700
0	2.63251900	0.60555700	4.50596200
Η	-7.00683700	-1.47635200	1.34758000
С	-1.97059100	-1.02455200	3.37950200
Η	4.79447400	1.65838900	6.12017000

С	-6.32081200	-0.65868000	1.62512300
С	3.84156400	2.17639500	5.90675400
Η	0.71448600	1.14137900	4.46685600
Η	-3.47094000	1.44514700	2.16429500
Н	-1.06513600	4.13637100	2.89115000
Н	-5.97237900	-0.82382800	2.66133600
С	2.67906400	1.15235000	5.89037100
Н	-7.60352100	0.78340700	0.57752200
Н	3.67902500	2.95195800	6.67268100
С	-5.76523600	1.68399700	1.34730300
С	-6.98498200	0.73943800	1.49172700
N	0.35512600	1.64711900	5.29382300
С	-1.77260700	1.84810500	4.11014100
Н	-2.60091500	-0.73574400	4.22010700
N	-3.92213600	1.94468300	2.94832300
C	1.31866500	1.79527100	6.27629400
Н	-5.98932300	2.56030300	0.71087800
Н	2.85354700	0.30776700	6.58221900
C	-1 00357700	2 06441800	5 29979400
C	-5.26365500	2.20338000	2.71971000
C	-3.15394900	2.21451800	4.11551100
Н	-7 60365800	1 02150500	2 35872800
0	1 17604300	2 36425000	7 40363900
õ	-6 06437400	2 80534800	3 50204500
C	-1 58828100	2 66489300	6 44767000
C	-3.74064600	2.81860600	5.26238700
Н	-0.97125800	2.82614000	7.33403100
С	-2.94679700	3.03922800	6.40632100
Н	-4.79416700	3.10255000	5.23073700
Н	-3.39942500	3.50806100	7.28767200
Н	-0.37309300	1.23497500	-1.32397100
Н	-8.02403800	0.11734300	-3.50753500
С	5.94479000	-2.37490700	-1.36755600
С	7.05035700	-3.00030000	-2.00596200
Н	7.92473400	-3.27007200	-1.40999700
С	6.98145100	-3.26236600	-3.39019700
Н	7.83074300	-3.74823200	-3.88425900
С	5.84956200	-2.91387900	-4.15674200
Н	5.79432700	-3.11870100	-5.22773300
N	5.95069100	-2.08248200	0.02292700
С	6.89918100	-2.39865900	0.97952400
Н	5.15086000	-1.56080400	0.41915200
0	7.99023700	-3.02219400	0.79188300
С	6.51856300	-1.88907800	2.39721700
0	5.17265600	-1.25097600	2.38616300
С	6.43470500	-3.03798900	3.43279100

Η	7.25499500	-1.11625000	2.68361600
С	4.19304400	-2.16637200	3.08138600
С	4.95212900	-3.49214500	3.31691900
Η	6.64424600	-2.64586900	4.44478400
Η	7.15932300	-3.83597300	3.20349300
Η	3.88551300	-1.66581100	4.01643300
Η	4.60086400	-4.01342100	4.22401700
Η	4.82893800	-4.17406900	2.45573600
Η	0.56381000	-0.12657400	-2.08356400

# **Transition State TS3**

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -452.7493 cm-1 G_corr: 1.42995 Hartree H_corr: 1.685717 Hartree SCF: -6075.317018 Hartree S: 538.307 Cal/Mol-Kelvin H: -6073.631301 Hartree G: -6073.887068 Hartree

7.69208300	-5.20412500	-1.81904700
5.83508600	-6.25970100	-0.48937100
7.36589900	-4.75258100	-0.87458600
6.31934800	-5.34600900	-0.12501000
8.81431800	-3.12332400	-0.96536700
7.99744900	-3.58005400	-0.39429700
5.90695300	-4.77375700	1.10021500
7.58163100	-3.00574900	0.83055600
5.10018900	-5.22630100	1.68848600
6.53473400	-3.59493000	1.58784900
8.08420400	-2.10551900	1.20941000
-4.39564600	-6.60630100	1.68706500
-4.21273700	-6.58655600	4.22433900
6.12700500	-3.00075600	2.92171800
-1.88820200	-5.90289000	2.89149600
-3.51783500	-5.74817600	4.39135300
-2.76264400	-6.04778300	5.14049200
-3.75322100	-5.52106500	1.84442600
-2.82025800	-5.33727300	3.07390600
5.94345200	-3.78828700	3.67337600
4.87068400	-2.16449600	2.83635500
-3.77415100	-4.42228400	1.00139800
-4.58211700	-4.20487200	-0.15254100
0.99367800	-4.06505000	1.93617900
	7.69208300 5.83508600 7.36589900 6.31934800 8.81431800 7.99744900 5.90695300 7.58163100 5.10018900 6.53473400 8.08420400 -4.39564600 -4.21273700 6.12700500 -1.88820200 -3.51783500 -2.76264400 -3.75322100 -2.82025800 5.94345200 4.87068400 -3.77415100 -4.58211700 0.99367800	7.69208300 $-5.20412500$ $5.83508600$ $-6.25970100$ $7.36589900$ $-4.75258100$ $6.31934800$ $-5.34600900$ $8.81431800$ $-3.12332400$ $7.99744900$ $-3.58005400$ $5.90695300$ $-4.77375700$ $7.58163100$ $-3.00574900$ $5.10018900$ $-5.22630100$ $6.53473400$ $-3.59493000$ $8.08420400$ $-2.10551900$ $-4.39564600$ $-6.60630100$ $-4.21273700$ $-6.58655600$ $6.12700500$ $-3.00075600$ $-1.88820200$ $-5.90289000$ $-3.51783500$ $-5.74817600$ $-2.76264400$ $-6.04778300$ $-3.75322100$ $-5.52106500$ $-2.82025800$ $-5.33727300$ $5.94345200$ $-3.78828700$ $4.87068400$ $-2.16449600$ $-3.77415100$ $-4.20487200$ $0.99367800$ $-4.06505000$

С	-4.22765400	-4.43258300	4.81842300
Ο	-2.44365100	-3.90415500	3.24350800
Η	-5.17744800	-4.31653200	4.26524500
С	3.68204400	-2.85061400	3.15869300
Н	-4.45139300	-4.40402300	5.89858000
Н	-0.45401100	-3.08681200	2.28845900
0	3.61923000	-4.09872000	3.34937800
С	0.64362900	-3.04077100	2.16719200
Ċ	0.79137800	-0.63180000	1.21145800
C	-3.21411600	-3.33994600	4.40551100
Н	-3.18044900	-3.64423700	1.32838600
Н	-2.48562900	-3.12508200	5.21032500
C	-4 35260300	-3 02188100	-0 92137400
Н	-5 29008600	-1 61879000	1 21444800
C	1 03105800	-2 11863000	0 97493200
N	2 60647100	-1 97907400	3 29542200
C	1 25811900	-2 58995400	3 51788500
н	0.47317300	-2 47138800	0.09076900
н	-2 54007200	-4 16459500	-2 52341400
C	-4 40208100	-0.98786100	1 25083800
н	-3 68204300	-2 40028200	4 06504200
н	1 36174400	-3 44309500	4 20813500
C	-3 18851400	-2.09568500	-0.65877300
н	2 10111600	-2.07500500	0.72508400
C	2.10111000	-0.59478300	2 96887700
н	-4 71150900	0.43286800	2.90087700
C	-1 82849800	-3 35389000	-2 37179500
$\frac{C}{C}$	-4 10959900	0.04247100	2 12090600
н	0.63546600	-1 81708500	3 99771200
C	-3 27865900	-1 09778600	0 32922400
Н	-1 51513100	0 56638300	6 06487800
Н	3 36116700	-0 29169600	2 19040400
C	-2 03645900	-2 23628100	-1 46075500
Н	-5 98704200	4 44777100	6 74744700
C	-0 58969600	-3 17600400	-2 95321400
н	-3 08615800	2 10107400	6 90744600
н	-0.08599500	-3 80626900	-3 68529400
н	-4 97583200	3 44111900	7 82630400
C	-1 66429000	0 77460400	4 99121500
C	-3 32649800	2 47307500	5 90238800
$\frac{c}{c}$	-2 82969700	0.60250900	1 70517000
$\frac{c}{c}$	_4 94930800	4 13987200	6 97084400
N	-2 30750700	-0 10707000	0.60100300
C	2.30730700	0.30008300	3 97987700
$\frac{c}{c}$	-2 67035300	1 89670000	4 78504400
Ч	-2.07033300	-0 16003300	4 50080200
11	1.777/1/00	0.10005500	7.50007500

С	-4.27982700	3.51128700	5.75669400
Ν	-0.93317600	-1.35449200	-1.48934000
Η	2.32948100	2.07040100	2.56974600
Н	2.21500000	-0.99986500	5.66207800
Н	-0.68004400	1.03119100	4.55860300
Н	2.39650600	-3.79880500	-1.57638100
C	-0.04800100	-1.92807600	-2.43286600
C	-2 97182900	2 37732700	3 47464900
C	2 20862700	1 78284200	3 62059600
C	2 10769700	0.04771400	5 35798100
C	-4 58045000	3 95859000	4 44534800
C	-2 28622400	1 76931400	2 27544000
$\frac{c}{c}$	-3 94426800	3 41131500	3 30262400
н	-4 40277600	5.04557900	7 30152500
н	-5 15436900	4 63791000	1 97226000
н	-5 33005300	4.03771000	4 30636300
C	3 23898700	-3 24984800	-2 04002000
$C_{0}$	-0 64243700	0 27298700	-0.39725800
со н	3 3/727000	-2 30//6100	-0.57725000 -1.47705700
C	-4 32002700	3 91687800	1 91627500
C	2 98035100	-2 99275400	-3 51727100
C	1 13070100	-1.34557900	-2 911/2/100
C	3 7/0//300	3 67/68/00	-2.711 <del>4</del> 8000 1 10608700
с и	1 53130800	4 37017000	4 16334200
и П	4.55159800	4.57017000	7 60440700
C	1 96327500	-2 08448700	-3 93882100
н	-4 62623700	3 08554100	1 25323500
C	1 96641600	2 76929200	4 59589900
C	1.90041000	1 0/20200	6 33015200
н	-2 06955200	0.32400100	-3 58692400
н	0.58283300	-0.91845700	-6 91834600
C	3 53426200	-3 48463600	-5 88261100
$\frac{c}{c}$	-1 20256700	2 43931400	1 67765900
н	4 15571400	-3 84558300	-1 88970000
н	4 99530400	-5 00277800	-6 45670000
C	1 72564700	-1 88093000	-5 33471100
$\frac{c}{c}$	4 37790800	-4 21288900	-6 91983200
C	2 51275800	-2 58385900	-6 27979700
C	0.63737200	-0.93096200	-5.81563100
N	-0.50348100	2 01759700	0 52322200
$\hat{\mathbf{C}}$	1 79804500	2.01757700	5 95830800
н	1 90974600	3 82165300	4 29581700
Н	1 75925400	0 75526000	7 38283100
Н	-0 35450300	-1 22807000	-5 42649900
Н	-3 46564400	4 42092400	1 42540000
C	1 51675700	-0.04185700	-2 53701700
$\sim$	1.51075700	0.04103/00	2.55701700
Η	2.32276000	-2.42510300	-7.35000400
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Ν	0.86425600	0.77018200	-1.58180100
Η	-2.61992000	3.35624900	-1.22311000
Η	-4.53434900	5.41826600	-1.42616900
С	-0.74059500	3.75278300	2.10527700
Н	-3.89147500	3.29393100	-2.49455900
Н	-1.11941800	4.27969100	2.98029900
С	-3.06790600	3.89757800	-2.07380400
Н	4.61135900	0.07569300	-1.03867000
Η	5.06246900	-3.51474600	-7.44058500
Н	5.32139700	-0.84807400	0.35061700
Н	1.61782800	3.17942700	6.71568100
С	-3.48005400	5.35057900	-1.74513800
Н	0.81951400	0.10426700	-5.46930900
С	0.36642100	3.08928300	0.22431800
Ċ	5.22687700	0.14494900	-0.12592200
С	2.59123700	0.69495700	-3.19009000
С	1.53371200	2.01201300	-1.65738700
Н	-2.84675600	5.76582100	-0.93974100
С	0.20843800	4.16673700	1.19340200
Н	3.23280900	0.28219100	-3.96797800
0	4.48918600	1.03709900	0.83915800
С	1.29410600	3.13178700	-0.83717700
0	-2.00345500	4.00617100	-3.14018000
Н	7.38047700	0.07828700	-0.59738400
С	2.58110800	1.97372000	-2.67003600
Н	-4.04555200	5.94635000	-3.78838000
С	6.59949600	0.81627900	-0.34682600
С	-3.21462100	6.10655900	-3.07723400
Η	-0.20730100	4.28419200	-2.32966500
Η	3.47774000	2.59837500	0.10875700
Η	0.76742100	5.10155100	1.16994500
Η	6.54768800	1.55817100	-1.16460700
С	-1.93161600	5.41874100	-3.60622800
Η	7.22956700	0.81039100	1.76435700
Η	-3.06188800	7.19018500	-2.94666500
С	5.45238100	2.02959300	1.40607600
С	6.86105900	1.52880200	1.00964500
Ν	0.18880800	5.23707800	-2.39557000
С	2.05739100	4.41398700	-1.06739900
Η	3.21536000	2.82059000	-2.93101500
N	3.95306900	3.51369600	0.16601100
С	-0.63550900	6.10369700	-3.09100600
Н	5.30050800	2.05109200	2.50065700
Н	-1.88488200	5.40711700	-4.71061200
С	1.45922300	5.48080600	-1.80876600

С	5.15542500	3.44896100	0.84656900
С	3.35872200	4.61337900	-0.51195300
Η	7.57113100	2.36832200	0.93695000
0	-0.41420700	7.33064700	-3.33702600
0	5.96746500	4.40546900	1.04591200
С	2.12505900	6.72886100	-1.95332200
С	4.02682700	5.86060400	-0.64768900
Η	1.63713600	7.52868200	-2.51443700
С	3.39404700	6.90045300	-1.36138600
Η	5.01087200	5.98959000	-0.19224100
Η	3.90526100	7.86453700	-1.46504600
Η	1.50915100	0.10546100	0.84196200
Η	6.87361300	-2.27934800	3.29176200
С	-5.22935900	-2.74284800	-2.02015600
С	-6.26150600	-3.65104700	-2.38050900
Η	-6.90196100	-3.41470600	-3.23281700
С	-6.42271900	-4.83440900	-1.63127400
Η	-7.20894700	-5.54389600	-1.91400400
С	-5.60656800	-5.12160400	-0.51797600
Η	-5.73498100	-6.03462000	0.06618700
Ν	-5.03797700	-1.52208200	-2.72170200
С	-5.70057900	-1.04169500	-3.83760800
Η	-4.33307100	-0.85414600	-2.36679200
0	-6.61928200	-1.62994400	-4.48971200
С	-5.22925700	0.37843300	-4.25513300
0	-4.11078500	0.84114700	-3.38798100
С	-4.68901000	0.41812700	-5.70656700
Н	-6.08112500	1.06689300	-4.10470600
С	-2.83372700	0.83614700	-4.19620200
С	-3.17873400	0.10613000	-5.51384100
Η	-4.82652400	1.42984300	-6.13021000
Η	-5.21249100	-0.31000000	-6.34722000
Η	-2.53752300	1.88949000	-4.34708300
Η	-2.56426200	0.46495500	-6.35738100
Н	-3.02628700	-0.98419200	-5.41157500
Η	0.11293500	-0.31085500	2.01053300

Intermediate E E_[Co(II)(P16)]=A_[Co(II)(P16)]

 $E_{[(S)-2a]}$ Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.304073 Hartree H_corr: 0.375911 Hartree SCF: -942.3408591 Hartree

S: 151.195 Cal/Mol-Kelvin H: -941.9649481 Hartree G: -942.0367861 Hartree

Η	-5.15112400	-3.33750900	-0.47050100
С	-4.41000300	-2.56170400	-0.24460000
Η	-4.43329600	-1.71508100	-2.25407500
Η	-4.13616300	-3.18680500	1.82859200
С	-4.00620500	-1.64686500	-1.24620300
С	-3.84000000	-2.47432800	1.04921000
С	-3.04643800	-0.64778300	-0.95600200
С	-2.88183600	-1.47477400	1.33501200
Η	-2.72071400	0.04172500	-1.74372900
С	-2.47849200	-0.54726100	0.33879200
Η	-2.43410000	-1.42270200	2.33687800
Η	-2.31661300	2.48350500	-1.16127100
С	-1.48086100	0.56094900	0.71113100
Η	-3.14319300	1.96562900	1.05771200
Ν	-0.80351700	1.17551000	-0.46629900
Η	-0.69235200	0.11553000	1.34770700
С	-1.24188800	2.52552500	-0.89565400
Η	-0.67674300	2.78347300	-1.80582300
С	-2.18012100	1.69292400	1.53265500
Η	-2.42619500	1.28981900	2.53249700
0	0.58572500	-0.65691400	-0.50302000
С	0.26828900	0.58129300	-1.10256500
Η	1.49357200	-2.43794800	-0.97541600
Η	-1.69385100	4.42361300	0.06678100
С	-1.01038800	3.56836500	0.23556500
С	1.76414400	-1.37714100	-1.10229100
0	0.89368800	1.05985600	-2.09407500
С	-1.26061800	2.93362200	1.64203300
Η	1.80903500	-1.11071300	-2.17250700
Η	-0.29602700	2.62066400	2.08856800
Η	0.02197000	3.95977000	0.17775500
Η	-1.70235200	3.67946300	2.32928100
Η	2.96845100	-2.74683100	0.96065300
С	3.05446600	-1.03152000	-0.38359500
С	3.53719600	-1.85457300	0.66777700
Η	3.41879900	0.75737800	-1.56583100
С	3.79859000	0.12022100	-0.75837500
С	4.74078100	-1.53336200	1.33896000
Η	5.10629300	-2.17726900	2.14774200
С	5.00064300	0.44297500	-0.08724000
С	5.47525600	-0.38269300	0.96211200

H 5.56900100 1.33262800 -0.38354200 H 6.40982100 -0.13368300 1.47905200

### **Transition State TS2'**

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -1198.6029 cm-1 G_corr: 1.425421 Hartree H_corr: 1.681987 Hartree SCF: -6075.275085 Hartree S: 539.989 Cal/Mol-Kelvin H: -6073.593098 Hartree G: -6073.849664 Hartree

Η	7.83617000	-1.97568200	1.09970900
Η	7.41451200	-0.62192500	-0.98950100
С	7.07494900	-2.29705000	0.37882200
С	6.84119700	-1.53469400	-0.79289900
Η	6.51231100	-4.07525700	1.51297000
С	6.32803900	-3.47701800	0.61247000
С	5.86736900	-1.95499500	-1.72623300
С	5.34548000	-3.88845300	-0.31931000
Η	5.69538600	-1.38205900	-2.64400600
С	5.10435900	-3.13046800	-1.49494400
Η	4.77315900	-4.80721200	-0.13768600
0	6.86829900	1.53766500	-2.45345400
Η	6.48696900	2.48903000	-4.77920100
С	4.09414200	-3.60614900	-2.51841000
Η	4.67105200	0.52180000	-4.05709500
С	5.38660800	2.48499100	-4.83406700
Η	5.07478100	2.12950500	-5.83269300
С	5.59910500	1.63214700	-2.42782200
С	4.77948200	1.57693600	-3.74329100
Η	4.50711500	-3.54753300	-3.54134800
0	2.80844500	-2.80551200	-2.51593700
Ν	4.81726300	1.76831200	-1.29181100
С	5.24015700	1.95828700	0.05535900
Η	0.21131800	1.65397000	-4.59758300
С	4.75212300	3.86789300	-4.51143100
0	3.39416500	2.09335100	-3.52202200
Η	5.31585500	4.36184900	-3.69909200
С	2.68058400	-1.87000100	-3.54857500
Η	4.74228300	4.54354400	-5.38399500
Η	-0.94182300	0.35562600	-4.23577300
0	3.66575300	-1.43906800	-4.22564300

С	0.09050500	0.70604900	-4.03576200
С	-0.41463400	-0.00024400	-1.56278000
С	3.32199500	3.50176200	-4.04379300
Η	3.80660200	1.80666100	-1.49181100
Н	2.59848400	3.50159800	-4.88102600
С	4.25898500	1.88565700	1.09131400
Н	3.21029000	4.23295200	0.35197800
С	0.26596000	1.00163700	-2.51105900
Ň	1.36409900	-1.48091800	-3.76475200
C	1.14622100	-0.26867200	-4.62170300
Н	-0.19101500	1.99152800	-2.30836900
Н	4.47506300	-0.67918600	1.78606600
C	2.20936900	3.81604100	0.24334700
Н	2.94434300	4.14544900	-3.22965200
Н	2.11594700	0.24806100	-4.69328000
C	2.83523000	1.45224400	0.82496900
Н	1.34281300	1.10585600	-2.28630500
С	0.19071800	-1.99293500	-3.08157800
Н	0.86747200	5.52564500	-0.25168400
C	3.40563100	-0.84830300	1.67175700
Ċ	1.03084300	4.46677600	-0.05458100
Н	0.84625900	-0.59070200	-5.63993100
C	1.87925000	2.41064600	0.44126700
Н	-1.54898900	5.35741100	-4.05577000
Н	0.07123800	-1.29119400	-1.97054500
C	2.46849800	0.12532800	1.12615500
Н	-3.99571200	9.49238300	-1.25788000
С	2.67497700	-1.96782800	2.00019700
Н	-2.19660300	7.43327900	-3.17410000
Н	3.02459000	-2.89723600	2.44689800
Н	-2.44117600	10.06274600	-0.60163300
С	-1.46385900	4.80565700	-3.10312300
С	-2.16382800	7.05622600	-2.14313500
С	-0.02412000	3.46349000	-0.04477000
С	-2.89686900	9.37272400	-1.33566300
Ν	0.49678500	2.18655000	0.26434400
С	-0.01332600	-3.47500600	-2.89687700
С	-1.80133400	5.70280500	-1.91984000
Н	-0.43273400	4.40981400	-3.02628400
С	-2.48404300	7.93070300	-1.07545800
Ν	1.16169600	-0.41078900	1.06973000
Н	1.44209300	-4.14350500	-4.38313900
Н	-1.58167900	-3.16858600	-1.41299400
Н	-2.14024300	3.93099000	-3.15232600
Н	2.36116200	-4.07167900	0.22075300
С	1.29282800	-1.70859300	1.61794400

С	-1.75399600	5.20627700	-0.58230600
С	0.66175200	-4.45294800	-3.67948700
С	-1.04962900	-3.90614000	-2.02368200
С	-2.42708000	7.41742900	0.24540600
С	-1.36933500	3.77246900	-0.31028700
Ċ	-2.07002300	6.07201400	0.51085600
Н	-2.59866500	9.70351300	-2.34640500
Н	-2 30192300	6 36719600	2 65615500
Н	-2 66611000	8 07850000	1 08940800
C	1 65425300	-4 90673600	0.37603700
Č0	-0 48844400	0 45342600	0.35967000
н	0.79359200	-4 70876300	-0 28676800
C	-2 02705300	5 56700900	1 94683800
C	1 22374800	-5 02478700	1.94005000
C	0.26312400	-2 65193200	1.02507100
C	1 47298600	-6 23000/00	2 53645300
н	1.98122200	-7.05136400	2.01376500
ц	0.51283400	8 0304400	5 22080100
C	0.51285400	3 05323700	2 40632600
ч	1 02002800	5 10800100	2.49032000
C	-1.02002800	5 81007000	2.21890700
C	1 40144600	-3.81907900 5 27177000	1 02072600
с u	-1.40144000	-3.2/1//900	-1.93073000
и П	0.02097800	2 25500400	5.53332800
$\Gamma$	-0.70122700	-3.23390400	3.07274300
C	-2 369//200	2 78560000	-0.30682600
н	2 135/15000	-5.84021200	0.03328200
н	2.13343700	-7 56/01000	5 33634700
C	0.1/698/00	-/ 110/000	3 860/2600
C	1 38249900	-7 69727400	4 63331200
C	0.41892100	-5 32981500	4 52888400
C	-0 56467300	-2 99141400	4 60924900
N	-2 18771100	1 41155000	-0.02535900
C	-0 72260900	-6 23878200	-2 71156700
н	0.83656300	-6 55606300	-4 20543600
н	-2 20026600	-5 57637900	-1 24455500
н	0.00897200	-2 04750200	4 55910700
н	-2 72640000	4 72176900	2 09511000
C	-1.04766200	-2 39756900	1 34077200
н	0 10487100	-5 44065000	5 57572900
N	-1 52292500	-1 19290300	0.78012100
Н	-2 93128300	1 86005900	3 28083400
Н	-4 54179000	3 15994300	5 17533500
C	-3 76602600	3 06934200	-0.60344300
н	-2 60524100	1 65904400	5 04411300
Н	-4 15413200	4 06051000	-0 83488800
11	115715200	1.000001000	0.00400000

С	-3.36137800	1.53905100	4.24432600
Η	-2.79162500	-3.59229100	-4.17745500
Η	1.65244200	-8.51263000	3.93890300
Η	-2.01860800	-2.78528100	-5.59131000
Η	-0.99285200	-7.29881900	-2.64077600
С	-4.69602700	2.22598600	4.60803700
Н	-1.56137600	-2.77530400	4.18123900
С	-3.47626200	0.84944800	-0.17157000
С	-2.93127600	-2.83763600	-4.96838600
С	-2.11277600	-3.38907700	1.40657800
С	-2.88563800	-1.44316300	0.50646800
Н	-5.27505800	2.46351000	3.69712600
С	-4.44872300	1.87357000	-0.52670900
Н	-1.98562300	-4.39635800	1.80120700
0	-3.08672800	-1.49378000	-4.29592000
С	-3.82334500	-0.49930500	0.04244100
0	-3.67751100	0.07489300	4.12336800
Н	-4.00136200	-3.64077900	-6.72577400
С	-3.24452000	-2.80729300	0.87503000
Η	-5.03940700	1.11932500	6.47941900
С	-4.20288100	-3.04259100	-5.82075200
С	-5.42355100	1.13491900	5.44293200
Η	-4.46164100	-0.44694900	2.36140100
Η	-4.05719800	-1.18153000	-2.39735200
Η	-5.51073800	1.68719700	-0.68077800
Η	-4.99006800	-3.55385800	-5.23716400
С	-5.02529500	-0.17413100	4.71239400
Η	-4.04814000	-1.17673300	-6.98521500
Η	-6.51820000	1.25694100	5.47287000
С	-4.31372200	-0.84192100	-4.84208800
С	-4.64421000	-1.58998500	-6.15137900
Ν	-5.47651500	-0.64230700	2.34830800
С	-5.27762300	-0.88763200	-0.07485100
Η	-4.24030600	-3.23400300	0.75962400
Ν	-5.07645700	-1.08792300	-2.51161400
С	-6.04557600	-0.55913600	3.60662100
Η	-4.06760200	0.22365700	-5.00810600
Η	-4.94064900	-1.03224700	5.40364700
С	-6.09402600	-0.93784800	1.10256700
С	-5.48723700	-0.91321300	-3.82412100
С	-5.89294500	-1.13136100	-1.34063400
Н	-5.71579300	-1.50431500	-6.39231700
0	-7.26272200	-0.76746300	3.90752400
0	-6.68741500	-0.77649800	-4.21754600
С	-7.48054000	-1.23744700	1.01629400
С	-7.28271800	-1.41762900	-1.43020100

Η	-8.07479300	-1.26515300	1.93203500
С	-8.05340900	-1.46997800	-0.25090600
Η	-7.73413100	-1.57180000	-2.41138700
Η	-9.12417000	-1.69305300	-0.32129200
Η	-0.69757400	-1.59783000	-3.60464300
Η	3.74451100	-4.62820900	-2.29713500
С	4.65787000	2.20807300	2.42992200
С	6.01037200	2.51854100	2.73577300
Η	6.28187400	2.74431900	3.76901800
С	6.96226400	2.52350100	1.69587600
Η	8.00845700	2.75360700	1.92870700
С	6.59691500	2.25897600	0.36096400
Η	7.33148400	2.27559400	-0.44598500
Ν	3.65249300	2.22235900	3.43486500
С	3.78089300	2.45408700	4.79257500
Η	2.66971100	2.07298900	3.15198600
0	4.85808700	2.67817800	5.42930400
С	2.42318100	2.44044700	5.54481000
0	1.30524300	2.12365200	4.61024100
С	2.37808900	1.35592900	6.65208600
Η	2.25410200	3.45718900	5.94397000
С	0.76184000	0.76589700	4.95105600
С	1.79991100	0.12526100	5.89930600
Η	1.69477000	1.67248900	7.46154200
Η	3.37833300	1.18252500	7.08064900
Η	-0.21886600	0.90501200	5.44603400
Η	1.34219400	-0.61052000	6.58243600
Η	2.59304700	-0.38296900	5.32150400
Η	-1.47240800	-0.17913700	-1.86892800

### Intermediate D'

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 1.426151 Hartree H_corr: 1.686746 Hartree SCF: -6075.32398 Hartree S: 548.469 Cal/Mol-Kelvin H: -6073.637234 Hartree G: -6073.897829 Hartree

Η	-7.00919100	-1.35883700	3.07261800
Η	-7.08117700	-1.90355100	0.60824200
С	-6.83445700	-0.57218500	2.32872100
С	-6.87915800	-0.88103100	0.94673800
Η	-6.54334000	0.99914400	3.81668500

С	-6.57079900	0.75547400	2.74780900
С	-6.66787000	0.13757200	-0.01067900
С	-6.35405000	1.77084400	1.78681900
Η	-6.70711600	-0.09674100	-1.08013000
С	-6.39944600	1.47068800	0.39939800
Н	-6.15291800	2.79960500	2.11345600
0	-6.32208400	-3.24237800	-1.77400000
Н	-6.27190300	-3.14321900	-4.35281600
С	-6.20675000	2.56671400	-0.62854000
Н	-5.17386200	-0.94091600	-3.06176700
С	-5.32181400	-2.61237900	-4.52520900
Н	-5.47942900	-1.83100800	-5.29046300
С	-5.13540400	-2.82914400	-1.97621200
С	-4.79950600	-1.96822500	-3.22126400
Н	-6.78579000	2.36572100	-1.54644900
0	-4.75790300	2.74229800	-1.03396200
Ν	-4.04414300	-3.05449000	-1.15365300
С	-3.96352000	-3.85602300	0.01936900
Н	-0.63725500	0.89494800	-4.53758100
С	-4.15369400	-3.55580000	-4.92972400
0	-3.31705500	-1.85365800	-3.39570600
Η	-4.21082100	-4.49932900	-4.35655800
С	-4.35756800	2.04570600	-2.17728800
Η	-4.16200300	-3.80426700	-6.00505700
Η	-0.34585100	2.23319200	-3.42797900
0	-5.07664700	1.18225900	-2.76144500
С	-0.92744900	1.29711600	-3.54495000
С	0.01443600	0.87952600	-1.16722800
С	-2.89762200	-2.74591700	-4.53048500
Η	-3.18198200	-2.60264200	-1.49702100
Η	-2.54993400	-2.09212400	-5.35327900
С	-2.78212000	-3.75904800	0.81618300
Η	-1.40784800	-5.01276400	-1.15016500
С	-0.50524500	0.24473700	-2.45747400
Ν	-3.06875200	2.41100000	-2.58413200
С	-2.44390300	1.58187200	-3.67468600
Η	0.29823900	-0.37517400	-2.89312300
Η	-3.76775100	-1.73548700	2.25616400
С	-0.69283200	-4.19520300	-1.23944700
Η	-2.05949600	-3.36887500	-4.17227600
Η	-2.99312000	0.62554100	-3.70694300
С	-1.71582900	-2.73029100	0.53207400
Н	-1.35819500	-0.42900900	-2.25896200
С	-2.41701900	3.59517500	-2.15013100
Н	0.69572700	-4.75106000	-2.89300500
С	-2.87739700	-1.13005400	2.09471200

С	0.36400500	-4.06255800	-2.11646300
Η	-2.62252000	2.11528200	-4.63165300
С	-0.70480400	-2.99934200	-0.40735800
Η	1.45124000	-1.59089500	-6.34533500
Н	-0.77274200	1.34627000	-0.55005500
С	-1.76421500	-1.51199000	1.23650100
Н	4.46824000	-6.66644700	-6.60198100
С	-2.58763500	0.11931300	2.59900700
Н	2 77823700	-3 40553400	-7 00079100
Н	-3.18851700	0.73695100	3.26457300
Н	3.90583200	-5.44294600	-7.77892200
C	1 57102200	-1 69733400	-5 25282200
C	3 02033400	-3 66377900	-5 96115900
$\frac{c}{c}$	1 03103600	-2 80859900	-1 79450600
$\frac{c}{c}$	4 46610400	-5 58628500	-6 83788900
N	0 37470200	-2 14851200	-0 73135000
$\hat{\mathbf{C}}$	-2 99486400	4 90940200	-2 17080900
C	2.77435500	-2 87566200	-4 91524800
н	0 56554600	-1 82028400	-4 80664500
C	3 86760400	-4 76891400	-5 70162700
N	-0 77571100	-0 50362400	1 21713900
H	-4 81397400	4 41586300	-3 30221300
н	-1.32070900	5 70368/00	-1.06/2/800
н	1 98183600	-0.7/782200	-1.85973600
н Ц	2 0321/200	3 10555300	1 06832000
C	-1 28066900	0.49773000	2 08149400
C	2 78180000	-3 20542300	-3 56068900
C	-4 25125500	5 21470800	-2 80623200
C	-2 27324100	6 00147600	-1 56783600
C	4 15983200	-5 07932500	-4 34911100
C	2 21352600	-2 38543100	-2 42762800
C	3 63547900	-4 31723300	-3 27560500
Н	5 51699600	-5 29163500	-7 02929600
Н	4 65948200	-5 56554300	-1 81478500
н	4 81241800	-5 93338800	-4 12298700
C	-2 27970600	3 88554300	1 50494700
$C_{0}$	0.85530200	-0.41739200	0 11608900
н	-1 32493400	3 84724400	0.11000900
C	3 98862700	-4 68913800	-1 84167900
C	-2 05755000	3 67399000	2 99386100
C	-0 59502100	1 65294300	2.99900100
C	-2 66357300	4 55195300	3 93001600
н	-3 27729500	5 38180900	3 55478100
Н	-3 54515900	4 82623200	7 19344300
C	-1 25446100	2 59559400	3 47240900
н	3 08533100	-4 93020100	-1 24970900
**	2.00233100	1.75020100	1.2 17 10 7 00

С	-4.74696800	6.53052100	-2.82993500
С	-2.78062300	7.31252800	-1.59167700
Η	0.48235000	-1.70572800	3.60853400
Η	-0.19485000	1.28288700	6.52106400
С	-2.49722100	4.38824400	5.32694100
С	2.94354900	-1.28577200	-1.94289700
Н	-2.75598300	4.86056900	1.30389900
Н	-3.95183000	5.93208700	5.84791500
С	-1 07573400	2 40823700	4 87880800
C	-3.13294700	5.35652300	6.31501800
C	-1.70093900	3.30530600	5.78000100
C	-0 23528200	1 25880700	5 41789700
N	2 57951700	-0.46224600	-0.85134600
C	-4 02322200	7 59145900	-2 22166700
н	-5 70017900	6 74066900	-3 32995200
н	-2 21445000	8 12395300	-1 11861800
н	-0.65298000	0.28052600	5 11199400
н	4 49684800	-3 85278500	-1 32452200
C	0.73017900	1 89930800	2 09155900
н	-156173600	3 15521500	6 85921200
N	1 / 8105600	1 11706500	1 18638700
н	3 75638200	2 / 21// 200	1 3183/300
и П	5.73038200	1 30333600	2 11375600
$\Gamma$	<i>J</i> .0244 <i>J</i> 300 <i>A</i> 27600100	-4.39333000	2.113/3000
с u	4.27000100	2 48288700	2 74140500
н	<i>1</i> 77113800	-3.48388700 -1.44103400	-3 25822600
C	<i>A</i> 1/0/0000	-2 76612500	2 297/15/00
н	1 155////00	4 83234700	-10/913300
н	-2 38803000	6.08529700	6 6 9 3 5 / 0 0 0
н	0 58432800	5 14995500	-273684400
н	-4 41742200	8 61380200	-2 24189700
C	5 59130800	-3 29862600	2.24107700
н	0.80223200	1 29846200	5 03532200
C	3 70909000	0.36116200	-0 64453100
$\frac{c}{c}$	1 46330800	5 13771700	-2 06479300
C	1 53979000	2 98421800	2 62868100
C	2 75246600	1 73000200	1 16125900
н	6 15601100	-2 82968500	1 42325900
C	4 76032600	0.05130900	-1 60409400
н	1 17999000	3 71792200	3 34903900
$\hat{0}$	2 41245800	4 09286900	-2 57700000
$\frac{0}{C}$	3 83257600	1 36336200	0.33630700
$\tilde{0}$	4 17682700	-1 55401300	3 19898200
н	1 56214700	7 33801300	-2 25644600
C	2 79707100	2 86558100	2.23044000
н	5 86696100	-3 54823400	4 42058500
11	2.00070100	5.5-1025-100	1.72020200

С	2.23476400	6.47660600	-2.10473700
С	6.17959900	-2.85452400	3.61880700
Η	4.68560800	0.07644200	2.16141100
Н	3.70857600	3.20990300	-1.27544800
Н	5.72960300	0.54722200	-1.63603700
Η	2.79736300	6.63450700	-1.16653500
С	5.52004400	-1.47009400	3.83709200
Н	2.70230600	6.43584700	-4.25613800
Н	7 27898000	-2 77852500	3 62233000
C	3 60768900	4 78127900	-3 14820700
C	3 21635600	6 27087100	-3 29188200
N	5 65236800	0.42112700	2 28508300
$\hat{\mathbf{C}}$	5 17024800	2 03621500	0.52305100
ч	3.67833600	2.03021300	2 24027000
N	<i>J.</i> 07855000	2 62252200	1 25525200
	4.03400300	0.21462000	-1.23333300
	0.33707100	-0.31408000	3.22093000
п	5.82882000	4.30314/00	-4.1211/300
Н	5.36928400	-1.23881500	4.90/34/00
C	6.0/881500	1.53135100	1.506/6600
C	4.84633500	4.58601400	-2.23097700
C	5.5//46600	3.1412/200	-0.2849/400
H	4.10622000	6.918/5800	-3.24388600
0	7.55599800	-0.11932100	3.59580800
0	5.90604400	5.25760000	-2.43409900
С	7.36278000	2.11657900	1.67839300
С	6.86167000	3.72878000	-0.11753000
Η	8.03792600	1.70424900	2.43119900
С	7.73223600	3.20560900	0.86156300
Η	7.15113800	4.56501000	-0.75684600
Η	8.72291000	3.65686100	0.98975600
Η	-1.37396000	3.47167000	-1.83779500
Η	-6.45439900	3.56026100	-0.22067500
С	-2.64324900	-4.62261800	1.94917800
С	-3.67838000	-5.52932900	2.30459500
Η	-3.54723600	-6.16897300	3.17990800
С	-4.84977000	-5.56922600	1.51943600
Η	-5.65458600	-6.26021500	1.79611700
С	-5.00754100	-4.75216200	0.38057100
Η	-5.91382300	-4.78530900	-0.22713500
Ν	-1.43548000	-4.53713800	2.69372200
С	-1.07173800	-5.20130800	3.85166600
Н	-0.69187500	-3.90539000	2.35217000
0	-1.77101900	-6.04434900	4.49616400
С	0.36024400	-4.83540200	4.33340300
0	0.95297000	-3.76655800	3.48250900
Ċ	0.36711200	-4.27781500	5.77768200

Η	0.99019900	-5.73709700	4.22475100
С	0.94509900	-2.46889600	4.25740400
С	0.14871500	-2.75530200	5.55177400
Η	1.35001500	-4.46608100	6.24707400
Η	-0.41757000	-4.74905900	6.39145400
Η	1.99819700	-2.19391700	4.44517600
Н	0.50926300	-2.14906100	6.40043800
Н	-0.92658900	-2.54369400	5.40682900
Н	0.84546300	1.58374500	-1.35407800

#### **Transition State TS3'**

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -512.035 cm-1 G_corr: 1.429097 Hartree H_corr: 1.6858 Hartree SCF: -6075.305653 Hartree S: 540.278 Cal/Mol-Kelvin H: -6073.619853 Hartree G: -6073.876556 Hartree

Η	7.64612300	-2.33624900	0.32078400
Η	7.30677800	-0.61359600	-1.49318000
С	6.89088500	-2.49751100	-0.45756000
С	6.70509600	-1.52893700	-1.47487000
Η	6.24967400	-4.42951300	0.33076100
С	6.10330400	-3.67413700	-0.45050200
С	5.73799100	-1.74186800	-2.48279700
С	5.12773400	-3.87783900	-1.45456800
Η	5.60616200	-1.00487300	-3.28319500
С	4.93441400	-2.91267100	-2.47727500
Η	4.52342600	-4.79410600	-1.45174500
0	6.75930700	1.76904800	-2.35682600
Η	6.25563800	3.17637100	-4.42340500
С	3.92409300	-3.15733700	-3.57565800
Η	4.49322000	1.08206600	-4.01161700
С	5.15402100	3.16463000	-4.42517400
Η	4.79524000	2.99912300	-5.45703600
С	5.49554100	1.87314200	-2.23962000
С	4.60747900	2.05519000	-3.49921900
Η	4.31718400	-2.85443100	-4.56212800
0	2.62487300	-2.39767400	-3.36775100
Ν	4.78104500	1.82970800	-1.05428600
С	5.27872300	1.80112700	0.28137300
Η	-0.05143400	2.34796500	-4.20831800

С	4.52477100	4.44989800 -3.81639700
0	3.23288600	2.49725300 -3.11862800
Η	5.12116700	4.79027100 -2.95031800
С	2.42326600	-1.29984100 -4.20083000
Η	4.46883400	5.27875200 -4.54287600
Н	-1.19930400	1.00233500 -4.07878500
0	3.34877000	-0.72079900 -4.84660700
Ċ	-0.14443800	1.28849000 -3.89774200
Ċ	-0.19260400	-0.11292100 -1.67240700
C	3.12121500	3.97740200 -3.36556700
Н	3.76157000	1.91103200 -1.18698800
Н	2 36161300	4 11869400 -4 15784700
C	4 34925100	1 59435300 1 34800900
н	3 36068800	4 04776400 0 91021800
C	0.18873600	1 19165400 -2 37631800
N	1 08008000	-0.89353000 -4.25149500
$\hat{\mathbf{C}}$	0.82651600	0.48576900 -4.79716300
н	-0.35520700	2 01708800 -1 88288000
и П	-0.33320700 A A7635700	1 02085600 1 81010000
C	2 35021100	3 67102300 0 75100500
ц	2.33021100	<i>A A</i> <b>A 8 3 1 5 0 0 1 7 3 10 0 0 10 10 0 0 10 1</b>
и П	1 80200700	0.00236100 / 83870300
C	2 00272400	1.24177000 + 1.00032100
с u	1 26401000	1.24177000 1.09032100
$\Gamma$	0.02258800	1.57858000 -2.24012400
н	1 06150700	5 46026100 0 41901200
C	3 40368800	-1.15691300 - 1.67369200
C	1 193/0100	4 38257200 0 50926200
н	0.43282400	0.39726600 -5.82940900
C	1 97817800	2 26365500 0 80622000
н	-1 31274900	5 73531700 -3 40066200
н	0.56189300	-0.83881600 -1.35352100
C	2 49474700	-0.09723100 1.25623100
н	-2 05521800	10 11128500 0 45778100
C	2 64470000	-2 29254000 1 85149600
н	-1.92887700	7.73042400 -2.32583800
Н	2 96842800	-3 28229700 2 17054200
Н	-2 30894700	9 90795400 -1 30053200
C	-1 28184000	5 08466200 -2 50890200
C	-1 91143100	7 25432700 -1 33627400
C	0 11060600	3 41338800 0 41110600
C	-2 57353500	9 50375800 -0 30732800
Ň	0.59155500	2.09746600 0.59749500
C	-0.23320100	-2.96895600 -3.50164400
C	-1.59184000	5.87488000 -1.24461400
H	-0.27997400	4.61709700 -2.45527700

С	-2.20899200	8.03031800	-0.18907000
Ν	1.17211600	-0.58668500	1.13869200
Η	1.17384400	-3.60701900	-5.05487000
Η	-1.76334200	-2.71842600	-1.95993200
Н	-2.00957300	4.26400200	-2.65818500
Н	2.34595200	-4.12933400	-0.28067700
С	1.27113000	-1.95086700	1.50722700
C	-1.56400500	5.24930000	0.03788900
C	0 42043000	-3 93358000	-4 33125000
C	-1 25298500	-3 43319700	-2 61507300
C	-2 17359300	7 38932700	1 07574500
C	-1 22324200	3 78466900	0 16791200
C	-1 85815500	6 01440900	1 20960400
н	-3 66132300	9 65801500	-0 16413200
н	-2 09963700	6 10380800	3 37237000
н	-2 39668000	7 97221300	1 97968100
C	1 53119700	-4 87763100	-0.27726900
$C_{0}$	-0.44946100	0 39835200	0.55749900
н	0.44940100	-4 43244900	-0.84485900
C	-1 83624700	5 37102800	2 58959900
C	1 11671900	-5 24772700	1 13844700
C	0.22264500	-2 88778300	1.13044700
C	1 3/608100	-6 56511600	1.52504800
ч	1.81047700	7 20/36000	0.04402000
н	0.31837200	-7.29 <del>4</del> 30000 8 8/085500	3 80510000
C	0.31837200	-// 29519200	2 00057400
н	-0.838/7100	4 95668800	2.00037400
C	0.06/63900	-5 29810000	- <i>A</i> 27226600
C	1 5083/600	-5.29810000	2 55155800
н	0.85002900	-0.46073700	4 16683300
н	-0.70751500	-// 13885900	5 26970500
$\Gamma$	-0.70751500	6 06380000	2 02424700
C	2 25108700	2 83236800	0.06000000
с u	-2.23138700	2.83230800	0.009999000
и П	1.87872400	-3.70829300 8.28160000	4 25188500
$\Gamma$	0.11250100	-8.38100000	4.23188300
C	0.11230100	-4.07750100	3.32089300
C	0.26201100	-0.5/052500	3.42213400
C	0.50591100	-0.00128000	3.70413000
U N	-0.34629700	-3.08318000	4.2/3/9800
	-2.1108/100	1.43033900	0.20491/00
U U	-0.94282800	-3./4383800	-3.36018300
п	0.300/0200	-0.01323000	-4.93310000
H IT	-2.3/443000	-3.1243/200	-1.84930900
П	0.0/8/3/00	-2./8193300	4.40048800
H	-2.55308900	4.52988800	2.030/1400
U	-1.08251300	-2.5449/900	1.12648/00

Η	0.07125300	-6.28647600	4.78393100
Ν	-1.52932000	-1.25979900	0.74962000
Η	-2.63261300	1.30639300	3.61955200
Η	-4.11161500	2.45960900	5.70384600
С	-3.64093200	3.19169100	-0.18001700
Н	-2.29509000	0.85029200	5.33236900
Н	-3.99705700	4.21325700	-0.30759700
С	-3.07042000	0.89259500	4.54316400
Н	-3.19771000	-2.91057500	-4.57900400
Η	1.64380000	-9.02191500	2.61874300
Н	-2.36757100	-1.96587100	-5.86786700
Н	-1.21249800	-6.80520500	-3.33317300
C	-4.34414300	1.61889200	5.02781800
Н	-1.52436300	-3.33548200	3.89627100
C	-3 42317300	0.93250600	0.03269700
C	-3.27375100	-2.02976300	-5.23742900
C	-2 16908400	-3 51180300	1 05411200
C	-2 90186700	-1 43707200	0.45878300
н	-4 92119000	2 01041900	4 17037400
C	-4 36474000	2 01887100	-0 20596300
Н	-2 06386800	-4 56892400	1 29496800
$\mathbf{O}$	-3 28583900	-0 79960000	-4 35946500
C	-3 81576600	-0.41810900	0 12176200
$\mathbf{O}$	-3 49869100	-0 51507800	4 23829600
Н	-4 43527100	-2 41920500	-7 07599800
C	-3.29158600	-2.83095700	0.63170600
Н	-4.73042300	0.30799300	6.75282500
C	-4.57252600	-1.97922900	-6.07348700
C	-5.13283500	0.48298700	5.73809800
Н	-4.36758000	-0.71770300	2.45058900
Н	-4.16662300	-0.72588500	-2.37416200
Н	-5.43503300	1.88568800	-0.35770300
Н	-5.38882600	-2.52471000	-5.56615500
С	-4.84496700	-0.74706400	4.83681000
Н	-4.29662000	0.04846300	-6.89799400
Η	-6.21484500	0.67628800	5.81546600
С	-4.48513500	0.01310500	-4.72215900
С	-4.90380700	-0.46202200	-6.12862000
N	-5.39222500	-0.85350400	2.44886100
С	-5.28631600	-0.74207000	0.00861000
Н	-4.29890500	-3.21284500	0.46809400
Ν	-5.18346800	-0.57970300	-2.43573800
С	-5.91698800	-0.92181000	3.72711500
Η	-4.17174100	1.07361200	-4.70063500
Н	-4.80443200	-1.68918400	5.41310500
С	-6.06243700	-0.93898900	1.19848200

С	-5.62867200	-0.17717000	-3.68458500
С	-5.95865700	-0.77203500	-1.25135200
Η	-5.97102300	-0.25786600	-6.31016500
0	-7.13396500	-1.10989100	4.04237600
0	-6.83372400	0.06887600	-4.00274400
С	-7.46258300	-1.17247800	1.12567000
С	-7.36054900	-0.99358500	-1.32612200
Η	-8.02461400	-1.31228200	2.05147900
С	-8.09027700	-1.19383700	-0.13657100
Η	-7.85243300	-0.98372900	-2.29992400
Η	-9.17120000	-1.36571200	-0.19410300
Η	-0.96055500	-0.96358100	-3.82198200
Η	3.58823900	-4.20775300	-3.59061000
С	4.82646000	1.70056700	2.69521100
С	6.20218200	1.92969900	2.96874800
Η	6.53302400	1.99139000	4.00752300
С	7.09953000	2.06858300	1.89028600
Η	8.16304300	2.23583600	2.09687000
С	6.65836500	2.01788600	0.55268500
Η	7.35089200	2.13964500	-0.28203300
Ν	3.87417300	1.58453100	3.74415500
С	4.07932300	1.59451600	5.11202900
Η	2.87518900	1.49521300	3.49424200
0	5.19437700	1.69696600	5.71419700
С	2.76332700	1.48098000	5.92771500
0	1.59185700	1.31851800	5.02006500
С	2.75917900	0.24242600	6.86057700
Η	2.63295300	2.42740600	6.48363300
С	1.04726500	-0.07196700	5.17974600
С	2.12354500	-0.85593500	5.96326700
Η	2.12372700	0.43855600	7.74378400
Η	3.77764700	-0.00106500	7.20359100
Η	0.09585100	-0.00243900	5.74232100
Η	1.69131500	-1.68393700	6.55093900
Η	2.87701400	-1.27541200	5.27233200
Η	-1.24087100	-0.42680000	-1.67326800

Intermediate E' E'_[Co(II)(P16)]=A_[Co(II)(P16)]

 $E'_{[(R)-2a]}$ Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.304991 Hartree H⁻corr: 0.375721 Hartree SCF: -942.3325619 Hartree

S: 148.864 Cal/Mol-Kelvin H: -941.9568409 Hartree G: -942.0275709 Hartree

Η	-0.63379100	4.60360300	0.72152600
С	-1.14442600	3.62282500	0.75101500
Η	-0.34642900	3.37260100	-1.30708600
Η	-0.28326900	3.10972700	2.70633000
Η	-2.13187400	3.33228200	-1.21468700
С	-1.19942100	3.02957100	-0.69924100
Η	0.62719900	2.48999700	1.31448400
Η	-2.16710700	3.80854000	1.13557700
С	-0.39598300	2.65170300	1.70572800
0	0.73651600	1.46828000	-2.08230000
Ν	-1.11917300	1.54062000	-0.63956200
С	-1.12869600	1.27442900	1.82495800
Η	3.18056400	0.86156300	-1.53082400
Η	5.33042700	1.09744900	-0.23465000
С	-0.01004300	0.93465000	-1.20895600
Η	-2.81450400	1.61852000	0.52053700
Η	-1.84469400	1.28592100	2.67036500
С	-1.90518400	0.97883600	0.50729500
С	3.46803700	0.11081500	-0.78530500
С	4.66882100	0.24266100	-0.05023600
Η	-0.38986300	0.47575900	2.02001200
0	0.19022200	-0.36471600	-0.70864800
Η	1.42922800	-0.77494800	-2.38035500
С	1.31628400	-1.13838700	-1.34426100
С	2.60406200	-0.99453900	-0.55595000
С	5.02217300	-0.72912400	0.91883300
С	-2.41972000	-0.44923300	0.29991100
Η	5.95577100	-0.62794500	1.48525400
Η	-3.00334700	-0.07441300	-1.76371200
Η	-1.97866000	-1.15387400	2.31838000
С	-3.00493900	-0.80482300	-0.94492300
С	-2.42314500	-1.40183900	1.34833600
С	2.96543600	-1.96459700	0.41579700
С	4.16750700	-1.83418100	1.15096500
Η	0.93974900	-2.17403500	-1.33129000
С	-3.56547400	-2.08557500	-1.14188100
С	-2.98854700	-2.68611900	1.15600400
Η	2.30296500	-2.82113200	0.59663900
Η	4.43868500	-2.59049900	1.89716300
Η	-4.00709000	-2.34449200	-2.11180400
С	-3.56078500	-3.03481900	-0.08926100

H -2.98122800 -3.41182900 1.97841600 H -3.99817000 -4.02915700 -0.23873100

#### **Transition State TS4**

Temperature: 298.15 Kelvin Pressure: 1.0 Atm Imaginary Frequency: -891.097 cm-1 G_corr: 1.422137 Hartree H_corr: 1.682058 Hartree SCF: -6075.306416 Hartree S: 547.05 Cal/Mol-Kelvin H: -6073.624358 Hartree G: -6073.884279 Hartree

Η	-9.28691900	-0.91508100	-3.07055200
Η	-8.26601500	1.36918700	-2.80247800
С	-8.98131400	-0.56185100	-2.07837700
С	-8.40812200	0.72551200	-1.92649300
Η	-9.61932300	-2.38098100	-1.05521500
С	-9.17033800	-1.38707900	-0.94287700
С	-8.02848800	1.18838000	-0.64523700
С	-8.78995000	-0.92283800	0.33846000
Η	-7.57941000	2.18052700	-0.51934600
С	-8.21583700	0.36613200	0.49894900
Η	-8.94405200	-1.56026000	1.21899500
0	-0.97325800	7.76688300	0.33475700
Η	-1.11484200	7.91503900	2.90062600
С	-7.83322600	0.86947200	1.87688900
Η	-2.54567500	5.85549800	1.73067400
С	-1.18986300	6.85129300	3.17735000
Η	-2.00482900	6.72800100	3.91362000
С	-0.79641600	6.54986100	0.65466400
С	-1.46520500	5.97394900	1.93323900
Η	-8.00094500	1.95605600	1.97369300
0	-6.38288400	0.60496700	2.20057700
Ν	-0.05032000	5.61343900	-0.04059900
С	0.74235200	5.79536300	-1.20930500
Η	-2.94549800	1.85161600	-0.20110100
С	0.13918700	6.25661500	3.72179100
0	-0.91702600	4.62385800	2.25459700
Η	0.99802800	6.66082700	3.15559100
С	-5.49715200	1.65599300	1.89178900
Η	0.29523000	6.47855500	4.79140100
Η	-1.42408800	2.36833700	0.57004400
0	-5.84633200	2.73545600	1.33703700

С	-2.31598000	1.72960300	0.69956100
С	-0.51306700	-0.07496600	1.29055400
С	-0.02519400	4.74194000	3.46043600
Н	-0.03633700	4.68789600	0.41625900
Н	-0.53245600	4.22834700	4.29950200
С	1.39900400	4.64968000	-1.75654000
Н	2.98178500	4.71203300	0.45697500
С	-1.88979600	0.25314000	0.86739600
N	-4 20387200	1 35466700	2 31299400
C	-3.09533300	2 27962400	1 93389700
н	-2 25453100	-0 40007600	0.05521000
н	-0.69701700	3 93243800	-3 26660000
C	2 77476900	3 66797700	0.69095800
ч	0.01086600	1 22486600	3 22168400
и П	3 52080200	4.22480000	1 71758600
$\Gamma$	-3.32980200	3.20003000	1.71738000
	2 62806200	0.12625900	1 70492100
П	-2.03800200	-0.12023800	1./9483100
	-3.80344900	0.03/82600	2.78211900
Н	3.90335100	3.20909600	2.55859100
C	-0.6/568400	2.90931000	-2.89325500
C	3.2358/300	2.9121/100	1./5019400
H	-2.41011400	2.37808500	2.79194700
C	1.91509300	2.80224000	-0.10634200
Н	1.46852300	1.08004300	6.0403/100
Η	-4.45961500	-0.76157300	2.45728500
С	0.23324800	2.42955000	-1.86004900
Η	7.27436100	2.02692300	6.96185100
С	-1.47218700	1.84314300	-3.25669000
Η	3.60904200	1.31490900	6.98764200
Η	-2.27147600	1.81312400	-3.99630300
Η	5.79495900	1.99881700	7.96695200
С	1.78666800	0.86582800	5.00533400
С	4.10290200	1.16603500	6.01798900
С	2.71165200	1.56391700	1.57306000
С	6.35979800	1.45047700	7.19174700
Ν	1.88648100	1.49525600	0.42842800
С	-3.15035900	-0.08413100	4.09782300
С	3.30209500	0.92251400	4.87280300
Η	1.29215300	1.59569800	4.33662900
С	5.51713600	1.21969100	5.94500900
Ν	0.01218700	1.05940400	-1.59506200
Н	-2.52664500	-2.14634400	3.70143200
Н	-3.60999600	1.92905100	4.81201200
Η	1.39010300	-0.12973500	4.73009000
Н	-4.04291600	0.31151100	-1.55660400
C	-1.02854700	0.69256800	-2.48020800
-	1.02001,00		

С	3.93753900	0.73171800	3.60795000
С	-2.53351400	-1.32908600	4.43672600
С	-3.11218000	0.98117000	5.04871100
С	6.12623300	1.02888900	4.67888900
С	3.09665300	0.47770600	2.38133800
Ċ	5.36254800	0.78315500	3.51002600
Н	6.68210600	0.48817800	7.63711500
Н	7 16113800	0.66541100	2 28812900
Н	7 22074500	1 07126400	4 59752600
C	-4 39248300	-0.66201400	-1 94857000
Č0	0.91634100	-0.08505800	-0 25585900
н	-3 97973400	-1 44542800	-1 28374800
C	6 06627600	0 57801000	2 17502300
C	-3 9/890900	-0.87078600	-3 38892000
C	1 50001/00	0.61/73300	2 70/27800
C	-1.30091400	1 00004700	-2.70+27800
С Ц	-4.92431400	-1.09904700	-4.39288800
	-3.98303000	-1.11/38/00	-4.10327300
П	-3.92460400	-0.33223000	-7.26407700
	-2.30/39300	-0.84/42000	-3./4944000
П	3./3802300	1.52517000	1.42440000 5.60422000
C	-1.91/33300	-1.30132900	5.09455000
	-2.49804400	0.79824100	6.30652500
H	2.0283/300	0.238/2900	-3.64903500
H	-0.62189300	-1.24320600	-6.61335500
C	-4.56992300	-1.30316800	-5./4921600
C	2./9/38100	-0.84698600	2.01503800
H	-5.49211900	-0.68402800	-1.86098700
H	-6.54983100	-1.96439/00	-6.39297700
C	-2.18/26500	-1.05/38300	-5.1123/200
C	-5.63409900	-1.51439500	-6.81/06/00
С	-3.1930/000	-1.2826/800	-6.08553100
С	-0.72381500	-1.04549600	-5.53191100
Ν	2.02541400	-1.24330600	0.90001800
С	-1.89686700	-0.44222300	6.63780600
Η	-1.44653600	-2.46040500	5.94174300
Η	-2.50006900	1.61719000	7.03623100
Η	-0.25333100	-0.06714400	-5.31730500
Η	5.84357400	-0.42000100	1.75329400
С	-0.89511700	-1.72706400	-2.08770200
Η	-2.89241700	-1.44575400	-7.12921100
Ν	0.13833700	-1.68428400	-1.12494100
Η	4.39935600	-1.12160800	-0.84966800
Η	7.20511800	-1.37343000	-0.96517300
С	3.34830400	-2.01004800	2.69801000
Н	5.31033400	-0.47527800	-2.26081500
Н	3.97669900	-1.95951500	3.58651000

С	5.10670700	-1.37723800	-1.65683900
Η	-2.93217800	-3.48499100	0.06322700
Η	-5.27116400	-2.17414100	-7.62598000
Η	-4.13286200	-3.05147400	1.34497100
Η	-1.42337600	-0.58342500	7.61647700
С	6.38866100	-2.08795400	-1.16747600
Н	-0.13873100	-1.81001400	-4.98621800
С	2.10236600	-2.65258900	0.89875800
С	-3.38521300	-3.79939000	1.01918900
С	-1.19379400	-3.10442700	-2.46049500
С	0.48120600	-3.03809500	-0.91374700
Н	6.19151500	-2.66103500	-0.24295000
С	2.94114800	-3.12436800	1.99354400
Н	-1.94594700	-3.39150300	-3.19458600
0	-2.29037300	-3.81698600	2.05355200
Ċ	1.42698700	-3.51735300	0.01318400
0	4.40915000	-2.37048200	-2.55688700
H	-5.04404000	-5.23228900	0.73408800
C	-0.32478500	-3.91309300	-1.75610200
Н	7.23783100	-2.51556100	-3.15361600
C	-3.97017400	-5.23040200	0.98745300
C	6.72617900	-3.05600800	-2.33629000
Н	3.28824200	-3.62255100	-1.49626400
Н	-0.46093000	-4.38823000	1.32297600
Н	3.16109900	-4.17257400	2.19260300
Н	-3.43877500	-5.85368900	0.24564400
С	5.32476800	-3.51760700	-2.80947800
Н	-4.45379800	-5.37440900	3.13296500
Н	7.34824800	-3.91385400	-2.03350600
С	-2.31513400	-5.13827800	2.74666600
С	-3.69624800	-5.75588500	2.42442600
Ν	3.64405700	-4.58715500	-1.38597700
С	1.73700400	-4.99339900	0.07639800
Н	-0.22390900	-4.99760000	-1.78963400
Ν	-0.22540000	-5.38140800	1.47754600
С	4.83572500	-4.78766100	-2.05992000
Η	-2.17812300	-4.94675800	3.82663200
Η	5.29851700	-3.73946900	-3.89207800
С	2.88716600	-5.50230000	-0.60655000
С	-1.14811000	-6.04453600	2.26684400
С	0.93551200	-5.90052800	0.83603100
Н	-3.66333900	-6.85502900	2.49101900
0	5.50547700	-5.86605800	-2.11689100
0	-1.10769000	-7.26542400	2.61820300
С	3.24472300	-6.87463700	-0.50936700
С	1.29026400	-7.27402000	0.93691100

Η	4.13505000	-7.22951000	-1.03277400
С	2.44204000	-7.73635000	0.26642700
Η	0.66748200	-7.94099200	1.53608400
Η	2.71747400	-8.79418900	0.34756300
Η	-0.37879600	-1.12296500	1.59731700
Η	-8.36115500	0.32176700	2.67422500
С	2.26309900	4.83231200	-2.88413400
С	2.42189900	6.11200700	-3.48225600
Η	3.07712300	6.21419500	-4.34984800
С	1.72516700	7.21133600	-2.94003700
Η	1.83855100	8.19788200	-3.40406900
С	0.89508900	7.07642900	-1.80810200
Η	0.36196000	7.92823200	-1.38162700
Ν	2.96380100	3.69457800	-3.36751500
С	3.80850500	3.58634100	-4.45833900
Η	2.88328900	2.80648500	-2.84414600
0	4.10775300	4.50509000	-5.28380900
С	4.43184900	2.17045900	-4.60020900
0	3.88645400	1.24678700	-3.56769100
С	4.11279800	1.51753200	-5.96797600
Η	5.51893500	2.26878900	-4.42457500
С	2.96111600	0.25833800	-4.23839500
С	2.77831800	0.77280400	-5.68443600
Η	4.90988900	0.79994100	-6.23555000
Η	4.03481100	2.27597500	-6.76378200
Η	3.44754700	-0.73180800	-4.18541800
Η	2.59753500	-0.05140300	-6.39567400
Н	1.92922900	1.47814200	-5.74744700
Η	-0.04821600	0.63628200	1.99417700

### Intermediate F

 $\overline{F_{[Co(II)(P16)]}} = A_{[Co(II)(P16)]}$ 

#### F[3a]

Temperature: 298.15 Kelvin Pressure: 1.0 Atm G_corr: 0.296938 Hartree H_corr: 0.373916 Hartree SCF: -942.313589 Hartree S: 162.013 Cal/Mol-Kelvin H: -941.939673 Hartree G: -942.016651 Hartree

Cartesian Coordinates:

H -6.40386400 0.29051800 -1.57275400 H -5.30365600 -1.76473300 -2.51482000

С	-5.51880100	-0.13232500	-1.08250300
С	-4.89863500	-1.28967800	-1.61346100
С	-4.99441300	0.47479800	0.08533000
Η	-5.47408700	1.36919600	0.50010000
С	-3.75894600	-1.83537700	-0.97731200
Η	-3.27901700	-2.73255800	-1.38972000
С	-3.85577100	-0.07216400	0.72108400
С	-3.22612700	-1.23153200	0.19232000
Η	-3.43784800	0.39647800	1.61982800
Η	-1.86785600	-2.88260900	0.63085900
С	-2.00292300	-1.81565600	0.87186500
Η	-2.02895500	-1.66200700	1.96448000
0	-0.72960600	-1.19532700	0.35562400
0	-0.85663100	0.37748200	2.11422000
С	-0.29888700	-0.04168700	1.05755000
Η	0.70660300	-0.66592700	-1.24649100
Η	1.98442800	-2.12064300	0.83978900
Ν	0.80024000	0.54101400	0.46359400
Η	-0.37980300	2.91993700	1.36728800
С	1.42348200	0.03503000	-0.78835500
Η	1.44761000	1.51608600	2.22586700
С	2.87918900	-1.76406700	0.31498200
С	1.40627100	1.71718400	1.14005800
С	0.62197800	3.04523100	0.91136900
Η	1.55662300	0.89073100	-1.47772000
С	2.77350600	-0.65165700	-0.56245000
Η	4.19141200	-3.26760200	1.18755400
Η	-0.10330600	2.75174500	-1.17660600
С	4.12151700	-2.40890500	0.50909200
Η	1.14371800	3.84593000	1.47185600
С	0.48396200	3.43933900	-0.54676400
Η	2.44062000	1.81171900	0.76386000
С	3.93445400	-0.19775000	-1.23839800
С	5.27721200	-1.95120000	-0.17281500
С	1.00002700	4.56490500	-1.10656000
Η	3.86435300	0.66318500	-1.91660400
С	5.18075200	-0.84400500	-1.04805600
Н	6.24072300	-2.45217500	-0.02173400
Н	0.85559100	4.79915000	-2.16818200
Н	1.57699000	5.28904200	-0.51586600
Η	6.07084700	-0.48271600	-1.57675100

# **3.5. NMR/HPLC SPECTRAL DATA**



## ¹H NMR of **s2-a**, 600 MHz, CDCl₃







## ¹H NMR of **s2-b**, 600 MHz, CDCl₃





.17	.27	342 234 234 234 234 234 234 234 234 234	16	0	7 <u>7</u> 4900	<u>5</u> 5 5 5
201	156 156	138 137 128 128 128 124	77.3 77.1 76.9	67.2	40.00 40.00 40.00	20.4 20.4
Ŷ	$\checkmark$		$\sim$	Ī	S V P	SK

## ¹³C NMR of **s2-b**, 150 MHz, CDCl₃









¹H NMR of **s2-d**, 600 MHz, CDCl₃













Note: The spectrum contains a mixture of amide rotamers.





¹H NMR of **s2-f**, 600 MHz,  $CDCI_3$ 


230	55 20	97	62 54 11 81 83 33 81	49 17 07	5 0 2	8	228-22	~ 23
201	156. 156.	147 147	12,0,0,0,0,0,0,0,0,0,0,0,0,0,0,0,0,0,0,0	108 107 107	77.3 77.1 76.9	67.3	50.3 45.0 40.8	20.5 20.1
$\mathbf{V}$	$\searrow$	$\mathbf{Y}$	$\searrow \checkmark \lor \lor \lor$	$\langle \mathcal{V} \rangle$	$\checkmark$			$\mathbf{\nabla}$

### ¹³C NMR of **s2-f**, 150 MHz, CDCl₃





#### ¹H NMR of **s2-g**, 500 MHz, CDCl₃







# ¹H NMR of **s2-h**, 500 MHz, $CDCI_3$





Note: The spectrum contains a mixture of amide rotamers.



¹H NMR of **s2-i**, 500 MHz, CDCl₃







¹H NMR of **s2-j**, 600 MHz,  $CDCI_3$ 









# ¹⁹F NMR of **s2-j**, 564 MHz, $CDCI_3$



30	20	10	0	-10	-20	-30	-40	-50	-60	-70	-80	-90	-100	-110	-120	-130	-140	-150	-160	-170	-180	-190	-20
											f1 (p	opm)											612



¹H NMR of **s2-k**, 500 MHz,  $CDCI_3$ 







#### ¹³C NMR of **s2-k**, 125 MHz, CDCl₃



# ¹⁹F NMR of **s2-k**, 564 MHz, CDCl₃



30 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -20 20 10 -10 -20 -30 -50 -60 -70 -80 -90 Ò -40 f1 (ppm) 615



¹H NMR of **s2-I**, 600 MHz,  $CDCI_3$ 







¹⁹F NMR of **s2-I**, 470 MHz, CDCl₃



-65 -70 -75 -80 -85 -90 -95 -100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -160 -165 -170 -175 -180 -185 f1 (ppm)



¹H NMR of **s2-m**, 500 MHz,  $CDCI_3$ 









¹H NMR of **s2-n**, 500 MHz, CDCl₃







72 59	262 262 262 262 262 262 262 262 262 262	26 24	70 67	39 29 29	$\begin{array}{c} 45 \\ 445 \\ 333 \\ 322 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ 333 \\ $
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17		$\mathbf{Y}$	$\mathbf{Y}$	$\lor$	

¹H NMR of **s2-o**, 600 MHz,  $CDCI_3$ 







#### ¹³C NMR of **s2-o**, 150 MHz, CDCl₃





¹H NMR of **s2-p**, 500 MHz,  $CDCI_3$ 



.19	.96 .08	.06 51 38 92 73	32 42 95	298 N	0,720,720	2 2
201	155 151	142 136 128 127 127	110. 108 107	77.4 77.1 76.5 67.2	46.0 43.6 40.7 40.7	20.5
$\vee$		$    \vee$	$\leq \vee$		$\searrow$	$\checkmark$

# ¹³C NMR of **s2-p**, 125 MHz, CDCl₃





¹H NMR of **s2-q**, 500 MHz,  $CDCI_3$ 



Note: The spectrum contains a mixture of amide rotamers.





¹H NMR of **s2-r**, 600 MHz,  $CDCI_3$ 



	35	660 56 31 36 36 36 36	0 9 4 9	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	က္တစ္သ
201.	156. 156.	122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.6.77 122.77 122.77 123.6.77 123.6.77 123.6.77 123.6.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 123.77 177 177 177 177 177 177 177 177 177	77.4 77.1 76.9 67.3	46.2 45.4 40.9 40.0	20.6 20.2
Ý	Y				$\mathbf{\nabla}$

### ¹³C NMR of **s2-r**, 150 MHz, CDCl₃





¹H NMR of **s2-s**, 600 MHz,  $CDCI_3$ 





16	63 11 12 14 14 14 14 14 14 14 14 14 14 14 14 14	72 69 69 69 69 69 69 69 69 60 69 60 60 60 60 60 60 60 60 60 60 60 60 60	30 22	00000 <u>7</u> 0	<u>0</u> 5
201	157 157 156 156 149	136 122 127 127 127 127 127 127 127 127 127	77.3 77.1 76.9 67.1	440.22 440.25 40.08 40.08	20.9
$\nabla$	$\searrow$			VVV	$\mathbf{i}$

 $^{13}\text{C}$  NMR of **s2-s**, 150 MHz, CDCl_3



· · · · ·	· · ·		· · ·	· · · ·	· · ·	· · ·	· · · · ·	· · · · ·			· · ·		·	· · ·	· · ·	· · ·	· · ·		·				-
220	210	200	190	180	170	160	150	140	130	120	110	100	90	80	70	60	50	40	30	20	10	0	
	f1 (ppm)														~								



### ¹H NMR of **s2-t**, 600 MHz, CDCl₃



¹³C NMR of **s2-t**, 150 MHz, CDCl₃







Note: The spectrum contains a mixture of amide rotamers.





¹H NMR of **s2-v**, 600 MHz,  $CDCI_3$ 



Note: The spectrum contains a mixture of amide rotamers.




¹H NMR of **s2-w**, 600 MHz, CDCl₃























¹H NMR of **1c**, 600 MHz,  $CDCI_3$ 







¹H NMR of **1d**, 600 MHz, CDCl₃







¹H NMR of **1e**, 500 MHz,  $CDCI_3$ 







651













¹H NMR of **1i**, 600 MHz, CDCl₃













## $^{19}\mathsf{F}$ NMR of 1j, 564 MHz, $\mathsf{CDCI}_3$



-10 -70 -5 -15 -20 -25 -30 -35 -40 -45 -50 -55 -60 -65 -75 -80 -85 -90 -95 -100 -105 -110 -115 f1 (ppm)



¹H NMR of **1k**, 600 MHz, CDCl₃





## ¹⁹F NMR of **1k**, 470 MHz, CDCl₃



30 -70 -20 20 10 -10 -20 -30 -50 -60 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -40 0 f1 (ppm) 664



¹H NMR of **1I**, 500 MHz, CDCl₃





## $^{19}\mathsf{F}$ NMR of 1j, 470 MHz, CDCl_3



														· · ·									
30	20	10	0	-10	-20	-30	-40	-50	-60	-70	-80	-90	-100	-110	-120	-130	-140	-150	-160	-170	-180	-190	-20
											f1 (p	opm)											667



¹H NMR of 1m, 600 MHz, CDCl₃











¹H NMR of **1o**, 600 MHz, CDCl₃







¹H NMR of **1p**, 500 MHz,  $CDCI_3$ 






Note: The spectrum contains a mixture of amide rotamers.





¹H NMR of **1r**, 500 MHz, CDCl₃











¹H NMR of **1t**, 600 MHz, CDCl₃







Note: The spectrum contains a mixture of amide rotamers.





Note: The spectrum contains a mixture of amide rotamers.





¹H NMR of **1w**, 600 MHz, CDCl₃



Note: The spectrum contains a mixture of amide rotamers.





¹H NMR of **2a**, 600 MHz,  $CDCI_3$ 





### ¹³C NMR of (–)-2a, 150 MHz, CDCl₃





UV Spectrum Retention time = 26.626

н

(±)-2a



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	24.860	10797507	50.146
2	26.626	10734683	49.854
Total		21532190	100.000



Peak Table

190 200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	23.661	8927216	96.065
2	26.311	365687	3.935
Total		9292902	100.000

H

693

nm



¹H NMR of **2b**, 500 MHz, CDCl₃





# ¹³C NMR of **2b**, 125 MHz, CDCl₃





Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	12.723	7502487	49.865
2	15.621	7543044	50.135
Total		15045530	100.000

Н

(±)-2b

Me



mAU

Me













Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	12.859	6092003	96.774
2	15.835	203112	3.226
Total		6295115	100.000







nm

Peak Table

PDA Ch1	PDA Ch1 220nm				
Peak#	Ret. Time	Area	Area%		
1	10.866	1973567	49.657		
2	12.109	2000871	50.343		
Total		3974438	100.000		

H,

(±)-2c

^tBu

Cbz



mAU











Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	10.548	149550	3.991
2	11.761	3597176	96.009
Total		3746726	100.000



¹H NMR of **2d**, 600 MHz, CDCl₃





# ¹³C NMR of **2d**, 150 MHz, CDCl₃





mAU



Chromatogram

UV Spectrum Retention time = 17.152



### UV Spectrum Retention time = 19.322



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	17.152	6877100	49.898
2	19.322	6905106	50.102
Total		13782206	100.000



Peak Table

PDA Ch1	220nm		MAD-INCOME WITH I GIVE
Peak#	Ret. Time	Area	Area%
1	17.165	3546226	95.014
2	19.368	186098	4.986
Total		3732324	100.000





# ¹³C NMR of **2e**, 125 MHz, CDCl₃





nm

Peak Table

1	PDA Ch1	220nm		
	Peak#	Ret. Time	Area	Area%
	1	15.633	8380509	49.912
1	2	19.824	8409976	50.088
8	Total		16790485	100.000

Н

(±)-2e

MeO



mAU



UV Spectrum Retention time = 15.598









Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	15.598	3566978	95.339
2	19.816	174387	4.661
Total		3741365	100.000



¹H NMR of **2f**, 500 MHz, CDCl₃


	137.00 133.96 128.56 128.56 127.89	— 119.70	~ 108.34 ~ 107.38		77.41 77.16 76.91	-67.30	53.45	-40.50	—28.34 —25.53	—19.39
--	------------------------------------------------	----------	----------------------	--	-------------------------	--------	-------	--------	------------------	--------

## ¹³C NMR of **2f**, 125 MHz, CDCl₃





<b>D</b>			1
Laa		o b.	
FEA	K	20	
	n 1	uv	

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	22.423	2609653	49.997
2	23.699	2609949	50.003
Total		5219602	100.000

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H.

(±)-2f



<b>D</b>				
Lag			-	
PEA	ĸ	21	• •	
			~	

	C1 1	220
PDA	CnI	220nm

Н

Peak#	Ret. Time	Area	Area%
1	22.537	2046103	91.924
2	23.939	179772	8.076
Total		2225875	100.000



¹H NMR of **2g**, 500 MHz, CDCl₃





#### ¹³C NMR of **2g**, 125 MHz, CDCl₃





Peak Table

PDA Chi	220nm	72	
Peak#	Ret. Time	Area	Area%
1	28.750	3052678	50.199
2	37.870	3028462	49.801
Total		6081141	100.000

DD 1 01 1 000

 $O_2N$ 



mAU



UV Spectrum Retention time = 28.303







Peak Table

PDA Ch1	220nm	226	
Peak#	Ret. Time	Area	Area%
1	28.303	999921	97.507
2	37.215	25565	2.493
Total		1025486	100.000



¹H NMR of **2h**, 500 MHz, CDCl₃





## ¹³C NMR of **2h**, 125 MHz, CDCl₃







Н

(±)-2h

NC



#### Peak Table

PDA Ch1	220nm		2
Peak#	Ret. Time	Area	Area%
1	32.513	16992358	50.026
2	41.115	16974473	49.974
Total		33966830	100.000

	Sample Information
Sample Name	: XXW119220%-IC-0.8mL
Sample ID	: XXW119220%-IC-0.8mL
Data File	: XXW119220%-IC-0.8mL001.lcd
Method File	: XXW-20%.0.8.mL.lcm
	Chromatogram

mAU



UV Spectrum Retention time = 32.125



UV Spectrum Retention time = 40.784



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	32.125	7208016	96.800
2	40.784	238257	3.200
Total		7446274	100.000







-167.01	-156.20	- 145.62	- 136.87 130.05 130.05 128.69 128.61 126.69	77.37 77.16 76.95	-67.43	~53.82 ~52.19	-40.80	-28.28 -25.40 -19.44	
	I	1				۱ ۲	1		

# 13 C NMR of **2i**, 150 MHz, CDCl₃





mAU









UV Spectrum Retention time = 9.980



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.698	2151633	49.889
2	9.980	2161248	50.111
Total		4312881	100.000

	Sample Information
Data File	: XXW-1459-IA-20%-0.8mL002.lcd
Sample Name	: XXW-1459-IA-20%-0.8mL001
Sample ID	: XXW-1459-IA-20%-0.8mL001
Method File	: XXW-20%.0.8.mL.lcm
	Chromatogram

mAU



UV Spectrum Retention time = 8.699



UV Spectrum Retention time = 9.982



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	8.699	67412	7.190
2	9.982	870110	92.810
Total		937522	100.000



¹H NMR of **2j**, 600 MHz, CDCl₃







¹⁹F NMR of **2j**, 564 MHz,  $CDCI_3$ 



		'						· · ·			·		·	·									
30	20	10	0	-10	-20	-30	-40	-50	-60	-70	-80	-90	-100	-110	-120	-130	-140	-150	-160	-170	-180	-190	-20
											f1 (p	opm)											728



mAU



UV Spectrum Retention time = 10.238



UV Spectrum Retention time = 10.709



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	10.238	6123071	49.829
2	10.709	6165160	50.171
Total		12288231	100.000





Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	10.159	51591	2.986
2	10.623	1675959	97.014
Total		1727550	100.000

н

(–)-**2j** 

F₃C

Ċbz



¹H NMR of **2k**, 500 MHz, CDCl₃



		—156.18	— 143.00	-136.93 $-136.93$ $-136.93$ $-128.62$ $-128.62$ $-122.25$ $-113.23$ $-113.52$	77.41 77.16 76.91	-67.43	53.47	40.70	—28.25 —25.45	— 19.42
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## ¹³C NMR of **2k**, 125 MHz, CDCl₃



מירוי הלכונה לי הנגונים המירוגוליוי ב

# ¹⁹F NMR of **2k**, 470 MHz, CDCl₃



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واجتل بجنير وارتجاب وارتي وارتشدها

20 -10 -20 -30 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -20 30 10 -40 -50 -60 0 f1 (ppm)

مالاهم الأثلا الالاتجاد القال البات المتكمر أطاقا أطعاده فصلك

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Peak Table

F	PDA Ch1	220nm		
	Peak#	Ret. Time	Area	Area%
	1	14.389	5839034	50.095
	2	16.326	5816784	49.905
Γ	Total		11655818	100.000

Н

(±)-2k



nm

Peak Table

PDA Ch1 220nm

н

(–)-**2k** 

Ċbz

1 DA Chi	220mm		
Peak#	Ret. Time	Area	Area%
1	14.461	4437193	96.158
2	16.479	177291	3.842
Total		<u>4614484</u>	100.000



¹H NMR of **2I**, 500 MHz, CDCl₃



736



## 13 C NMR of **2I**, 150 MHz, CDCl₃



---137.46 ---141.03

¹⁹F NMR of **2I**, 470 MHz, CDCl₃





Peak Table

PDA Ch1	220nm	100	
Peak#	Ret. Time	Area	Area%
1	9.497	2162800	49.360
2	10.596	2218913	50.640
Total		4381712	100.000

н

(±)-2I



mAU



UV Spectrum Retention time = 9.464







Peak Table

			I Cak Table
PDA Ch1	220nm	10 10 10 10 10 10 10 10 10 10 10 10 10 1	
Peak#	Ret. Time	Area	Area%
1	9.464	1529239	96.758
2	10.567	51241	3.242
Total		1580481	100.000



¹H NMR of **2m**, 500 MHz, CDCl₃





— 156.19	139.18 136.91 131.82 128.62 128.13 128.13	—120.60			77.41 77.16 76.91	67.41		40.62	—28.14 —25.46	— 19.35
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# ¹³C NMR of **2m**, 125 MHz, CDCl₃







Peak Table

PDA Ch1 220nm							
Peak#	ŧ	Ret. Time	Area	Area%			
	1	13.037	8066614	49.572			
	2	14.076	8205989	50.428			
Tot	tal	an an an Anna Anna Anna Anna Anna Anna	16272603	100.000			

B



Peak Table

PDA Ch1 220nm								
Peak#	Ret. Time	Area	Area%					
1	13.090	3475373	95.699					
2	14.153	156179	4.301					
Total		3631552	100.000					

Br



¹H NMR of **2n**, 600 MHz, CDCl₃






mAU

Data File



UV Spectrum Retention time = 10.659







Peak Table

	PDA Ch1 220nm							
Peak#		Ret. Time	Area	Area%				
	1	10.659	1561767	50.343				
	2	11.307	1540479	49.657				
	Total		3102245	100.000				

	Sample Information
Sample Name	: XXW-1477-ID-10%-0.8mL
Sample ID	: XXW-1477-ID-10%-0.8mL
Data File	: XXW-1477-ID-10%-0.8mL.lcd
Method File	: XXW-10%.0.8.mL.lcm
	Chromatogram

mAU

Н

Br



UV Spectrum Retention time = 10.641







Peak Table

PDA Ch1	I can Iaoic		
Peak#	Ret. Time	Area	Area%
1	10.641	2415278	96.534
2	11.315	86721	3.466
Total		2501999	100.000



¹H NMR of **20**, 500 MHz, CDCl₃



749



### ¹³C NMR of **20**, 125 MHz, CDCl₃







Н

(±)-20



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	14.924	14579106	50.069
2	17.183	14538752	49.931
Total		29117857	100.000



nm

Peak Table

P	PDA Ch1 220nm						
	Peak#	Ret. Time	Area	Area%			
	1	14.911	19221549	95.285			
	2	17.193	951066	4.715			
	Total		20172615	100.000			

H.

(–)**-20** 



¹H NMR of **2p**, 500 MHz,  $CDCI_3$ 







¹³C NMR of **2p**, 125 MHz, CDCl₃











Peak Table

	I can Inoie		
PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	15.178	9796266	49.903
2	16.493	9834169	50.097
Total		19630436	100.000

Sample Information : XXW-1396-IE-10%-0.8mL : XXW-1396-IE-10%-0.8mL : XXW-1396-IE-10%-0.8mL001.lcd : XXW-10%.0.8.mL.lcm Chromatogram

mAU

Sample Name Sample ID

Data File Method File



UV Spectrum Retention time = 15.487





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mAU 150 100 50 0 200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390 mm

Peak Table

PDA Ch1 220nm							
Peak#	Ret. Time	Ret. Time Area					
	1 15.48	68087	4.463				
	2 16.80	04 1457622	95.537				
Tot	al	1525709	100.000				



¹H NMR of **2q**, 600 MHz, CDCl₃







Peak Table

PDA Chi 220nm							
Peak#	Ret. Time	Area	Area%				
1	15.148	5611931	50.011				
2	16.648	5609361	49.989				
Total		11221292	100.000				

DD 4 01 1 000

759



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	15.120	1906567	95.864
2	16.638	82261	4.136
Total		1988827	100.000

Н

(–)-2q

760



¹H NMR of 2r, 600 MHz, CDCl₃





### ¹³C NMR of **2r**, 150 MHz, $CDCI_3$





Peak Table

PD.	PDA Ch1 235nm						
P	eak#	Ret. Time	Area	Area%			
	1	17.861	7968103	50.010			
	2	19.740	7965054	49.990			
2	Total		15933157	100.000			

763



Peak Table

PDA Ch1	235nm		
Peak#	Ret. Time	Area	Area%
1	17.851	6402109	96.280
2	19.776	247338	3.720
Total		6649447	100.000

н

(–)-**2r** 



¹H NMR of **2s**, 500 MHz, CDCl₃





	<ul> <li></li></ul>	77.41 77.16 76.91	67.29	55.81		27.82 25.42	—19.75	
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### ¹³C NMR of **2s**, 125 MHz, $CDCI_3$







UV Spectrum Retention time = 15.327



UV Spectrum Retention time = 25.868

mAU



Peak Table

PDA Ch1	254nm		
Peak#	Ret. Time	Area	Area%
1	15.327	812418	50.057
2	25.868	810557	49.943
Total	5 4 4 5 4 5 4 5 4 5 4 5 4 5 4 5 4 5 4 5	1622975	100.000

	Sample Information
Sample Name	: XXW-1203-20%-IC-0.8mL
Sample ID	: XXW-1203-20%-IC-0.8mL
Data File	: XXW-1203-20%-IC-0.8mL.lcd
Method File	: XXW-20%.0.8.mL.lcm
	Chromatogram

mAU



UV Spectrum Retention time = 15.346



UV Spectrum Retention time = 25.814



Peak Table

PDA Ch1			
Peak#	Ret. Time	Area	Area%
1	15.346	125314	3.864
2	25.814	3117683	96.136
Total		3242997	100.000



¹H NMR of **2t**, 500 MHz, CDCl₃





67	34	88388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 14388 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 148888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 143888 1438888 143888 143888 143888 143888 1438888 1438888 1438888 1438888 1438888 1438888 14388888 1438888 14388888 1438888 1438888 14388888 1438888 143888888 14388888 143888888 14388888888 14388888888 1438888888888	- 0 -	<b>0</b>	<del>~</del>	~	2 0 2
2	.0		4 - 0	č.	.5	2	0 7 0
0			77 76	67	55	41	25 25 19

## ¹³C NMR of **2t**, 125 MHz, CDCl₃





Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	7.469	2335725	49.442
2	7.935	2388422	50.558
Total		4724148	100.000

Br







UV Spectrum Retention time = 7.927



Peak Table

PDA Ch1	220nm		and the second
Peak#	Ret. Time	Area	Area%
1	7.463	47650	2.230
2	7.927	2088645	97.770
Total		2136294	100.000



¹H NMR of **2u**, 600 MHz, CDCl₃





# ¹³C NMR of 2u, 150 MHz, CDCl₃









Peak Table

PDA Ch1 220nm						
Peak#	Ret. Time	Area	Area%			
1	11.785	3126871	50.139			
2	17.464	3109576	49.861			
Total		6236447	100.000			

- - -



mAU

N∕∕



UV Spectrum Retention time = 11.783







Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	11.783	259636	4.946
2	17.357	4989752	95.054
Total		5249388	100.000



¹H NMR of **2v**, 500 MHz, CDCl₃





### ¹³C NMR of 2v, 125 MHz, CDCl₃





Peak Table

PDA Ch1	220nm	0	
Peak#	Ret. Time	Area	Area%
1	15.695	22384674	50.145
2	21.725	22255160	49.855
Total		44639833	100.000





Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	15.655	240903	3.011
2	21.942	7760545	96.989
Total		8001448	100.000

(-)-2v



¹H NMR of **2w**, 500 MHz, CDCl₃







#### ¹³C NMR of **2w**, 125 MHz, CDCl₃




Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	13.079	20695279	50.076
2	14.947	20632340	49.924
Total		41327619	100.000





Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	13.124	385480	3.610
2	14.985	10292731	96.390
Total		10678211	100.000

Н

(-)-**2w** 

### **CHAPTER 4**

## DESIGN AND SYNTHESIS OF A NOVEL *D*₂-SYMMETRIC CHIRAL PORPHYRIN FOR COBALT(II)-BASED METALLORADICAL CATALYSIS 4.1. INTRODUCTION

The past decades have witnessed renaissance of radical reactions in view of their rich reactivities and attractive characteristics.¹ Despite recent advancements, the enduring issues associated with controlling reactivity and stereoselectivity of radical species remain a formidable challenge and largely unaddressed.² Among considerable efforts in surmounting this challenge,³ metalloradical catalysis (MRC) involving catalytic generation and regulation of metal-stabilized organic radicals has emerged as a conceptually different

¹ For selected books, see: (a) Zard, S. Z. *Radical Reactions in Organic Synthesis*; Oxford University Press, 2003. (b) Chatgilialoglu, C.; Studer, A., *Encyclopedia of Radicals in Chemistry, Biology, and Materials*; John Wiley & Sons, 2012. For selected reviews, see: (c) Zard, S. Z. *Chem. Soc. Rev.* **2008**, *37*, 1603–1618. (d) Narayanam, J. M.; Stephenson, C. R. *Chem. Soc. Rev.* **2011**, *40*, 102–113. (e) Quiclet-Sire, B.; Zard, S. Z. *Pure Appl. Chem.* **2011**, *83*, 519–551. (f) Prier, C. K.; Rankic, D. A.; MacMillan, D. W. *Chem. Rev.* **2013**, *113*, 5322–5363. (g) Studer, A.; Curran, D. P. *Angew. Chem., Int. Ed.* **2016**, *55*, 58–102.

² For selected reviews, see: (a) Bar, G.; Parsons, A. F. *Chem. Soc. Rev.* **2003**, *32*, 251–263. (b) Sibi, M. P.; Manyem, S.; Zimmerman, J. *Chem. Rev.* **2003**, *103*, 3263–3295. (c) Brimioulle, R.; Lenhart, D.; Maturi, M. M.; Bach, T. *Angew. Chem., Int. Ed.* **2015**, *54*, 3872–3890.

³ For selected examples on approaches to controlling radical reactivity and stereoselectivity, see: (a) Du, J. N.; Skubi, K. L.; Schultz, D. M.; Yoon, T. P. *Science* **2014**, *344*, 392–396. (b) Huo, H.; Shen, X.; Wang, C.; Zhang, L.; Röse, P.; Chen, L.-A.; Harms, K.; Marsch, M.; Hilt, G.; Meggers, E. *Nature* **2014**, *515*, 100–103. (c) Kainz, Q. M.; Matier, C. D.; Bartoszewicz, A.; Zultanski, S. L.; Peters, J. C.; Fu, G. C. *Science* **2016**, *351*, 681–684. (d) Zhang, W.; Wang, F.; McCann, S. D.; Wang, D. H.; Chen, P. H.; Stahl, S. S.; Liu, G. S. *Science* **2016**, *353*, 1014–1018. (e) Kern, N.; Plesniak, M. P.; McDouall, J. J. W.; Procter, D. J. *Nat. Chem.* **2017**, *9*, 1198–1204. (f) Morrill, C.; Jensen, C.; Just-Baringo, X.; Grogan, G.; Turner, N. J.; Procter, D. J. *Angew. Chem., Int. Ed.* **2018**, *57*, 3692–3696. (g) Proctor, R. S. J.; Davis, H. J.; Phipps, R. J. *Science* **2018**, *360*, 419–422. (h) Biegasiewicz, K. F.; Cooper, S. J.; Gao, X.; Oblinsky, D. G.; Kim, J. H.; Garfinkle, S. E.; Joyce, L. A.; Sandoval, B. A.; Scholes, G. D.; Hyster, T. K. *Science* **2019**, *364*, 1166–1169. (i) Huang, H.-M.; McDouall, J. J. W.; Procter, D. J. *Nat. Catal.* **2019**, *2*, 211–218. (j) Nakafuku, K. M.; Zhang, Z.; Wappes, E. A.; Stateman, L. M.; Chen, A. D.; Nagib, D. A. *Nat. Chem.* **2020**, *12*, 697–704.

approach for the development of stereoselective radical transformations.^{4,5,6} To this end, we have demonstrated that Co(II) complexes of porphyrins [Co(Por)] as stable 15emetalloradicals exhibit unique capability of homolytically activating diazo compounds to generate the fundamentally new  $\alpha$ -Co(III)-alkyl radicals⁷ as key intermediates for catalytic reactions involving carbon-centered radicals. Through the introduction of modularlydesigned  $D_2$ -symmetric chiral amidoporphyrins as the supporting ligand system whose environments can be fine-tuned, the initially-formed Co-stabilized C-centered radicals, which are situated inside the confined chiral ligand environment, can be precisely governed to perform subsequent homolytic radical reactions, such as radical addition and H-atom abstraction as well as radical substitution, giving rise to the discovery of new catalytic

⁴ For selected reviews and highlights on Co(II)-based MRC, see: (a) Lu, H. J.; Zhang, X. P. *Chem. Soc. Rev.* **2011**, *40*, 1899–1909. (b) Pellissier, H.; Clavier, H. *Chem. Rev.* **2014**, *114*, 2775–2823. (c) Demarteau, J.; Debuigne, A.; Detrembleur, C. *Chem. Rev.* **2019**, *119*, 6906–6955. (d) Huang, H.-M.; Garduño-Castro, M. H.; Morrill, C.; Procter, D. J. *Chem. Soc. Rev.* **2019**, *48*, 4626–4638. (e) Singh, R.; Mukherjee, A. *ACS Catal.* **2019**, *9*, 3604–3617.

⁵ For selected examples of Ti(III)-based radical processes, see: (a) Nugent, W. A.; RajanBabu, T. V. J. Am. Chem. Soc. 1988, 110, 8561-8562. (b) Rajanbabu, T. V.; Nugent, W. A. J. Am. Chem. Soc. 1994, 116, 986-997. (c) Gansäuer, A.; Hildebrandt, S.; Michelmann, A.; Dahmen, T.; von Laufenberg, D.; Kube, C.; Fianu, G. D.; Flowers II, R. A. Angew. Chem., Int. Ed. 2015, 54, 7003-7006. (d) Funken, N.; Mühlhaus, F.; Gansäuer, A. Angew. Chem., Int. Ed. 2016, 55, 12030–12034. (e) Hao, W.; Wu, X.; Sun, J. Z.; Siu, J. C.; MacMillan, S. N.; Lin, S. J. Am. Chem. Soc. 2017, 139, 12141-12144. (f) Yao, C. B.; Dahmen, T.; Gansäuer, A.; Norton, J. Science 2019, 364, 764–767. (g) Ye, K. Y.; McCallum, T.; Lin, S. J. Am. Chem. Soc. 2019, 141, 9548–9554. ⁶ For selected examples of metalloradical-mediated radical processes, see: (a) Wayland, B. B.; Poszmik, G.; Mukerjee, S. L.; Fryd, M. J. Am. Chem. Soc. 1994, 116, 7943-7944. (b) Zhang, X.-X.; Wayland, B. B. J. Am. Chem. Soc. 1994, 116, 7897-7898. (c) Chan, K. S.; Li, X. Z.; Dzik, W. I.; de Bruin, B. J. Am. Chem. Soc. 2008, 130, 2051–2061. (d) Chan, Y. W.; Chan, K. S. J. Am. Chem. Soc. 2010, 132, 6920–6922. (e) Li, G.; Han, A.; Pulling, M. E.; Estes, D. P.; Norton, J. R. J. Am. Chem. Soc. 2012, 134, 14662–14665. (f) Kuo, J. L.; Hartung, J.; Han, A.; Norton, J. R. J. Am. Chem. Soc. 2015, 137, 1036-1039. (g) Roy, S.; Khatua, H.; Das, S. K.; Chattopadhyay, B. Angew. Chem., Int. Ed. 2019, 58, 11439-11443. (h) Das, S. K.; Roy, S.; Khatua, H.; Chattopadhyay, B. J. Am. Chem. Soc. 2020, 142, 16211-16217. (i) Zhang, Z.; Gevorgyan, V. Org. Lett. 2020, 22, 8500-8504.

⁷ (a) Dzik, W. I.; Xu, X.; Zhang, X. P.; Reek, J. N. H.; de Bruin, B. *J. Am. Chem. Soc.* **2010**, *132*, 10891–10902. (b) Belof, J. L.; Cioce, C. R.; Xu, X.; Zhang, X. P.; Space, B.; Woodcock, H. L. Organometallics **2011**, *30*, 2739–2746. (c) Lu, H.; Dzik, W. I.; Xu, X.; Wojtas, L.; de Bruin, B.; Zhang, X. P. J. Am. Chem. Soc. **2011**, *133*, 8518–8521.

systems for stereoselective radical transformations.⁸ The key to the success of controlling reactivity and stereoselectivity in these radical- mediated asymmetric catalytic systems lies in the catalyst





⁸ (a) Chen, Y.; Fields, K. B.; Zhang, X. P. J. Am. Chem. Soc. 2004, 126, 14718–14719. (b) Caselli, A.; Gallo, E.; Ragaini, F.; Ricatto, F.; Abbiati, G.; Cenini, S. Inorg. Chim. Acta. 2006, 359, 2924–2932. (c) Chen, Y.; Ruppel, J. V.; Zhang, X. P. J. Am. Chem. Soc. 2007, 129, 12074–12075. (d) Fantauzzi, S.; Gallo, E.; Rose, E.; Raoul, N.; Caselli, A.; Issa, S.; Ragaini, F.; Cenini, S. Organometallics 2008, 27, 6143-6151. (e) Zhu, S.; Ruppel, J. V.; Lu, H.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2008, 130, 5042–5043. (f) Zhu, S.; Xu, X.; Perman, J. A.; Zhang, X. P. J. Am. Chem. Soc. 2010, 132, 12796-12799. (g) Xu, X.; Lu, H. J.; Ruppel, J. V.; Cui, X.; de Mesa, S. L.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2011, 133, 15292-15295. (h) Paul, N. D.; Mandal, S.; Otte, M.; Cui, X.; Zhang, X. P.; de Bruin, B. J. Am. Chem. Soc. 2014, 136, 1090-1096. (i) Reddy, A. R.; Hao, F.; Wu, K.; Zhou, C. Y.; Che, C. M. Angew. Chem., Int. Ed. 2016, 55, 1810-1815. (j) Chirila, A.; Gopal Das, B.; Paul, N. D.; de Bruin, B. ChemCatchem 2017, 9, 1413–1421. (k) Wang, Y.; Wen, X.; Cui, X.; Wojtas, L.; Zhang, X. P. J. Am. Chem. Soc. 2017, 139, 1049-1052. (1) Wang, Y.; Wen, X.; Cui, X.; Zhang, X. P. J. Am. Chem. Soc. 2018, 140, 4792–4796. (m) Wen, X.; Wang, Y.; Zhang, X. P. Chem. Sci. 2018, 9, 5082-5086. (n) Lee, W.-C. C.; Wang, D.-S.; Zhang, C.; Xie, J.; Li, B.; Zhang, X. P. Chem 2021, 7, 1588-1601. (o) Wang, X.; Ke, J.; Zhu, Y.; Deb, A.; Xu, Y.; Zhang, X. P. J. Am. Chem. Soc. 2021, 143, 11121-11129. (p) Xie, J.; Xu, P.; Zhu, Y.; Wang, J.; Lee, W.-C. C.; Zhang, X. P. J. Am. Chem. Soc. 2021, 143, 11670-11678. (q) Zhang, C.; Wang, D.-S.; Lee, W.-C. C.; McKillop, A. M.; Zhang, X. P. J. Am. Chem. Soc. 2021, 143, 11130–11140.

design and development that maximize noncovalent attractive interactions through finetuning of the steric, electronic, and chiral environments of  $D_2$ -symmetric chiral amidoporphyrin ligands.^{8n-8q} While existing  $D_2$ -symmetric chiral amidoporphyrin ligands have been successfully applied in a number of catalytic systems, there is an emerging need to design novel [Co( $D_2$ -Por*)] complexes that may address unsolved limitations of the existing systems and also lead to the discovery of new stereoselective radical transformations.

Since the first introduction in 2004,^{8a} Co(II) complexes of  $D_2$ -symmetric chiral amidoporphyrins  $[Co(D_2-Por^*)]$  have emerged as a new class of catalysts for controlling reactivity and stereoselectivity in various radical catalytic systems based on Co(II)-MRC. Previous studies have revealed that multiple noncovalent attractive interactions, including H-bonding and  $\pi$ -stacking interactions between the catalyst and the substrates, could be exploited for effective stereochemical control.^{8i, 8n-8q} In 2011, our group reported the first construction of [Co(3,5-Di'Bu-QingPhyrin)] based on the Co(II)-catalyzed asymmetric cyclopropanation of alkenes.^{8g} The incorporation of phenyl moieties in the chiral amide units significantly enhanced the catalytic reactivity and stereoselectivity as a result of potential  $\pi$ -stacking interactions,^{8n-q} enabling the development of a new Co(II)-based catalytic system for asymmetric intramolecular cyclopropanation of diazoacetates. Inspired by the positive outcome of introducing phenyl units in the chiral amides, we sought to develop novel porphyrin catalysts bearing diphenyl moieties that might further rigidify the conformation of radical intermediates through additional  $\pi$ -stacking interactions, allowing for effective control of asymmetric induction (Scheme 4.1). We herein report the design and synthesis of the first example of  $[Co(D_2-Por^*)]$  complex bearing chiral amides with

diphenyl moieties, ([Co(XPhyrin)]), which could be essentially derived from the corresponding optically pure diphenyl-cyclopropanecarboxamide as a chiral building block. As an initial demonstration, the Co(II) complex of 3,5-Di'Bu-XPhyrin has proven to be an effective catalyst for asymmetric olefin cyclopropanation with in situ-generated  $\alpha$ -quinolinyldiazomethane, affording the valuable chiral quinolinyl cyclopropanes in high yield with excellent stereoselectivity.

### **4.2. RESULTS AND DISCUSSION**

### 4.2.1. Synthetic Approach to Construct [Co(XPhyrin)]

For construction of diphenyl porphyrin catalyst [Co(P18)], our first endeavor is to identify the optimal catalyst for highly asymmetric radical cyclopropanation of 1,1,diphenylethylene (1) with tert-butyl diazoacetate (t-BDA) (2) (Figure 4.1). It was found that first generation chiral metalloradical catalyst [Co(P3)] (P3 = 3,5-Di'Bu-ChenPhyrin)]^{8a} could catalyze the formation of *tert*-butyl 2,2-diphenylcyclopropane-1carboxylate (3) in 43% yield with 16% ee (Figure 4.1, entry 1). With the employment of [Co(P17)] (P17 = 3,5-Di'Bu-RuppelPhyrin)]^{8e} bearing chiral acyclic amide units that possess intramolecular H-bonding interaction, the catalytic cyclopropanation reaction afforded **3** in higher yield (80%) but with diminished enantioselectivity (12%) in favor of the opposite enantiomer (entry 2). To improve the enantioselectivity of the catalytic process, we then applied metalloradical catalyst [Co(P15)] (P15 = 3,5-Di'Bu-ZhuPhyrin),^{8e} which is devised to achieve conformational rigidity through intramolecular H-bonding interaction in (S)-(-)-2-tetrahydrofurancarboxamide units. To our delight, the catalytic reaction by [Co(P15)] indeed delivered the desired chiral diphenyl cyclopropanecarboxylate 3 in excellent yield (98%) with significantly enhanced

enantioselectivity (94% ee) (entry 3). Using [Co(P15)], we then examined the effect of solvent and temperature on the catalytic reaction (entries 4–8). Gratifyingly, we revealed that [Co(P15)] could effectively catalyze the asymmetric radical cyclopropanation in PhCF₃ even at 4 °C, delivering the corresponding diphenyl cyclopropanecarboxylate **3** in 81% yield with 96% ee (entry 8).

Figure 4.1. Optimization of Asymmetric Radical Cyclopropanation of 1,1-Diphenylethylene with *t*-BDA^{*a*}

Ph Ph

I	Ph Ph +	H CO₂ ^t Bu −	[Co(Por)] (2 mol %) solvent; temp; 24 h		Du
	1	2		<b>3</b>	Bu
entry	[Co(Por)]	temp (°C)	solvent	yield (%)	ee (%)
1	[Co( <b>P3</b> )]	22	PhCF ₃	43	16
2	[Co( <b>P17</b> )]	22	PhCF ₃	80	-12
3	[Co( <b>P15</b> )]	22	PhCF ₃	98	94
4	[Co( <b>P15</b> )]	22	PhCl	99	93
5	[Co( <b>P15</b> )]	22	PhH	99	92
6	[Co( <b>P15</b> )]	22	PhCH ₃	43	93
7	[Co( <b>P15</b> )]	22	$C_6F_6$	60	92
8 ^b	[Co( <b>P15</b> )]	4	PhCF ₃	81	96
( <b>P3</b> = 3	[Co( <b>P3</b> )] 5-Di ^t Bu-ChenPhyrin)	( <b>P17</b> = 3 !	[Co( <b>P17</b> )] 5-Di ^t Bu-RuppelPhyrin)	[C ( <b>P15 =</b> 3 5-1	o( <b>P15</b> )] Di ^t Bu-ZhuPhvrin)

^aCarried out with **1** (0.10 mmol) and **2** (0.12 mmol) by [Co(Por)] (2 mol %) in solvent (1.0 mL) for 24 h; Isolated yields; Enantiomeric excess (ee) determined by chiral HPLC. ^bReaction performed in 10.0 mmol scale for 72 h.

### Scheme 4.2. Synthesis of [Co(P18)]: Asymmetric Synthesis of Diphenyl Cyclopropane



### **Carboxamide 4**

Following the previously established procedures,^{8g} the obtained carboxylate **3** could be readily transformed into the corresponding cyclopropanecarboxamide **4** over three steps in 78% yield with retention of the original enantiomeric purity (Scheme 4.2). Further recrystallization of chiral amide **4** from CH₃Cl/pentane led to an improved enantiopurity (>99% ee). The absolute configuration of the newly-generated stereogenic center was established as (*S*) by X-ray crystallography.

With the optically pure chiral amide **4** in hand, it could then be utilized as a chiral building block for the construction of new metalloradical catalyst [Co(P18)] (P18 = 3,5-Di'Bu-XPhyrin) via a two-step procedure (Scheme 4.3). According to our previously reported procedures,^{8a} the free-base porphyrin  $[H_2(P18)]$ ) was prepared in 79% yield by the Pd-catalyzed quadruple amidation reaction of tetrabromoporphyrin **6** with chiral amide **4**.

Lastly, metalation of free-base  $[H_2(P18)]$  was achieved using cobalt(II) chloride in the presence of sterically-hindered base 2,6-lutidine, resulting in the formation of [Co(P18)] in 85% yield.



### Scheme 4.3. Synthesis of [H₂(P18)] and [Co(P18)]

4.2.2. Asymmetric Radical Cyclopropanation Catalyzed by [Co(XPhyrin)]

Chiral heteroaryl cyclopropanes represent a class of prevalent structural motifs in biologically important compounds. However, there remained limited methods for their stereoselective synthesis. In an effort to explore a general synthesis to access these molecules as well as to examine the catalytic performance of the new metalloradical catalyst, [Co(P18)], we decided to study the asymmetric cyclopropanation of styrene with in situ-generated  $\alpha$ -quinolinyldiazomethane as a model reaction (Scheme 4.4).

### Scheme 4.4. Ligand Effect on Co(II)-Catalyzed Asymmetric Olefin Cyclopropanation



with 3-Quinolinyldiazomethanes^a

^aCarried out with **7** (0.10 mmol), **8** (0.15 mmol), DMAP (0.05 mmol) and  $Cs_2CO_3$  (0.20 mmol) using [Co(Por)] (2 mol %) in toluene (1.0 mL) at 22 °C for 16 h; Isolated yields; Enantiomeric excess (ee) determined by chiral HPLC.

Gratifyingly, [Co(P18)] was shown to be an effective catalyst for the cyclopropanation reaction in the presence of DMAP (4-(dimethylamino)pyridine) as an additive, furnishing the cyclopropane product in high yield (89%) with excellent diastereoselectivity (97:3 dr) and enantioselectivity (86% ee). As comparison, the catalytic cyclopropanation reaction was also conducted under the same reaction conditions using [Co(P6)] (P6 = 3,5-Di'Bu-QingPhyrin)^{8g} as the catalyst, which features one phenyl moiety in the chiral amides. While the reaction with [Co(P6)] could afford the desired cyclopropane in a similarly high diastereoselectivity (96:4 dr), both the yield (53%) and enantioselectivity (62% ee) significantly decreased. It is noted that the sense of asymmetric induction also switched in favor of the opposite enantiomer with the use of [Co(P6)]. Evidently, the difference in catalytic performance between [Co(P18)] and [Co(P6)] implies a positive effect of installing two phenyl groups in the chiral amides, signifying the importance of judicious tuning of ligand environment in controlling reactivity and stereoselectivity of the Co(II)-based metalloradical system.

### **4.3. CONCLUSIONS AND OUTLOOK**

In summary, we have introduced a novel  $D_2$ -symmetric chiral amidoporphyrin XPhyrin containing chiral amides with diphenyl moieties for the first time. The newly synthesized metalloradical catalyst, which is derived from optically pure diphenyl cyclopropanecarboxamide, can be effectively constructed based on highly effective Co(II)catalyzed asymmetric cyclopropanation of diphenyl ethylene with diazoacetate. The optically pure chiral amide was utilized as a chiral building block for Pd-catalyzed amidation and base-assisted metalation, enabling the synthesis of Co(II) complex of 3,5-Di'Bu-XPhyrin in high overall yields. Furthermore, the Co(II) complex of 3,5-Di'Bu-XPhyrin has proven to be an effective catalyst for asymmetric olefin cyclopropanation with in situ-generated  $\alpha$ -quinolinyldiazomethane, affording the valuable chiral quinolinyl cyclopropanes in high yield with high stereoselectivity. Considering the modular design of  $D_2$ -symmetric chiral amidoporphyrins, we envision the possibilities of rendering chiral diphenyl cyclopropanecarboxamide as a new class of synthon for the synthesis of other chiral catalyst derivatives. Inspired by the successful demonstration of Co(II) complex of 3,5-Di'Bu-XPhyrin for asymmetric radical cyclopropanation, we are currently working to explore new applications in asymmetric radical processes.

#### **4.4. EXPERIMENTAL SECTION**

### 4.4.1. General Considerations

All catalytic reactions were performed under nitrogen in oven-dried glassware following standard Schlenk techniques Gas tight syringes were used to transfer liquid reagents and solvents in catalytic reactions. Solvent was freshly distilled/degassed prior to use unless otherwise noted. Thin layer chromatography was performed on Merck TLC plates (silica gel 60 F254). Flash column chromatography was performed with ICN silica gel (60 Å, 230-400 mesh, 32-63 μm). ¹H NMR spectra were acquired using Bruker 500 (500 MHz) or Varian INOVA 600 (600 MHz) spectrometer. Chemical shifts were internally referenced to residual solvent peak (CHCl₃  $\delta$  = 7.26 ppm). Data were reported as follows: chemical shift (ppm), integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, p = pentet, hept = heptet, br = broad, m = multiplet), and coupling constants J (Hz). ¹³C NMR spectra were acquired using Bruker 500 (125 MHz) or INOVA 600 (150 MHz) spectrometer with complete proton decoupling. Chemical shifts were reported in ppm with residual solvent peak (CDCl₃ $\delta$  = 77.16 ppm) as the internal standard. Infrared spectra were measured with a Nicolet Avatar 320 spectrometer with a Smart Miracle accessory. Optical rotations were measured on a Rudolph Research Analytical AUTOPOL® IV digital polarimeter. HPLC measurements were carried out on a Shimadzu HPLC system with Chiralcel IA, AD-H and ID columns. High-resolution mass spectrometry (DART and ESI) was performed at the Mass Spectrometry Facility, Boston College, Chestnut Hill, MA. The X-ray diffraction data were collected using Bruker-AXS SMART-APEXII CCD diffractometer. Anhydrous cobalt(II) chloride, palladium(II) acetate, and 9-dimethyl-4,5bis(diphenylphosphino) xanthenes (Xantphos) were purchased from Strem Chemical Co.. Anhydrous cesium carbonate was purchased from Oakwood Chemical.

### 4.4.2. Synthesis and Characterization of Catalysts

4.4.2.1. Experimental Procedure for Preparation of Diphenylcyclopropanecarboxylate 3



To a 100 mL oven-dried Schlenk tube was charged with [Co(P15)] (P15 = 3,5-Di'Bu-ZhuPhyrin) (2 mol %). The Schlenk tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen 3 times. Under nitrogen atmosphere, diphenylethylene 1 (10 mmol, 1.0 equiv), *t*-BDA 2 (1.2 equiv), and anhydrous PhCF₃ (0.1 M) were added. The Schlenk tube was then purged with nitrogen for 1 min and sealed with the Teflon screw cap. The reaction mixture was stirred at 4 °C for 72 h. Following completion of the reaction, the reaction mixture was concentrated under vacuum and purified by flash column chromatography to give the title compound. Yield: 81%. R_f = 0.30 (1:1 Hexanes/DCM).  $[\alpha]_D^{20} = (+)-150.2^{\circ}$  (*c* = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.36 (d, *J* = 6.8 Hz, 2H), 7.29 – 7.13 (m, 8H), 2.48 – 2.44 (m, 1H), 2.13 – 2.09 (m, 1H), 1.50 (dd, *J* = 8.8, 5.6 Hz, 1H), 1.21 (s, 9H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  169.78, 145.35, 140.42, 130.18, 128.51, 128.31, 127.70, 126.96, 126.50, 80.40, 77.18, 39.57, 30.21, 27.95, 20.15. IR (neat, cm⁻¹): 3058.40, 2976.40, 1726.91, 1601.45, 1447.15, 1366.87, 1149.49, 847.25. HPLC analysis: ee = 96%. (99.8% hexanes: 0.2% isopropanol, 1.0 mL/min): *t_{mator}* = 7.04 min,  $t_{minor} = 10.30 \text{ min. HRMS (DART) ([M+H]^+) Calcd. for C_{20}H_{23}O_2^+: 295.16926, Found: 295.16906.$ 

# 4.4.2.2. Experimental Procedure for Preparation of Diphenylcyclopropanecarboxamide 4



After the asymmetric cyclopropanation reaction, the obtained carboxylate further underwent 3 steps^{8g} to afford the corresponding optically pure amide. To a 100-mL roundbottom flask was charged with tert-butyl (S)-2,2-diphenylcyclopropane-1-carboxylate 3 (10 mmol, 1.0 equiv), trifluoroacetic acid (10.0 equiv) and DCM (40 mL). The reaction was stirred at room temperature overnight. After evaporation of the solvent, the free acid was afforded and used directly for next step without purification. The solution of the free acid and oxalyl chloride (5.0 equiv) in DCM (45 mL) was cooled to 0 °C, then 2-3 drops of DMF was added. The reaction mixture was stirred for 2 h at 0 °C, then 5 h at room temperature. After removal of oxalyl chloride and DCM under reduced pressure, the residue was then dissolved in DCM again and dropwise added to NH₃/MeOH solution (10.0 equiv). After the reaction was completed, the resulting mixture was concentrated under reduced pressure and purified by flash column chromatography to afford with 78% overall yield. The product was recrystallized from CHCl₃/pentane to further improve the enantiopurity to >99% ee.  $R_f = 0.30$  (5:1 Hexanes/EtOAc).  $[\alpha]_D^{20} = (+)-151.0^{\circ}$  (c = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃)  $\delta$  7.39 (d, J = 7.8 Hz, 2H), 7.35 – 7.16 (m, 8H), 5.62 (d, J = 33.8 Hz, 2H), 2.36 – 2.29 (m, 1H), 2.16 – 2.09 (m, 1H), 1.62 (dd, J = 7.8, 4.4 Hz,

1H). ¹³C NMR (125 MHz, CDCl₃) δ 171.96, 145.29, 140.06, 129.95, 128.55, 128.52,



127.45, 127.13, 126.53, 39.24, 30.65, 20.14. IR (neat, cm⁻¹): 3457.25, 3314.49, 3182.89, 1656.24, 1599.75, 1414.54, 1305.22, 1026.86. HPLC analysis: ee > 99%. (95% hexanes: 5% isopropanol, 0.8 mL/min):  $t_{major} = 23.84$  min,  $t_{minor}$ 

= 17.43 min. HRMS (DART) ( $[M+H]^+$ ) Calcd. for C₁₆H₁₆NO⁺: 238.12264, Found: 238.12247. The absolute configuration of the amide was assigned as (*S*) by X-ray crystallography.

### 4.4.2.3. Experimental Procedure for Preparation of [H₂(P18)]



[H₂(**P18**)] was synthesized according to our previous reported procedure^{8a} with 79% yield. The 5,15-bis(2,6-dibromophenyl)-10,20-bis(3,5-di-*t*-butylphenyl)porphyrin (0.20 mmol), (*S*)-2,2-diphenylcyclopropane-1-carboxamide (3.20 mmol), Pd(OAc)₂ (0.08 mmol), Xantphos (0.16 mmol), and Cs₂CO₃ (3.20 mmol) were placed in an oven-dried, resealable Schlenk tube. The tube was capped with a Teflon screwcap, evacuated, and backfilled with nitrogen. The screwcap was replaced with a rubber septum, and dioxane (10 mL) was added via a gastight syringe. The tube was purged with nitrogen for 2 minutes, and then the septum was replaced with the Teflon screwcap. The tube was sealed and stirred at 80 °C for 72 h. The resulting mixture was cooled down to room temperature, diluted in ethyl acetate, filtrated through a silica pad and concentrated under vacuum. The pure compound was obtained as a purple solid after purification by flash column chromatography (eluent: 4:1 to 3:1 Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 9.12 (d, J = 4.7 Hz, 4H), 8.92 (d, J = 4.7 Hz, 4H), 8.25 (d, J = 8.4 Hz, 4H), 7.92 (d, J = 1.8 Hz, 4H), 7.80 – 7.79 (m, 2H), 7.66 (t, J = 8.4 Hz, 2H), 7.13 – 7.12 (m, 12H), 6.97 – 6.87 (m, 8H), 6.69 (brs, 4H), 6.46 – 6.43 (m, 4H), 5.96 (brs, 8H), 5.77 (brs, 8H), 1.64 (brs, 4H), 1.30 (s, 36H), 0.78 (brs, 8H), -2.11 (s, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 166.93, 149.52, 144.37, 139.74, 139.30, 139.06, 130.72, 129.94, 129.50, 128.32, 127.52, 127.07, 126.89, 125.87, 123.62, 122.04, 120.39, 116.68, 108.22, 39.54, 35.04, 31.61, 31.42, 18.86. HRMS (ESI) ([M+H]⁺) Calcd. for C₁₂₄H₁₁₅N₈O₄⁺: 238.12264, Found: 1779.90295. UV-Vis (CHCl₃), λ_{max} nm (log ε): 425 (5.61), 516 (4.29), 551 (3.85), 590 (3.81), 646 (3.85).

4.4.2.4. Experimental Procedure for Preparation of [Co(P18)]



[Co(**P18**)] was synthesized according to our previous reported procedure^{8a} with 85% yield. Free base porphyrin [H₂(**P18**)] and anhydrous CoCl₂ (8.0 equiv) were placed in an ovendried Schlenk tube. The tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen. Under nitrogen atmosphere, 2,6-Lutidine (5.0 equiv) and anhydrous dioxane

(0.02 M) were added. The tube was purged with nitrogen for 1 min and sealed with Teflon screw cap. The reaction mixture was stirred at 80 °C for 16 h prior to being cooled down to room temperature. The reaction mixture was diluted with EtOAc, washed with water 2 times and then concentrated under vacuum. The compound [Co(**P18**)] was isolated as a purple solid after purification by flash column chromatography (eluent: 3:1 Hexanes/EtOAc). HRMS (ESI) ([M]⁺) Calcd. for C₁₂₄H₁₁₂N₈O₄Co⁺, Found: 1835.81384. UV-Vis (CHCl₃),  $\lambda_{max}$  nm (log  $\varepsilon$ ): 416 (5.22), 530 (4.06).

### 4.4.3. Synthesis and Characterization of 3-Quinolinyl Cyclopropanes

## 4.4.3.1 Experimental Procedure for [Co(Por)]-Catalyzed Asymmetric Cyclopropanation

A 10 mL oven-dried Schlenk tube was charged with *N*-sulfonyl hydrazone (0.10 mmol, 1.0 equiv), [Co(Por)] (2 mol %), DMAP (0.5 equiv) and Cs₂CO₃ (0.20 mmol, 2.0 equiv). The Schlenk tube was capped with a Teflon screw cap, evacuated and backfilled with nitrogen 3 times. Under nitrogen atmosphere, olefin (1.5 equiv) and anhydrous toluene (1.0 mL) were added. The Schlenk tube was then purged with nitrogen for 1 min and sealed with the Teflon screw cap. The reaction mixture was stirred at 22 °C for 16 h. Following completion of the reaction, the reaction mixture was filtered through a pad of silica gel, concentrated under vacuum and purified by flash column chromatography.



**3-((1***R***,2***R***)-2-Phenylcyclopropyl)quinoline (9)** was obtained through the general procedure using 1.5 equiv of styrene (8) with 3-quinolinecarboxaldehyde 2,4,6-

triisopropylbenzenesulfonyl hydrazone (7) at 22 °C.  $R_f = 0.30$  (3:1 Hexanes/EtOAc). ¹H NMR (500 MHz, CDCl₃)  $\delta$  8.82 (d, J = 2.3 Hz, 1H), 8.09 (d, J = 8.4 Hz, 1H), 7.81 (d, J =

2.3 Hz, 1H), 7.77 (d, J = 8.1 Hz, 1H), 7.65 (ddd, J = 8.4, 6.9, 1.5 Hz, 1H), 7.53 (ddd, J = 8.2, 6.9, 1.2 Hz, 1H), 7.37 – 7.30 (m, 2H), 7.25 – 7.17 (m, 3H), 2.36 (ddd, J = 8.3, 6.3, 4.5 Hz, 1H), 2.31 (ddd, J = 8.6, 6.5, 4.5 Hz, 1H), 1.65 – 1.58 (m, 2H). ¹³C NMR (125 MHz, CDCl₃)  $\delta$  150.55, 146.97, 141.82, 135.47, 131.11, 129.31, 128.71, 128.68, 128.20, 127.38, 126.93, 126.27, 125.97, 28.32, 25.71, 18.08. IR (neat, cm⁻¹): 3025.24, 1602.76, 1494.46, 1342.61, 1126.47, 1073.30, 912.98, 786.17. HRMS (DART) ([M+H]⁺) Calcd. for C₁₈H₁₆N⁺: 246.12773, Found: 246.12896. With [Co(**P18**)]: Yield: 89%. dr: 97:3. HPLC analysis (*E*)-isomer: ee = 86%. ID(95% hexanes: 5% isopropanol, 0.8 mL/min): *t_{major}* = 45.69 min, *t_{minor}* = 41.15 min. With [Co(**P15**)]: Yield: 53%. dr: 96:4. HPLC analysis (*E*)-isomer: ee = 63%. ID(95% hexanes: 5% isopropanol, 0.8 mL/min): *t_{major}* = 43.63 min.

### 4.4.4. X-Ray Crystallography

The X-ray diffraction data were collected using Bruker-AXS SMART-APEXII CCD diffractometer (CuK $\alpha$ ,  $\lambda$  = 1.54178 Å). Indexing was performed using *APEX2*⁹ (Difference Vectors method). Data integration and reduction were performed using SaintPlus.¹⁰ Absorption correction was performed by multi-scan method implemented in SADABS.¹¹ Space groups were determined using XPREP implemented in APEX2.⁹ The structure was solved using SHELXS-97 (direct methods) and refined using SHELXL97 contained in WinGX v1.70.01^{12,13,14} program.

⁹ Bruker (2012). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.

¹⁰ Bruker (**2012**). SAINT. Data Reduction Software.

¹¹ Sheldrick, G. M. (1996). SADABS. Program for Empirical Absorption Correction. University of Gottingen, Germany.

¹² Farrugia, L. J. J. Appl. Cryst. 1999, 32, 837–838.

¹³ Sheldrick, G. M. (**2012** Beta) SHELXL-97. Program for the Refinement of Crystal.

¹⁴ Sheldrick, G. M. Acta Cryst. **1990**, A46, 467–473.



⁰ ⁴ Table S1. Crystal data and stru	acture refinement for 4				
Identification code	C16H15NO				
Empirical formula	C16 H15 N O				
Formula weight	237.29				
Temperature	100(2) K				
Wavelength	1.54178 ≈				
Crystal system	Orthorhombic				
Space group	P212121				
Unit cell dimensions	$a = 6.5868(2) \approx$	$\alpha = 90 \infty$ .			
	$b = 8.9299(3) \approx$	$\beta = 90 \infty$ .			
	$c = 21.6132(8) \approx$	$\gamma = 90  \infty$ .			
Volume	$1271.28(7) \approx ^{3}$				
Z	4				
Density (calculated)	1.240 Mg/m ³				
Absorption coefficient	0.606 mm ⁻¹				
F(000)	504				
Crystal size	0.600 x 0.420 x 0.280 mm ³				
Theta range for data collection	$5.360$ to $70.612 \infty$ .				
Index ranges	-8<=h<=7, -10<=k<=10, -	25<=l<=26			
Reflections collected	16879				
Independent reflections	2392 [R(int) = 0.0331]				
Completeness to theta = $67.679 \infty$	98.6 %				
Absorption correction	Semi-empirical from equivalents				
Max. and min. transmission	0.7534 and 0.6671				
Refinement method	Full-matrix least-squares on F ²				
Data / restraints / parameters	2392 / 0 / 163				
Goodness-of-fit on F ²	1.080				
Final R indices [I>2sigma(I)]	R1 = 0.0274, wR2 = 0.0686				
R indices (all data)	R1 = 0.0275, wR2 = 0.0689				
Absolute structure parameter	0.05(5)				
Extinction coefficient	n/a				
Largest diff. peak and hole	0.177 and -0.135 e. $\approx^{-3}$				

## 4.5. NMR/HPLC SPECTRAL DATA

37 20 4 8 0









## ¹³C NMR of (+)-3, 125 MHz, $CDCI_3$



	- I I	- I I		, j j		· · · ·	ייןיי		· · · · ·		·	·			· · · ·			· · · · ·	
190	180	170	160	150	140	130	120	110	100	90	80	70	60	50	40	30	20	10	0
									f1 (p	pm)									805



1. Table

PDA Ch1 220mm

Ph

Η

P	eak	La	b.	¢

11	DACIII	22VIIIII		
	Peak#	Ret. Time	Area	Area%
	1	7.386	1810111	49.698
	2	10.137	1832085	50.302
	Total		3642196	100.000



### Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	7.042	4577135	97.948
2	10.295	95896	2.052
Total		4673031	100.000

Ph

Н





### ¹H NMR of **(+)-4**, 500MHz, CDCl₃



71.96	45.29 40.06 28.55 28.55 28.55 27.13 27.13 26.53 26.53	7.41 6.91 6.91	9.24	0.65 0.14
<del>~</del>	- $        -$		ო	m N

## ¹³C NMR of **(+)-6**, 125 MHz, CDCl_{3 |}





mAU

Sample ID

Data File



UV Spectrum Retention time = 17.803



(±)-4

NH₂

Н

റ്

UV Spectrum Retention time = 24.145

~ . - -



Peak Table

PDA Chl	220nm		
Peak#	Ret. Time	Area	Area%
1	17.803	7712958	50.044
2	24.145	7699362	49.956
Total	and Colored and South	15412319	100.000



mAU



UV Spectrum Retention time = 17.434



(+)-4

NH₂

Η'

o″





Peak Table

PDA Ch1	220nm		Peak Table
Peak#	Ret. Time	Area	Area%
1	17.434	219939	0.352
2	23.842	62206174	99.648
Total		62426113	100.000







¹H NMR of **9**, 500 MHz, CDCl₃





f1 (ppm)







UV Spectrum Retention time = 45.192



Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	40.549	17696645	49.768
2	45.192	17861769	50.232
Total		35558414	100.000

50.0

min



UV Spectrum Retention time = 45.691

**C1** 1

Н



Peak Table

PDA ChI	220nm		8
Peak#	Ret. Time	Area	Area%
1	41.148	4456535	6.957
2	45.691	59604662	93.043
Total		64061196	100.000



50 1 0 200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390 nm

Peak Table

PDA Ch1	220nm		
Peak#	Ret. Time	Area	Area%
1	39.140	52288930	81.437
2	43.625	11918590	18.563
Total		64207520	100.000

Η.